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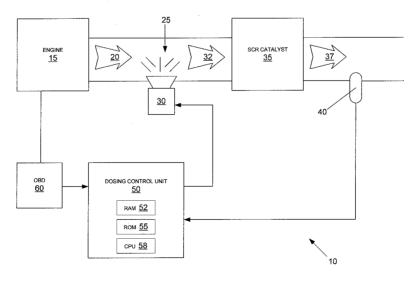


FIG. 1

(57) Abstract: An exhaust aftertreatment system comprising a selective reduction catalyst, a NO<sub>x</sub> sensor or an NH<sub>3</sub> sensor, a urea injector, and a dosing control unit, wherein the dosing control unit calculates the rate of urea injection by estimating the concentrations of NO and NO2 in the exhaust downstream of the SCR catalyst.



# NO<sub>X</sub> CONTROL SYSTEMS AND METHODS FOR CONTROLLING NO<sub>X</sub> EMISSIONS

#### CROSS-REFERENCE TO RELATED APPLICATIONS

[0001] This application claims priority to U.S. Provisional Patent Applications Serial Nos. 60/990,211, filed November 26, 2007, and 61/052,759, filed May 13, 2008, both of which are incorporated herein by reference in their entirety.

#### FIELD OF INVENTION

**[0002]** The present invention relates to methods for reducing nitrogen oxide emissions from combustion processes. In particular, the invention relates to the use of catalysts in combination with an ammonia-precursor injector to reduce nitrogen oxide emissions from engines, such as those used by diesel-powered vehicles, equipment, and generators.

#### **BACKGROUND**

[0003] Nitrogen oxides (NO<sub>x</sub>) are a well-known class of atmospheric pollutants that pose risks to human health, agriculture, and property. Nitrogen oxides are key ingredients in the formation of photochemical smog, which damages respiratory functions and limits visibility. Additionally, nitrogen oxides form nitric acid when they react with moisture in the atmosphere, thus forming acid rain. They are also involved in the production of tropospheric ozone. Environmental agencies around the world have instituted standards for nitrogen oxide control in an effort to reduce nitrogen oxides' harmful effects.

[0004] Nitrogen oxides broadly include compounds such as NO<sub>3</sub> and N<sub>2</sub>O<sub>5</sub>, but the bulk of atmospheric nitrogen oxides are N<sub>2</sub>O, NO and NO<sub>2</sub>. When dealing with reactive atmospheric nitrogen, it is often sufficient to deal only with NO and NO<sub>2</sub>. (See Seinfeld and Pandis, Atmospheric Chemistry and Physics.)

[0005] While there are some natural sources of  $NO_x$ , most of the  $NO_x$  in urban environments is a by-product of fossil-fuel combustion. At temperatures found in engines, free-radical chemistry converts atmospheric nitrogen into nitrogen oxides. In particular, lean burning internal combustion engines, such as those used in diesel-powered vehicles, may produce  $NO_x$  at a many parts-per-million level. While combustion processes typically

produce more NO than NO<sub>2</sub>, the two easily interconvert in the presence of oxygen species. Thus, an effective emission control systems must account for both species. Additionally, NO<sub>2</sub>-producing processes present a greater health risk than NO-producing processes, as is reported in the literature.

[0006] In view of the health risks posed by atmospheric  $NO_x$ , the U.S. Environmental Protection Agency regulates acceptable levels of  $NO_x$  emissions for U.S. cars and trucks.

#### SUMMARY OF THE INVENTION

[0007] In one embodiment, the invention comprises an exhaust aftertreatment control system for exhaust gases. The exhaust aftertreatment control system comprises a selective catalytic reduction (SCR) catalyst in fluid communication with an exhaust stream from an engine, a urea solution injector, upstream of the SCR catalyst, in fluid communication with the exhaust stream, a gas sensor, downstream of the SCR catalyst, in fluid communication with the exhaust stream, a diagnostic unit, operatively connected to the engine, and a dosing control unit, operatively connected to the gas sensor, the urea solution injector, and the diagnostic unit. The dosing control unit receives a signal indicative of NO, NO<sub>2</sub> or NH<sub>3</sub> concentrations from the gas sensor, receives a signal indicative of engine parameters from the diagnostic unit, converts the engine parameters to NO and NO<sub>2</sub> concentrations, estimates NO and NO<sub>2</sub> concentrations downstream from the SCR catalyst, estimates the fractional surface coverage of NH<sub>3</sub> in the SCR catalyst, calculates a quantity of urea solution needed to minimize NO<sub>x</sub> emissions, and signals the urea solution injector to inject the quantity of the urea solution into the exhaust stream.

[0008] In another embodiment, the invention comprises a method for controlling emission of  $NO_x$  from an engine with an exhaust aftertreatment system having a  $NO_x$  sensor. The method comprises inputting engine parameters into a dosing control unit, measuring  $NO_x$  concentration in the exhaust, estimating  $NO_x$  and  $NO_x$  concentrations in the exhaust using the engine parameters and the measured  $NO_x$  concentration, estimating fractional surface coverage of  $NH_3$  in the catalyst, calculating an amount of urea solution needed to minimize  $NO_x$  emissions using the estimate of  $NO_x$  and  $NO_x$  concentrations in the exhaust and the estimate of fractional surface coverage of  $NH_3$  in the catalyst, and injecting an amount of urea solution.

[0009] In another embodiment, the invention comprises a method for controlling emission of  $NO_x$  from an engine with an exhaust aftertreatment system having a  $NH_3$  sensor. The method comprises inputting engine parameters into a dosing control unit, measuring  $NH_3$  concentration in the exhaust, estimating NO and  $NO_2$  concentrations in the exhaust using the engine parameters and the measured  $NH_3$  concentration, estimating fractional surface coverage of  $NH_3$  in the catalyst, calculating an amount of urea solution needed to minimize  $NO_x$  emissions using the estimate of NO and  $NO_2$  concentrations in the exhaust and the estimate of fractional surface coverage of  $NH_3$  in the catalyst, and injecting an amount of urea solution.

[0010]In another embodiment, the invention comprises an exhaust aftertreatment control system for exhaust gases from an industrial burner. The burner may be fueled with a variety of fuels, including natural gas, heating oil, biomass, or municipal solid waste. The exhaust aftertreatment control system comprises a selective catalytic reduction (SCR) catalyst in fluid communication with an exhaust stream from a burner, a urea solution injector, upstream of the SCR catalyst, in fluid communication with the exhaust stream, a gas sensor, downstream of the SCR catalyst, in fluid communication with the exhaust stream, a diagnostic unit. operatively connected to the burner, and a dosing control unit, operatively connected to the gas sensor, the urea solution injector, and the diagnostic unit. The dosing control unit receives a signal indicative of NO, NO<sub>2</sub> or NH<sub>3</sub> concentrations from the gas sensor, receives a signal indicative of burner parameters from the diagnostic unit, converts the burner parameters to NO and NO<sub>2</sub> concentrations, estimates NO and NO<sub>2</sub> concentrations downstream from the SCR catalyst, estimates the fractional surface coverage of NH<sub>3</sub> in the SCR catalyst, calculates a quantity of urea solution needed to minimize NO<sub>x</sub> emissions, and signals the urea solution injector to inject the quantity of the urea solution into the exhaust stream.

[0011] Other aspects of the invention will become apparent by consideration of the detailed description and accompanying drawings.

#### BRIEF DESCRIPTION OF THE DRAWINGS

[0012] FIG. 1 is a block diagram of an embodiment of an exhaust aftertreatment control system using a NO<sub>x</sub> sensor.

[0013] FIG. 2 is a block diagram of an embodiment of a dosing control unit for the exhaust aftertreatment control system of FIG. 1.

