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Analysis of Change of Red Tide Species in Yodo River Estuary by the Numerical Ecosystem Model

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# ABSTRACT

Occurrence number of red tides in Osaka Bay in Japan is more than 20 cases every year. Diatom red tide was dominant in Osaka Bay, but the non-diatom red tide was dominant in early 1990's. Therefore, the material cycling in Yodo River estuary in Osaka Bay in August from 1991 to 2000 was analyzed by using the numerical ecosystem model and field observation data to clarify the reasons of change in red tide species. Limiting nutrient of primary production is phosphate over the period. Diatom dominated from 1991 to 1993, but it was difficult for non-diatom to grow due to the limitation by physical condition. Non-diatom was able to grow because of good physical and nutrient conditions from 1994 to 1996. And diatom dominated again under the good physical condition, and phosphorus supply was not enough for non-diatom to grow from 1998 to 2000.

KEY WORDS: Red tide; Estuary; Ecosystem model; Box model; Yodo River; Osaka Bay

## **INTRODUCTION**

Red tides in Osaka Bay in Japan occur more than 20 cases every year. Osaka Bay is the semi-enclosed bay and located in the west of Japan as shown in Fig.1. There are many factories and residential areas around Osaka Bay, moreover, the major rivers, Yodo River, Yamato River and so on, empty into Osaka Bay through the big cities, Kyoto, Osaka and so on. Much nutrient is loaded to Osaka Bay, and organic matter accumulates in the bottom and nutrients are released from the bottom sediment. Osaka Bay is one of the eutrophic areas in Japan, and the lower trophic level ecosystem of Osaka Bay is complex.

Diatom red tide was dominant in Osaka Bay but the occurrence number of non-diatom red tide was dominant in the early 1990's as shown in Fig.2. The silica deficiency hypothesis may be related to this phenomenon ((Billen *et al.*, 1991, Humborg C. *et al.*, 1997, Yamamoto and Hatta, 2004). Therefore, the reason of this remarkable change was investigated using the numerical ecosystem model including nitrogen, phosphorus and silicate cycling in August, summer in Japan, from 1991 to 2000.

#### **OBSERVED DATA**

Analyzed area is the surface layer of Yodo River estuary in Osaka Bay shown in Fig.1, where the red tide has occurred every summer. Analyzed month is August, summer in Japan, and the period is from 1991 to 2000. It is assumed as the one box with 3 m depth which is the mixed layer depth in summer in this area.

Figure 3 shows the Yodo River discharge in August observed by the Ministry of Land Infrastructure and Transport (MLIT). Loading nitrogen and phosphorus from Yodo River were estimated by the river discharge, Total Nitrogen (TN) and Total Phosphorus (TP) concentrations in the downstream of Yodo River in August observed by MLIT. Dissolved Silicon (DSi) concentration in Yodo River was observed in August in 1973, 1995 and 2002 (Harashima *et al.*, 2006, Mishima *et al.*, 1999). And the year-to-year variation of DSi concentration is small. Therefore DSi concentration in Yodo River is assumed constant of 108.8 μM during the analyzed period. Figure 4 shows TN, TP and Total Silicon (TSi) fluxes from land area in August, which are the sum of the loading fluxes from Yodo River and the direct fluxes from plant around Osaka Bay.

Figure 5 shows water temperature (a) and salinity (b) in Osaka Bay in August. And Figure 6 shows Dissolved Inorganic Nitrogen (DIN) concentration (a), Dissolved Inorganic Phosphorus (DIP) concentration (b) and DSi concentration (c) in Osaka Bay in August. Figure 7 shows chl.a concentration in Osaka Bay in August. These data were observed by Osaka Prefectural Fisheries Experimental Station and National Institute for Environmental Studies (Harasima *et al.*, 1997). Photon in the water shown in Fig. 8 was estimated by light intensity observed by Japan Metrological Agency (Parsons *et al.*, 1984).

## NUMERICAL MODEL

## **Ecosystem Model**

The numerical ecosystem model, shown in Fig.9, has five compartments, nutrients (DIN, DIP and DSi), phytoplankton (diatom and non-diatom), zooplankton, detritus and Dissolved Organic Nitrogen and Phosphorus (DON and DOP). Material cycling in the box is based on the bio-chemical processes between compartments. And also, the model includes material

load from the land area, sinking of diatom and detritus, diurnal motion of non-diatom and advection and diffusion related to the estuarine circulation. Temporal change in concentration of each compartment is represented by the equations. For example, equation 1 represents the temporal change in DIP concentration. It consists of three parts, biochemical process, boundary condition and physical process.

