



(51) International Patent Classification:

*C08K 3/013* (2018.01)      *C08K 3/38* (2006.01)  
*C08K 3/22* (2006.01)      *C08K 3/24* (2006.01)  
*C08K 3/28* (2006.01)      *H01M 10/653* (2014.01)

(21) International Application Number:

PCT/US2021/014100

(22) International Filing Date:

20 January 2021 (20.01.2021)

(25) Filing Language:

English

(26) Publication Language:

English

(30) Priority Data:

63/010,396      15 April 2020 (15.04.2020)      US  
PCT/US2020/064499  
11 December 2020 (11.12.2020)      US

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(54) Title: COMPOSITIONS CONTAINING THERMALLY CONDUCTIVE FILLERS

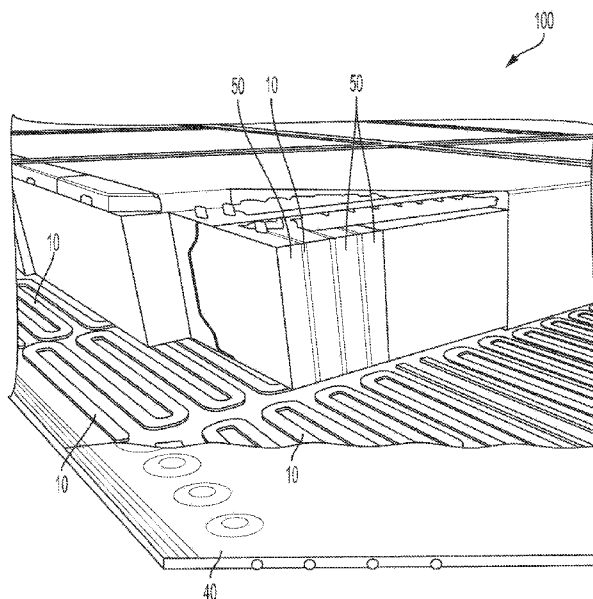


FIG. 2

(57) **Abstract:** Disclosed herein is a composition comprising a thiol-terminated compound; an oxidant; and a thermally conductive filler package comprising thermally conductive, electrically insulative filler particles. The thermally conductive, electrically insulative filler particles have a thermal conductivity of at least 5 W/m·K (measured according to ASTM D7984) and a volume resistivity of at least 1  $\Omega \cdot m$  (measured according to ASTM D257, C611, or B193) and may be present in an amount of at least 50% by volume based on total volume of the filler package. The thermally conductive filler package may be present in an amount of 15% by volume to 90% by volume based on total volume of the composition. The present invention also is directed to a method for treating a substrate and to substrates comprising a layer formed from a composition disclosed herein.



**(81) Designated States** (*unless otherwise indicated, for every kind of national protection available*): AE, AG, AL, AM, AO, AT, AU, AZ, BA, BB, BG, BH, BN, BR, BW, BY, BZ, CA, CH, CL, CN, CO, CR, CU, CZ, DE, DJ, DK, DM, DO, DZ, EC, EE, EG, ES, FI, GB, GD, GE, GH, GM, GT, HN, HR, HU, ID, IL, IN, IR, IS, IT, JO, JP, KE, KG, KH, KN, KP, KR, KW, KZ, LA, LC, LK, LR, LS, LU, LY, MA, MD, ME, MG, MK, MN, MW, MX, MY, MZ, NA, NG, NI, NO, NZ, OM, PA, PE, PG, PH, PL, PT, QA, RO, RS, RU, RW, SA, SC, SD, SE, SG, SK, SL, ST, SV, SY, TH, TJ, TM, TN, TR, TT, TZ, UA, UG, US, UZ, VC, VN, WS, ZA, ZM, ZW.

**(84) Designated States** (*unless otherwise indicated, for every kind of regional protection available*): ARIPO (BW, GH, GM, KE, LR, LS, MW, MZ, NA, RW, SD, SL, ST, SZ, TZ, UG, ZM, ZW), Eurasian (AM, AZ, BY, KG, KZ, RU, TJ, TM), European (AL, AT, BE, BG, CH, CY, CZ, DE, DK, EE, ES, FI, FR, GB, GR, HR, HU, IE, IS, IT, LT, LU, LV, MC, MK, MT, NL, NO, PL, PT, RO, RS, SE, SI, SK, SM, TR), OAPI (BF, BJ, CF, CG, CI, CM, GA, GN, GQ, GW, KM, ML, MR, NE, SN, TD, TG).

**Declarations under Rule 4.17:**

- *as to applicant's entitlement to apply for and be granted a patent (Rule 4.17(ii))*
- *as to the applicant's entitlement to claim the priority of the earlier application (Rule 4.17(iii))*

**Published:**

- *with international search report (Art. 21(3))*

## COMPOSITIONS CONTAINING THERMALLY CONDUCTIVE FILLERS

### CROSS-REFERENCE TO RELATED APPLICATION

[0001] This application claims priority to U.S. Provisional Application No. 63/010,396, filed on April 15, 2020, and entitled “Compositions Containing Thermally Conductive Fillers,” and PCT Application No. PCT/US2020/064499, filed on December 11, 2020, and entitled “Compositions Containing Thermally Conductive Fillers,” both of which are incorporated herein by reference in their entireties.

### FIELD OF THE INVENTION

[0002] The present invention relates to compositions containing a thermally conductive filler component, for example sealants, adhesives, putties, and coating compositions.

### BACKGROUND OF THE INVENTION

[0003] Coating compositions, including sealants and adhesives, are utilized in a wide variety of applications to treat a variety of substrates or to bond together two or more substrate materials.

[0004] The present invention is directed toward one-component and two-component compositions that contain thermally conductive fillers.

### SUMMARY OF THE INVENTION

[0005] The present invention is directed to a composition comprising: a thiol-terminated compound; an oxidant; and a thermally conductive filler package comprising thermally conductive, electrically insulative filler particles, the thermally conductive, electrically insulative filler particles having a thermal conductivity of at least 5 W/m·K (measured according to ASTM D7984) and a volume resistivity of at least 1  $\Omega\cdot\text{m}$  (measured according to ASTM D257); wherein the thermally conductive, electrically insulative filler particles are present in an amount of at least 50% by volume based on total volume of the filler package.

[0006] The present invention also is directed to a composition comprising: a thiol-terminated compound; an oxidant; and a thermally conductive filler package comprising thermally conductive, electrically insulative filler particles, the thermally conductive, electrically

insulative filler particles having a thermal conductivity of at least 5 W/m·K (measured according to ASTM D7984) and a volume resistivity of at least 1  $\Omega\cdot\text{m}$  (measured according to ASTM D257); wherein the thermally conductive, electrically insulative filler particles are present in an amount of at least 50% by volume based on total volume of the filler package; and wherein at least a portion of the thermally conductive, electrically insulative filler particles comprise thermally stable filler particles.

**[0007]** The present invention also is directed to a composition comprising: a thiol-terminated compound; an oxidant; and a thermally conductive filler package comprising thermally conductive, electrically insulative filler particles, the thermally conductive, electrically insulative filler particles having a thermal conductivity of at least 5 W/m·K (measured according to ASTM D7984) and a volume resistivity of at least 1  $\Omega\cdot\text{m}$  (measured according to ASTM D257); wherein the thermally conductive, electrically insulative filler particles are present in an amount of at least 50% by volume based on total volume of the filler package; and wherein at least a portion of the thermally conductive, electrically insulative filler particles comprise thermally unstable filler particles.

**[0008]** The present invention also is directed to a method of treating a substrate comprising contacting at least a portion of a surface of the substrate with a composition of the present invention.

**[0009]** The present invention also is directed to a coating formed on a surface of a substrate, wherein the coating, in an at least partially cured state:

(a) has a thermal conductivity of at least 0.4 W/m·K (measured according to ASTM D7984);

(b) has a dielectric strength of at least 1kV/mm measured according to ASTM D149 on a dielectric meter (Sefetec RMG12AC-DC) connected to two copper electrodes with 1-inch diameter;

(c) has a dielectric strength of at least 1kV/mm measured according to ASTM D149 on a dielectric meter (Sefelec RMG12AC-DC) connected to two copper electrodes with 1-inch diameter;

(c) has a Shore A hardness 5 to 95 measured according to ASTM D2240 with a Type A durometer (Model 2000, Rex Gauge Company, Inc.) at room temperature;

(d) has a 180° peel strength of at least 1 lbf/in (measured according to SAE AS5127/1B with an Instron universal testing machine, model 3345);

(e) has a tensile stress at break of 1 MPa to 100 MPa, as determined according to ISO 37-2 using an Instron universal testing machine, model 3345 with a pull rate of 2 inches/min;

(f) has an elongation of 1% to 900%, as determined according to ISO 37-2 using an Instron universal testing machine, model 3345 with a pull rate of 2 inches/min;

(g) has a Young's modulus of 0.01 MPa to 100 MPa, as determined according to ISO 37-2 using an Instron universal testing machine, model 3345 with a pull rate of 2 inches/min;

(h) maintains a temperature of the substrate that is at least 100°C lower following exposure of the coating on the surface of the substrate to 1000°C for a time of at least 90 seconds than a surface temperature of a bare substrate exposed to 1000°C for the time; and/or

(i) does not smoke upon exposure of the substrate to 1000°C for 500 seconds.

**[0010]** The present invention also is directed to a coating formed on a surface of a substrate, wherein the coating, in an at least partially cured state, has a thermal conductivity of at least 0.5 W/m·K (measured according to ASTM D7984) and maintains a temperature of the substrate that is at least 100°C lower following exposure of the coating on the surface of the substrate to 1000°C for a time of at least 90 seconds than a surface temperature of a bare substrate exposed to 1000°C for the time, wherein the coating is formed from a coating composition of the present invention.

**[0011]** The present invention also is directed to a battery assembly comprising: a battery cell; and a coating formed on a surface of the battery cell from a composition of the present invention.

**[0012]** The present invention also is directed to a substrate comprising a surface at least partially coated with a layer formed from a composition of the present invention.

**[0013]** The present invention also is directed to a method of forming an article comprising extruding a composition of the present invention.

**[0014]** The present invention also is directed to a use of a composition of the present invention for making a coating that, in an at least partially cured state has a thermal conductivity of at least 0.5 W/m·K (measured according to ASTM D7984) and maintains a temperature of the substrate that is at least 100°C lower following exposure of the coating on the surface of the

substrate to 1000°C for at a time of at least 90 seconds than a surface temperature of a bare substrate exposed to 1000°C for the time.

**[0015]** The present invention also is directed to a use of a coating formed from a composition of the present invention to provide a substrate with thermal and fire protection.

#### BRIEF DESCRIPTION OF THE DRAWINGS

**[0016]** FIGS. 1 and 2 are schematic perspective views illustrating a thermally conductive member utilized in a battery pack.

**[0017]** FIG. 3 is a schematic showing the setup used in the fire protection test of the Examples.

**[0018]** FIG. 4 is a graph illustrating the fire performance of a substrate having a coating formed from the compositions of Examples 10 and 11 compared to a bare (uncoated) substrate.

#### DETAILED DESCRIPTION OF THE INVENTION

**[0019]** For purposes of this detailed description, it is to be understood that the invention may assume alternative variations and step sequences, except where expressly specified to the contrary. Moreover, other than in any operating examples, or where otherwise indicated, all numbers expressing, for example, quantities of ingredients used in the specification and claims are to be understood as being modified in all instances by the term “about”. Accordingly, unless indicated to the contrary, the numerical parameters set forth in the following specification and attached claims are approximations that may vary depending upon the desired properties to be obtained by the present invention. At the very least, and not as an attempt to limit the application of the doctrine of equivalents to the scope of the claims, each numerical parameter should at least be construed in light of the number of reported significant digits and by applying ordinary rounding techniques.

**[0020]** Notwithstanding that the numerical ranges and parameters setting forth the broad scope of the invention are approximations, the numerical values set forth in the specific examples are reported as precisely as possible. Any numerical value, however, inherently contains certain errors necessarily resulting from the standard variation found in their respective testing measurements.

**[0021]** Also, it should be understood that any numerical range recited herein is intended to include all sub-ranges subsumed therein. For example, a range of “1 to 10” is intended to include all sub-ranges between (and including) the recited minimum value of 1 and the recited maximum value of 10, that is, having a minimum value equal to or greater than 1 and a maximum value of equal to or less than 10.

**[0022]** As used herein, “including,” “containing” and like terms are understood in the context of this application to be synonymous with “comprising” and are therefore open-ended and do not exclude the presence of additional undescribed or unrecited elements, materials, ingredients or method steps. As used herein, “consisting of” is understood in the context of this application to exclude the presence of any unspecified element, ingredient or method step. As used herein, “consisting essentially of” is understood in the context of this application to include the specified elements, materials, ingredients or method steps “and those that do not materially affect the basic and novel characteristic(s)” of what is being described. As used herein, open-ended terms include closed terms such as consisting essentially of and consisting of.

**[0023]** In this application, the use of the singular includes the plural and plural encompasses singular, unless specifically stated otherwise. For example, although reference is made herein to “an” oxide, “a” polysulfide, or “a” filler material, a combination (i.e., a plurality) of these components may be used.

**[0024]** In addition, in this application, the use of “or” means “and/or” unless specifically stated otherwise, even though “and/or” may be explicitly used in certain instances.

**[0025]** As used herein, the terms “on,” “onto,” “applied on,” “applied onto,” “formed on,” “deposited on,” “deposited onto,” and the like mean formed, overlaid, deposited, or provided on, but not necessarily in contact with, a substrate surface. For example, a composition “applied onto” a substrate surface does not preclude the presence of one or more other intervening coating layers or films of the same or different composition located between the composition and the substrate surface.

**[0026]** As used herein, a “coating composition” refers to a composition, e.g., a solution, mixture, or a dispersion, that, in an at least partially dried or cured state, is capable of producing a film, layer, or the like on at least a portion of a substrate surface.

**[0027]** As used herein, a “sealant composition” refers to a coating composition, e.g., a solution, mixture, or a dispersion, that, in an at least partially dried or cured state, has the ability

to resist atmospheric conditions such as moisture and temperature gradients and particulate matter, such as moisture and temperature and at least partially block the transmission of materials, such as particulates, water, fuel, and other liquids and gasses.

**[0028]** As used herein, a “gap filler composition” refers to a coating composition, e.g., a solution, mixture, or a dispersion, that, in an at least partially dried or cured state, fills a gap.

**[0029]** As used herein, an “adhesive composition” refers to a coating composition, e.g., a solution, mixture, or a dispersion, that, in an at least partially dried or cured state, produces a load-bearing joint, such as a load-bearing joint having a lap shear strength of at least 0.05 MPa, as determined according to ASTM D1002-10 using an Instron universal testing machine, model 3345 in tensile mode with a pull rate of 1 mm per minute.

**[0030]** As used herein, the term “one component” or “1K” refers to a composition in which all of the ingredients may be premixed and stored at ambient conditions or optionally may be premixed and frozen and stored (“pre-mixed frozen” or “PMF” as described below), and wherein the reactive components do not readily react at stored conditions and remain “workable” for at least 10 days after mixing, but instead react only upon activation by an external energy source, under pressure, and/or under shear force, and in the case of PMFs, thawing. External energy sources that may be used to promote curing include, for example, radiation (i.e., actinic radiation such as ultraviolet light) and/or heat. The viscosity of the composition does not double or more for at least 10 days after mixing the ingredients (i.e., the composition remains “workable”).

**[0031]** As further defined herein, ambient conditions generally refer to room temperature (e.g. 23 °C) and humidity conditions or temperature and humidity conditions that are typically found in the area in which the composition is applied to a substrate, e.g., at 10°C to 40°C and 5% to 80% relative humidity, while slightly thermal conditions are temperatures that are slightly above ambient temperature.

**[0032]** As used herein, the term “two-component” or “2K” refers to a composition in which at least a portion of the reactive components readily associate to form an interaction or react to form a bond (physically or chemically), and at least partially cure without activation from an external energy source, such as at ambient or slightly thermal conditions, when mixed. One of skill in the art understands that the two components of the composition are stored



separately from each other and mixed just prior to application of the composition. Two-component compositions may optionally be heated or baked, as described below.

**[0033]** As used herein, the term “cure” or “curing”, means that the components that form the composition are crosslinked to form a film, layer, or bond. As used herein, the term “at least partially cured” means that at least a portion of the components that form the composition interact, react, and/or are crosslinked to form a film, layer, or bond. In the case of a 1K composition, the composition is at least partially cured or cured when the composition is subjected to curing conditions that lead to the reaction of the reactive functional groups of the components of the composition. In the case of a 2K composition, the composition is at least partially cured or cured when the components of the composition are mixed to lead to the reaction of the reactive functional groups of the components of the composition.

**[0034]** As used herein, the term “monofunctional” means an atom or molecule that is only capable of reacting to form one new bond.

**[0035]** As used herein, the term “polyfunctional” means an atom or a molecule that is capable of reacting to form more than one new bond more than one time through the same atom and/or through multiple single reactions of atoms within the molecule. For clarity, polyfunctional includes difunctional.

**[0036]** As used herein, the term “thermally conductive filler” or “TC” filler means a pigment, filler, or inorganic powder that has a thermal conductivity of at least 5 W/m·K at 25°C (measured according to ASTM D7984).

**[0037]** As used herein, the term “non-thermally conductive filler” or “NTC filler” means a pigment, filler, or inorganic powder that has a thermal conductivity of less than 5 W/m·K at 25°C (measured according to ASTM D7984).

**[0038]** As used herein, the term “electrically insulative filler” or “EI filler” means a pigment, filler, or inorganic powder that has a volume resistivity of at least 1  $\Omega\cdot\text{m}$  (measured according to ASTM D257).

**[0039]** As used herein, the term “electrically conductive filler” or “EC filler” means a pigment, filler, or inorganic powder that has a volume resistivity of less than 1  $\Omega\cdot\text{m}$  (measured according to ASTM D257).

**[0040]** As used herein, the term “thermally stable” means a pigment, filler, or inorganic powder that, when tested using the thermal gravimetric analysis (TGA) test under air (according

to ASTM E1131), has no more than 5% weight loss of the total weight of the pigment, filler, or powder occurring before 600°C.

**[0041]** As used herein, the term “thermally unstable” means a pigment, filler, or inorganic powder that, when tested using the TGA test under air (according to ASTM E1131), has a weight loss of the total weight of the pigment of more than 5% occurring before 600°C.

**[0042]** As used herein, the term “smoke” means a suspension of airborne particles and/or gasses, visible to the naked eye, that are emitted when a material undergoes combustion.

**[0043]** As used herein, the term “combustion” refers to the rapid oxidation of materials resulting from exposure to heat or flame.

**[0044]** As used herein, the term “accelerator” means a substance that increases the rate or decreases the activation energy of a chemical reaction in comparison to the same reaction in the absence of the accelerator. An accelerator may be either a “catalyst,” that is, without itself undergoing any permanent chemical change, or may be reactive, that is, capable of chemical reactions and includes any level of reaction from partial to complete reaction of a reactant.

**[0045]** As used herein, the term “solvent” refers to a molecule or a compound that has a high vapor pressure such as greater than 2 mm Hg at 25°C determined by differential scanning calorimetry according to ASTM E1782 and is used to lower the viscosity of a resin but that does not have a reactive functional group capable of reacting with a functional group(s) on molecules or compounds in a composition.