- [0014] FIG. 3 is a block diagram of an embodiment of an exhaust aftertreatment control system using a NH<sub>3</sub> sensor.
- [0015] FIG. 4 is a block diagram of an embodiment of a dosing control unit for the exhaust aftertreatment control system FIG. 3.
- [0016] FIG. 5 depicts an experimental setup used to verify an estimator.
- [0017] FIG. 6 shows the engine conditions used to test the four-state model in Example 1, described hereafter.
- [0018] FIG. 7 shows the measured NO, NO<sub>2</sub>, and NH<sub>3</sub> input concentrations in Example 1, described hereafter.
- [0019] FIG. 8 compares the estimated NO, NO<sub>2</sub>, and NH<sub>3</sub> output concentrations to experimental data in Example 1, described hereafter.
- [0020] FIG. 9 compares urea consumption and fractional surface coverage of the SCR catalyst by  $NH_3$  for a  $NO_x$ -sensor-based and a  $NH_3$ -sensor-based exhaust aftertreatment system.
- [0021] FIG. 10 is a block diagram of an embodiment of an exhaust aftertreatment control system using a  $NO_x$  sensor.

#### **DETAILED DESCRIPTION**

[0022] Before any embodiments of the invention are explained in detail, it is to be understood that the invention is not limited in its application to the details of construction and the arrangement of components set forth in the following description or illustrated in the following drawings. The invention is capable of other embodiments and of being practiced or of being carried out in various ways. While the following detailed description describes the invention through reference to embodiments involving an internal combustion engine exhaust treatment system, it should be understood that other systems, both stationary and mobile, where exhaust is treated for  $NO_x$  emissions, may also be suitable for a use in accordance with principles of the invention.

[0023] Also, it is to be understood that the phraseology and terminology used herein is for the purpose of description and should not be regarded as limiting. The use of "including," "comprising," or "having" and variations thereof herein is meant to encompass the items listed thereafter and equivalents thereof as well as additional items. Unless specified or limited otherwise, the terms "mounted," "connected," "supported," and "coupled" and variations thereof are used broadly and encompass both direct and indirect mountings, connections, supports, and couplings. Further, "connected" and "coupled" are not restricted to physical or mechanical connections or couplings.

[0024] Further, no admission is made that any reference, including any patent or patent document, cited in this specification constitutes prior art. In particular, it will be understood that, unless otherwise stated, reference to any document herein does not constitute an admission that any of these documents form part of the common general knowledge in the art in the United States or in any other country. Any discussion of the references states what their authors assert, and the applicant reserves the right to challenge the accuracy and pertinency of any of the documents cited herein.

[0025] One approach for controlling  $NO_x$  emissions from combustion processes involves adding ammonia (NH<sub>3</sub>) to  $NO_x$  in the presence of a catalyst to reduce the  $NO_x$  to nitrogen and water. This process, known as selective catalytic reduction (SCR), is described with the balanced equation:

$$4 \text{ NH}_3 + 2 \text{NO} + 2 \text{NO}_2 \rightarrow 4 \text{N}_2 + 6 \text{ H}_2 \text{O}$$

A number of known catalysts may be used in the selective reduction process. See, for example, US 5,451,387, which describes catalysts that may be used for this application.

[0026] Unfortunately, ammonia poses its own risks to human health and the environment. Ammonia is caustic, and irritates the lungs. Excess ammonia released into the environment often finds its way into the water system, where it threatens the health of plants and animals that contact the water. Because of this risk, selective catalytic reduction with ammonia requires careful monitoring of ammonia emissions as well as  $NO_x$  emissions. The unintended release of ammonia is known as ammonia "slip."

[0027] For large-scale applications, such as power plants, compressed ammonia gas is used for the catalytic reduction of NO<sub>x</sub> to nitrogen and water. However, for small-scale or mobile processes, it is often impractical to store/transport ammonia gas for selective catalytic

reduction. In some small-scale systems, it is feasible to use highly concentrated ammonia solutions. Highly concentrated ammonia solutions are caustic, however, and present storage and handling issues.

[0028] Urea ( $NH_2CONH_2$ ) solutions provide a suitable alternative to ammonia gas or ammonia solutions for SCR-based  $NO_x$  control systems. In a urea SCR system, an aqueous urea solution is injected into a hot exhaust stream, where the urea thermally decomposes into ammonia and carbon dioxide. The resultant ammonia mixes with the exhaust gas prior to entering a catalytic unit where the  $NO_x$  is converted to nitrogen and water. Because urea solutions are unreactive, many of the storage and handling concerns are alleviated.

[0029] Urea-SCR systems have seen limited use in diesel trucks and diesel-powered marine vessels. Commercially-available urea-SCR systems typically comprise a urea injector (Bosch GmbH, Stuttgart, Germany), an SCR catalyst, and a NO<sub>x</sub> sensor (Siemens VGO Trading Ltd., Schwalbach, Germany). State of the art systems employ a rudimentary look-up table based upon engine parameters to provide a guess for the appropriate urea injection rate. These look-up tables are often subject to error due to engine load and exhaust temperature. The NO<sub>x</sub> sensor provides feedback as to when the system is exceeding NO<sub>x</sub> limits, allowing the injection rate to be adjusted in an iterative fashion. Alternatively, or in addition, an NH<sub>3</sub> sensor may be used to provide feedback to minimize the amount of ammonia slip.

[0030] Because the current systems do not model the catalytic reactions in real time, current systems use more urea solution than necessary and have unacceptable levels of ammonia slip. Additionally, the current systems may be ineffective for controlling NOx emissions for vehicles that stop and start frequently, e.g., busses and garbage trucks.

[0031] The German Association of the Automobile Industry has championed urea-SCR NO<sub>x</sub> control, and advocates for a network of filling stations where diesel vehicles can replenish urea solutions. Suitable urea solutions are now commercially-available throughout Europe under the trade name AdBlue. AdBlue solutions comprise 32.5% urea in deionized water, and are sold by BASF, Yara, and GreenChem, among others. These solutions typically cost \$4-7/gal. The rate of consumption varies tremendously depending upon the type of engine and the loading, however, it is not uncommon for an engine to consume 5 gallons of urea solution for each 100 gallons of diesel consumed. The high cost of urea solutions is motivating equipment manufacturers to maximize NO<sub>x</sub> control per unit of solution.

[0032] The invention described herein provides a closed-loop system for minimizing NO<sub>x</sub> emissions from combustion processes while minimizing the consumption of urea solution and producing negligible ammonia slip. In one embodiment, the aftertreatment system comprises a selective catalytic reduction (SCR) unit, a urea injector, a gas sensor, and a dosing control unit. The gas sensor is sensitive to one or more gases that are to be controlled, including, but not limited to, NO, NO<sub>2</sub>, or NH<sub>3</sub>. The catalyst within the SCR unit facilitates the conversion of ammonia and NO<sub>x</sub> to nitrogen and water. The catalyst may comprise any of a number of catalysts known to work for selective catalytic reduction, including, but not limited to, compounds of vanadium, iron, or copper, and metal zeolites. The urea injector is responsive to the dosing control unit which regulates the concentration of ammonia in the SCR unit by increasing or decreasing the rate of urea injection.

[0033] In one embodiment, the gas sensor is a  $NO_x$  sensor. The  $NO_x$  sensor provides  $NO_x$  concentrations to a dosing control unit, thus allowing the dosing control unit to estimate the  $NO_x$  and  $NO_2$  concentrations upstream from the SCR. In an alternate embodiment, the gas sensor is an  $NH_3$  sensor that provides the  $NH_3$  concentration to the dosing control unit, thus allowing the dosing control unit to estimate the  $NO_x$  and  $NO_y$  concentrations upstream from the SCR. In both embodiments, the dosing control unit is capable of calculating the optimum urea injection rate to minimize  $NO_x$  emissions with minimal ammonia slip.