$$\frac{dDIP_{u}}{dt} = -A_{1dtm}DTMP_{u} - A_{1ndm}NDMP_{u} + B_{2}ZOOP_{u} + C_{1}DETP_{u} + D_{1}DOP_{u} 
+ \frac{1}{V_{u}}(DIP_{load} - F_{u}UDIP_{u} + F_{s}WDIP_{l} 
+ F_{u}\frac{K_{h}}{L}(DIP_{o} - DIP_{u}) + F_{s}\frac{K_{v}}{H}(DIP_{l} - DIP_{u})),$$
(1)

where  $A_l$ ,  $B_2$ ,  $C_l$  and  $D_l$  are the coefficients of biochemical processes, subscript *dtm* refers to diatom, subscript *ndm* refers to non-diatom, *DTMP* is phosphorus concentration in diatom, *NDMP* is phosphorus concentration in non-diatom, *ZOOP* is phosphorus concentration in zooplankton, *DETP* is phosphorus concentration in detritus, *DOP* is dissolved inorganic phosphorus concentration, subscript *u* refers to the upper layer, subscript *l* refers to the lower layer, subscript *o* refers to the adjacent area.  $V_u$  is the volume of box,  $F_u$  is the surface area of the box,  $F_s$  is the boundary area, *L* is the horizontal length between the box and the adjacent area, *H* is the vertical length between the upper and lower layers, *U* is the horizontal advection speed, *W* is the vertical advection speed,  $K_h$  is the horizontal eddy diffusivity and  $K_v$  is the vertical eddy diffusivity.

Biochemical processes are represented by other equations. For example,  $A_1$  represents photosynthesis speed and is the functions of nutrients concentration, water temperature (*T*) and photon in the water (*I*), as shown in equation 2.

$$A_{1} = V_{m} \cdot Min(\frac{DIN}{DIN + k_{n}}, \frac{DIP}{DIP + k_{p}}, \frac{DSi}{DSi + k_{s}})$$
$$\cdot (\frac{T}{T_{o}})^{2} \exp(1 - (\frac{T}{T_{o}})^{2}) \cdot (\frac{I}{I_{o}})^{2} \exp(1 - (\frac{I}{I_{o}})^{2}), \qquad (2)$$

where  $V_m$  denotes the maximum specific nutrient uptake rate, k is the half saturation constant, subscript n refers to nitrogen, subscript p refers to phosphorus, subscript s refers to silicon,  $T_{opt}$  is the optimum water temperature,  $I_{opt}$  is the optimum photon.

Parameters of the ecosystem model are referred to Kawamiya *et al.* (1995), Okaichi (1997) and Yamamoto (1998), and they are shown in Table 1. The half saturation constants are selected from the references when calculation results reproduced the observation values.

## **Physical Model**

*U*, *W*,  $K_h$  and  $K_v$  are estimated by the physical model shown in Fig.10. The governing equations based on salt and water balances are given in equation 3 and 4.

$$R + F_{s}W = U F_{u}$$

$$S_{l}W F_{s} - S_{u}U F_{u} + K_{v} \frac{S_{l} - S_{u}}{H} F_{s} + K_{h} \frac{S_{o} - S_{u}}{L} F_{u} = 0,$$
(4)

where R is river discharge and S is salinity. We suppose that advection is driven by an estuarine circulation. So U is directly related to river discharge, defined by equation 5.

$$U = m \frac{R}{R_m},$$
 (5)

where *m* is unknown quantity,  $R_m$  is average river discharge.  $K_v$  is inversely proportional to stratification. So,  $K_v$  is inversely proportional to river discharge, as defined by equation 6.

$$K_{v} = \frac{l}{R/R_{m}},$$
 (6)

where l is unknown quantity. The unknown quantities,  $K_h$ , m and k are calculated by the least squares method using monthly observed values.

#### **RESULTS AND DISCUSIONS**

The time step of the calculation was 1 hour. Calculations were carried out for August at each year. We obtained the quasi-steady state on the 10th days after the calculation beginning in every year. The year-to-year variations in calculated Total Phosphorus (TP) concentration and chl.a concentration in the box in August by the ecosystem model well reproduced the observed values as shown in Fig.11. Calculated DIN, DIP and DSi concentrations also reproduced the observed ones. According to the later description, phosphorus is the most dominant parameter of photosynthesis shown in equation 2. Therefore the sensitivities of  $k_p$  to TP and chl.a concentrations are confirmed. The range of  $k_p$  shown in the references is double from half of the number used by the calculation. The ranges of calculated TP and chl.a concentrations are from - 1.5 % to + 1.8 % and from - 21.4 % to + 15.9 %, respectively. It doesn't have the influence to the conclusion qualitatively.

Figure 12 shows the year-to-year variations in calculated concentration ratio and observed total red tide days ratio in diatom (a) and non-diatom (b) in August. The concentration ratio means the rate of diatom or non-diatom concentrations in chl.a concentration. And the total red tide day ratio means the ratio of diatom or non-diatom red tide days to total red tide days. The year-to-year variations in calculated concentration ratio agree with the observed total red tide day ratio. The diatom ratios from 1991 to 1993 and from 1998 to 2000 are higher than the non-diatom ratio. But the non-diatom ratios from 1994 to 1996 are relatively high than other years. Figure 13 shows the year-to-year variations in limiting factor to diatom (a) and non-diatom (b) in August calculated by the ecosystem model. Small factor is more effective to primary production. Limiting nutrient of primary production for diatom and non-diatom was phosphate over the period. Figure 14 shows the year-to-year variations in inflow fluxes (a) and outflow fluxes (b) for TP concentration in the box in August calculated by the ecosystem model. Bar charts mean advection plus diffusion, and details (DIP, DOP and Particulate Organic Phosphorus) are also shown.