**[0046]** As used herein, the term “reactive diluent” refers to a molecule or a compound that has a low vapor pressure such as 2 mm Hg or less at 25°C determined by differential scanning calorimetry according to ASTM E1782 and is used to lower the viscosity of a resin but that has at least one functional group capable of reacting with a functional group(s) on molecules or compounds in a composition.

**[0047]** As used herein, the term “plasticizer” refers to a molecule or a compound that does not have a functional group capable of reacting with a functional group(s) on molecules or compounds in a composition and that is added to the composition to adjust viscosity, adjust glass transition temperature (T<sub>g</sub>), facilitate application, and impart flexibility.

**[0048]** As used herein, a dash (“—”) that is not between two letters or symbols is used to indicate a point of bonding for a substituent or between two atoms. For example, —CONH<sub>2</sub> is bonded to another chemical moiety through the carbon atom.

**[0049]** As used herein, “polymer” refers to oligomers, homopolymers, and copolymers.

**[0050]** As used herein, unless indicated otherwise, the term “substantially free” means that a particular material is not purposefully added to a mixture or composition, respectively, and is only present as an impurity in a trace amount of less than 5% by weight based on a total weight of the mixture or composition, respectively. As used herein, unless indicated otherwise, the term “essentially free” means that a particular material is only present in an amount of less than 2% by weight based on a total weight of the mixture or composition, respectively. As used herein, unless indicated otherwise, the term “completely free” means that a mixture or composition, respectively, does not comprise a particular material, i.e., the mixture or composition comprises 0% by weight of such material.

**[0051]** As used herein, the volume percentage of each ingredient is calculated using below equation:

$$\text{vol\% (ingredient)} = \frac{\text{volume of ingredient}}{\text{volume of total composition}} \times 100\%$$

wherein the volume of the ingredient is calculated by  $\frac{\text{Weight of ingredient}}{\text{True Density of ingredient}}$

### Compositions of the Present Invention

**[0052]** Disclosed herein is a composition comprising, or consisting essentially of, or consisting of: a thiol-terminated compound; an oxidant; and a thermally conductive filler package comprising thermally conductive, electrically insulative filler particles, the thermally conductive, electrically insulative filler particles having a thermal conductivity of at least 5 W/m·K (measured according to ASTM D7984) and a volume resistivity of at least 1 Ω·m (measured according to ASTM D257).

**[0053]** Also disclosed herein is a composition comprising, or consisting essentially of, or consisting of: a thiol-terminated compound; an oxidant; and a thermally conductive filler package comprising thermally conductive, electrically insulative filler particles, the thermally conductive, electrically insulative filler particles having a thermal conductivity of at least 5 W/m·K (measured according to ASTM D7984) and a volume resistivity of at least 1 Ω·m (measured according to ASTM D257); wherein the thermally conductive, electrically insulative filler particles are present in an amount of at least 50% by volume based on total volume of the

filler package; and wherein at least a portion of the thermally conductive, electrically insulative filler particles comprise thermally stable filler particles.

**[0054]** Also disclosed herein is a composition comprising, or consisting essentially of, or consisting of: a thiol-terminated compound; an oxidant; and a thermally conductive filler package comprising thermally conductive, electrically insulative filler particles, the thermally conductive, electrically insulative filler particles having a thermal conductivity of at least 5 W/m·K (measured according to ASTM D7984) and a volume resistivity of at least 1  $\Omega\cdot\text{m}$  (measured according to ASTM D257); wherein the thermally conductive, electrically insulative filler particles are present in an amount of at least 50% by volume based on total volume of the filler package; and wherein at least a portion of the thermally conductive, electrically insulative filler particles comprise thermally unstable filler particles.

**[0055]** The composition may be a coating composition, such as a sealant composition, an adhesive composition, a gap filler composition, a putty, a molding compound, a potting compound, and/or a 3D-printable composition or may be used in its at least partially dried or cured state to form a film, layer, or the like, or a part, such as a casted, molded, extruded, or machined part.

**[0056]** The composition may be provided as a one-component composition, or as a two-component composition, or as a three-component or higher composition.

**[0057]** The compositions disclosed herein may be a 1K composition comprising, or consisting essentially of, or consisting of, a thiol-terminated compound, an oxidant, and a thermally conductive filler package as described below, and optionally a curing agent, an accelerator, a dispersant and/or any of the additives described below. As described in more detail below, at least a portion of the thermally conductive, electrically insulative filler particles may comprise thermally stable filler particles and/or at least a portion of the thermally conductive, electrically insulative filler particles may comprise thermally unstable filler particles.

**[0058]** The compositions disclosed herein may be a 2K composition comprising, or consisting essentially of, or consisting of: a first component comprising, or consisting essentially of, or consisting of, a thiol-terminated compound; a second component comprising, or consisting essentially of, or consisting of, an oxidant; and a thermally conductive filler package that may be present in the first component and/or the second component, and optionally a curing agent, an accelerator, a dispersant and/or any of the additives described below. As described in more

detail below, the filler package optionally also may further comprise at least one thermally stable filler material and/or at least one thermally unstable filler material. Such accelerator and/or a dispersant and/or any of the additives described below may be present in the first component and/or the second component. The first and second components may be mixed together immediately prior to use.

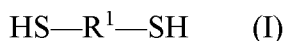
**[0059]** The compositions disclosed herein may be a 3K or higher composition comprising, or consisting essentially of, or consisting of: a first component comprising, or consisting essentially of, or consisting of, a thiol-terminated compound; a second component comprising, or consisting essentially of, or consisting of, an oxidant; and a third component comprising, or consisting essentially of, or consisting of, a thermally conductive filler package, and optionally, a curing agent, an accelerator, a dispersant and/or any of the additives described below. As described in more detail below, the filler package optionally also may further comprise at least one thermally stable filler material and/or at least one thermally unstable filler material. Such accelerator and/or a dispersant and/or any of the additives described below may be present in the first component and/or the second component and/or the third component.

**[0060]** In the case of a 2K composition, one of the components may be substantially free, or essentially free, or completely free, of filler materials, and in the case of a 3K composition, one or two of the components may be substantially free, or essentially free, or completely free, of filler materials.

#### Thiol-Terminated Compounds

**[0061]** As stated above, the composition comprises, or consists essentially of, or consists of, a thiol-terminated compound. The thiol-terminated compound may be a monomer, a polymer, and/or an oligomer. Suitable thiol-terminated compounds are disclosed in U.S. Patent No. 7,858,703B2, incorporated herein by reference in its entirety. The thiol-terminated compound may comprise a liquid.

**[0062]** Suitable thiol-terminated compounds useful in the compositions of the present invention for preparing the thiol-terminated polymer include compounds, monomers or polymers having at least two thiol groups. Useful polythiols include dithiols having the formula (I):



where R<sup>1</sup> can be a C<sub>2-6</sub> n-alkylene group; C<sub>3-6</sub> branched alkylene group, having one or more pendant groups which can be, for example, hydroxyl groups, alkyl groups such as methyl or

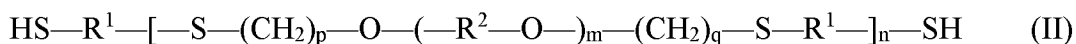
ethyl groups; alkoxy groups, C<sub>6-8</sub> cycloalkylene; C<sub>6-10</sub> alkylcycloalkylene group;  $—[(—CH_2)_p—X]_q—(—CH_2)_r—$ ; or  $—[(—CH_2)_p—X]_q—(—CH_2)_r—$  in which at least one  $—CH_2—$  unit is substituted with a methyl group and in which p is an independently selected integer ranging from 2 to 6, q is an independently selected integer ranging from 1 to 5 and r is an independently selected integer ranging from 2 to 10. Further useful dithiols include one or more heteroatom substituents in the carbon backbone, that is, dithiols in which X includes a heteroatom such as O, S, S-S or another bivalent heteroatom radical; a secondary or tertiary amine group, i.e.,  $—NR^6—$ , where R<sup>6</sup> is hydrogen or methyl; or another substituted trivalent heteroatom. In an example, X is O or S, and thus R<sup>1</sup> is  $—[(—CH_2—)_p—O—]_q—(—CH_2—)_r—$  or  $—[(—CH_2—)_p—S—]_q—(—CH_2—)_r—$ . In examples, p and r may be equal, and in an example both have the value of 2.

**[0063]** Useful polythiols include but are not limited to dithiols such as 1,2-ethanedithiol, 1,2-propanedithiol, 1,3-propanedithiol, 1,3-butanedithiol, 1,4-butanedithiol, 2,3-butanedithiol, 1,3-pentanedithiol, 1,5-pentanedithiol, 1,6-hexanedithiol, 1,3-dimercapto-3-methylbutane, dipentenedimercaptan, ethylcyclohexyldithiol (ECHDT), dimercaptodiethylsulfide, methyl-substituted dimercaptodiethylsulfide, dimethyl-substituted dimercaptodiethylsulfide, dimercaptodioxaoctane, tris[2-(3-mercaptopropionyloxy)ethyl]isocyanurate, 2,3-di(2-mercптоethylthio)-1-propane-thiol, (2,2'-thiodiethanethiol), (2,2'-(ethylenedioxy)diethanethiol 1,8-dimercapto-3,6-dioxaoctane, 1,5-dimercapto-3-oxapentane and mixtures thereof. The polythiol material can have one or more pendant groups selected from lower alkyl groups, lower alkoxy groups and hydroxyl groups. Suitable alkyl pendant groups include C<sub>1</sub>-C<sub>6</sub> linear alkyl, C<sub>3</sub>-C<sub>6</sub> branched alkyl, cyclopentyl, and cyclohexyl.

**[0064]** Useful dithiols include dimercaptodiethylsulfide (DMDS) (p=2, r=2, q=1, X=S); dimercaptodioxaoctane (DMDO) (p=2, q=2, r=2, X=O); and 1,5-dimercapto-3-oxapentane (p=2, r=2, q=1, X=O). It is also possible to use dithiols that include both heteroatom substituents in the carbon backbone and pendant alkyl groups, such as methyl groups. Such compounds include methyl-substituted DMDS, such as HS—CH<sub>2</sub>CH(CH<sub>3</sub>)—S—CH<sub>2</sub>CH<sub>2</sub>—SH, HS—CH(CH<sub>3</sub>)CH<sub>2</sub>—S—CH<sub>2</sub>CH<sub>2</sub>—SH and dimethyl substituted DMDS such as HS—CH<sub>2</sub>CH(CH<sub>3</sub>)—S—CH(CH<sub>3</sub>)CH<sub>2</sub>—SH and HS—CH(CH<sub>3</sub>)CH<sub>2</sub>—S—CH<sub>2</sub>CH(CH<sub>3</sub>)—SH.

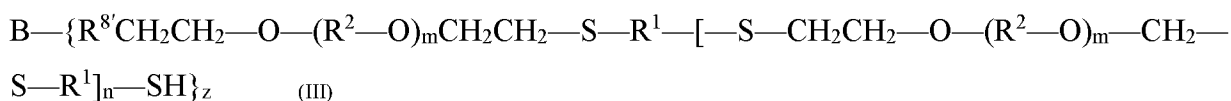
**[0065]** Two or more different polythiols can be used if desired in the compositions of the present invention.

**[0066]** A thiol-terminated compound useful in the compositions of the present invention may have the structure of formula (II):

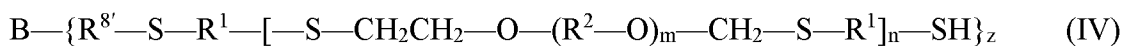


wherein R<sup>1</sup> denotes a C<sub>2-10</sub> n-alkylene, C<sub>2-6</sub> branched alkylene, C<sub>6-8</sub> cycloalkylene or C<sub>6-10</sub> alkylcycloalkylene group, heterocyclic, —[(—CH<sub>2</sub>)<sub>p</sub>—X]<sub>q</sub>—(—CH<sub>2</sub>)<sub>r</sub>; or —[(—CH<sub>2</sub>)<sub>p</sub>—X]<sub>q</sub>—(—CH<sub>2</sub>)<sub>r</sub>— in which at least one —CH<sub>2</sub>— unit is substituted with a methyl group; R<sup>2</sup> denotes a C<sub>2-10</sub> n-alkylene, C<sub>2-6</sub> branched alkylene, C<sub>6-8</sub> cycloalkylene or C<sub>6-14</sub> alkylcycloalkylene group, heterocyclic, —[(—CH<sub>2</sub>)<sub>p</sub>—X]<sub>q</sub>—(—CH<sub>2</sub>)<sub>r</sub>; X denotes one selected from the group consisting of O, S, S-S and —NR<sup>6</sup>—; R<sup>6</sup> denotes H or methyl; m is an independently selected rational number from 1 to 50; and n is an independently selected integer from 1 to 60; p is an independently selected integer ranging from 2 to 6; q is an independently selected integer ranging from 1 to 5; and r is an independently selected integer from 2 to 10. In an example of the foregoing polymer, R<sup>1</sup> is C<sub>2</sub>-C<sub>6</sub> alkyl and R<sup>2</sup> is C<sub>2</sub>-C<sub>6</sub> alkyl.

**[0067]** Polyfunctional thiol-terminated polymers according to the present invention may have the formula (III):



or formula (IV):



wherein B denotes a z-valent residue of a polyfunctionalizing agent B—(R<sup>8</sup>)<sub>z</sub>, R<sup>1</sup> denotes a C<sub>2-10</sub> n-alkylene, C<sub>2-6</sub> branched alkylene, C<sub>6-8</sub> cycloalkylene or C<sub>6-10</sub> alkylcycloalkylene group, heterocyclic, —[(—CH<sub>2</sub>)<sub>p</sub>—X]<sub>q</sub>—(—CH<sub>2</sub>)<sub>r</sub>; or —[(—CH<sub>2</sub>)<sub>p</sub>—X]<sub>q</sub>—(—CH<sub>2</sub>)<sub>r</sub>— in which at least one —CH<sub>2</sub>— unit is substituted with a methyl group; R<sup>2</sup> denotes a C<sub>2-10</sub> n-alkylene, C<sub>2-6</sub> branched alkylene, C<sub>6-8</sub> cycloalkylene or C<sub>6-14</sub> alkylcycloalkylene group, heterocyclic, —[(—CH<sub>2</sub>)<sub>p</sub>—X]<sub>q</sub>—(—CH<sub>2</sub>)<sub>r</sub>; X denotes one selected from the group consisting of O, S, and —NR<sup>6</sup>—; R<sup>6</sup> denotes H or methyl; m is an independently selected rational number from 1 to 50; n is an independently selected integer ranging from 1 to 60; p is an independently selected integer ranging from 2 to 6; q is an independently selected integer ranging from 1 to 5; and r is an independently selected integer ranging from 2 to 10; R<sup>8</sup> denotes a moiety which is reactive with a terminal vinyl group or mercapto group; z is an integer from 3 to 6; and R<sup>8'</sup> is the product of the reactants.

**[0068]** Other suitable thiol-terminated monomers for use in the compositions of the present invention include, for example, mercapto-propionates, mercapto-acetates, and combinations of any of the foregoing.

**[0069]** Examples of suitable mercapto-propionates for use in the compositions of the present invention include pentaerythritol tetra(3-mercapto-propionate) (PETMP), trimethylolpropane tri(3-mercaptopropionate) (TMPMP), glycol di(3-mercaptopropionate) (GDMP), tris[2-(3-mercapto-propionyloxy)ethyl]isocyanurate (TEMPIC), di-pentaerythritol hexa(3-mercaptopropionate) (di-PETMP), tri(3-mercaptopropionate) pentaerythritol, triethylolethane tri(3-mercaptopropionate), and combinations of any of the foregoing.

**[0070]** Examples of suitable polymeric thiols for use in the compositions of the present invention include ethoxylated trimethylolpropane tri(3-mercaptopropionate), polycaprolactone tetra-3-mercaptopropionate, and combinations thereof.

**[0071]** Examples of suitable mercapto-acetates for use in the compositions of the present invention include pentaerythritol tetramercaptoacetate (PRTMA), trimethylolpropane trimercaptoacetate (TMPMA), glycol dimercaptoacetate (GDMA), ethyleneglycol dimercaptoacetate, di-trimethylolpropane tetramercaptoacetate, and combinations of any of the foregoing.

**[0072]** Suitable thiol-terminated monomers for use in compositions of the present invention are commercially available, for example, from Bruno Bock Thiochemicals under the Thiocure® tradename. Suitable thiol-terminated polymers for use in compositions of the present invention are commercially available, for example, from Toray Industries, Inc. under the Thiokol® LP tradename or from Nouryon under the Thioplast® tradename.

**[0073]** The thiol-terminated compound of the present invention may have the structure of formula (V):



wherein R<sup>1</sup> is selected from C<sub>2-6</sub> alkanediyl, C<sub>6-8</sub> cycloalkanediyl, C<sub>6-10</sub> alkanecycloalkanediyl, C<sub>5-8</sub> heterocycloalkanediyl, substituted C<sub>2-6</sub> alkanediyl, substituted C<sub>6-8</sub> cycloalkanediyl, substituted C<sub>6-10</sub> alkanecycloalkanediyl, substituted C<sub>5-8</sub> heterocycloalkanediyl and  $\text{---}[(\text{CHR}^3)_p\text{---X}]_q\text{---}(\text{CHR}^3)_r\text{---}$ ; where, each R<sup>3</sup> is selected from hydrogen and methyl; each X is independently selected from O, S, S—S, NH, and N(—CH<sub>3</sub>); p is an integer from 2 to 6; q is an integer from 1



to 5; and r is an integer from 2 to 10. In examples, each p can independently be 2, 3, 4, 5, and 6. In examples, each p can be the same and can be 2, 3, 4, 5, or 6.

**[0074]** Other suitable thiol-terminated compounds useful in the present invention include those which are commercially available under the trade name Thiokol polysulfides (commercially available from Toray Chemical). Such polysulfide polymers are disclosed in U.S. Patent No. 2,466,963. Other suitable thiol-terminated compounds useful in the present invention include Thioplasts (commercially available from AkzoNobel).

**[0075]** The thiol-terminated compound may have a weight average molecular weight ( $M_w$ ) of at least 80 g/mol, such as at least 100 g/mol, such as at least 150 g/mol, and may have a weight average molecular weight of no more than 40,000 g/mol, such as no more than 30,000 g/mol, such as no more than 20,000 g/mol. The thiol-terminated compound may have a weight average molecular weight of 80 g/mol to 40,000 g/mol, such as 100 g/mol to 30,000 g/mol, such as 150 g/mol to 20,000 g/mol. As used herein, the term “weight average molecular weight” or “( $M_w$ )” means the weight average molecular weight ( $M_w$ ) as determined by gel permeation chromatography (GPC) using polystyrene standards for calibration. The GPC determination can be performed using a Waters 2695 separation module with a Waters 410 differential refractometer (RI detector), linear polystyrene standards having molecular weights of from 580 Da to 365,000 Da, tetrahydrofuran (THF) as the eluent at a flow rate of 0.5 mL/min, and an Agilent PLgel Mixed-C column (300 x 7.5 mm, 5  $\mu$ m) for separation.

**[0076]** The thiol-terminated compound may be present in the composition in an amount of at least 9.9% by volume based on total volume of the composition, such as at least 24.5% by volume, and may be present in the composition in an amount of no more than 84.9% by volume based on total volume of the composition, such as no more than 65% by volume. The thiol-terminated compound may be present in the composition in an amount of 9.9% by volume to 84.9% by volume based on total volume of the composition, such as 24.5% by volume to 65% by volume.