[0034] In many embodiments, the dosing control unit is suitably a single electronic unit. Nonetheless, the dosing control unit can be functionally described as comprising two components, an estimator and a controller. Using engine or burner operating parameters and total NO<sub>x</sub> or total NH<sub>3</sub> output, the estimator employs a mathematical model to estimate the concentrations of NO, NO<sub>2</sub>, and NH<sub>3</sub> downstream from the SCR catalyst as well as the fractional surface coverage of NH<sub>3</sub> within the SCR catalyst. The estimator then passes these estimates to the controller which calculates the amount of NH<sub>3</sub> needed to minimize NO<sub>x</sub> output with minimal NH<sub>3</sub> slip. Having calculated the optimum NH<sub>3</sub> concentration, the controller then directs the urea injector to inject urea at an appropriate rate.

[0035] FIG. 1 is a block diagram of an embodiment of an exhaust aftertreatment system 10 employing a NO<sub>x</sub> sensor according to the principles of the invention. The exhaust aftertreatment system 10 includes a urea injector 30, a selective catalytic reduction (SCR) catalyst 35, a NO<sub>x</sub> sensor 40, and a dosing control unit 50. An internal combustion engine 15 produces an exhaust stream 20 containing nitrogen oxides, including NO and NO<sub>2</sub>. As

exhaust stream 20 moves through aftertreatment system 10, exhaust stream 20 is mixed with an atomized urea solution 25 from urea injector 30 to create urea-enriched exhaust 32. Urea-enriched exhaust 32 then enters SCR catalyst 35. SCR catalyst 35 contains metal zeolites arranged in a honeycomb such that the passing exhaust gasses will interact with a multitude of catalytic sites. As urea-enriched exhaust 32 is swept into SCR catalyst 35, the urea breaks down into ammonia (NH<sub>3</sub>) and carbon dioxide (CO<sub>2</sub>). Once inside SCR catalyst 35, the nitrogen oxides react with ammonia in the presence of the metal zeolite to produce nitrogen gas (N<sub>2</sub>) and water vapor (H<sub>2</sub>O). NO<sub>x</sub> sensor 40 measures the NO<sub>x</sub> in a scrubbed exhaust stream 37 exiting SCR catalyst 35, and provides a signal indicating the concentration of NO<sub>x</sub> to dosing control unit 50. Dosing control unit 50 uses the measured NO<sub>x</sub> concentration, as well as operating parameters from engine 15, which originate from an onboard diagnostic unit (OBD) 60. Operating parameters may include exhaust gas temperature, speed, load, airflow rate, fuel-flow rate and/or exhaust gas pressure. Dosing control unit 50 then calculates the rate at which atomized urea solution 25 should be mixed with exhaust stream 20, and adjusts urea injector 30 accordingly.

[0036] Dosing control unit 50 is an electronic control unit comprising random access memory (RAM) 52, read only memory (ROM) 55, and a microprocessor (CPU) 58. Functionally, dosing control unit 50 may be described as comprising an estimator 70 and a controller 80, as shown in FIG. 2. Estimator 70 collects information about the engine from OBD 60, as well as information from NO<sub>x</sub> sensor 40. Using this information, estimator 70 estimates the concentration of NO, NO<sub>2</sub>, and NH<sub>3</sub> in exhaust stream 20, prior exhaust stream 20 entering SCR catalyst 35. Estimator 70 also estimates the NH<sub>3</sub> loading,  $\theta$ , in the catalyst using equation VIII, discussed hereafter. Controller 80 then uses the estimated NH<sub>3</sub> loading and the NO, NO<sub>2</sub> and NH<sub>3</sub> concentrations as inputs to equation XIV, described hereafter. Controller 80 then adjusts the rate at which atomized urea solution 25 is injected exhaust stream 20 to minimize NO, NO<sub>2</sub> and NH<sub>3</sub> emissions in scrubbed exhaust stream 37.

[0037] The estimator employs a four-state model to estimate the concentrations of NO, NO<sub>2</sub> and NH<sub>3</sub> within SCR catalyst 35. The four-state model encompasses six chemical equations (I-VI) which describe the simultaneous reactions within SCR catalyst 35

$$4 \text{ NH}_3 + 2 \text{NO} + 2 \text{NO}_2 \rightarrow 4 \text{N}_2 + 6 \text{ H}_2 \text{O}$$
 (I)

$$4 \text{ NH}_3 + 4 \text{NO} + \text{O}_2 \rightarrow 4 \text{N}_2 + 6 \text{ H}_2 \text{O}$$
 (II)

$$4 \text{ NH}_3 + 3\text{NO}_2 \rightarrow 7/2\text{N}_2 + 6 \text{ H}_2\text{O}$$
 (III)

$$4 NH3 + 3O2 \rightarrow 2N2 + 3H2O$$
 (IV)  

$$NH3 + S \rightarrow NH3*$$
 (V)  

$$NH3* \rightarrow NH3 + S$$
 (VI)

wherein S represents the free active sites in the catalyst, and NH<sub>3</sub>\* is NH<sub>3</sub> adsorbed at an active site in the catalyst. Incorporating the appropriate rate constants for chemical reactions I-VI, the time-dependent states of NO, NO<sub>2</sub>, NH<sub>3</sub>, and the ammonia storage in the catalyst can be represented as

$$\begin{split} \dot{C}_{NO} &= -\overline{Q}C_{NO} - k_1\Omega\theta C_{NO}C_{NO_2} - k_2\Omega\theta C_{NO}C_{O_2} + \overline{Q}C_{NO,in} \\ \dot{C}_{NO_2} &= -\overline{Q}C_{NO_2} - k_1\Omega\theta C_{NO}C_{NO_2} - k_3\Omega\theta C_{NO_2} + \overline{Q}C_{NO_2,in} \\ \dot{\theta} &= -(k_6 + k_4)\theta + k_5C_{NH_3} - k_1C_{NO}C_{NO_2}\theta - k_2C_{NO}C_{O_2}\theta - k_3C_{NO_2}\theta - k_5\theta C_{NH_3} \\ \dot{C}_{NH_3} &= k_6\Omega\theta - \left(k_5\Omega + \overline{Q}\right)C_{NH_3} + k_5\Omega\theta C_{NH_3} + \overline{Q}k_5C_{NH_3,in} \end{split}$$
 (VII)

wherein

 $\overline{Q}$  is the normalized flow rate into the SCR catalyst 35 in sec<sup>-1</sup>, wherein  $\overline{Q} = Q/V$ , and Q is the actual volumetric flow rate in actual m<sup>3</sup>/sec, and V is the volume of the exhaust gas in the substrate channel in m<sup>3</sup>,

 $C_i$  is the concentration of species i in moles/m<sup>3</sup>, where i = NO,  $NO_2$ ,  $NH_3$ , or  $O_2$ ,

 $C_{i,in}$  is the concentration of species *i* entering SCR catalyst 35, where  $i = NO, NO_2, NH_3$ ,

 $\theta$  is the fractional surface coverage of SCR catalyst 35 by NH<sub>3</sub>,

 $\Omega$  is SCR catalyst 35 adsorption capacity in moles of NH<sub>3</sub>/m<sup>3</sup>,

 $k_l$  is the rate constant for equation I,

 $k_2$  is the rate constant for equation II,

 $k_3$  is the rate constant for equation III,

 $k_4$  is the rate constant for equation IV,

 $k_5$  is the rate constant for equation V,

 $k_6$  is the rate constant for equation VI.