From 1991 to 1993, water temperature and photon limiting were relatively strong, and phosphorus limiting was small. In these years water temperature and light intensity were low, and photosynthesis was not active, because much photon was necessary for non-diatom than diatom. Therefore diatom ratios in these years were relatively high than other years.

Physical conditions were good after 1994. But phosphorus supply from the lower layer was small from 1998 to 2000 because of weak estuarine circulation and low phosphorus concentration. Therefore phosphorus was exhausted by diatom, and non-diatom could not grow, because non-diatom has higher half-saturation concentration of phosphate compared to diatom (Justić, D. *et al.*, 1995). On the other hand, phosphorus supply from the lower layer was large from 1994 to 1996. Therefore phosphorus concentration in the upper layer was high, and it was good nutrient condition for non-diatom.

## CONCLUSIONS

Diatom dominated from 1991 to 1993, but it was difficult for non-diatom to grow due to the limitation by physical condition. Non-diatom was able to grow because of good physical and nutrient conditions from 1994 to 1996. Diatom again dominated under the good physical condition, and phosphorus supply was not enough for non-diatom to grow from 1998 to 2000. Phosphate concentration in the lower layer of Yodo River estuary was important to the variation in red tide species in the upper layer of Yodo River estuary.

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Fig.1 Study area.

Fig.2 Year-to-year variations in occurrence number of red tide in a year in Osaka Bay.

Fig. 3 Year-to-year variations in Yodo River discharge in August.

Fig.4 TN, TP and TSi fluxes from land area in August.

- Fig.5 Year-to-year variations in water temperature (a) and salinity (b) in Osaka Bay in August.
- Fig.6 Year-to-year variations in DIN (a), DIP (b) and DSi (c) concentrations in Osaka Bay in August

Fig.7 Year-to-year variations in chl.a concentration in Osaka Bay in August

Fig.8 Year-to-year variations in light intensity in Osaka Bay in August

Fig.9 Ecosystem model.

- Fig.10 Physical model.
- Fig.11 Comparison between calculation and observation in TP and chl.a concentrations in August.
- Fig.12 Comparison between calculated concentration ratio and observed total red tide days ratio in diatom (a) and non-diatom (b) in August.
- Fig.13 Year-to-year variations in limiting factor to diatom (a) and non-diatom (b) in August calculated by the ecosystem model.
- Fig.14 Year-to-year variations in inflow flux (a) and outflow flux (b) of phosphorus in August calculated by the ecosystem model.

Table 1 Parameters of the ecosystem model



Fig.1



Fig.2



Fig.3



Fig.4



Fig.5



Fig.6



Fig.7



Fig.8





Fig.10



Fig.11



Fig.12



Fig.13



Fig.14

Parameters	Value	unit
Surface layer depth		m
Length of the boundary line		km
Surface area of study area		km <sup>2</sup>
Length between the upper and lower layer		m
Length between the box and outer area		km
Maximum specific nutrients uptake rate by diatom		day <sup>-1</sup>
Maximum specific nutrients uptake rate by non-diatom		day-1
Ratio of extracellular excretion of DON and DOP by photosynthesis		g m <sup>-2</sup> s <sup>-1</sup>
Mortality speed of phytoplankton at 0 deg-C		m <sup>3</sup> gP <sup>-1</sup> day <sup>-1</sup>
Temperature dependency of mortality of phytoplankton		°C-1
Threshold of phytoplankton density for grazing		mg chl.a <sup>-1</sup>
Ivlev constant		$(\mu g chl.a^{-1})^{-1}$
Grazing speed of phytoplankton by zooplankton at 0 deg-C		day-1
Temperature dependency of grazing by zooplankton		°C-i
Constant for urine generation		
Constant for fecal pellet generation		
Mortality speed of zooplankton at 0 deg-C		m <sup>3</sup> gP <sup>-1</sup> day <sup>-1</sup>
Temperature dependency of mortality of zooplankton		°C <sup>-1</sup>
Decomposition speed of detritus to Dissolved Inorganic Matter at 0 deg-C		day <sup>-1</sup>
Decomposition speed of detritus to Dissolved Organic Matter at 0 deg-C		day-1
Decomposition speed of DOM to DIM at 0 deg-C		day-1
Temperature dependency of decomposition of detritus to DIM		°C <sup>-1</sup>
Temperature dependency of decomposition of detritus to DOM		°C-1
Temperature dependency of decomposition of DOM to DIM		°C-1
Half saturation constant of nitrogen to diatom		$\mu$ M
Half saturation constant of phosphorus to diatom		$\mu$ M
Half saturation constant of silicon to diatom		$\mu$ M
Half saturation constant of nitrogen to non-diatom		$\mu$ M
Half saturation constant of phosphorus to non-diatom		$\mu$ M
Optimum water temperature		°C
Optimum photon		Em <sup>2</sup> s <sup>-1</sup>

Table 1