#### Oxidants

**[0077]** The composition of the present invention also comprises, or consists essentially of, or consists of, an oxidant that polymerizes the thiol-terminated polymers of the present invention to rubbery solids by oxidizing the thiol functional group of the thiol-terminated

compound to form sulfur-sulfur bonds. For example, a mechanism of cure may comprise  $2 - \text{RSH} + (\text{O}) \rightarrow \text{RSSR} + \text{H}_2\text{O}$ .

**[0078]** The curing rate may be modified by the addition of additives, such as acidic materials to slow the curing rate or alkaline materials to accelerate the curing rate. Compositions of the present invention can comprise a polysulfide cure retarder or a combination of polysulfide cure retarders. A polysulfide cure retarder can comprise an acid such as a fatty acid, an organic acid, an inorganic acid, a fatty acid salt, or combinations thereof. Examples of suitable polysulfide cure retarders include phenylphosphonic acid and itaconic acid. Cure retarders can improve the stability of the polysulfide cure activator and polysulfide cure accelerator.

**[0079]** Suitable oxidants that may be used in the compositions of the present invention may comprise a reactive metal oxide (i.e., a metal oxide that can promote an oxidation pathway in another species in the composition) and/or an organic peroxide. Suitable examples of reactive metal oxides and/or organic peroxides include lead oxide, lead dioxide, lead peroxide, manganese dioxide, sodium dichromate, sodium perborate, sodium perborate monohydrate, potassium permanganate, calcium dioxide, calcium peroxide, barium peroxide, lithium peroxide, zinc peroxide, zinc chromate, barium oxide, alkaline dichromate, or combinations thereof. The oxidant may comprise an organic peroxide such as cumene hydroperoxide, t-butyl hydroperoxide, or combinations thereof. In examples, the oxidants useful in the composition of the present invention are not thermally conductive or thermally stable. In examples, the thermally conductive filler particles (described below) do not include reactive metal oxides.

**[0080]** The oxidant may be present in the composition in an amount of at least 0.1% by volume based on total volume of the composition, such as at least 0.5% by volume, and may be present in the composition in an amount of no more than 10% by volume based on total volume of the composition, such as no more than 5% by volume. The oxidant may be present in the composition in an amount of 0.1% by volume to 10% by volume based on total volume of the composition, such as 0.5% by volume to 5% by volume.

#### Thermally Conductive Filler Package

**[0081]** The compositions disclosed herein also comprise a thermally conductive filler package comprising, or consisting essentially of, or consisting of, particles of a thermally conductive, electrically insulative filler material (referred to herein as “TC/EI filler material” and described in more detail below). The TC/EI filler material may comprise organic or inorganic

material and may comprise particles of a single type of filler material or may comprise particles of two or more types of TC/EI filler materials. That is, the thermally conductive filler package may comprise particles of a first TC/EI filler material and may further comprise particles of at least a second (i.e., a second, a third, a fourth, etc.) TC/EI filler material that is different from the first TC/EI filler material. In an example, the particles of the first TC/EI filler material may have an average particle size that is at least one order of magnitude greater than an average particle size of the particles of the second TC/EI filler material, such as at least two orders of magnitude greater, such as at least three orders of magnitude greater, wherein the particle sizes may be measured by methods known to those skilled in the art, for example, using a scanning electron microscope (SEM). For example, powders may be dispersed on segments of carbon tape attached to aluminum stubs and coated with Au/Pd for 20 seconds. Samples then may be analyzed in a Quanta 250 FEG SEM under high vacuum (accelerating voltage 10 kV and spot size 3.0), measuring 30 particles from three different areas to provide an average particle size for each sample. One skilled in the art will recognize that there can be variations in this procedure that retain the essential elements of microscopic imaging and averaging of representative size. As used herein with respect to types of filler material, reference to “first,” “second”, etc. is for convenience only and does not refer to order of addition to the filler package or the like. Optionally, as discussed in more detail below, the filler package also may comprise particles of thermally conductive, electrically conductive filler material (referred to herein as “TC/EC” filler material) and/or particles of non-thermally conductive, electrically insulative filler material (referred to herein as “NTC/EI” filler material). The filler materials may be organic or inorganic.

**[0082]** The TC/EC filler material may comprise particles of a single type of filler material or may comprise particles of two or more types of TC/EC filler materials. That is, the thermally conductive filler package may comprise particles of a first TC/EC filler material and may further comprise particles of at least a second (i.e., a second, a third, a fourth, etc.) TC/EC filler material that is different from the first TC/EC filler material. In an example, the particles of the first TC/EC filler material may have an average particle size that is at least one order of magnitude greater than an average particle size of the particles of the second TC/EC filler material, such as at least two orders of magnitude greater, such as at least three orders of magnitude greater, wherein the particle sizes may be measured, for example, using a SEM as described above.

**[0083]** Likewise, the NTC/EI filler material may comprise particles of a single type of filler material or may comprise particles of two or more types of NTC/EI filler materials. That is, the thermally conductive filler package may comprise particles of a first NTC/EI filler material and may further comprise particles of at least a second (i.e., a second, a third, a fourth, etc.) NTC/EI filler material that is different from the first NTC/EI filler material. In an example, the particles of the first NTC/EI filler material may have an average particle size that is at least one order of magnitude greater than an average particle size of the particles of the second NTC/EI filler material, such as at least two orders of magnitude greater, such as at least three orders of magnitude greater, wherein the particle sizes may be measured, for example, using a SEM as described above.

**[0084]** Particles of filler material used in the thermally conductive filler package may have a reported Mohs hardness of at least 1 (based on the Mohs Hardness Scale), measured according to ASTM D2240, such as at least 2, such as at least 3, and may have a reported Mohs hardness of no more than 10, such as no more than 8, such as no more than 7. Particles of filler used in the thermally conductive filler package may have a reported Mohs hardness of 1 to 10, such as 2 to 8, such as 3 to 7.

**[0085]** Particles of filler material used in the thermally conductive filler package may have a reported average particle size in at least one dimension of at least 0.01  $\mu\text{m}$ , as reported by the manufacturer, such as at least 2  $\mu\text{m}$ , such as at least 10  $\mu\text{m}$ , and may have a reported average particle size in at least one dimension of no more than 500  $\mu\text{m}$  as reported by the manufacturer, such as no more than 400  $\mu\text{m}$ , such as no more than 300  $\mu\text{m}$ , such as no more than 100  $\mu\text{m}$ . The particles of filler material used in the thermally conductive filler package may have a reported average particle size in at least one dimension of 0.01  $\mu\text{m}$  to 500  $\mu\text{m}$  as reported by the manufacturer, such as 0.1  $\mu\text{m}$  to 400  $\mu\text{m}$ , such as 2  $\mu\text{m}$  to 300  $\mu\text{m}$ , such as 10  $\mu\text{m}$  to 100  $\mu\text{m}$ . Suitable methods of measuring average particle size include measurement using an instrument such as the Quanta 250 FEG SEM or an equivalent instrument.

**[0086]** Particles of filler material used in the thermally conductive filler package may comprise a plurality of particles each having, for example, a platy, spherical, or acicular shape, and agglomerates thereof. As used herein, “platy” refers to a two-dimensional material having a substantially flat surface and that has a thickness in one direction that is less than 25% of the largest dimension.

**[0087]** Particles of filler material used in the thermally conductive filler package may be thermally conductive. The particles of thermally conductive filler material may have a thermal conductivity of at least 5 W/m·K at 25°C (measured according to ASTM D7984), such as at least 18 W/m·K, such as at least 55 W/m·K, and may have a thermal conductivity of no more than 3,000 W/m·K at 25°C, such as no more than 1,400 W/m·K, such as no more than 450 W/m·K. The particles of a thermally conductive filler material may have a thermal conductivity of 5 W/m·K to 3,000 W/m·K at 25°C (measured according to ASTM D7984), such as 18 W/m·K to 1,400 W/m·K, such as 55 W/m·K to 450 W/m·K.

**[0088]** Particles of filler material used in the thermally conductive filler package may be non-thermally conductive. The particles of non-thermally conductive filler material may have a thermal conductivity of less than 5 W/m·K at 25°C (measured according to ASTM D7984), such no more than 3 W/m·K, such as no more than 1 W/m·K, such as no more than 0.1 W/m·K, such as no more than 0.05 W/m·K, such as 0.02 W/m·K at 25°C to 5 W/m·K at 25°C. Thermal conductivity may be measured as described above.

**[0089]** Particles of filler material used in the thermally conductive filler package may be electrically insulative. The particles of electrically insulative filler material may have a volume resistivity of at least 1  $\Omega\cdot\text{m}$  (measured according to ASTM D257), such as at least 10  $\Omega\cdot\text{m}$ , such as at least 100  $\Omega\cdot\text{m}$ .

**[0090]** Particles of filler material used in the thermally conductive filler package may be electrically conductive. The particles of electrically conductive filler material may have a volume resistivity of less than 1  $\Omega\cdot\text{m}$  (measured according to ASTM D257), such as less than 0.1  $\Omega\cdot\text{m}$ .

**[0091]** The thermally conductive filler package may be present in the composition in an amount of at least 15% by volume based on total volume of the composition, such as at least 30% by volume, and may be present in the composition in an amount of no more than 90% by volume based on total volume of the composition, such as no more than 75% by volume. The thermally conductive filler package may be present in the composition in an amount of 15% by volume to 90% by volume based on total volume of the composition, such as 30% by volume to 75% by volume.

**[0092]** As noted above, the thermally conductive filler package may comprise particles of TC/EI filler material.

**[0093]** Suitable TC/EI filler materials include boron nitride (for example, commercially available as CarboTherm from Saint-Gobain, as CoolFlow and PolarTherm from Momentive, and as hexagonal boron nitride powder available from Panadyne), silicon nitride, or aluminum nitride (for example, commercially available as aluminum nitride powder available from Micron Metals Inc., and as Toyalnite from Toyal), metal oxides such as aluminum oxide (for example, commercially available as Microgrit from Micro Abrasives, as Nabalox from Nabaltec, as Aeroxide from Evonik, and as Alodur from Imerys), magnesium oxide, beryllium oxide, silicon dioxide, titanium oxide, zinc oxide, nickel oxide, copper oxide, or tin oxide, metal hydroxides such as aluminum trihydrate, aluminum hydroxide or magnesium hydroxide, arsenides such as boron arsenide, carbides such as silicon carbide, minerals such as agate and emery, ceramics such as ceramic microspheres (for example, commercially available from Zeeospheres Ceramics or 3M), silicon carbide, and diamond. These fillers can also be surface modified, such as PYROKISUMA 5301K available from Kyowa Chemical Industry Co., Ltd. These thermally conductive fillers may be used alone or in a combination of two or more.

**[0094]** The TC/EI filler particles may be present in an amount of at least 50% by volume based on total volume of the filler package, such as at least 60% by volume, such as at least 70% by volume, such as at least 80% by volume, such as at least 90% by volume, and may be present in an amount of no more than 100% by volume based on total volume of the filler package, such as no more than 90% by volume, such as no more than 80% by volume. The TC/EI filler particles may be present in an amount of 50% by volume to 100% by volume based on total volume of the filler package, such as 60% by volume to 100% by volume, such as 70% by volume to 100% by volume, such as 80% by volume to 100% by volume, such as 90% by volume to 100% by volume, such as 50% by volume to 90% by volume, such as 60% by volume to 90% by volume, such as 70% by volume to 90% by volume, such as 80% by volume to 90% by volume, such as such as 50% by volume to 80% by volume, such as 60% by volume to 80% by volume, such as 70% by volume to 80% by volume, such as 50% by volume to 70% by volume, such as 50% by volume to 60% by volume, such as 60% by volume to 70% by volume.

**[0095]** The filler package may comprise thermally stable filler materials. In an example, at least a portion of the TC/EI filler particles may be thermally stable. For example, at least 0.1% by volume of the TC/EI filler particles may be thermally stable based on total volume of the TC/EI fillers present in the thermally conductive filler package, such as at least 1% by

volume, such as at least 10% by volume such as at least 15% by volume, such as at least 20% by volume, such as at least 25% by volume, such as at least 30% by volume, such as at least 35% by volume, such as at least 40% by volume, such as at least 45% by volume, such as at least 50% by volume, such as at least 55% by volume, such as at least 60% by volume, such as at least 65% by volume, such as at least 70% by volume, such as at least 75% by volume, such as at least 80% by volume, such as at least 85% by volume, such as at least 90% by volume, such as at least 91% by volume, such as at least 92% by volume, such as at least 93% by volume, such as at least 94% by volume, such as at least 95% by volume, such as at least 96% by volume, such as at least 97% by volume, such as at least 98% by volume, such as at least 99% by volume, such as 100% by volume. For example, 0.1% by volume to 100% by volume of the TC/EI filler particles may be thermally stable based on total volume of the TC/EI fillers present in the thermally conductive filler package, such as 1% by volume to 90% by volume, such as 10% by volume to 80% by volume, such as 20% by volume to 70% by volume, such as 30% by volume to 60% by volume, such as 90% by volume to 100% by volume, such as 93% by volume to 98% by volume.

**[0096]** In an example, the composition may comprise at least a portion of TC/EI filler particles that are thermally unstable. For example, at least 0.1% by volume of the TC/EI filler particles may be thermally unstable based on total volume of the TC/EI fillers present in the thermally conductive filler package, such as at least 1% by volume, such as at least 10% by volume such as at least 15% by volume, such as at least 20% by volume, such as at least 25% by volume, such as at least 30% by volume, such as at least 35% by volume, such as at least 40% by volume, such as at least 45% by volume, such as at least 50% by volume, such as at least 55% by volume, such as at least 60% by volume, such as at least 65% by volume, such as at least 70% by volume, such as at least 75% by volume, such as at least 80% by volume, such as at least 85% by volume, such as at least 90% by volume, such as at least 91% by volume, such as at least 92% by volume, such as at least 93% by volume, such as at least 94% by volume, such as at least 95% by volume, such as at least 96% by volume, such as at least 97% by volume, such as at least 98% by volume, such as at least 99% by volume, such as 100% by volume. For example, 0.1% by volume to 100% by volume of the TC/EI filler particles may be thermally stable based on total volume of the TC/EI fillers present in the thermally conductive filler package, such as 1% by volume to 90% by volume, such as 10% by volume to 80% by volume, such as 20% by volume to 70% by volume, such as 30% by volume to 60% by volume, such as 90% by volume to 100%

by volume, such as 93% by volume to 98% by volume. In other examples, no more than 10% by volume of the TC/EI filler particles may be thermally unstable based on total volume of the TC/EI fillers present in the thermally conductive filler package, such as no more than 9% by volume, such as no more than 8% by volume, such as no more than 7% by volume, such as no more than 6% by volume, such as no more than 5% by volume, such as no more than 4% by volume, such as no more than 3% by volume, such as no more than 2% by volume, such as no more than 1% by volume. For example, up to 10% by volume of the TC/EI filler particles may be thermally unstable based on total volume of the TC/EI fillers present in the thermally conductive filler package, such as 2% by volume to 7% by volume.

**[0097]** Suitable thermally stable TC/EI fillers include boron nitride, silicon nitride, or aluminum nitride, arsenides such as boron arsenide, metal oxides such as aluminum oxide, magnesium oxide, beryllium oxide, silicon dioxide, titanium oxide, zinc oxide, nickel oxide, copper oxide, or tin oxide, carbides such as silicon carbide, minerals such as agate and emery, ceramics such as ceramic microspheres, and diamond. The silica ( $\text{SiO}_2$ ) may comprise fumed silica which comprises silica that has been treated with a flame to form a three-dimensional structure. The fumed silica may be untreated or surface treated with a siloxane, such as, for example, polydimethylsiloxane. Exemplary non-limiting commercially available fumed silica includes products sold under the trade name AEROSIL®, such as AEROSIL® R 104, AEROSIL® R 106, AEROSIL® R 202, AEROSIL® R 208, AEROSIL® R 972 commercially available from Evonik Industries and products sold under the trade name HDK® such as HDK® H17 and HDK® H18 commercially available from Wacker Chemie AG. These fillers can also be surface modified, such as PYROKISUMA 5301K available from Kyowa Chemical Industry Co., Ltd. These thermally stable, TC/EI fillers may be used alone or in a combination of two or more.

**[0098]** Suitable thermally unstable TC/EI filler materials include metal hydroxides such as aluminum trihydrate, aluminum hydroxide or magnesium hydroxide. These fillers can also be surface modified, such as Hymod®M9400 SF available from J.M. Huber Corporation. These thermally unstable, TC/EI fillers may be used alone or in a combination of two or more.

**[0099]** As noted above, the thermally conductive filler package may comprise particles of TC/EC filler material.



**[0100]** Suitable TC/EC filler materials include metals such as silver, zinc, copper, gold, or metal coated hollow particles, carbon compounds such as, graphite (such as Timrex commercially available from Imerys or ThermoCarb commercially available from Asbury Carbons), carbon black (for example, commercially available as Vulcan from Cabot Corporation), carbon fibers (for example, commercially available as milled carbon fiber from Zoltek), graphene and graphenic carbon particles (for example, xGnP graphene nanoplatelets commercially available from XG Sciences, and/or for example, the graphene particles described below), carbonyl iron, copper (such as spheroidal powder commercially available from Sigma Aldrich), zinc (such as Ultrapure commercially available from Purity Zinc Metals and Zinc Dust XL and XLP available from US Zinc), and the like. Examples of “graphenic carbon particles” include carbon particles having structures comprising one or more layers of one-atom-thick planar sheets of sp<sup>2</sup>-bonded carbon atoms that are densely packed in a honeycomb crystal lattice. The average number of stacked layers may be less than 100, for example, less than 50. The average number of stacked layers may be 30 or less, such as 20 or less, such as 10 or less, such as 5 or less. The graphenic carbon particles may be substantially flat; however, at least a portion of the planar sheets may be substantially curved, curled, creased, or buckled. The particles typically do not have a spheroidal or equiaxed morphology. Suitable graphenic carbon particles are described in U.S. Publication No. 2012/0129980, at paragraphs [0059]-[0065], the cited portion of which is incorporated herein by reference. Other suitable graphenic carbon particles are described in U.S. Pat. No. 9,562,175, at 6:6 to 9:52, the cited portion of which are incorporated herein by reference. As used herein, the term “substantially flat” means planar; “curved” or “curled” materials deviate from planarity by having a non-zero curvature; and “creased” or “buckled” indicates that at least a portion of the area is thicker than one sheet, such that the plane is doubled or folded upon itself.

**[0101]** The TC/EC filler particles, if present at all, may be present in an amount of no more than 50% by volume based on total volume of the filler package, such as no more than 40% by volume, such as no more than 30% by volume, such as no more than 20% by volume, such as no more than 10% by volume, and may be present in an amount of at least 0.1% by volume based on total volume of the filler package, such as at least 0.5% by volume, such as at least 1% by volume, such as at least 5% by volume, such as at least 10% by volume. The TC/EC filler particles may be present in an amount of 0.1% by volume to 50% by volume based on total

volume of the filler package, such as 0.1% by volume to 40% by volume, such as 0.1% by volume to 30% by volume, such as 0.1% by volume to 20% by volume, such as 0.1% by volume to 10% by volume, such as 0.5% by volume to 50% by volume, such as 0.5% by volume to 40% by volume, such as 0.5% by volume to 30% by volume, such as 0.5% by volume to 20% by volume, such as 0.5% by volume to 10% by volume, such as 1% by volume to 50% by volume, such as 1% by volume to 40% by volume, such as 1% by volume to 30% by volume, such as 1% by volume to 20% by volume, such as 1% by volume to 10% by volume, such as 5% by volume to 50% by volume, such as 5% by volume to 40% by volume, such as 5% by volume to 30% by volume, such as 5% by volume to 20% by volume, such as 5% by volume to 10% by volume, such as 10% by volume to 50% by volume, such as 10% by volume to 40% by volume, such as 10% by volume to 30% by volume, such as 10% by volume to 20% by volume.