[0038] In one embodiment, an estimator would employ a converging linear state equation of the form

$$\begin{cases} \dot{\boldsymbol{C}}_{NO,est} \\ \dot{\boldsymbol{C}}_{NO_{2},est} \\ \dot{\boldsymbol{\theta}}_{est} \\ \dot{\boldsymbol{C}}_{NH_{3},est} \end{cases} = \vec{f} \left( \boldsymbol{C}_{NO,est}, \boldsymbol{C}_{NO_{2},est}, \boldsymbol{\theta}_{est}, \boldsymbol{C}_{NH_{3},est}, \boldsymbol{C}_{NO,in}, \boldsymbol{C}_{NO_{2},in} \right) + \vec{L} \left( \boldsymbol{C}_{NO,meas} + \boldsymbol{C}_{NO_{2},meas} - \boldsymbol{C}_{NO,est} - \boldsymbol{C}_{NO_{2},est} \right)$$

$$(VIII)$$

to estimate  $\theta$  and the concentrations of NO, NO<sub>2</sub>, and NH<sub>3</sub> downstream from the SCR catalyst.  $\vec{L}$  serves to stabilize the estimator. That is, when the system begins, it is necessary to employ an estimate for the NO and NO<sub>2</sub>, however, once the system is running, the measured and estimated values of NO and NO<sub>2</sub> should converge, and  $\vec{L}$  will consequently disappear. The estimator's ability to predict concentrations of NO, NO<sub>2</sub>, and NH<sub>3</sub> was verified experimentally, as described hereafter in Example 1.

[0039] For some applications, it may be computationally expensive to estimate the four states using equation VIII. In an alternative embodiment, an acceptable estimate may be obtained quicker and/or with less computational resources by solving  $\dot{C}_{NO}$ ,  $\dot{C}_{NO_2}$ , and  $\dot{C}_{NH_H}$  as steady state expressions. This approximation results in a single state model for the fractional NH<sub>3</sub> coverage in the catalyst

$$\dot{\theta} = -(k_6 + k_4)\theta + k_5C_{NH_3} - k_1C_{NO}C_{NO_2}\theta - k_2C_{NO}C_{O_2}\theta - k_3C_{NO_2}\theta - k_5\theta C_{NH_3}$$
(IX)

and a corresponding linear state estimator of the form

$$\dot{\theta}_{est} = \vec{f} \Big( C_{NO}, C_{NO_2}, \theta, C_{NH_3}, C_{NO,in}, C_{NO_2,in}, C_{NH_3,in} \Big) + \vec{L} \Big( C_{NO_x,meas} - C_{NO_x,est} \Big)$$
 (X)

where  $C_{NO_x,meas}$  is the measurement from the NO<sub>x</sub> sensor 16. Because the NO<sub>x</sub> sensor is cross-sensitive to NH<sub>3</sub>,  $C_{NO_x,est}$  must actually incorporate an estimate of the downstream NH<sub>3</sub> concentration. Thus,  $C_{NO_x,est}$  is

$$C_{NO_x,est} = A_1 C_{NO} + A_2 C_{NO_2} + A_3 (\alpha) C_{NH_3}$$
(XI)

where  $A_1$ ,  $A_2$ , and  $A_3(\alpha)$  are specific to the aftertreatment control system.

[0040] The initial concentrations of NO and NO<sub>2</sub> are calculated by estimator 70 using a model specific for internal combustion engine 15 and additional emission equipment that may lie between internal combustion engine 15 and SCR 35 (e.g. a catalyzed particulate filter). Onboard diagnostic unit 60 reports engine parameters, such as engine speed, load, exhaust gas temperatures, flow rates and exhaust gas pressure, to estimator 70. Estimator 70 uses these engine parameters to produce an estimate of NO and NO<sub>2</sub> concentrations for the purpose of estimating  $C_{NO_x}$ . Estimator 70 also estimates the NH<sub>3</sub> entering SCR catalyst 35 based upon the rate that urea injector 30 is injecting atomized urea solution 25 into exhaust stream 20 previous to urea-enriched exhaust 32 entering SCR catalyst 35.

[0041] Controller 80 functions to minimize the NO, NO<sub>2</sub>, and NH<sub>3</sub> in scrubbed exhaust stream 37 while using as little atomized urea solution 25 as necessary. The controller's efficiency at converting NO and NO<sub>2</sub> to N<sub>2</sub> per mole of NH<sub>3</sub> can be expressed as

$$\eta_{T} = \frac{C_{NO,in} + C_{NO_{2},in} - C_{NO,out} - C_{NO_{2},out} - \lambda C_{NH_{3},out}}{C_{NO,in} + C_{NO_{2},in}}$$
(XIII)

wherein  $\lambda$  accounts for NH<sub>3</sub> slipping out of the catalyst.

[0042] Defining a new function  $e_{\bar{p}} = \overline{p}_{des} - \overline{p}$ , where  $\overline{p}$  is a linear combination of  $C_{NO}$ ,  $C_{NO_2}$ ,  $C_{NH_3}$ , and  $\overline{p}_{des}$  is a linear combination of the desired concentrations, i.e.,  $\overline{p}_{des} = C_{NO,des} + C_{NO_2,des} + \alpha C_{NH_3,des}$ , the controller's goal can be represented as  $e_{\bar{p}} = \dot{e}_{\bar{p}} = 0$ .

[0043] Substituting  $\eta_T = 1 - p_{des}$ , the dynamic concentration of NH<sub>3</sub> within the SCR catalyst 15 for the goal  $e_{\bar{p}} = \dot{e}_{\bar{p}} = 0$  can be expressed as,

$$\begin{split} C_{NH_3,in,dyn} &= C_{NH_3,est} + \frac{1}{\lambda} \Big( C_{NO,est} + C_{NO_2,est} - C_{NO,in} - C_{NO_2,in} \Big) \\ &+ \frac{1}{\overline{Q}} \Big( k_5 \Omega \Big( 1 - \theta_{est} \Big) C_{NH_3,est} - k_6 \Omega \theta_{est} \Big) \\ &+ \frac{1}{\lambda \overline{Q}} \Big( \dot{\overline{p}}_{des} + 2\Omega \theta_{est} k_1 C_{NO,est} C_{NO_2,est} + \Omega \theta_{est} k_2 C_{NO,est} C_{O_2} + \Omega \theta_{est} k_3 C_{NO_2,est} \Big) \end{split}$$
 (XIV)

[0044] As the system, in accordance with the invention, is subject to nonlinear dynamics, controller 80 must employ an additional correction term to ensure stability in the event of measurement errors or other transient disturbances. Therefore, an additional term is added that penalizes deviations from the objective  $e_{\bar{p}} = 0$ ,

$$C_{NH_{2},in} = C_{NH_{2},in,dyn} - \Gamma sgn(e_{\overline{p}})$$
 (XV)

where  $\Gamma$  is a control variable that can be tuned to meet the control objective. Based upon the sign of  $sgn(e_{\overline{p}})$ , the signum function changes:

$$sgn(x) = \begin{cases} -1: x < 0 \\ 0: x = 0 \\ 1: x > 0 \end{cases}$$
 (XVI)

Thus, given estimates of NO, NO<sub>2</sub>, NH<sub>3</sub> and free catalytic sites from estimator 70, controller 80 calculates the optimum concentration of ammonia to achieve its efficiency goals. Upon calculating this value, controller 80 adjusts the rate of injection of atomized urea solution 25 from injector 30 assuming a 2:1 molar ratio of ammonia to atomized urea solution 25.

[0045] As an alternative to NO<sub>x</sub> sensor 40 downstream of SCR catalyst 35, an NH<sub>3</sub> sensor 140 can be used for SCR control applications. See FIG. 3. NH<sub>3</sub> sensor 140 has the additional, beneficial, characteristic of having minimal cross-sensitivity toward NO and NO<sub>2</sub> species, as is reported in the literature (*See* Wang, DEER 2007; Moos, <u>Sensors and Actuators</u>, 2002.) Because of the minimal cross-sensitivity, ammonia-sensor-based exhaust aftertreatment system 110 requires fewer variables than exhaust aftertreatment system 10, employing NO<sub>x</sub> sensor 40.