**[0102]** As noted above, the thermally conductive filler package may comprise particles of NTC/EI filler material.

**[0103]** Suitable NTC/EI filler materials include but are not limited to mica, wollastonite, calcium carbonate, glass microspheres, clay, or combinations thereof.

**[0104]** As used herein, the term “mica” generally refers to sheet silicate (phyllosilicate) minerals. The mica may comprise muscovite mica. Muscovite mica comprises a phyllosilicate mineral of aluminum and potassium with the formula  $KAl_2(AlSi_3O_{10})(F,OH)_2$  or  $(KF)_2(Al_2O_3)_3(SiO_2)_6(H_2O)$ . Exemplary non-limiting commercially available muscovite mica include products sold under the trade name DakotaPURE™, such as DakotaPURE™ 700, DakotaPURE™ 1500, DakotaPURE™ 2400, DakotaPURE™ 3000, DakotaPURE™ 3500 and DakotaPURE™ 4000, available from Pacer Minerals.

**[0105]** Wollastonite comprises a calcium inosilicate mineral ( $CaSiO_3$ ) that may contain small amounts of iron, aluminum, magnesium, manganese, titanium and/or potassium. The wollastonite may have a B.E.T. surface area of 1.5 to 2.1 m<sup>2</sup>/g, such as 1.8 m<sup>2</sup>/g and a median particle size of 6 microns to 10 microns, such as 8 microns. Non-limiting examples of commercially available wollastonite include NYAD 400 available from NYCO Minerals, Inc.

**[0106]** The calcium carbonate ( $CaCO_3$ ) may comprise a precipitated calcium carbonate or a ground calcium carbonate. The calcium carbonate may or may not be surface treated, such as treated with stearic acid. Non-limiting examples of commercially available precipitated calcium carbonate include Ultra-Pflex®, Albafil®, and Albacar HO® available from Specialty

Minerals and Winnofil® SPT available from Solvay. Non-limiting examples of commercially available ground calcium carbonate include Duramite™ available from IMERYS and Marblewhite® available from Specialty Minerals.

**[0107]** Useful clay minerals include a non-ionic platy filler such as talc, pyrophyllite, chlorite, vermiculite, or combinations thereof.

**[0108]** The glass microspheres may be hollow borosilicate glass. Non-limiting examples of commercially available glass microspheres include 3M Glass bubbles type VS, K series, and S series available from 3M.

**[0109]** The NTC/EI filler particles, if present at all, may be present in an amount of no more than 10% by volume based on total volume of the filler package, such as no more than 5% by volume, such as no more than 1% by volume, and may be present in an amount of at least 0.1% by volume based on total volume of the filler package, such as at least 0.5% by volume. The NTC/EI filler particles may be present in an amount of 0.1% by volume to 10% by volume based on total volume of the filler package, such as 0.5% by volume to 5% by volume, such as 0.5% by volume to 1% by volume.

#### Reactive Diluents

**[0110]** Optionally, the composition may comprise a reactive diluent. The reactive diluent may be a monomer or a polymer, and may be mono-functional, bi-functional, or multi-functional. The reactive diluent, in some instances, may be an adhesion promoter or a surface active agent. Suitable examples of reactive diluent include 1,4-butanediol diglycidyl ether (available as Heloxy modifier BD from Hexion), 1,6-hexanediol diglycidyl ether, mono-functional aliphatic diluents (Epotec RD 108, RD 109, RD 188 available from Aditya Birla), and mono-functional aromatic reactive diluents (Epotec RD 104, RD 105, and RD 136 available from Aditya Birla). Other suitable examples of the reactive diluent include saturated epoxidized oils, unsaturated oils such as glycerides of polyunsaturated fatty acids such as nut oils or seed oils, including as examples cashew nut oil, sunflower oil, safflower oil, soybean oil, linseed oil, castor oil, orange oil, rapeseed oil, tall oil, vegetable processing oil, vulcanized vegetable oil, high oleic acid sunflower oil, tung oil, and combinations thereof. The reactive diluent of the present invention also may be homopolymers of 1,2-butadiene or 1,4-butadiene or combinations thereof, copolymers of butadiene and acrylic or olefin monomers, or combinations thereof.

[0111] The reactive diluent may have a boiling point of greater than 100°C at 1 atm, such as greater than 130°C, such as greater than 150°C, for example, and the reactive diluent may have a boiling point of less than 425°C at 1 atm, such as less than 390°C, such as less than 360°C, for example.

[0112] The reactive diluent can lower the viscosity of the mixture. According to the present invention, the reactive diluent may have a viscosity of from 1 mPa·s to 4,000 mPa·s at 298K and 1 atm according to ASTM D789, such as for example, from 1 mPa·s to 3,000 mPa·s, 1 mPa·s to 2,000 mPa·s, 1 mPa·s to 1,000 mPa·s, 1 mPa·s to 100 mPa·s, or 2 mPa·s to 30 mPa·s.

#### Accelerators

[0113] Any accelerator capable of accelerating a reaction of the thiol-terminated compound and the oxidant may be used in the present invention. Suitable accelerators that may be used in accordance with the present invention thus include for example thiazoles, thiurams, sulfenamides, guanidines, dithiocarbamates, xanthates, thioureas, aldehydeamines, and combinations of any of the foregoing. Examples of suitable thiazoles include bis(2-benzothiazole) disulfide (MBTS), 2-mercaptobenzothiazole (MBT), and the zinc salt of mercaptobenzothiazole (ZMBT). Examples of suitable thiurams include tetramethyl thiuram monosulfide, tetramethyl thiuram disulfide (TMTD), tetraethyl thiuram disulfide, tetrabutyl thiuram disulfide, dipentamethylene thiuram hexasulfide, dicyclohexamethylene thiuram disulfide, diisopropyl thiuram disulfide, bis(morpholinothiocarbonyl) sulfide, tetramethyl thiuram monosulfide (TMTM), dipentamethylene thiuram tetrasulfide (DPTT), and compounds having the structure  $(R)_2N-C(=S)-S_x-C(=S)-N(R)_2$  where each R can be C<sub>1-6</sub> alkyl and x is an integer from 1 to 4, and combinations of any of the foregoing. Examples of suitable sulfenamides include *N*-cyclohexyl-2-benzothiazolsulfenamide, tertbutyl-2-benzothiazolsulfenamide (TBBS), dicyclohexyl-2-benzothiazolsulfenamide (DCBS), and combinations of any of the foregoing. Examples of suitable guanidines include dimethyl guanidine, trimethylguanidine, tetramethylguanidine, pentamethylguanidine, phenyl guanidine, diphenyl guanidine (DPG), butyl guanide, *N,N'*-diorthotolyl guanidine (DOTG), 1-*o*-tolylbiguanide, 1-phenylbiguanide, 1-methyl-3-nitroguanidine, 1,8-bis(tetramethylguanidino)-naphthalene, *N,N,N',N'*-tetramethyl-*N''*-[4-morpholinyl(phenylimino)methyl]guanidine, and compounds having the structure  $R-NH-C(=NH)-NH-R$  where each R is selected from C<sub>1-6</sub>

alkyl, phenyl and toluoyl, and combinations of any of the foregoing. Examples of suitable dithiocarbamates include zinc dialkyl dithiocarbamates such as dimethyl- dithiocarbamate (ZDMC), diethyl-dithiocarbamate (ZDEC) and dibutyl-dithiocarbamate (ZDBC), other metal or ammonium salts of dithiocarbamoic acid, compounds having the structure  $Zn(-S-C(=S)-N(R)_2)$  where each R is selected from C<sub>1-6</sub> alkyl, phenyl and toluoyl, and combinations of any of the foregoing. Examples of suitable xanthates include zinc salts of xanthic acid. Examples of suitable thioureas include ethylene thiourea (ETU), dipentamethylene thiourea (DPTU), dibutyl thiourea (DBTU), and compounds having the structure  $R-NH-C(=S)-NH-R$  where each R is selected from C<sub>1-6</sub> alkyl, phenyl and toluoyl, and combinations of any of the foregoing. Examples of suitable aldehydeamines include condensation products of aldehydes and amines, such as aniline, ammonia or their derivatives and also butyraldehyde, crotonylaldehyde or formaldehyde such as butyraldehydeaniline and tricrotonylidenetetramine, and combinations of any of the foregoing. Examples of other suitable cure accelerators include triazines and sulfides or metallic and amine salts of dialkyldithiophosphoric acids and dithiophosphates such as triazines and sulfides or metallic and amine salts of dialkyldithiophosphoric acids, and combinations of any of the foregoing. Examples of non-sulfur-containing polysulfide cure accelerators include sodium hydroxide (NaOH), water, and amines. Examples of amines include quaternary amines, tertiary amines, cyclic tertiary amines, or secondary amines. Some examples include N,N-bis(N,N-dimethyl-2-aminoethyl)methylamine; N,N-dimethylcyclohexylamine; N-methylmorpholine; N-ethylmorpholine; piperidine; piperazine; pyrrolidine; homopiperazine; 1,2-dimethyl-1,4,5,6-tetrahydropyrimidine; 1,4,5,6-tetrahydropyrimidine; 1,8-diazabicyclo[5.4.0]undec-7-ene; 1,5,7-triazabicyclo[4.4.0]dec-5-ene; 7-methyl-1,5,7-triazabicyclo[4.4.0]dec-5-ene; 1,5-diazabicyclo[4.3.0]non-5-ene; 6-(dibutylamino)-1,8-diazabicyclo(5,4,0)undec-7-ene; 1,4-diazabicyclo[2.2.2]octane; 7-azabicyclo[2.2.1]heptane; N,N-dimethylphenylamine; and 4,5-dihydro-1H-imidazole.

**[0114]** The accelerator may be present in the composition in an amount of at least 0.01% by volume based on the total volume of the composition, such as at least 0.02% by volume, such as at least 0.03% by volume, and may be present in an amount of no more than 30% by volume based on the total volume of the composition, such as no more than 20% by volume, such as no more than 10% by volume. The accelerator may be present in the composition in an amount of

0.01% to 30% by volume based on the total volume of the composition, such as 0.02% to 20% by volume, such as 0.03% to 10% by volume.

#### Dispersants

**[0115]** The composition optionally may further comprise a dispersant. As used herein, the term “dispersant” refers to a substance that may be added to the composition in order to improve the separation of the thermally conductive filler particles by wetting the particles and breaking apart agglomerates.

**[0116]** The dispersant, if present at all, may be present in the composition in an amount of at least 0.05% by volume based on total volume of the composition, such as at least 0.2% by volume, such as at least 1% by volume, and may be present in an amount of no more than 20% by volume based on total volume of the composition, such as no more than 10% by volume. The dispersant, if present at all, may be present in the composition in an amount of 0.05% by volume to 2 % by volume based on total volume of the composition, such as 0.2% by volume to 10% by volume, such as 1% by volume to 10% by volume.

**[0117]** Suitable dispersants for use in the composition include fatty acid, phosphoric acid esters, polyurethanes, polyamines, polyacrylates, polyalkoxylates, sulfonates, polyethers, and polyesters, or any combination thereof. Non-limiting examples of commercially available dispersants include ANTI-TERRA-U100, DISPERBYK-102, DISPERBYK-103, DISPERBYK-111, DISPERBYK-171, DISPERBYK-2151, DISPERBYK-2059, DISPERBYK-2000, DISPERBYK-2117, and DISPERBYK-2118 available from BYK Company; and SOLSPERSE 24000SC, SOLSPERSE 16000 and SOLSPERSE 8000 hyperdispersants available from The Lubrizol Corporation.

#### Additives

**[0118]** The composition may optionally comprise at least one additive. As used herein, an “additive” refers to a rheology modifier, a tackifier, a thermoplastic polymer, a surface active agent (other than the reactive diluent described above), a flame retardant, a corrosion inhibitor, a UV stabilizer, a colorant, a tint, a solvent, a plasticizer, an adhesion promoter (other than the reactive diluent described above), an antioxidant, a moisture scavenger, a silane, a silane terminated polymer, and/or a silyl terminated polymer.

**[0119]** Examples of suitable corrosion inhibitors include, for example, zinc phosphate-based corrosion inhibitors, for example, micronized Halox® SZP-391, Halox® 430 calcium phosphate, Halox® ZP zinc phosphate, Halox® SW-111 strontium phosphosilicate Halox® 720 mixed metal phosphor-carbonate, and Halox® 550 and 650 proprietary organic corrosion inhibitors commercially available from Halox. Other suitable corrosion inhibitors include Heucophos® ZPA zinc aluminum phosphate and Heucophos® ZMP zinc molybdenum phosphate, commercially available from Heucotech Ltd.

**[0120]** A corrosion inhibitor can comprise a lithium silicate such as lithium orthosilicate ( $\text{Li}_4\text{SiO}_4$ ) and lithium metasilicate ( $\text{Li}_2\text{SiO}_3$ ), MgO, an azole, or a combination of any of the foregoing. The corrosion inhibiting component may further comprise at least one of magnesium oxide (MgO) and an azole.

**[0121]** A corrosion inhibitor can comprise a monomeric amino acid, a dimeric amino acid, an oligomeric amino acid, or a combination of any of the foregoing. Examples of suitable amino acids include histidine, arginine, lysine, cysteine, cystine, tryptophan, methionine, phenylalanine, tyrosine, and combinations of any of the foregoing.

**[0122]** A corrosion inhibitor can comprise a nitrogen-containing heterocyclic compound. Examples of such compounds include azoles, oxazoles, thiazoles, thiazolines, imidazoles, diazoles, pyridines, indolizines, and triazines, tetrazoles, tolyltriazone, and combinations of any of the foregoing.

**[0123]** Examples of suitable triazoles include 1,2,3-triazole, 1,2,4-triazole, benzotriazole, derivatives thereof, and combinations of any of the foregoing. Derivatives of 1,2,3-triazole include 1-methyl-1,2,3-triazole, 1-phenyl-1,2,3-triazole, 4-methyl-2-phenyl-1,2,3-triazole, 1-benzyl-1,2,3-triazole, 4-hydroxy-1,2,3-triazole, 1-amino-1,2,3-triazole, 1-benzamido-4-methyl-1,2,3-triazole, 1-amino-4,5-diphenyl-1,2,3-triazole, 1,2,3-triazole aldehyde, 2-methyl-1,2,3-triazole-4-carboxylic acid, and 4-cyano-1,2,3-triazole, or combinations thereof. Derivatives of 1,2,4-triazole include 1-methyl-1,2,4-triazole, 1,3-diphenyl-1,2,4-triazole, 5-amino-3-methyl-1,2,4-triazole, 3-mercapto-1,2,4-triazole, 1,2,4-triazole-3-carboxylic acid, 1-phenyl-1,2,4-triazole-5-one, 1-phenylurazole, and combinations of any of the foregoing. Examples of diazoles include 2,5-dimercapto-1,3,4-thiadiazole.

**[0124]** A corrosion inhibitor can include an azole or combination of azoles. Azoles are five-membered *N*-heterocyclic compounds that contain in the heterocyclic ring two double

bonds, one to three carbon atoms and optionally a sulfur or oxygen atom. Examples of suitable azoles include benzotriazole, 5-methyl benzotriazole, tolyltriazole, 2,5-dimercapto-1,3,4-thiazole, 2-mercaptobenzothiazole, 2-mercaptobenzimidazole, 1-phenyl-5-mercaptotetrazole, 2-amino-5-mercapto-1,3,4-thiadiazole, 2-mercapto-1-methylimidazole, 2-amino-5-ethyl-1,3,4-thiadiazole, 2-amino-5-ethylthio-1,3,4-thiadiazole, 5-phenyltetrazole, 7*H*-imidazo(4,5-*d*)pyrimidine, and 2-amino thiazole. Salts of any of the foregoing, such as sodium and/or zinc salts, can also be used as effective corrosion inhibitors. Other suitable azoles include 2-hydroxybenzothiazole, benzothiazole, 1-phenyl-4-methylimidazole, and 1-(*p*-tolyl)-4-methylimidazole.

**[0125]** The rheology modifier may be present in the composition in an amount of at least 0.01% by volume based on total volume of the composition, such as at least 0.2% by volume, such as at least 0.3% by volume, and in some instances may be present in the composition in an amount of no more than 5% by volume based on total volume of the composition, such as no more than 3% by volume, such as no more than 1% by volume. The rheology modifier may be present in the composition in an amount of 0.01 % by volume to 5% by volume based on total volume of the composition, such as 0.2% by volume to 3% by volume, such as 0.3% by volume to 1% by volume.

**[0126]** Useful rheology modifiers that may be used include polyamide, amide waxes, polyether phosphate, oxidized polyolefin, Castor wax and organoclay. Commercially available thixotropes useful in the present invention include Disparlon 6500 available from King Industries, Garamite 1958 available from BYK Company, Bentone SD2 and Thxatrol@ST available from Elementis, and Crayvallac SLX available from Palmer Holland.

**[0127]** Useful colorants or tints may include phthalocyanine blue.

**[0128]** Compositions provided by the present disclosure can comprise a flame retardant or combination of flame retardants. Certain TC materials described above such as aluminum hydroxide and magnesium hydroxide, for example, also may be flame retardants. As used herein, "flame retardant" refers to a material that slows down or stops the spread of fire or reduces its intensity. Flame retardants may be available as a powder that may be mixed with a composition, a foam, or a gel. In examples, when the compositions of the present invention include a flame retardant, such compositions may form a coating on a substrate surface and such coating may function as a flame retardant.



**[0129]** As set forth in more detail below, a flame retardant can include a mineral, an organic compound, an organohalogen compound, an organophosphorous compound, or a combination thereof.

**[0130]** Suitable examples of minerals include huntite, hydromagnesite, various hydrates, red phosphorous, boron compounds such as borates, carbonates such as calcium carbonate and magnesium carbonate, and combinations thereof.

**[0131]** Suitable examples of organohalogen compounds include organochlorines such as chlorendic acid derivatives and chlorinated paraffins; organobromines such as decabromodiphenyl ether (decaBDE), decabromodiphenyl ethane (a replacement for decaBDE), polymeric brominated compounds such as brominated polystyrenes, brominated carbonate oligomers (BCOs), brominated epoxy oligomers (BEOs), tetrabromophthalic anhydride, tetrabromobisphenol A (TBBPA) and hexabromocyclododecane (HBCD). Such halogenated flame retardants may be used in conjunction with a synergist to enhance their efficiency. Other suitable examples include antimony trioxide, antimony pentoxide, and sodium antimonate.

**[0132]** Suitable examples of organophosphorous compounds include triphenyl phosphate (TPP), resorcinol bis(diphenylphosphate) (RDP), bisphenol A diphenyl phosphate (BADP), and tricresyl phosphate (TCP); phosphonates such as dimethyl methylphosphonate (DMMP); and phosphinates such as aluminium diethyl phosphinate. In one important class of flame retardants, compounds contain both phosphorus and a halogen. Such compounds include tris(2,3-dibromopropyl) phosphate (brominated tris) and chlorinated organophosphates such as tris(1,3-dichloro-2-propyl)phosphate (chlorinated tris or TDCPP) and tetrakis(2-chlorethyl)dichloroisopentyldiphosphate (V6).

**[0133]** Suitable examples of organic compounds include carboxylic acid, dicarboxylic acid, melamine, and organonitrogen compounds.

**[0134]** Other suitable flame retardants include ammonium polyphosphate and barium sulfate.

**[0135]** The composition optionally may comprise at least one plasticizer. Examples of plasticizers include diisononylphthalate (Jayflex™ DINP available from Exxon Mobil), diisodecylphthalate (Jayflex™ DIDP available from Exxon Mobil), and alkyl benzyl phthalate (Santicizer 278 available from Valtris); benzoate-based plasticizers such as

dipropylene glycol dibenzoate (K-Flex® available from Emerald Performance Materials); and other plasticizers including terephthalate-based dioctyl terephthalate (DEHT available from Eastman Chemical Company), alkylsulfonic acid ester of phenol (Mesamoll available from Borchers), and 1,2-cyclohexane dicarboxylic acid diisononyl ester (Hexamoll DINCH available from BASF). Other plasticizers may include isophthalic hydrogenated terphenyls, quarterphenyls and higher or polyphenyls, phthalate esters, chlorinated paraffins, modified polyphenyl, tung oil, naphthalene sulfonates, trimellitates, adipates, sebacates, maleates, sulfonamide, organophosphates, polybutene, and combinations of any of the foregoing. These plasticizers can be polymers such as polyacrylates.

**[0136]** The plasticizer, if present at all, may be present in the composition in an amount of at least 0.5% by volume based on the total volume of the composition, such as at least 2% by volume, such as at least 3% by volume, and may be present in an amount of no more than 30% by volume based on total volume of the composition, such as no more than 20% by volume, such as no more than 16% by volume. The plasticizer, if present at all, may be present in the composition in an amount of 0.5% to 30% by volume based on total volume of the composition, such as 2% to 20% by volume, such as 3% to 16% by volume.

**[0137]** Suitable moisture scavengers include vinyltrimethoxysilane (Silquest A-171 from Momentive), vinyltriethoxysilane (Silquest A-151NT from Momentive), gamma-methacryloxypropyltrimethoxysilane (Silquest A-174NT available from Evonik), molecular sieves, calcium oxide (POLYCAL OS325 available from Mississippi Lime), or combinations thereof.

**[0138]** The composition also may comprise a solvent. Suitable solvents include toluene, acetone, ethyl acetate, methyl ethyl ketone, benzene, n-hexane, xylene, and combinations thereof.

**[0139]** The solvent, if present at all, may be present in the composition in an amount of at least 1% by volume based on the total volume of the composition, such as at least 2% by volume, such as at least 5% by volume, and may be present in an amount of no more than 60% by volume, such as no more than 40% by volume, such as no more than 20% by volume. The solvent, if present at all, may be present in the composition in an amount of 1% to 60% by volume based on total volume of the composition, such as 2% to 40% by volume, such as 5% to 20% by volume.

[0140] The skilled person will understand to select a particular antioxidant and/or an amount of antioxidant in order to avoid competition with the oxidants discussed above, thereby avoiding any detrimental effects to curing the composition.

### Compositions

[0141] Optionally, the compositions disclosed herein may be substantially free, or essentially free, or completely free, of epoxy-containing compounds. As used herein, a composition is “substantially free” of an epoxy-containing compound if the epoxy-containing compound is present in an amount of less than 0.001% by weight of the total weight of the composition. As used herein, a composition is “essentially free” of an epoxy-containing compound if the epoxy-containing compound is present in an amount of less than 0.0005% by weight of the total weight of the composition. As used herein, a composition is “completely free” of an epoxy-containing compound if the epoxy-containing compound is not present in the composition, i.e., 0.0000% by weight.

[0142] The composition of the present invention may comprise, or consist essentially of, or consist of, a thiol-terminated compound, an oxidant, and a thermally conductive filler package as described above, and optionally an accelerator and/or a dispersant. As used herein, the composition “consists essentially of” a thiol-terminated compound, an oxidant, a thermally conductive filler package as described above, and optionally an accelerator and/or a dispersant, when the maximum amount of other components is 5% by volume or less based on total volume of the composition.

[0143] The compositions disclosed herein may be 1K compositions comprising, or consisting essentially of, or consisting of, a thiol-terminated compound, an oxidant, and a thermally conductive filler package, and optionally an accelerator and/or a dispersant and/or any of the additives described hereinabove.

[0144] In examples, the compositions of the present invention may be provided as pre-mixed frozen compositions (PMF). PMFs may be cured by external factors, such as temperature. In examples, the PMF may be stored at temperatures between and including -100°C and -15°C, such as -100°C to -25°C, to retard hardening, such as at a minimum of -75°C, such as at a maximum of -40°C. When applying the composition to the substrate, the composition may be exposed to temperatures sufficient to at least partially cure the composition.

**[0145]** In the examples described above, the composition may be at least partially cured by the addition of heat. In another non-limiting example, the composition may be at least partially cured by oxidation. The term “inhibiting” refers to restraining, impeding, slowing or interfering with a particular reaction or function. This can be accomplished in a number of ways, for example, controlling the environment to which the composition is exposed. In the case of oxidation, inhibiting refers to restraining, impeding, slowing or interfering with the oxidation of the thiol-terminated compound. In a non-limiting example, oxidation is at least partially inhibited by limiting the composition’s exposure to air or ambient conditions.

**[0146]** PMFs may be packaged, for example, in a cartridge, a cartridge and plunger, a syringe, or may be supplied as a tape, a cap, or any pre-formed geometry.

**[0147]** The compositions disclosed herein may be 1K compositions comprising, or consisting essentially of, or consisting of, a thiol-terminated compound, an oxidant, and a thermally conductive filler package, and optionally an accelerator and/or a dispersant and/or any of the additives described herein above. It has been surprisingly discovered that the 1K coating compositions of the present invention are workable for at least 10 days, such as at least 20 days, such as at least 30 days, when stored at ambient conditions. It also has been surprising discovered that the PMF compositions of the present invention are workable for at least 10 days, such as at least 20 days, such as at least 30 days, when thawed and stored at ambient conditions.

**[0148]** The compositions disclosed herein may be 2K compositions comprising, or consisting essentially of, or consisting of, a first component comprising, or consisting essentially of, or consisting of, a thiol-terminated compound, a second component comprising, or consisting essentially of, or consisting of, a oxidant, and a thermally conductive filler package that may be present in the first component and/or the second component, and optionally an accelerator and/or a dispersant and/or any of the additives described herein above may be present in the first component and/or the second component. The first and second components are mixed together immediately prior to use.

**[0149]** The compositions disclosed herein may be 3K or higher compositions comprising, or consisting essentially of, or consisting of, a first component comprising, or consisting essentially of, or consisting of, a thiol-terminated compound, a second component comprising, or consisting essentially of, or consisting of, an oxidant, and a third component comprising, or consisting essentially of, or consisting of, a thermally conductive filler package, and optionally

an accelerator and/or a dispersant and/or any of the additives described herein above may be present in the first component and/or the second component and/or the third component.

**[0150]** The composition may have a total solids content of at least 40% by volume based on total volume of the composition, such as at least 60%, such as at least 80% by volume, and may have a total solids content of no more than 100% by volume based on total volume of the composition. The composition may have a total solids content of 40% to 100% by volume based on total volume of the composition, such as 60% to 100% by volume, such as 80% to 100% by volume. As used herein, “total solids” refers to the non-volatile content of the composition, i.e., materials which will not volatilize when heated to 110°C and standard atmospheric pressure (101325 Pa) for 60 minutes.

**[0151]** In the case of a 2K composition, one of the components may be substantially free, or essentially free, or completely free, of filler materials, and in the case of a 3K composition, one or two of the components may be substantially free, or essentially free, or completely free, of filler materials.

**[0152]** The composition may be a low-VOC composition. As used herein, the term “low-VOC” refers to a composition having a theoretical VOC volume % of less than 7% by volume, such as less than 3% by volume, such as less than 2% by volume, based on total volume of the composition. The theoretical volatile organic content (“VOC”) may be less than 105 g/L, such as less than 75 g/L, such as less than 30 g/L. As used herein, the term “volatile organic compounds” or “VOC” means any compound of carbon, excluding carbon monoxide, carbon dioxide, carbonic acid, metallic carbides or carbonates, and ammonium carbonate, which participates in atmospheric photochemical reactions.

#### Methods and Systems

**[0153]** The present invention may also be a method for preparing a composition comprising, or in some cases consisting of, or in some cases consisting essentially of, a thiol-terminated compound, an oxidant that reacts with the thiol-terminated compound, a thermally conductive filler package, and optionally an accelerator, and optionally a dispersant, and any of the optional further components, if used, described above, the method comprising, or in some cases consisting of, or in some cases consisting essentially of, mixing such ingredients at a

temperature of less than 50°C, such as from 0°C to 50°C, such as from 15°C to 35°C, such as at ambient temperature.

**[0154]** The composition described above may be applied alone or as part of a system that can be deposited in a number of different ways onto a number of different substrates. The system may comprise a number of the same or different films, coatings, or layers. A film, coating, or layer is typically formed when a composition that is deposited onto at least a portion of the substrate surface is at least partially dried or cured by methods known to those of ordinary skill in the art (e.g., under ambient conditions or by exposure to thermal heating).

**[0155]** The composition can be applied to the surface of a substrate in any number of different ways, non-limiting examples of which include brushes, rollers, films, pellets, trowels, spatulas, dips, spray guns, applicator guns, and pneumatic guns to form a coating on at least a portion of the substrate surface.

**[0156]** After application to the substrate(s), the composition may be cured. For example, the composition may be allowed to cure at room temperature or slightly thermal conditions (e.g., above ambient temperature to 70°C) and/or the composition may be cured by baking and/or curing at elevated temperature, such as at a temperature of 180°C or below, such as 130°C or below, such as 110°C or below, such as 100°C or below, such as 90°C or below, such as 80°C or below, such as 70°C or below, but greater than ambient, such as greater than 40°C, such as greater than 50°C, and for any desired time period (e.g., from 5 minutes to 24 hours) sufficient to at least partially cure the composition on the substrate(s). Alternatively, the composition of the present invention may cure at ambient or slightly above ambient conditions.

**[0157]** The present invention also is directed to a method for treating a substrate comprising, or consisting essentially of, or consisting of, contacting at least a portion of a surface of the substrate with one of the compositions of the present invention described hereinabove. The composition may be cured to form a coating, layer or film on the substrate surface under ambient conditions or by exposure to an external energy source, for example such as by heating the substrate to a temperature of less than 180°C, such as less than 130°C, such as less than 90°C. The coating, layer or film, may be, for example, a sealant, a putty, a molding compound, a potting compound, a gap filler, or an adhesive.

**[0158]** The present invention is also directed to a method for forming a bond between two substrates for a wide variety of potential applications in which the bond between the

substrates provides particular mechanical properties related to lap shear strength. The method may comprise, or consist essentially of, or consist of, applying the composition described above to a first substrate; contacting a second substrate to the composition such that the composition is located between the first substrate and the second substrate; and curing the composition under ambient conditions or by exposure to an external energy source, for example such as by heating to a temperature of less than 180°C, such as less than 130°C, such as less than 90°C. For example, the composition may be applied to either one or both of the substrate materials being bonded to form an adhesive bond therebetween and the substrates may be aligned and pressure and/or spacers may be added to control bond thickness. The composition may be applied to cleaned or uncleaned (i.e., including oily or oiled) substrate surfaces.

**[0159]** As stated above, the composition of the present disclosure also may form a sealant on a substrate or a substrate surface. The sealant composition may be applied to substrate surfaces, including, by way of non-limiting example, a vehicle body or components of an automobile frame or an airplane. The sealant formed by the composition of the present invention provides sufficient sound damping, tensile strength and tensile elongation. The sealant composition may be applied to cleaned or uncleaned (i.e., including oily or oiled) substrate surfaces. It may also be applied to a substrate that has been pretreated, coated with an electrodepositable coating, coated with additional layers such as a primer, basecoat, or topcoat. The coating composition may dry or cure at ambient conditions once applied to a substrate or substrates coated with coating compositions may optionally subsequently be baked in an oven to cure the coating composition.

**[0160]** The composition may be injected or otherwise placed in a die caster or a mold and at least partially dried or cured under ambient conditions or by exposure to moisture or water, and optionally additionally by heating to a temperature of less than 180°C, such as less than 130°C, such as less than 90°C to form a part or a member and optionally may be machined to a particular configuration.

#### Substrates

**[0161]** The substrates that may be coated by the compositions of the present invention are not limited. Suitable substrates useful in the present invention include, but are not limited to, materials such as metals or metal alloys, polymeric materials such as hard plastics including filled and unfilled thermoplastic materials or thermoset materials, or composite materials. For

example, suitable substrates include rigid metal substrates such as ferrous metals, aluminum, aluminum alloys, magnesium titanium, copper, and other metal and alloy substrates. The ferrous metal substrates used in the practice of the present invention may include iron, steel, and alloys thereof. Non-limiting examples of useful steel materials include cold rolled steel, galvanized (zinc coated) steel, electrogalvanized steel, stainless steel, pickled steel, zinc-iron alloy such as GALVANNEAL, and combinations thereof. Combinations or composites of ferrous and non-ferrous metals can also be used. Aluminum alloys of the 1XXX, 2XXX, 3XXX, 4XXX, 5XXX, 6XXX, 7XXX, or 8XXX series as well as clad aluminum alloys and cast aluminum alloys of the A356, 1XX.X, 2XX.X, 3XX.X, 4XX.X, 5XX.X, 6XX.X, 7XX.X, or 8XX.X series also may be used as the substrate. Magnesium alloys of the AZ31B, AZ91C, AM60B, or EV31A series also may be used as the substrate. The substrate used in the present invention may also comprise titanium and/or titanium alloys of grades 1-36 including H grade variants. Other suitable non-ferrous metals include copper and magnesium, as well as alloys of these materials. In examples, the substrate may be a multi-metal article. As used herein, the term “multi-metal article” refers to (1) an article that has at least one surface comprised of a first metal and at least one surface comprised of a second metal that is different from the first metal, (2) a first article that has at least one surface comprised of a first metal and a second article that has at least one surface comprised of a second metal that is different from the first metal, or (3) both (1) and (2). Suitable metal substrates for use in the present invention include those that are used in the assembly of vehicular bodies (e.g., without limitation, door, body panel, trunk deck lid, roof panel, hood, roof and/or stringers, rivets, landing gear components, and/or skins used on an aircraft), a vehicular frame, vehicular parts, motorcycles, wheels, and industrial structures and components. As used herein, “vehicle” or variations thereof includes, but is not limited to, civilian, commercial and military aircraft, and/or land vehicles such as cars, motorcycles, and/or trucks. The metal substrate also may be in the form of, for example, a sheet of metal or a fabricated part. It will also be understood that the substrate may be pretreated with a pretreatment solution including a zinc phosphate pretreatment solution such as, for example, those described in U.S. Patent Nos. 4,793,867 and 5,588,989, or a zirconium containing pretreatment solution such as, for example, those described in U.S. Patent Nos. 7,749,368 and 8,673,091. It will be understood the substrate may also be anodized, primed, organic-coated or chromate-coated. Other substrates may comprise epoxy, urethane, graphite, Kevlar®, acrylics,



polycarbonates, a composite material such as a plastic or a fiberglass composite. The substrate may be a fiberglass and/or carbon fiber composite. The compositions of the present invention are particularly suitable for use in various industrial or transportation applications including automotive, light and heavy commercial vehicles, marine, or aerospace.

#### Extrusion and the Like

**[0162]** Alternatively, the composition may be casted, extruded, molded, or machined to form a part or a member in at least partially dried or cured state.

**[0163]** The compositions disclosed herein surprisingly may be used in any suitable additive manufacturing technology, such as extrusion, jetting, and binder jetting.

**[0164]** The present disclosure is directed to the production of structural articles, such as by way of non-limiting example, sound damping pads, using three-dimensional printing. A three-dimensional article may be produced by forming successive portions or layers of an article by depositing the composition of the present invention onto a substrate and thereafter depositing additional portions or layers of the composition over the underlying deposited portion or layer and/or adjacent the previously deposited portion or layer. Layers can be successively deposited adjacent a previously deposited layer to build a printed article. First and second components of the composition can be mixed and then deposited or the first and second components of the composition can be deposited separately. When deposited separately, the first and second components can be deposited simultaneously, sequentially, or both simultaneously and sequentially.

**[0165]** By “portions of an article” is meant subunits of an article, such as layers of an article. The layers may be on successive horizontal parallel planes. The portions may be parallel planes of the deposited material or beads of the deposited material produced as discrete droplets or as a continuous stream of material. The first and second components may each be provided neat or may also include an organic solvent and/or other additives as described below. First and second components provided by the present disclosure may be substantially free of solvent. By substantially free is meant that the first and second components comprise less than 5% by volume, less than 4% by volume, less than 2% by volume, or less than 1% by volume of solvent, where % by volume is based on the total volume of the first component or the second component, as the case may be. Similarly, the composition provided by the present disclosure may be substantially free of solvent, such as having less than 5% by volume, less than 4% by

volume, less than 2% by volume, or less than 1% by volume of solvent, where % by volume is based on the total volume of the composition.

**[0166]** The first and second components of a 2K composition of the present invention may be mixed together and subsequently deposited as a mixture of components that react to form portions of an article. For example, two components may be mixed together and deposited as a mixture of components that react to form a thermoset by delivery of at least two separate streams of the components into a mixer such as a static mixer and/or a dynamic mixer to produce a single stream that is then deposited. The components may be at least partially reacted by the time a composition comprising the reaction mixture is deposited. The deposited reaction mixture may react at least in part after deposition and may also react with previously deposited portions and/or subsequently deposited portions of the article such as underlying layers or overlying layers of the article.

**[0167]** Two or more components can be deposited using any suitable equipment. The selection of suitable deposition equipment depends on a number of factors including the deposition volume, the viscosity of the composition and the complexity of the part being fabricated. Each of the two or more components can be introduced into an independent pump and injected into a mixer to combine and mix the two components. A nozzle can be coupled to the mixer and the mixed composition can be pushed under pressure or extruded through the nozzle.

**[0168]** A pump can be, for example, a positive displacement pump, a syringe pump, a piston pump, or a progressive cavity pump. The two pumps delivering the two components can be placed in parallel or placed in series. A suitable pump can be capable of pushing a liquid or viscous liquid through a nozzle orifice. This process can also be referred to as extrusion. A component can be introduced into the mixer using two pumps in series.

**[0169]** For example, the first and second components can be deposited by dispensing materials through a disposable nozzle attached to a progressive cavity two-component dosing system such as a ViscoTec eco-DUO 450 precision dosing system, where the first and second components are mixed in-line. A two-component dosing system can comprise, for example, two progressive cavity pumps that separately dose reactants into a disposable static mixer dispenser or into a dynamic mixer. Other suitable pumps include positive displacement pumps, syringe pumps, piston pumps, and progressive cavity pumps. Upon dispensing, the materials of the first

and second components form an extrudate which can be deposited onto a surface to provide an initial layer of material and successive layers on a base. The deposition system can be positioned orthogonal to the base, but also may be set at any suitable angle to form the extrudate such that the extrudate and deposition system form an obtuse angle with the extrudate being parallel to the base. The extrudate refers to the combined components, i.e., a composition, that have been mixed, for example, in a static mixer or in a dynamic mixer. The extrudate can be shaped upon passing through a nozzle.

**[0170]** The base, the deposition system, or both the base and the deposition system may be moved to build up a three-dimensional article. The motion can be made in a predetermined manner, which may be accomplished using any suitable CAD/CAM method and apparatus such as robotics and/or computerized machine tool interfaces.