[0046] FIG. 3 is a block diagram depicting an ammonia-sensor-based exhaust aftertreatment system 110 embodying the principles of the invention for catalytic reduction of NO<sub>x</sub> in the presence of ammonia. The ammonia-sensor-based exhaust aftertreatment system 110 includes urea injector 30, SCR catalyst 35, an NH<sub>3</sub> sensor 140, and an ammonia-sensorbased dosing control unit 150. Internal combustion engine 15 produces exhaust stream 20 containing nitrogen oxides, including NO and NO<sub>2</sub>. As exhaust stream 20 moves through the ammonia-sensor-based aftertreatment system 110, exhaust stream 20 is mixed with atomized urea solution 25 from urea injector 30 to create urea-enriched exhaust 32. Urea-enriched exhaust 32 then enters SCR catalyst 35. SCR catalyst 35 contains metal zeolites arranged in a honeycomb such that the passing exhaust gasses will interact with a multitude of catalytic sites. As urea-enriched exhaust 32 is swept into SCR catalyst 35, the urea breaks down into ammonia (NH<sub>3</sub>) and carbon dioxide (CO<sub>2</sub>). Once inside SCR catalyst 35, the nitrogen oxides react with ammonia in the presence of the metal zeolites to produce nitrogen gas (N<sub>2</sub>) and water vapor (H<sub>2</sub>O). NH<sub>3</sub> sensor 140 measures the NH<sub>3</sub> exiting SCR catalyst 35, providing a signal indicating the concentration of NH<sub>3</sub> in scrubbed exhaust stream 37 to the ammonia-

sensor-based dosing control unit 150. Ammonia-sensor-based dosing control unit 150 uses the measured NH<sub>3</sub> concentration, as well as internal combustion engine 15 operating parameters, which originate from OBD 60. Engine parameters may include exhaust gas temperature, speed, load, air-flow rate, fuel-flow rate and/or exhaust gas pressure. Ammonia-sensor-based dosing control unit 150 then calculates the rate at which the atomized urea solution 25 should be mixed with exhaust stream 20, and adjusts injector 30 accordingly.

[0047] Ammonia-sensor-based dosing control unit 150 is an electronic control unit comprising random access memory (RAM) 152, read only memory (ROM) 155, and a microprocessor (CPU) 158. Functionally, the ammonia-sensor-based dosing control unit 150 may be described as comprising an ammonia-sensor-based estimator 170 and an ammoniasensor-based controller 180. The functions of ammonia-sensor-based estimator 170 and ammonia-sensor-based controller 180 are illustrated in FIG. 4. Ammonia-sensor-based estimator 170 collects information about the engine from OBD 60, as well as information from NH<sub>3</sub> sensor 140. Using this information, ammonia-sensor-based estimator 170 estimates the concentration of NO, NO2, and NH3 in exhaust stream 20, prior to exhaust stream 20 entering SCR catalyst 35. Ammonia-sensor-based estimator 170 also estimates the NH<sub>3</sub> loading,  $\theta$ , in the catalyst using equation VIII, discussed above. Ammonia-sensor-based controller 180 then uses the estimated NH<sub>3</sub> loading and the NO, NO<sub>2</sub> and NH<sub>3</sub> concentrations as inputs to equation XIV, described previously. Ammonia-sensor-based controller 180 then adjusts the rate at which atomized urea solution 25 is injected into exhaust stream 20 to minimize NO, NO<sub>2</sub> and NH<sub>3</sub> in scrubbed exhaust stream 37.

[0048] Ammonia-sensor-based estimator 170 employs the same four-state model, described above with respect to equations (I) – (VII), to estimate the concentrations of NO, NO<sub>2</sub> and NH<sub>3</sub> within SCR catalyst 35. However, ammonia-sensor-based estimator 170 employs a modified converging linear state equation of the form

$$\begin{cases}
\dot{C}_{NO,est} \\
\dot{C}_{NO_{2},est} \\
\dot{\theta}_{est} \\
\dot{C}_{NH_{3},est}
\end{cases} = \vec{f}\left(C_{NO,est}, C_{NO_{2},est}, \theta_{est}, C_{NH_{3},est}, C_{NO,in}, C_{NO_{2},in}\right) + \vec{L}\left(C_{NH_{3},meas} - C_{NH_{3},est}\right)$$
(XVII)

to estimate  $\theta$  and the concentrations of NO, NO<sub>2</sub>, and NH<sub>3</sub> downstream from the SCR catalyst.  $\vec{L}$  serves to stabilize the estimator. That is, when the system begins, it is necessary to employ an estimate for the NH<sub>3</sub>, however, once the system is running, the measured and estimated values of NH<sub>3</sub> should converge, and  $\vec{L}$  will consequently disappear.

[0049] For some applications, it may be computationally expensive to estimate the four states using equation XVII. In an alternative embodiment, an acceptable estimate may be obtained quicker and/or with less computational resources by solving  $\dot{C}_{NO}$ ,  $\dot{C}_{NO_2}$ , and  $\dot{C}_{NH_H}$  as steady state expressions. Using equation (IX), above, to describe a single state model for the fractional NH<sub>3</sub> coverage in the catalyst, a corresponding linear state estimator for the use of an NH<sub>3</sub> sensor 140 can be determined

$$\dot{\theta}_{est} = \vec{f}(C_{NO}, C_{NO_2}, \theta_{est}, C_{NH_3}, C_{NO_iin}, C_{NO_2,in}, C_{NH_3,in}) + \vec{L}(C_{NH_{3x}, meas} - C_{NH_3, est})$$
(XVIII)

where  $C_{NH_3,meas}$  is the NH<sub>3</sub> concentration measured by the NH<sub>3</sub> sensor 140, and  $C_{NH_3,est}$  is calculated based upon an NH<sub>3</sub> sensor model given by

$$C_{NH_3,est} = A_3(\alpha)C_{NH_3} \tag{XIX}$$

where  $A_3$  is determined as a function of  $\alpha$ , which is specific to ammonia-sensor-based aftertreatment control system 110.

[0050] The initial concentrations of NO and NO<sub>2</sub> are calculated by ammonia-sensor-based estimator 170 using a model specific for internal combustion engine 15 and additional emission equipment that may lie between internal combustion engine 15 and SCR 35 (e.g. a catalyzed particulate filter). OBD 60 reports engine parameters, such as engine speed, load, exhaust gas temperatures, flow rates and exhaust gas pressure, to ammonia-sensor-based estimator 170. Ammonia-sensor-based estimator 170 uses these engine parameters to produce an estimate of NO and NO<sub>2</sub> concentrations for the purpose of estimating  $C_{NH_3}$ . The estimator also estimates the NH<sub>3</sub> entering SCR catalyst 35 based upon the rate that urea injector 30 is injecting atomized urea solution 25 into the exhaust stream previous to entering SCR catalyst 35.

[0051] Ammonia-sensor-based controller 180 functions to minimize the NO, NO<sub>2</sub>, and NH<sub>3</sub> leaving the aftertreatment system, while using as little atomized urea solution 25 as necessary, as is described in equation (XIII) above. However, because NH<sub>3</sub> sensor 140 measures the NH<sub>3</sub> slip directly, it is unnecessary to calculate a dynamic NH<sub>3</sub> concentration

after the SCR as is done in the NO<sub>x</sub>-sensor-based exhaust aftertreatment system 10. Thus, given estimates of NO, NO<sub>2</sub>, NH<sub>3</sub> and free catalytic sites from ammonia-sensor-based estimator 170, ammonia-sensor-based controller 180 can calculate the optimum concentration of ammonia to achieve its efficiency goals. Upon calculating this value, ammonia-sensor-based controller 180 adjusts the rate of atomized urea solution 25 injection from injector 130 assuming a 2:1 molar ratio of ammonia to atomized urea solution 25.

#### [0052] **EXAMPLE 1** – Verification of Four-State Model

[0053] FIG. 5 depicts an experimental setup in accordance with the invention. A Navistar I6 7.6L diesel engine 210 was connected to a diesel oxidation catalyst 230 and a catalyzed particulate filter 240, and then outfitted with urea injector 30 and a first SCR catalyst 35 and a second SCR catalyst 38. [Generally an exhaust aftertreatment system would have only one SCR catalyst, as is depicted in FIG. 1.] The exhaust concentrations of NO and NO<sub>2</sub> were measured before first SCR catalyst 35 with a first emission test bench 250 and after second SCR catalyst 38 with a second emission test bench 255. The NH<sub>3</sub> concentrations were measured before first SCR catalyst 35 with a first Fourier Transform Infrared Spectrometer (FTIR) 260 and after second SCR catalyst 38 with a second Fourier Transform Infrared Spectrometer (FTIR) 265.