**[0171]** An extrudate may be dispensed continuously or intermittently to form an initial layer and successive layers. For intermittent deposition, a dosing system may interface with a relay switch to shut off the pumps, such as the progressive cavity pumps and stop the flow of reactive materials. Any suitable switch such as an electromechanical switch that can be conveniently controlled by any suitable CAD/CAM methodology can be used.

**[0172]** A deposition system can include an in-line static and/or dynamic mixer as well as separate pressurized pumping compartments to hold the at least two components and feed the materials into the static and/or dynamic mixer. A mixer such as an active mixer can comprise a variable speed central impeller having high shear blades within a conical nozzle. A range of conical nozzles may be used which have an exit orifice dimension, for example, from 0.2 mm to 50 mm, from 0.5 mm to 40 mm, from 1 mm to 30 mm, or from 5 mm to 20 mm.

**[0173]** A range of static and/or dynamic mixing nozzles may be used which have, for example, an exit orifice dimension from 0.6 mm to 2.5 mm, and a length from 30 mm to 150 mm. For example, an exit orifice diameter can be from 0.2 mm to 4.0 mm, from 0.4 mm to 3.0 mm, from 0.6 mm to 2.5 mm, from 0.8 mm to 2 mm, or from 1.0 mm to 1.6 mm. A static mixer and/or dynamic can have a length, for example, from 10 mm to 200 mm, from 20 mm to 175 mm, from 30 mm to 150 mm, or from 50 mm to 100 mm. A mixing nozzle can include a static and/or dynamic mixing section and a dispensing section coupled to the static and/or dynamic mixing section. The static and/or dynamic mixing section can be configured to combine and mix the first and second components. The dispensing section can be, for example, a straight tube

having any of the above orifice diameters. The length of the dispensing section can be configured to provide a region in which the components can begin to react and build viscosity before being deposited on the article. The length of the dispensing section can be selected, for example, based on the speed of deposition, the rate of reaction of the first and second components, and the desired viscosity.

**[0174]** First and second components can have a residence time in the static and/or dynamic mixing nozzle, for example, from 0.25 seconds to 5 seconds, from 0.3 seconds to 4 seconds, from 0.5 seconds to 3 seconds, or from 1 seconds to 3 seconds. Other residence times can be used as appropriate based on the curing chemistries and curing rates.

**[0175]** In general, a suitable residence time is less than the gel time of the composition. A suitable gel time can be less than 7 days, such as less than 3 days, such as less than 2 days. A gel time of the composition can be, for example, from 10 min to 7 days, such as 12 hours to 3 days, such as 24 hours to 2 days. Gel time is considered as the time following mixing when the composition is no longer stirrable by hand. As used herein, “gel time” refers to the time it takes for a resin system, once all ingredients are combined or mixed together, to gel or become so viscous that it is no longer liquid.

**[0176]** Compositions provided by the present disclosure can have a volume flow rate, for example, from 0.1 mL/min to 20,000 mL/min, such as from 1 mL/min to 12,000 mL/min, from 5 mL/min to 8,000 mL/min, or from 10 mL/min to 6,000 mL min. The volume flow rate can depend, for example, on the viscosity of the composition, the extrusion pressure, the nozzle diameter, and the reaction rate of the first and second components.

**[0177]** A composition can be used at a print speed, for example, from 1 mm/sec to 400 mm/sec, such as from 5 mm/sec to 300 mm/sec, from 10 mm/sec to 200 mm/sec, or from 15 mm/sec to 150 mm/sec. The printed speed can depend, for example, on the viscosity of the composition, the extrusion pressure, the nozzle diameter, and the reaction rate of the components. The print speed refers to the speed at which a nozzle used to extrude a composition moves with respect to a surface onto which the composition is being deposited.

**[0178]** A static and/or dynamic mixing nozzle can be heated or cooled to control, for example, the rate of reaction between the first and second components and/or the viscosity of the first and second components. An orifice of a deposition nozzle can have any suitable shape and dimensions. A system can comprise multiple deposition nozzles. The nozzles can have a fixed

orifice dimension and shape, or the nozzle orifice can be controllably adjusted. The mixer and/or the nozzle may be cooled to control an exotherm generated by the reaction of the first and second components.

**[0179]** Methods provided by the present disclosure include printing the composition on a fabricated part. Methods provided by the present disclosure include directly printing parts.

**[0180]** Using the methods provided by the present disclosure parts can be fabricated. The entire part can be formed from one of the compositions disclosed herein, one or more portions of a part can be formed from one of the compositions disclosed herein, one or more different portions of a part can be formed using the compositions disclosed herein, and/or one or more surfaces of a part can be formed from a composition provided by the present disclosure. In addition, internal regions of a part can be formed from a composition provided by the present disclosure.

**[0181]** FIGS. 1 and 2 are schematic perspective views illustrating a thermally conductive member utilized as a gap filler in a battery pack **100**. As illustrated in FIG. 1, the thermally conductive matter **10** (formed from the compositions described herein in an at least partially cured state) is positioned between two battery cells/battery modules **50** which are interconnected in series or in parallel by interconnects (not shown). In other examples (FIG. 1), the thermally conductive matter may be positioned between cooling fin **30** and/or a battery cell / battery module **50**, between battery modules **50**, between a battery cell / battery module **50** and a surface of a wall of a battery box **20** or may be applied as a coating on at least a portion of the substrate of a wall of a battery box **20**. As shown in FIG. 2, the thermally conductive matter **10** may be positioned between a cooling plate **40** and a battery cell / battery module **50**. The battery pack may further comprise a thermal management system (not shown) comprising air or fluid circuits, which may be liquid based (for example glycol solutions) or direct refrigerant based.

#### Coatings and Formed Parts

**[0182]** The composition of the present invention, in an at least partially dried or cured state (i.e., a layer, coating, or film), surprisingly may:

(a) have a thermal conductivity of at least  $0.4 \text{ W/m}\cdot\text{K}$  (measured according to ASTM D7984), such as at least  $0.5 \text{ W/m}\cdot\text{K}$ , such as at least  $0.7 \text{ W/m}\cdot\text{K}$ ;

(b) have a dielectric strength of at least 1kV/mm measured according to ASTM D149 on a dielectric meter (Sefelec RMG12AC-DC) connected to two copper electrodes with 1 inch diameter, such as at least 1.5 kV/mm;

(c) have a Shore A hardness of at least 5 measured according to ASTM D2240 with a Type A durometer (Model 2000, Rex Gauge Company, Inc.) at room temperature, such as at least 10, such as at least 20, such as at least 30, such as at least 40, such as 5 to 95;

(d) have a 180° peel strength of at least 1 lbf/in (measured according to SAE AS5127/1B with an Instron universal testing machine, model 3345), such as at least 2 lbf/in, such as at least 3 lbf/in, such as at least 4 lbf/in;

(e) have a tensile stress at break of at least 1 MPa as determined according to ISO 37-2 using an Instron universal testing machine, model 3345 with a pull rate of 2 inches/min, such as at least 10 MPa, such as at least 25 MPa, such as at least 30 MPa, such as 1 MPa to 100 MPa;

(f) have an elongation of 1% to 900%, as determined according to ISO 37-2 using an Instron universal testing machine, model 3345 with a pull rate of 2 inches/min, such as at least 100%, such as at least 200%, such as at least 300%, such as at least 400%, such as at least 500%, such as at least 600%, such as at least 700%, such as at least 800%;

(g) have a Young's modulus of 0.01 MPa to 100 MPa, as determined according to ISO 37-2 using an Instron universal testing machine, model 3345 with a pull rate of 2 inches/min, such as at least 0.1 MPa, such as at least 0.7 MPa, such as at least 1 MPa, such as at least 10 MPa;

(h) maintain a temperature of the substrate that is at least 100°C lower following exposure of the coating on the surface of the substrate to 1000°C for a time of at least 90 seconds than a surface temperature of a bare substrate exposed to 1000°C for the time;

(i) not smoke upon exposure of the substrate to 1000°C for 500 sec; and/or

(j) exhibit no visible cracking or delamination upon exposure of the substrate to 1000°C for 500 sec.

**[0183]** Such coatings and/or formed parts may be formed from the compositions of the present invention.

**[0184]** In examples, coatings and the like and parts formed from the compositions of the present invention surprisingly may, in an at least partially cured state, have a thermal conductivity of at least 0.5 W/m·K (measured according to ASTM D7984), maintain a

temperature of the substrate that is at least 100°C lower following exposure of the coating on the surface of the substrate to 1000°C for at a time of at least 90 seconds than a surface temperature of a bare substrate exposed to 1000°C for the time, provide a substrate with thermal and fire protection, not smoke upon exposure of the substrate to 1000°C for 500 sec, and/or exhibit no visible cracking or delamination.

**[0185]** In examples, the compositions of the present invention surprisingly may be used for making a coating that, in at least partially cured state, have a thermal conductivity of at least 0.5 W/m·K (measured according to ASTM D7984), maintain a temperature of the substrate that is at least 100°C lower following exposure of the coating on the surface of the substrate to 1000°C for at a time of at least 90 seconds than a surface temperature of a bare substrate exposed to 1000°C for the time, provide a substrate with thermal and fire protection, not smoke upon exposure of the substrate to 1000°C for 500 sec, and/or exhibit no visible cracking or delamination.

**[0186]** Coatings and the like formed from compositions of the present invention may be used to provide a substrate with thermal and fire protection.

**[0187]** The coating compositions of the present invention may be used to make a coating having, in an at least partially cured state, a thermal conductivity of at least 0.5 W/m·K (measured according to ASTM D7984) and maintaining a temperature of the substrate that is at least 100°C lower following exposure of the coating on the surface of the substrate to 1000°C for at a time of at least 90 seconds than a surface temperature of a bare substrate exposed to 1000°C for the time.

**[0188]** The coating compositions of the present invention also may be used to make a coating that, in an at least partially cured state, provides a substrate with thermal and fire protection.

**[0189]** The coating compositions of the present invention also may be used to make a coating that, in an at least partially cured state, may not smoke upon exposure of the substrate to 1000°C for 500 sec.

**[0190]** The coating compositions of the present invention also may be used to make a coating that, in an at least partially cured state, exhibits no visible cracking or delamination.

**[0191]** Also disclosed are coatings that, in at least partially cured state, have a thermal conductivity of at least 0.5 W/m·K (measured according to ASTM D7984), maintain a

temperature of the substrate that is at least 100°C lower following exposure of the coating on the surface of the substrate to 1000°C for a time of at least 90 seconds than a surface temperature of a bare substrate exposed to 1000°C for the time, provide a substrate with thermal and fire protection, not smoke upon exposure of the substrate to 1000°C for 500 sec, and/or exhibit no visible cracking or delamination.

**[0192]** Coatings and the like formed from compositions of the present invention may be used to provide a substrate with thermal and fire protection.

**[0193]** As used herein, a “temperature of the substrate following exposure of the coating on the surface of the substrate to elevated temperatures such as 1000°C for a time” may be measured by applying a coating composition to a substrate surface and allowing such composition to cure (for example, for 2 days in an environmental chamber (50 % RH, 25°C) followed by 1 day at 140°F (60°C)). When the composition is at least partially cured, a thermocouple may be attached at a center point of the substrate to which the coating composition was applied to monitor the temperature through the coating. In order to determine the temperature at the back of the coated substrate, the center of the coated substrate may be positioned at a distance of 4 cm from a propane torch (diameter 3.5 cm, propane) with the coating in the direction of the torch. The temperature of the flame may be monitored through a second thermocouple placed close to the base of the flame.

**[0194]** As used herein, “thermal protection” of a substrate refers to a coating that has a thermal conductivity of at least 0.5 W/m·K (measured according to ASTM D7984).

**[0195]** As used herein, “fire protection” of a substrate refers to a coating that prevents a substrate from reaching its critical temperature or extends the time until the substrate reaches its critical temperature, and “critical temperature” means approximately the temperature where the substrate has lost approximately 50% of its yield strength from that at room temperature.

**[0196]** As used herein, “cracking and delamination” refers to an interruption of a coating such that at least a portion of the substrate surface is exposed.

### **ASPECTS**

**[0197]** In view of the foregoing the present invention thus relates inter alia, without being limited thereto, to the following aspects:

**[0198]** 1. A composition, comprising:



a thiol-terminated compound;  
an oxidant; and

a thermally conductive filler package comprising thermally conductive, electrically insulative filler particles, the thermally conductive, electrically insulative filler particles having a thermal conductivity of at least 5 W/m·K (measured according to ASTM D7984) and a volume resistivity of at least 1  $\Omega\cdot\text{m}$  (measured according to ASTM D257).

**[0199]** 2. The composition of Aspect 1, wherein the thermally conductive filler package is present in an amount of 15% by volume to 90% by volume based on total volume of the composition.

**[0200]** 3. The composition of Aspect 1 or Aspect 2, wherein the thiol-terminated compound comprises a liquid.

**[0201]** 4. The composition of any of the preceding Aspects, wherein the thiol-terminated compound has a weight average molecular weight of 80 g/mol to 40,000 g/mol.

**[0202]** 5. The composition of any of the preceding Aspects, wherein the thiol-terminated compound is present in an amount of 9.9% by volume to 84.9% by volume based on total volume of the composition.

**[0203]** 6. The composition of any of the preceding Aspects, wherein the oxidant comprises a metal oxide and/or an organic peroxide.

**[0204]** 7. The composition of Aspect 6, wherein the metal oxide comprises lead oxide, lead peroxide, manganese dioxide, sodium perborate, potassium permanganate, calcium peroxide, barium peroxide, lithium peroxide, zinc peroxide, barium oxide, or combinations thereof.

**[0205]** 8. The composition of Aspect 6 or Aspect 7, wherein the organic peroxide comprises cumene hydroperoxide, t-butyl hydroperoxide, or combinations thereof.

**[0206]** 9. The composition of any of the preceding Aspects, wherein the oxidant is present in an amount of 0.1% by volume to 10% by volume based on total volume of the composition.

**[0207]** 10. The composition of any of the preceding Aspects, wherein the thermally conductive, electrically insulative filler particles are present in an amount of at least 50% by volume based on total volume of the filler package.

**[0208]** 11. The composition of any of Aspects 1 to 9, wherein the thermally conductive, electrically insulative filler particles are present in an amount of at least 60% by volume based on total volume of the filler package.

**[0209]** 12. The composition of any of Aspects 1 to 9, wherein the thermally conductive, electrically insulative filler particles are present in an amount of at least 70% by volume based on total volume of the filler package.

**[0210]** 13. The composition of any of Aspects 1 to 9, wherein the thermally conductive, electrically insulative filler particles are present in an amount of at least 8% by volume based on total volume of the filler package.

**[0211]** 14. The composition of any of Aspects 1 to 9, wherein the thermally conductive, electrically insulative filler particles are present in an amount of at least 90% by volume based on total volume of the filler package.

**[0212]** 15. The composition of any of the preceding Aspects, wherein the filler package comprises thermally stable filler materials.

**[0213]** 16. The composition of any of the preceding Aspects, wherein at least a portion of the thermally conductive, electrically insulative filler particles are thermally stable.

**[0214]** 17. The composition of any of the preceding Aspects, wherein the filler package further comprises thermally conductive, electrically conductive filler particles having a thermal conductivity of at least 5 W/m·K (measured according to ASTM D7984) and a volume resistivity of less than 1  $\Omega\cdot\text{m}$  (measured according to ASTM D257), the thermally conductive, electrically conductive filler particles being present in an amount of no more than 30% by volume based on total volume of the filler package.

**[0215]** 18. The composition of any of the preceding Aspects, wherein the thermally conductive, electrically conductive filler particles have an average particle size in at least one dimension of 0.01  $\mu\text{m}$  to 500  $\mu\text{m}$ , as measured using SEM.

**[0216]** 19. The composition of any of the preceding Aspects, wherein the filler package further comprises non-thermally conductive, electrically insulative filler particles having a thermal of less than 5 W/m·K (measured according to ASTM D7984) and a volume resistivity of at least 1  $\Omega\cdot\text{m}$  (measured according to ASTM D257), the non-thermally conductive, electrically conductive filler particles being present in an amount of no more than 10% by volume based on total volume of the filler package.

[0217] 20. The composition of any of the preceding Aspects, further comprising a dispersant.

[0218] 21. The composition of Aspect 20, wherein the dispersant is present in an amount of 0.01% by volume to 70% by volume based on total volume of the composition.

[0219] 22. The composition of any of the preceding Aspects, further comprising a reactive diluent and/or at least one additive.

[0220] 23. The composition of any of the preceding Aspects, wherein the composition comprises a total solids content of 40 % by volume to 100 % by volume based on total volume of the composition.

[0221] 24. The composition of any of the preceding Aspects, wherein the composition comprises a one-component composition.

[0222] 25. The composition of any of the preceding Aspects, wherein the composition comprises a two-component composition.

[0223] 26. The composition of any of the preceding Aspects, wherein the composition comprises a pre-mixed frozen composition.

[0224] 27. The composition of any of the preceding Aspects, wherein the composition comprises a gap filler composition, a sealant composition, an adhesive composition, a putty, a molding compound, a potting compound, and/or a three-dimensionally printable composition.

[0225] 28. A method for treating a substrate comprising:  
contacting a surface of the substrate with the composition of Aspects 1 to 27.

[0226] 29. A substrate comprising a surface at least partially coated with a layer formed from the composition of any of Aspects 1 to 27.

[0227] 30. The substrate of Aspect 29, further comprising a film, a second layer, or a coating positioned between the substrate surface and the layer formed from the composition according to any of Aspects 1 to 27 and/or positioned over the layer formed from the composition according to any of Aspects 1 to 27.

[0228] 31. The substrate of Aspect 29 or Aspect 30, wherein the layer formed from the composition of any of Aspects 1 to 27, in an at least partially cured state:

(a) has a thermal conductivity of at least 0.4 W/m·K (measured according to ASTM D7984), such as at least 0.5 W/ m·K, such as at least 0.7 W/ m·K;

(b) has a dielectric strength of at least 1kV/mm measured according to ASTM D149 on a dielectric meter (Sefelec RMG12AC-DC) connected to two copper electrodes with 1 inch diameter, such as at least 1.5 kV/mm;

(c) has a Shore A hardness of at least 5 measured according to ASTM D2240 with a Type A durometer (Model 2000, Rex Gauge Company, Inc.) at room temperature, such as at least 10, such as at least 20, such as at least 30, such as at least 40, such as 5 to 95;

(d) has a 180° peel strength of at least 1 lbf/in (measured according to SAE AS5127/1B with an Instron universal testing machine, model 3345), such as at least 2 lbf/in, such as at least 3 lbf/in, such as at least 4 lbf/in;

(e) has a tensile stress at break of at least 1 MPa as determined according to ISO 37-2 using an Instron universal testing machine, model 3345 with a pull rate of 2 inches/min, such as at least 10 MPa, such as at least 25 MPa, such as at least 30 MPa, such as 1 MPa to 100 MPa;

(f) has an elongation of 1% to 900%, as determined according to ISO 37-2 using an Instron universal testing machine, model 3345 with a pull rate of 2 inches/min, such as at least 100%, such as at least 200%, such as at least 300%, such as at least 400%, such as at least 500%, such as at least 600%, such as at least 700%, such as at least 800%;

(g) has a Young's modulus of 0.01 MPa to 100 MPa, as determined according to ISO 37-2 using an Instron universal testing machine, model 3345 with a pull rate of 2 inches/min, such as at least 0.1 MPa, such as at least 0.7 MPa, such as at least 1 MPa, such as at least 10 MPa;

(h) maintains a temperature of the substrate that is at least 100°C lower following exposure of the coating on the surface of the substrate to 1000°C for a time of at least 90 seconds than a surface temperature of a bare substrate exposed to 1000°C for the time;

(i) does not smoke upon exposure of the substrate to 1000°C for 500 sec; and/or

(j) exhibits no visible cracking or delamination upon exposure of the substrate to 1000°C for 500 sec.

**[0229]** 32. A vehicle comprising the substrate of any of Aspects 29 to 31.

**[0230]** 33. A thermally conductive part comprising the substrate of any of Aspects 29 to 31.

**[0231]** 34. The thermally conductive part of Aspect 33, wherein the part comprises a three-dimensional part.

**[0232]** 35. A vehicle comprising the thermally conductive part of Aspect 33 or Aspect 34.

**[0233]** 36. A battery pack comprising:  
at least two battery cells; and  
the composition of any of Aspects 1 to 27 positioned between the two battery cells.

**[0234]** 37. The battery pack of Aspect 36, further comprising a cooling fin, a cooling plate, and/or a battery box.

**[0235]** 38. A circuit board comprising the composition of any of Aspects 1 to 27 positioned in or on the circuit board.

**[0236]** 39. A method of forming an article comprising extruding the composition of any of Aspects 1 to 27 onto a substrate.

**[0237]** 40. The method of Aspect 39, wherein the extruding comprises three-dimensional printing.

**[0238]** 41. An article formed by the method of Aspect 39 or Aspect 40.

**[0239]** 42. A coating, wherein the coating, in an at least partially cured state:

- (a) has a thermal conductivity of at least 0.4 W/m·K (measured according to ASTM D7984), such as at least 0.5 W/ m·K, such as at least 0.7 W/ m·K;
- (b) has a dielectric strength of at least 1kV/mm measured according to ASTM D149 on a dielectric meter (Sefelec RMG12AC-DC) connected to two copper electrodes with 1 inch diameter, such as at least 1.5 kV/mm;
- (c) has a Shore A hardness of at least 5 measured according to ASTM D2240 with a Type A durometer (Model 2000, Rex Gauge Company, Inc.) at room temperature, such as at least 10, such as at least 20, such as at least 30, such as at least 40, such as 5 to 95;
- (d) has a 180° peel strength of at least 1 lbf/in (measured according to SAE AS5127/1B with an Instron universal testing machine, model 3345), such as at least 2 lbf/in, such as at least 3 lbf/in, such as at least 4 lbf/in;
- (e) has a tensile stress at break of at least 1 MPa as determined according to ISO 37-2 using an Instron universal testing machine, model 3345 with a pull rate of 2 inches/min, such as at least 10 MPa, such as at least 25 MPa, such as at least 30 MPa, such as 1 MPa to 100 MPa;
- (f) has an elongation of 1% to 900%, as determined according to ISO 37-2 using an Instron universal testing machine, model 3345 with a pull rate of 2 inches/min, such as at least

100%, such as at least 200%, such as at least 300%, such as at least 400%, such as at least 500%, such as at least 600%, such as at least 700%, such as at least 800%;

(g) has a Young's modulus of 0.01 MPa to 100 MPa, as determined according to ISO 37-2 using an Instron universal testing machine, model 3345 with a pull rate of 2 inches/min, such as at least 0.1 MPa, such as at least 0.7 MPa, such as at least 1 MPa, such as at least 10 MPa;

(h) maintains a temperature of the substrate that is at least 100°C lower following exposure of the coating on the surface of the substrate to 1000°C for a time of at least 90 seconds than a surface temperature of a bare substrate exposed to 1000°C for the time;

(i) does not smoke upon exposure of the substrate to 1000°C for 500 sec; and/or

(j) exhibits no visible cracking or delamination upon exposure of the substrate to 1000°C for 500 sec.

[0240] Aspect 43. A vehicle, comprising the coating of Aspect 42.

[0241] Aspect 44. A part, comprising the coating of Aspect 42.

[0242] Aspect 45. A battery pack, comprising the coating of Aspect 42.

[0243] Aspect 46. A circuit board, comprising the coating of Aspect 42.

[0244] Illustrating the invention are the following examples, which, however, are not to be considered as limiting the invention to their details. Unless otherwise indicated, all parts and percentages in the following examples, as well as throughout the specification, are by volume.

## EXAMPLES

**Table 1. Abbreviation Description of Matrix Materials**

Abbreviation or trade designation of matrix materials	Description
Thioplast G1	Liquid Polysulfide Pre-polymer, end-capped by SH-functional end groups, available by Nouryon, Av. Molecular Weight: 3900 - 4400 g/mol
Thioplast G112	Liquid Polysulfide Pre-polymer, end-capped by SH-functional end groups, available by Nouryon, Av. Molecular Weight: 3400 - 4600 g/mol
Thioplast G4	Liquid Polysulfide Pre-polymer, end-capped by SH-functional end groups, available by Nouryon, Av. Molecular Weight: <1100 g/mol
PETMP	Pentaerythritol Tetra(3-mercaptopropionate) available by Bruno Bock Thiochemicals
Permapol P-3.1e	Liquid Polysulfide Pre-polymer, end-capped by SH-functional end groups, See US6372849B2, Number Av. Molecular Weight: 500 - 20,000 g/mol.
Cure accelerator	Organic amine base or sulfur-donor component
Stearic acid	Fatty acid supplied as cas 57-11-4, available by Lanxess, Chemical Associates, Acme-Hardesty, BASF, Croda
HB-40	Plasticizer available by EASTMAN
Methylon 75108	Coating intermediate containing a mixture of the allyl ethers of mono-, di-, and trimethylol phenols. See US3517082

P/S890 Class B Fuel Tank Sealant  
Part A

Part A of PS890 Class B Fuel Tank sealant, available by PPG

**Table 2. Abbreviation Description of Fillers Materials**

Abbreviation or trade designation of fillers	Particle Size ( $\mu\text{m}$ )*	True Density (g/ml)	Mohs Hardness	Description
Nabalox NO625-10	2.5	3.89	9	Alumina fillers available from Nabaltec AG
Nabalox 105RA	80	3.89	9	Alumina fillers available from Nabaltec AG
TFZ-N15P	16	3.28	5	Aluminum nitride fillers available from Toyol America, Inc.
CTS25M	300	2.1	2	Boron nitride fillers available from Saint-Gobain Ceramic Materials

\* Particle size as reported by supplier

**Table 3. Formulation of thermally conductive sealants (Unit: gram)**

	Ex 1	Ex 2	Ex 3	Ex 4	Ex 5	Ex 6	Ex 7	Ex 8	Ex 9
<b>Part A</b>									
Thioplast G1	10	10	10	10	10	5	5	5	10
Thioplast G112	20	20	20	20	20	20	20	20	20
Thioplast G4	—	—	—	—	—	5	—	—	—
PETMP	—	—	—	—	—	—	5	—	—
Permapol e3.1	—	—	—	—	—	—	—	5	—
Cure accelerator	0.05	0.05	0.05	0.05	0.05	0.05	0.05	0.05	0.05
Stearic acid	0.25	0.25	0.25	0.25	0.25	0.25	0.25	0.25	0.25
HB-40	—	—	—	—	4	—	4	—	—
Nabalox NO 625-10	40	—	—	1	45	40	45	40	—
Methylon 75108 (an adhesion promoter)	—	—	—	—	—	—	—	—	4.5
TFZ N15P	—	35	—	—	—	—	—	—	5
Nabalox 102RA	—	—	—	25	—	—	—	—	—
CST25M	—	—	10	—	—	—	—	—	—
<b>Part B</b>									
PS-890 B2 Part A (oxidant)	6.06	6.06	6.06	6.06	6.86	6.06	6.86	6.06	6.06
Nabalox NO 625-10	—	—	—	—	—	1	—	—	6

**Table 4. Composition of cured thermally conductive sealants (Unit: volume %)**

Component	Ex 1	Ex 2	Ex 3	Ex 4	Ex 5	Ex 6	Ex 7	Ex 8	Ex 9
Thioplast G1	20.74	20.53	24.34	22.95	18.00	10.27	9.00	10.26	25.00
Thioplast 112	41.70	41.28	48.94	46.15	36.20	41.32	36.18	41.27	50.27
Thioplast G4	—	—	—	—	—	10.53	—	—	—
PETMP	—	—	—	—	—	—	9.07	—	—
Permapol e3.1	—	—	—	—	—	—	—	11.31	—
Cure accelerator	0.12	0.11	0.14	0.13	0.10	0.11	0.10	0.11	0.14
Stearic acid	0.71	0.70	0.83	0.79	0.62	0.70	0.62	0.70	0.86
HB-40	—	—	—	—	9.17	—	9.16	—	—
Methylon 75108	—	—	—	—	—	—	—	—	11.1
PS-890 B2 Part A	9.21	9.12	10.81	10.19	9.05	9.13	9.05	9.12	12.65
Nabalox NO 625-10	27.51	—	—	0.76	26.87	27.93	26.85	27.22	—
TFZ N15P	—	28.26	—	—	—	—	—	—	—
Nabalox 102RA	—	—	—	19.03	—	—	—	—	—
CST25M	—	—	14.95	—	—	—	—	—	—
Total	100	100	100	100	100	100	100	100	100



**Table 5. Thermal conductivity and Shore A hardness of cured thermally conductive sealants**

Properties	Ex 1	Ex 2	Ex 3	Ex 4	Ex 5	Ex 6	Ex 7	Ex 8	Ex9
Thermal conductivity (W/mK)	0.55	0.416	0.796	0.53	0.489	0.439	0.54	0.573	0.23
Shore A Hardness	53.6	39.2	53.8	54.6	39.8	49.6	34.8	46	33

[0245] The cured samples were prepared according to the following procedure with all non-manual mixing performed using a Speedmixer DAC 600FVZ (commercially available from FlackTek Inc.). For each example, part A was mixed with part B for 5 s at 800 rpm, and then 15 s at 1850 rpm. The mixing step was repeated. If the mixing appeared non-homogenous, the mixing step was repeated. The composition was then transferred into an aluminum (Al) weighing dish (Fisherbrand, Catalog No. 08-732-101), and allowed to cure for 2 days in an environmental chamber (50 % RH, 25°C) followed by 1 day at 140°F (60°C).

[0246] **Thermal conductivity measurement.** The samples were tested using a modified transient plane source method (conform to ASTM D7984) with a TCi thermal conductivity analyzer. The sample size was at least 20 mm x 20 mm with a thickness of 5 mm. The weight placed on the sample during measurement was 500 g.

[0247] **Hardness test.** The samples were tested in accordance with ASTM D2240 with a Type A durometer at room temperature. The sample size was at least 20 mm x 20 mm with a thickness of 6 mm.

**Table 6. Mechanical Properties of cured thermally conductive sealants**

Property	Ex 1	Ex 3	Ex 9
Peel strength (lbf/in)	3.5	4.1	2.0
Elongation percentage (%)	326.5	112.6	272.1
Tensile stress at break (psi)	139.0	129.8	111.19
Young's modulus (MPa)	2.49	3.81	0.82

[0248] **Elongation percentage, tensile stress at break and Young's modulus measurement.** Dog bone tensile specimens were prepared using the mixed materials and the sample size is in accordance with ISO 37-2. The elongation percentage, tensile strength and Young's modulus were determined according to ISO 37 on an Instron universal testing machine, model 3345 with a pull rate at 2 inches/min.

[0249] **180° peel strength measurement.** The sealant composition was applied onto an aluminum substrate (AMS 3277) which was pre-abraded with a scotch bright pad and primed with PPG commercial PR-188 adhesion promoter. The samples were cured for 2 days in an environmental chamber (50 % RH, 25°C) followed by 1 day at 49°C. The samples were tested using an Instron universal testing machine, model 3345 in accordance with SAE AS5127/1B.

<b>Unit: g</b>	<b>Ex 10</b>	<b>Ex 11</b>
<b>Part A</b>		
<b>PETMP</b>	<b>69.46</b>	<b>69.46</b>
<b>Cure accelerator</b>	<b>0.15</b>	<b>0.15</b>
<b>Stearic acid</b>	<b>0.75</b>	<b>0.75</b>
<b>BYK 111</b>	<b>3</b>	<b>3</b>
<b>Nabalox NO 625-31</b>	<b>—</b>	<b>300</b>
<b>APRAL 20X</b>	<b>186.63</b>	<b>—</b>
<b>Part B</b>		
<b>PS-890 B2 Part A</b>	<b>26.64</b>	<b>26.64</b>
<b>Vol % of Fillers</b>	<b>43.00</b>	<b>43.00</b>
<b>Thermal conductivity (W/m•K)</b>	<b>1.80</b>	<b>2.00</b>
<b>Panel temperature at 500 s (oC)</b>	<b>355.46</b>	<b>455.71</b>
<b>Fire test performance</b>	<b>Char integrity maintained, smoking</b>	<b>Char integrity maintained, no smoking</b>

**Table 7. Thermal conductivity and fire test performance of Example 10 and 11**

[0250] **Fire protection test.** For each example, part A and part B were prepared separately using a Speedmixer DAC 600FVZ (commercially available from FlackTeck inc.). Equivalent mass amount of part A and part B were mixed using a Speedmixer DAC 600FVZ until the mixture appeared homogenous. The mixtures of Example 10 and 11 were trowel-applied to steel panel structures (7.8 mm to 7.9 mm thick coating). The steel panel structure had a dimension of depth 3/16", length 7" and width 3".

[0251] After application, the coated structures were allowed to cure for 2 days in an environmental chamber (50 % RH, 25°C) followed by 1 day at 140°F (60 °C), and final film thickness of coatings were measured and recorded before subjecting to fire tests.

[0252] On the back of the coated panel, a thermocouple was attached at the center point to monitor the temperature through the sample. The center of the coated panel was then positioned at a distance of 4 cm from a propane torch (diameter 3.5cm, propane) with the coating

in the direction of the torch. The temperature of the flame was monitored through a second thermocouple placed close to the base of the flame and found to remain stable between 900°C to 1000°C. See FIG. 3. The temperature at the back of the coated substrate and for comparison of an uncoated identical steel panel was measured for a prolonged period of time. Data are reported in FIG. 4.

**[0253]** Whereas specific aspects of the invention have been described in detail, it will be appreciated by those skilled in the art that various modifications and alternatives to those details could be developed in light of the overall teachings of the disclosure. Accordingly, the particular arrangements disclosed are meant to be illustrative only and not limiting as to the scope of the invention which is to be given the full breadth of the claims appended and any and all equivalents thereof.

We claim:

1. A composition, comprising:
  - a thiol-terminated compound;
  - an oxidant; and
  - a thermally conductive filler package comprising thermally conductive, electrically insulative filler particles, the thermally conductive, electrically insulative filler particles having a thermal conductivity of at least 5 W/m·K (measured according to ASTM D7984) and a volume resistivity of at least 1  $\Omega\cdot\text{m}$  (measured according to ASTM D257);wherein the thermally conductive, electrically insulative filler particles are present in an amount of at least 50% by volume based on total volume of the filler package; and
  - wherein at least a portion of the thermally conductive, electrically insulative filler particles comprise thermally stable filler particles.
2. The composition of Claim 1, wherein the thermally stable filler particles are present in an amount of at least 90% by volume based on total volume of the thermally conductive, electrically insulative filler particles.
3. The composition of Claim 1 or Claim 2, wherein the thermally conductive, electrically insulative filler particles further comprise thermally unstable filler particles.
4. The composition of Claim 3, wherein the thermally unstable filler particles are present in an amount of no more than 10% by volume based on a total volume of the thermally conductive, electrically insulative filler particles.
5. A composition, comprising:
  - a thiol-terminated compound;
  - an oxidant; and
  - a thermally conductive filler package comprising thermally conductive, electrically insulative filler particles, the thermally conductive, electrically insulative filler particles having a thermal conductivity of at least 5 W/m·K (measured according to ASTM D7984) and a volume resistivity of at least 1  $\Omega\cdot\text{m}$  (measured according to ASTM D257);

wherein the thermally conductive, electrically insulative filler particles are present in an amount of at least 50% by volume based on total volume of the filler package; and

wherein at least a portion of the thermally conductive, electrically insulative filler particles comprise thermally unstable filler particles.

6. The composition of Claim 5, wherein the thermally unstable filler particles are present in an amount of at least 90% by volume based on total volume of the thermally conductive, electrically insulative filler particles.

7. The composition of Claim 5 or Claim 6, wherein the thermally conductive, electrically insulative filler particles further comprise thermally stable filler particles.

8. The composition of Claim 7, wherein the thermally stable filler particles are present in an amount of no more than 10% by volume based on a total volume of the thermally conductive, electrically insulative filler particles.

9. The composition of any of the preceding Claims, wherein the thermally conductive filler package is present in an amount of 15% by volume percent to 90% by volume based on total volume of the composition.

10. The composition of any of the preceding Claims, wherein the thiol-terminated compound comprises a liquid.

11. The composition of any of the preceding Claims, wherein the thiol-terminated compound has an average molecular weight of 80 g/mol to 40,000 g/mol.

12. The composition of any of the preceding Claims, wherein the thiol-terminated compound is present in an amount of 9.9% by volume to 84.9% by volume based on total volume of the composition.

13. The composition of any of the preceding Claims, further comprising lead oxide, lead peroxide, manganese dioxide, sodium perborate, potassium permanganate, calcium peroxide, barium peroxide, lithium peroxide, zinc peroxide, barium oxide, or combinations thereof and/or cumene hydroperoxide, t-butyl hydroperoxide, or combinations thereof.

14. The composition of any of the preceding Claims, wherein the oxidant is present in an amount of 0.1% by volume to 10% by volume based on total volume of the resins.

15. The composition of any of the preceding Claims, wherein the filler package further comprises thermally conductive, electrically conductive filler particles having a thermal conductivity of at least 5 W/m·K (measured according to ASTM D7984) and a volume resistivity of less than 1  $\Omega\cdot\text{m}$  (measured according to ASTM D257), the thermally conductive, electrically conductive filler particles being present in an amount of no more than 30% by volume based on total volume of the filler package.

16. The composition of Claim 15, wherein the thermally conductive, electrically conductive filler particles have an average particle size in at least one dimension of 0.01  $\mu\text{m}$  to 500  $\mu\text{m}$ , as measured using SEM.

17. The composition of any of the preceding Claims, wherein the filler package further comprises non-thermally conductive, electrically insulative filler particles having a thermal conductivity of less than 5 W/m·K (measured according to ASTM D7984) and a volume resistivity of at least 1  $\Omega\cdot\text{m}$  (measured according to ASTM D257), the non-thermally conductive, electrically conductive filler particles being present in an amount of no more than 10% by volume based on total volume of the filler package.

18. The composition of any of the preceding Claims, further comprising a dispersant, a reactive diluent, and/or an additive.

19. The composition of Claim 18, wherein the dispersant is present in an amount of 0.05% by volume to 20% by volume based on total volume of the composition.

20. The composition of any of the preceding Claims, wherein the composition comprises a total solids content of 40% by volume to 100% by volume based on total volume of the composition.

21. The composition of any of the preceding Claims, wherein the composition comprises a one-component composition, a two-component composition, a three-component composition, a pre-mixed frozen composition, or combinations thereof.

22. The composition of any of the preceding Claims, wherein the composition comprises a gap filler composition, a sealant composition, an adhesive composition, a putty, a molding compound, a potting compound, and/or a three-dimensionally printable composition.