[0054] Engine 210 was run at a variety of speed and load conditions as shown in FIG. 6, resulting in a variety of engine temperatures and exhaust gas flow rates. The rate of urea injection was adjusted manually. FIG. 7 shows the NO and NO<sub>2</sub> concentrations measured with first emission test bench 250 during the engine conditions depicted in FIG. 6. The NH<sub>3</sub> concentration in FIG. 7 was calculated from the rate of urea injection in FIG. 6.

[0055] The resulting NO and NO<sub>2</sub> concentrations were measured with second emission test bench 255. The resulting NH<sub>3</sub> concentrations were measured with second FTIR 265. These measurements are illustrated with dashed lines in FIG. 8. For comparison, the NO, NO<sub>2</sub>, and NH<sub>3</sub> concentrations after first and second SCR catalysts 35 and 38 were estimated using the four state model (Equation VIII), engine 210 parameters, and the rate of urea solution injection. The estimated concentrations are depicted with a solid line.

[0056] As illustrated in FIG. 8, the four-state model provides a suitable method for estimating NO, NO<sub>2</sub>, and NH<sub>3</sub> concentrations.

[0057] As a final verification of the robustness of the invention described herein, the  $NO_x$  sensor model, described by equation (VIII) was used to control the rate of urea injection in the setup shown in FIG. 5, and then the  $NH_3$  sensor model, described by equation (XVII) was used to control the rate of urea injection in a similar experimental setup. As can be seen in FIG. 9, both sensor models result in similar rates of urea injection and fractional surface coverage,  $\theta$ , of the SCR catalyst by  $NH_3$ .

### [0058] EXAMPLE 2 – Controlling NO<sub>x</sub> Emissions from Industrial Burner

[0059] FIG. 10 is a block diagram of a prophetic embodiment of an exhaust aftertreatment system 310 employing a NO<sub>x</sub> sensor according to the principles of the invention. The exhaust aftertreatment system 310 will include a urea injector 30, a selective catalytic reduction (SCR) catalyst 35, a NO<sub>x</sub> sensor 40, and a dosing control unit 50. An oilfueled industrial burner 315 (used to power a 66 MW steam generator) will produce an exhaust stream 20 containing nitrogen oxides, including NO and NO<sub>2</sub>. Exhaust stream 20 will move through aftertreatment system 310, exhaust stream 20 will be mixed with an atomized urea solution 25 from urea injector 30 to create urea-enriched exhaust 32. Ureaenriched exhaust 32 will then enter SCR catalyst 35. SCR catalyst 35 will contain metal zeolites arranged in a honeycomb such that the passing exhaust gasses will interact with a multitude of catalytic sites. As urea-enriched exhaust 32 is swept into SCR catalyst 35, the urea will break down into ammonia (NH<sub>3</sub>) and carbon dioxide (CO<sub>2</sub>). Once inside SCR catalyst 35, the nitrogen oxides will react with ammonia in the presence of the metal zeolite to produce nitrogen gas (N<sub>2</sub>) and water vapor (H<sub>2</sub>O). NO<sub>x</sub> sensor 40 will measure the NO<sub>x</sub> in a scrubbed exhaust stream 37 exiting SCR catalyst 35, and will provide a signal indicating the concentration of NO<sub>x</sub> to dosing control unit 350. Dosing control unit 350 will use the measured NO<sub>x</sub> concentration, as well as operating parameters from industrial burner 315, which will originate from a burner diagnostics unit 360. Operating parameters will include exhaust gas temperature, steam load, steam temperature, air-flow rate, fuel-flow rate and exhaust gas pressure. Dosing control unit 350 then will calculate the rate at which atomized urea solution 25 will be mixed with exhaust stream 20, and will adjust urea injector 30 accordingly. Dosing control unit 350 is an electronic control unit comprising random access memory (RAM) 52, read only memory (ROM) 55, and a microprocessor (CPU) 58. Functionally, dosing control unit 350 will be nearly identical to dosing control unit 50, shown in FIG. 2. Dosing control unit will be programmed to estimate the concentration of NO<sub>x</sub> in

exhaust stream 20 based upon empirical measurements of NO<sub>x</sub> in exhaust stream 20 under various operating conditions.

[0060] Thus, the invention provides systems and methods for the reduction of  $NO_x$  emissions from a combustion process by measuring engine parameters and concentrations of  $NO_x$  or  $NH_3$ , estimating NO and  $NO_2$  concentrations, injecting a urea solution into the exhaust stream, and selectively reducing  $NO_x$  to  $N_2$  in the presence of a catalyst. The methods and systems minimize urea solution consumption at the same time they minimize  $NO_x$  and ammonia emissions.

[0061] The above embodiments and examples do not serve as limitations on the invention, as other embodiments may employ the principles of the invention described herein. For example, the principles of the invention may be employed to create an exhaust aftertreatment system for NO<sub>x</sub> reduction in industrial burners, such as those used in furnaces, boilers, or power plants. These burners may be powered by a variety of fuels, including, but not limited to, natural gas, heating oil, biomass, or municipal solid wastes.

[0062] Furthermore, Example 1 does not indicate that the invention is limited to diesel engines such as those used for over-the-road trucking. With only minor modifications, one skilled in the art could apply the principles of the invention described herein to reduce NO<sub>x</sub> emissions from any diesel powered vehicle or equipment, including locomotives, construction equipment, earthmoving equipment, or marine craft.

[0063] All publications, patents, and patent applications are herein expressly incorporated by reference to the same extent as if each individual publication, patent, or patent application was specifically and individually indicated by reference. In case of conflict between the present disclosure and the incorporated patents, publications and references, the present disclosure should control.

[0064] Various features and advantages of the invention are set forth in the following claims.

#### **CLAIMS**

#### What is claimed is:

- 1. An exhaust aftertreatment control system for exhaust gases, comprising:
- a selective catalytic reduction (SCR) catalyst in fluid communication with an exhaust stream from an engine,
- a urea solution injector, upstream of the SCR catalyst, in fluid communication with the exhaust stream;
- a gas sensor, downstream of the SCR catalyst, in fluid communication with the exhaust stream;
  - a diagnostic unit, operatively connected to the engine; and
- a dosing control unit, operatively connected to the gas sensor, the urea solution injector, and the diagnostic unit, wherein the dosing control unit receives a signal indicative of NO, NO<sub>2</sub> or NH<sub>3</sub> concentrations from the gas sensor, receives a signal indicative of engine operating parameters from the diagnostic unit, converts the engine operating parameters to NO and NO<sub>2</sub> concentrations, estimates NO and NO<sub>2</sub> concentrations downstream from the SCR catalyst, estimates the fractional surface coverage of NH<sub>3</sub> in the SCR catalyst, calculates a quantity of urea solution needed to minimize NO<sub>x</sub> emissions, and signals the urea solution injector to inject the quantity of the urea solution into the exhaust stream.
- 2. The exhaust aftertreatment control system of claim 1, wherein the gas sensor is a  $NO_x$  sensor.
- 3. The exhaust aftertreatment control system of claim 1, wherein the gas sensor is an NH<sub>3</sub> sensor.
- 4. The exhaust aftertreatment control system of claim 1, wherein the engine is a diesel engine.
- 5. The exhaust aftertreatment control system of claim 1, wherein the engine parameters are selected from the group consisting of engine speed, engine load, engine temperature, exhaust gas temperature, exhaust flow rate, and exhaust pressure.

6. The exhaust aftertreatment control system of claim 1, wherein the dosing control unit uses a converging linear state equation to calculate the quantity of urea solution needed to minimize  $NO_x$  emissions.

- 7. The exhaust aftertreatment control system of claim 1, wherein the selective catalytic reduction catalyst is chosen from the group consisting of vanadium compounds, iron compounds, copper compounds, and metal zeolites.
- 8. A method for minimizing NO<sub>x</sub> in an exhaust stream from an engine having an exhaust aftertreatment system comprising a urea solution injector and a catalyst, the method comprising:

collecting engine operating parameters;

measuring NO<sub>x</sub> concentrations in the exhaust stream;

estimating NO and NO<sub>2</sub> concentrations in the exhaust stream using the engine operating parameters and the measured NO<sub>x</sub> concentration;

estimating a fractional surface coverage of NH<sub>3</sub> in the catalyst;

calculating an amount of urea solution needed to minimize  $NO_x$  emissions using the estimate of NO and  $NO_2$  concentrations in the exhaust and the estimate of fractional surface coverage of  $NH_3$  in the catalyst; and

injecting an amount of urea solution into the exhaust stream.

- 9. The method of claim 8, wherein the engine is a diesel engine.
- 10. The method of claim 8, wherein the engine parameters are selected from the group consisting of engine speed, engine load, engine temperature, exhaust gas temperature, exhaust flow rate, and exhaust pressure.
- 11. The method of claim 8, wherein a converging linear state equation is used to calculate the quantity of urea solution needed to minimize NO<sub>x</sub> emissions.
- 12. A method for minimizing NO<sub>x</sub> in an exhaust stream from an engine having an exhaust aftertreatment system comprising a urea solution injector and a catalyst, the method comprising:

collecting engine operating parameters;

measuring NH<sub>3</sub> concentrations in the exhaust stream;

estimating NO and NO<sub>2</sub> concentrations in the exhaust stream using the engine operating parameters and the measured NH<sub>3</sub> concentration;

estimating a fractional surface coverage of NH<sub>3</sub> in the catalyst;

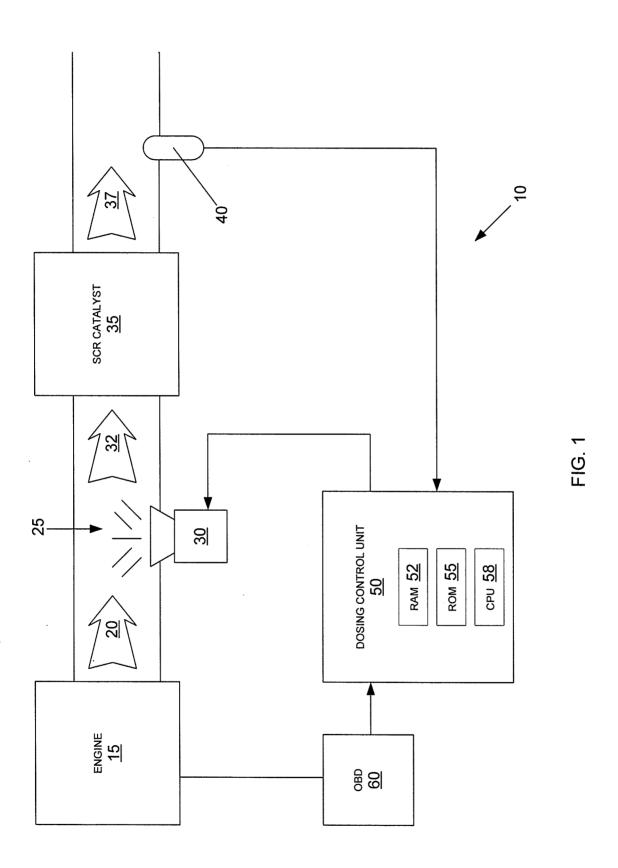
calculating an amount of urea solution needed to minimize  $NO_x$  emissions using the estimate of NO and  $NO_2$  concentrations in the exhaust and the estimate of fractional surface coverage of  $NH_3$  in the catalyst; and

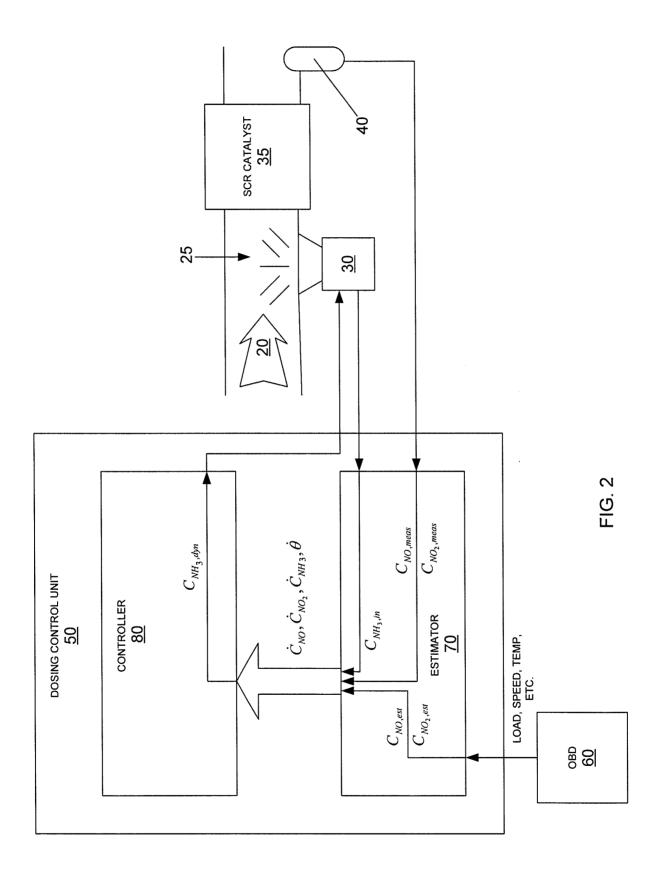
injecting an amount of urea solution into the exhaust stream.

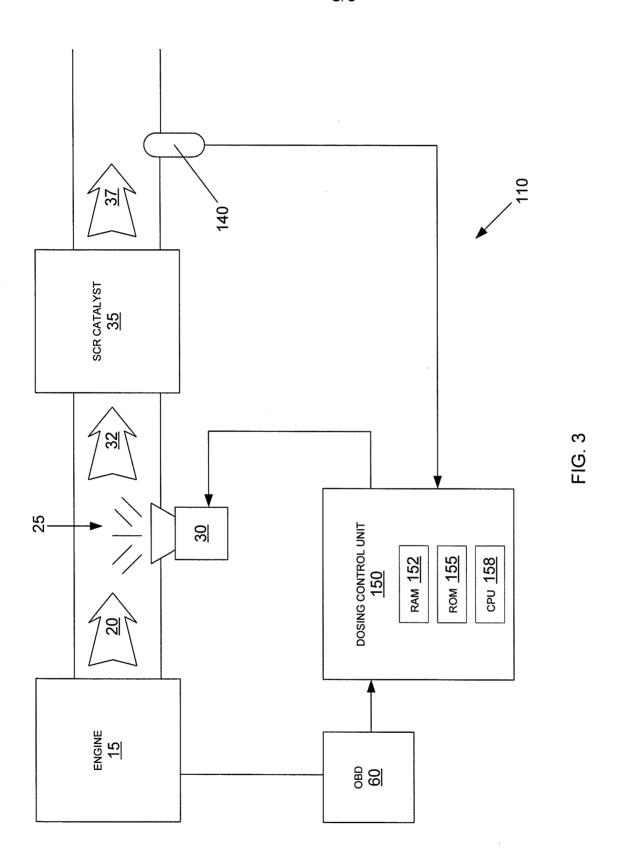
- 13. The method of claim 12, wherein the engine is a diesel engine.
- 14. The method of claim 12, wherein the engine parameters are selected from the group consisting of engine speed, engine load, engine temperature, exhaust gas temperature, exhaust flow rate, and exhaust pressure.
- 15. The method of claim 12, wherein a converging linear state equation is used to calculate the quantity of urea solution needed to minimize NO<sub>x</sub> emissions.
  - 16. An exhaust aftertreatment control system for exhaust gases, comprising:
- a selective catalytic reduction (SCR) catalyst in fluid communication with an exhaust stream from a burner,
- a urea solution injector, upstream of the SCR catalyst, in fluid communication with the exhaust stream;
- a gas sensor, downstream of the SCR catalyst, in fluid communication with the exhaust stream;
  - a burner diagnostic unit, operatively connected to the burner; and
- a dosing control unit, operatively connected to the gas sensor, the urea solution injector, and the burner diagnostic unit, wherein the dosing control unit receives a signal indicative of NO, NO<sub>2</sub> or NH<sub>3</sub> concentrations from the gas sensor, receives a signal indicative of burner operating parameters from the burner diagnostic unit, converts the burner operating parameters to NO and NO<sub>2</sub> concentrations, estimates NO and NO<sub>2</sub> concentrations downstream from the SCR catalyst, estimates the fractional surface coverage of NH<sub>3</sub> in the SCR catalyst, calculates a quantity of urea solution needed to minimize NO<sub>x</sub> emissions, and