23. A method of treating a substrate comprising:  
contacting at least a portion of a surface of the substrate with the composition of any of Claims 1 to 22.

24. The method of Claim 23, wherein the composition, in an at least partially cured state, forms a coating.

25. The method of Claim 23 or Claim 24, further comprising contacting a surface of a second substrate to the composition such that the composition is between the first and the second substrate.

26. A coating formed on a surface of a substrate, wherein the coating, in an at least partially cured state:

(a) has a thermal conductivity of at least 0.4 W/m·K (measured according to ASTM D7984);

(b) has a dielectric strength of at least 1kV/mm measured according to ASTM D149 on a dielectric meter (Sefetec RMG12AC-DC) connected to two copper electrodes with 1-inch diameter;

(c) has a shore A hardness 5 to 95 measured according to ASTM D2240 with a Type A durometer (Model 2000, Rex Gauge Company, Inc.) at room temperature;

(d) has a 180° peel strength of at least 1 lbf/in (measured according to SAE AS5127/1B with an Instron universal testing machine, model 3345);

(e) has a tensile stress at break of 1 MPa to 100 MPa, as determined according to ISO 37-2 using an Instron universal testing machine, model 3345 with a pull rate of 2 inches/min;

(f) has an elongation of 1% to 900%, as determined according to ISO 37-2 using an Instron universal testing machine, model 3345 with a pull rate of 2 inches/min;

(g) has a Young's modulus of 0.01 MPa to 100 MPa, as determined according to ISO 37-2 using an Instron universal testing machine, model 3345 with a pull rate of 2 inches/min;

(h) maintains a temperature of the substrate that is at least 100°C lower following exposure of the coating on the surface of the substrate to 1000°C for a time of at least 90 seconds than a surface temperature of a bare substrate exposed to 1000°C for the time; and/or

(i) does not smoke upon exposure of the substrate to 1000°C for 500 seconds.

27. A coating formed on a surface of a substrate, wherein the coating, in an at least partially cured state, has a thermal conductivity of at least 0.5 W/m·K (measured according to ASTM D7984) and maintains a temperature of the substrate that is at least 100°C lower following exposure of the coating on the surface of the substrate to 1000°C for a time of at least 90 seconds than a surface temperature of a bare substrate exposed to 1000°C for the time.

28. The coating of Claim 27, wherein the coating does not smoke upon exposure of the substrate to 1000°C for 500 seconds.

29. The coating of any of Claims 26 to 28 formed from the composition of any of Claims 1 to 22.

30. The coating of any of Claims 26 to 28, formed from a composition comprising:  
a thiol-terminated compound;  
an oxidant; and



a thermally conductive filler package comprising thermally conductive, electrically insulative filler particles, the thermally conductive, electrically insulative filler particles having a thermal conductivity of at least 5 W/m·K (measured according to ASTM D7984) and a volume resistivity of at least 1  $\Omega$ ·m (measured according to ASTM D257);

wherein the thermally conductive, electrically insulative filler particles are present in an amount of at least 50 % by volume based on total volume of the filler package.

31. The substrate of any of Claims 23 to 29 treated according to the method of any of Claims 23 to 25.

32. The substrate of any of Claims 23 to 31, further comprising a film, a layer, or a second coating positioned between the substrate surface and the coating.

33. The substrate of any of Claims 23 to 32, wherein the substrate comprises a vehicle, a part, an article, an appliance, a battery cell, a personal electronic device, a circuit board, a multi-metal article, or combinations thereof.

34. The substrate of Claim 33, wherein the vehicle comprises an automobile or an aircraft and/or the part comprises a thermally conductive part.

35. A battery assembly comprising:

a battery cell; and

a coating formed on a surface of the battery cell, wherein the coating, in an at least partially cured state:

(a) has a thermal conductivity of at least 0.4 W/m·K (measured according to ASTM D7984);

(b) has a dielectric strength of at least 1kV/mm measured according to ASTM D149 on a dielectric meter (Sefetec RMG12AC-DC) connected to two copper electrodes with 1-inch diameter;

(c) has a shore A hardness 5 to 95 measured according to ASTM D2240 with a Type A durometer (Model 2000, Rex Gauge Company, Inc.) at room temperature;

(d) has a 180° peel strength of at least 1 lbf/in (measured according to SAE AS5127/1B with an Instron universal testing machine, model 3345);

(e) has a tensile stress at break of 1 MPa to 100 MPa, as determined according to ISO 37-2 using an Instron universal testing machine, model 3345 with a pull rate of 2 inches/min;

(f) has an elongation of 1% to 900%, as determined according to ISO 37-2 using an Instron universal testing machine, model 3345 with a pull rate of 2 inches/min;

(g) has a Young's modulus of 0.01 MPa to 100 MPa, as determined according to ISO 37-2 using an Instron universal testing machine, model 3345 with a pull rate of 2 inches/min;

(h) maintains a temperature of the substrate that is at least 100°C lower following exposure of the coating on the surface of the substrate to 1000°C for a time of at least 90 seconds than a surface temperature of a bare substrate exposed to 1000°C for the time; and/or

(i) does not smoke upon exposure of the substrate to 1000°C for 500 seconds.

36. A battery assembly comprising:

a battery cell; and

a coating formed on a surface of the battery cell, wherein the coating, in an at least partially cured state, has a thermal conductivity of at least 0.5 W/m·K (measured according to ASTM D7984) and maintains a temperature of the substrate that is at least 100°C lower following exposure of the coating on the surface of the substrate to 1000°C for a time of at least 90 seconds than a surface temperature of a bare substrate exposed to 1000°C for the time.

37. The battery assembly of Claim 36, wherein the coating does not smoke upon exposure of the substrate to 1000°C for 500 seconds.

38. The battery assembly of any of Claims 35 to 37 wherein the coating is formed from the composition of any of Claims 1 to 22.

39. The battery assembly of any of Claims 35 to 37 wherein the coating is formed from a composition comprising:

a thiol-terminated compound;

an oxidant; and

a thermally conductive filler package comprising thermally conductive, electrically insulative filler particles, the thermally conductive, electrically insulative filler particles having a thermal conductivity of at least 5 W/m·K (measured according to ASTM D7984) and a volume resistivity of at least 1  $\Omega\cdot\text{m}$  (measured according to ASTM D257);

wherein the thermally conductive, electrically insulative filler particles are present in an amount of at least 50 % by volume based on total volume of the filler package.

40. The battery assembly of any of Claims 35 to 39, further comprising at least one second battery cell, a cooling fin, a cooling plate, and/or a battery box.

41. The battery assembly of Claim 40, wherein the coating is positioned between the battery cell and the at least one second battery cell and/or the cooling plate.

42. A method of forming an article comprising extruding the composition of any of Claims 1 to 22.

43. The method of Claim 42, wherein the extruding comprises three-dimensional printing.

44. The article formed by the method of Claim 42 or Claim 43.

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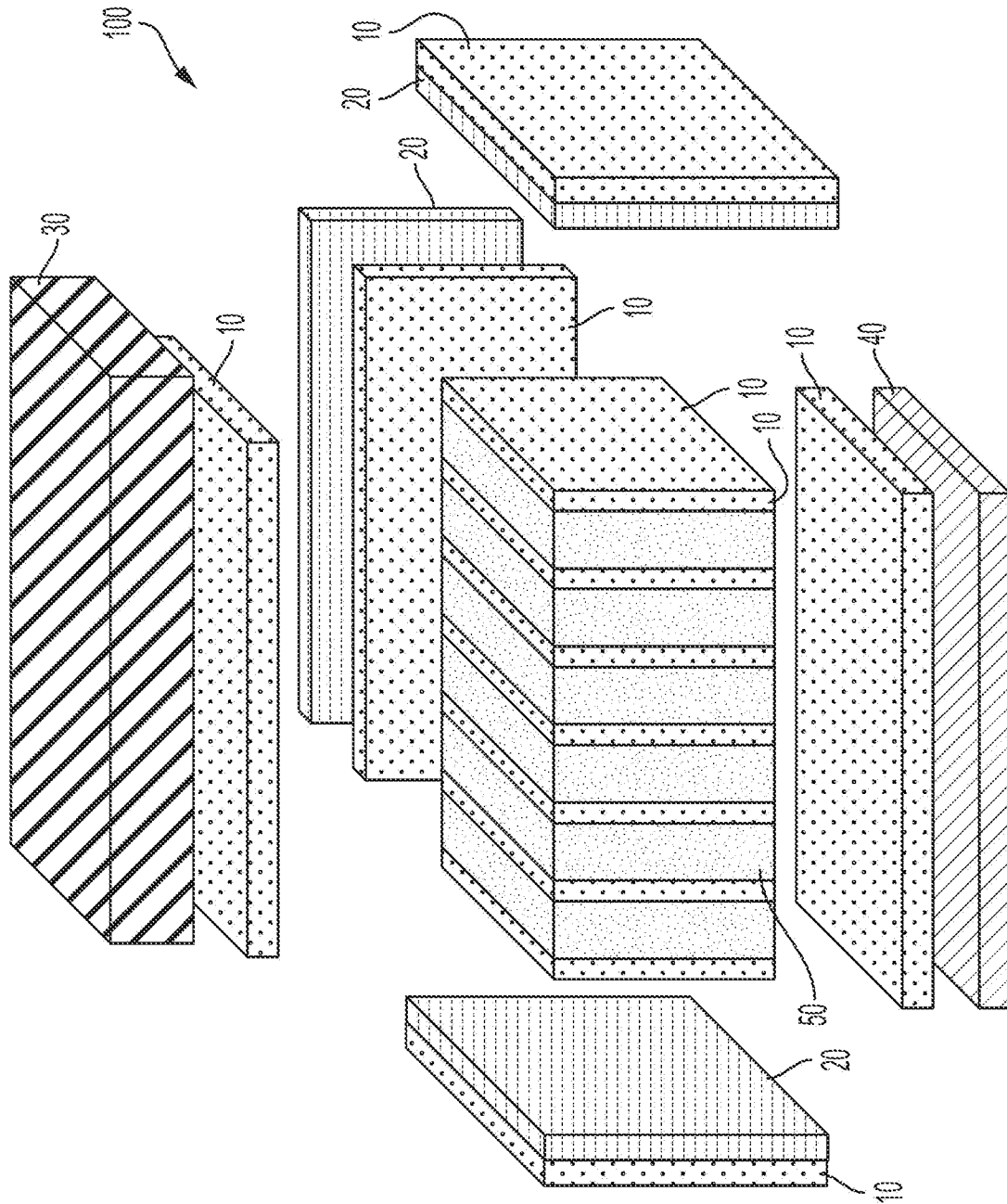


FIG. 1

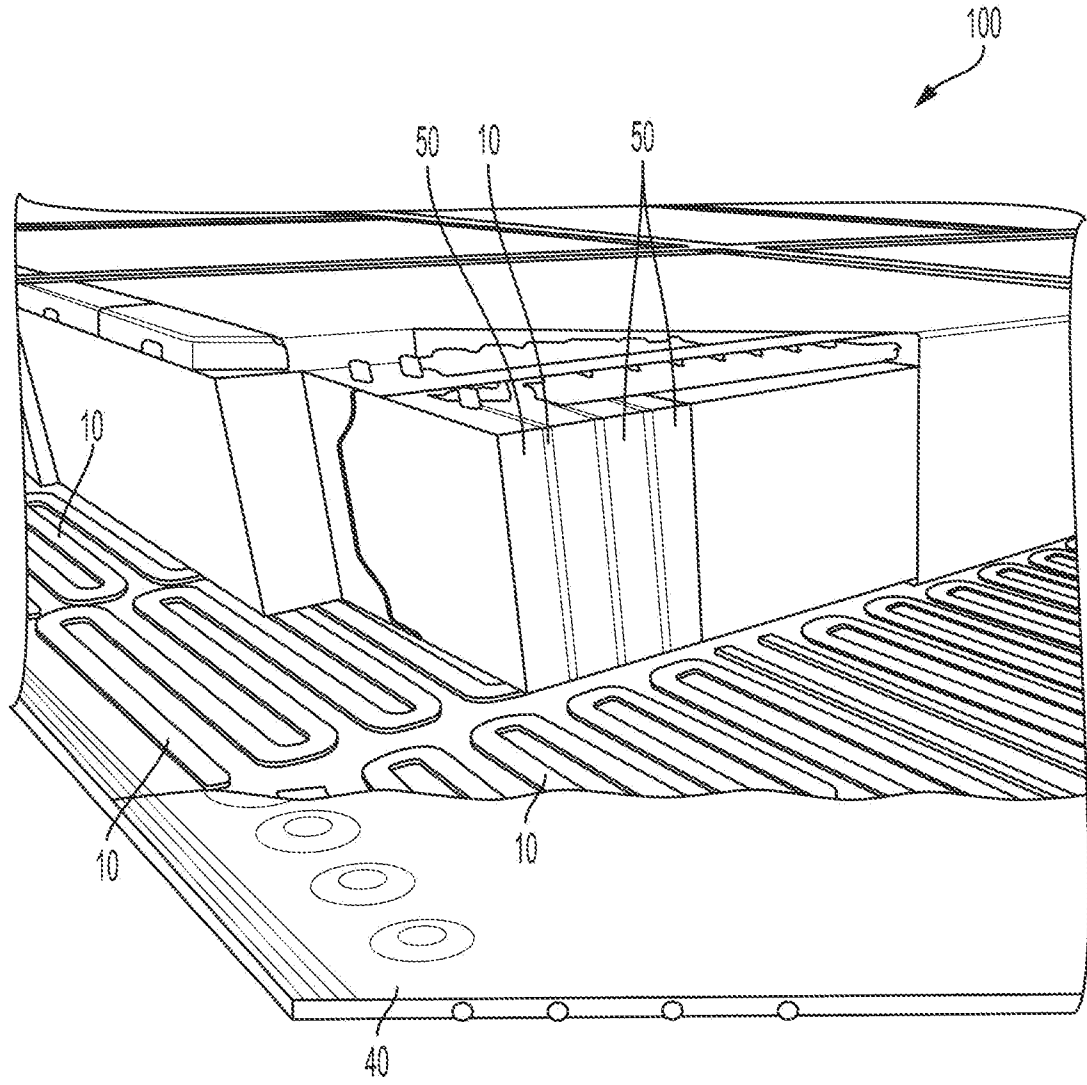


FIG. 2

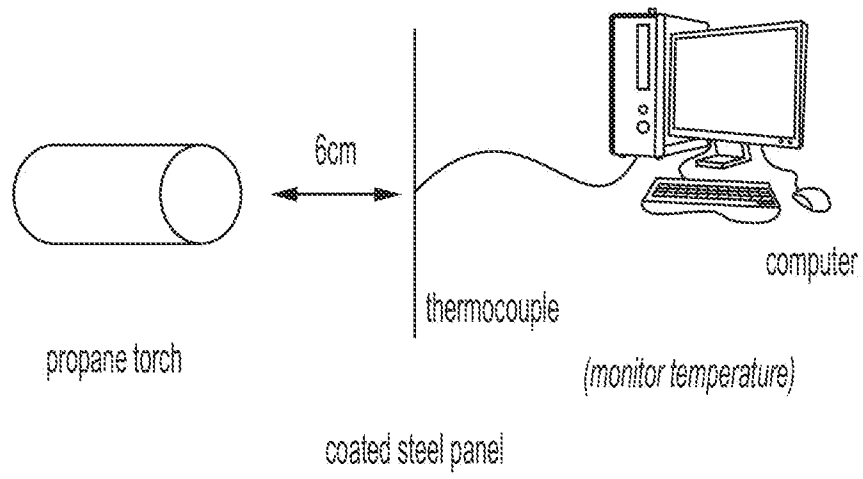


FIG. 3

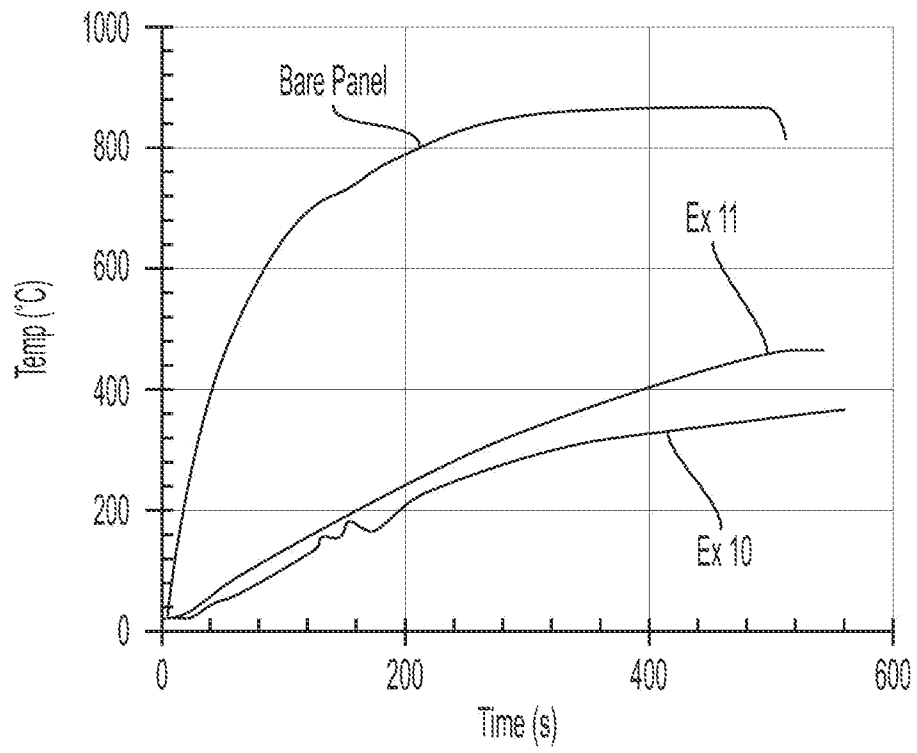


FIG. 4

INTERNATIONAL SEARCH REPORT

International application No  
PCT/US2021/014100

A. CLASSIFICATION OF SUBJECT MATTER  
 INV. C08K3/013 C08K3/22 C08K3/28 C08K3/38 C08K3/24  
 H01M10/653  
 ADD.  
 According to International Patent Classification (IPC) or to both national classification and IPC

B. FIELDS SEARCHED  
 Minimum documentation searched (classification system followed by classification symbols)  
 C08K H01M

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Electronic data base consulted during the international search (name of data base and, where practicable, search terms used)  
 EPO-Internal, WPI Data

C. DOCUMENTS CONSIDERED TO BE RELEVANT

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Further documents are listed in the continuation of Box C.

See patent family annex.

\* Special categories of cited documents :

<p>"A" document defining the general state of the art which is not considered to be of particular relevance</p> <p>"E" earlier application or patent but published on or after the international filing date</p> <p>"L" document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified)</p> <p>"O" document referring to an oral disclosure, use, exhibition or other means</p> <p>"P" document published prior to the international filing date but later than the priority date claimed</p>	<p>"T" later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention</p> <p>"X" document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step when the document is taken alone</p> <p>"Y" document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art</p> <p>"&amp;" document member of the same patent family</p>
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Date of the actual completion of the international search  26 March 2021	Date of mailing of the international search report  07/04/2021
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Name and mailing address of the ISA/ European Patent Office, P.B. 5818 Patentlaan 2 NL - 2280 HV Rijswijk Tel. (+31-70) 340-2040, Fax: (+31-70) 340-3016	Authorized officer  Masson, Patrick
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