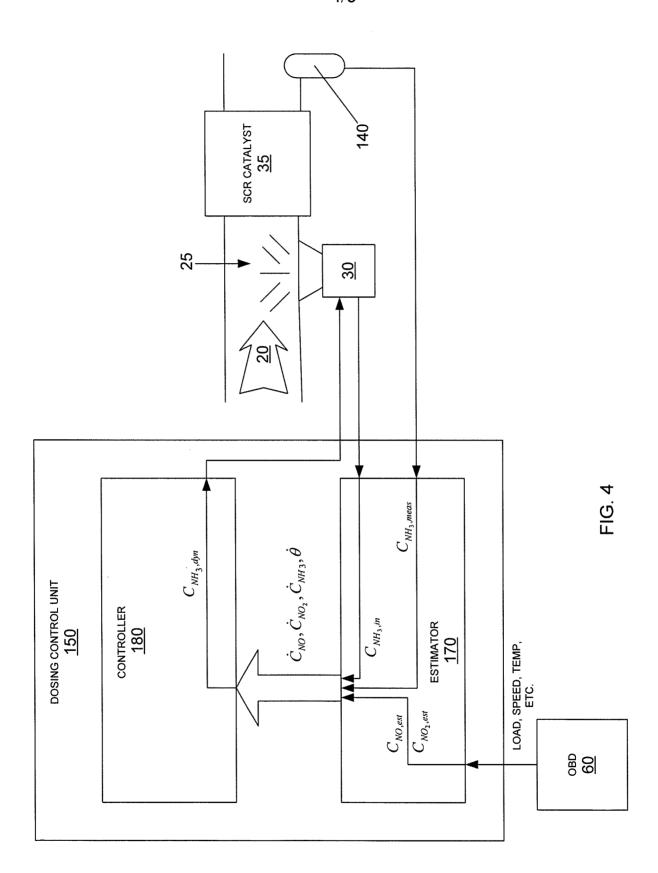
signals the urea solution injector to inject the quantity of the urea solution into the exhaust stream.

- 17. The exhaust aftertreatment control system of claim 16, wherein the gas sensor is a  $NO_x$  sensor.
- 18. The exhaust aftertreatment control system of claim 16, wherein the gas sensor is an NH<sub>3</sub> sensor.
- 19. The exhaust aftertreatment control system of claim 16, wherein the burner is supplied with a fuel selected from the group consisting of natural gas, heating oil, biomass, or municipal solid waste.









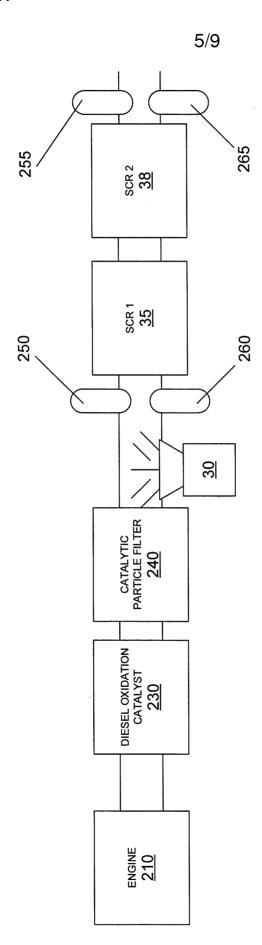
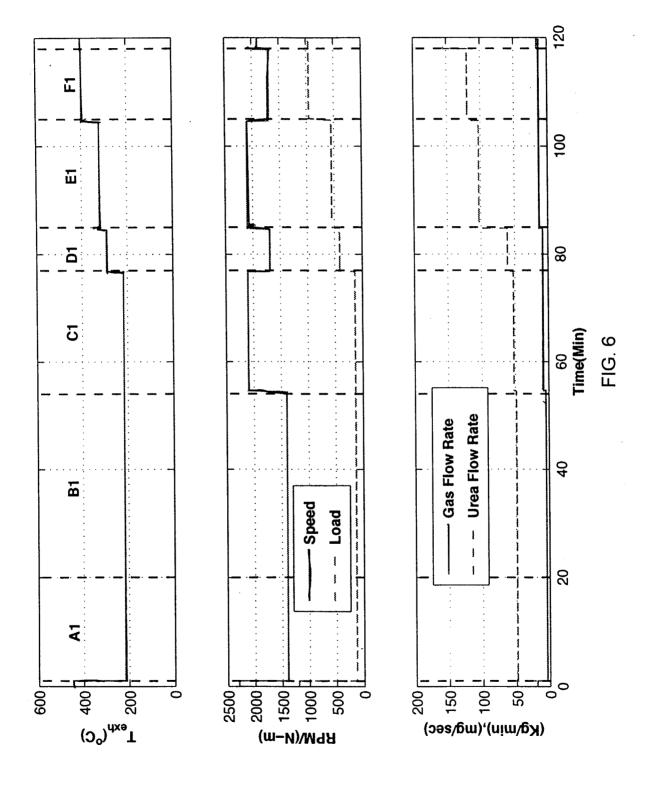
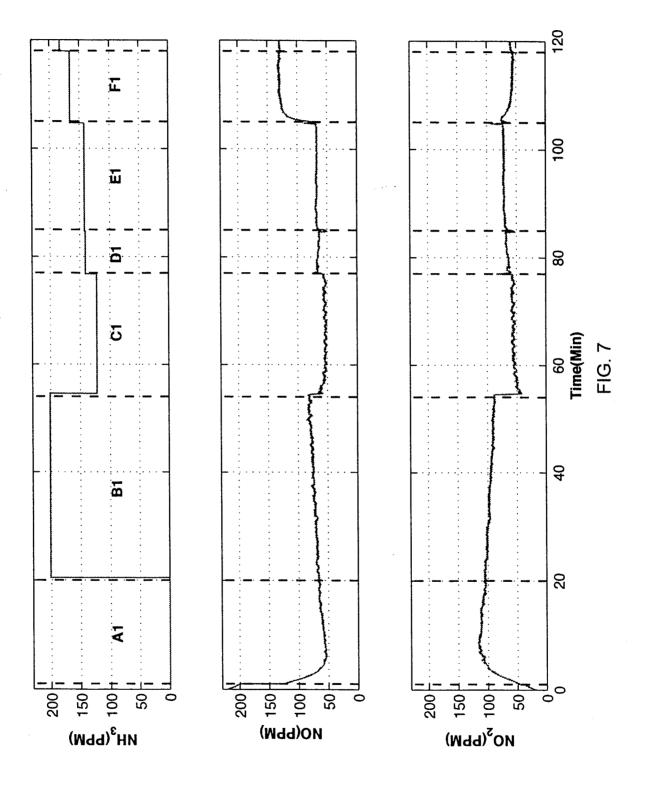
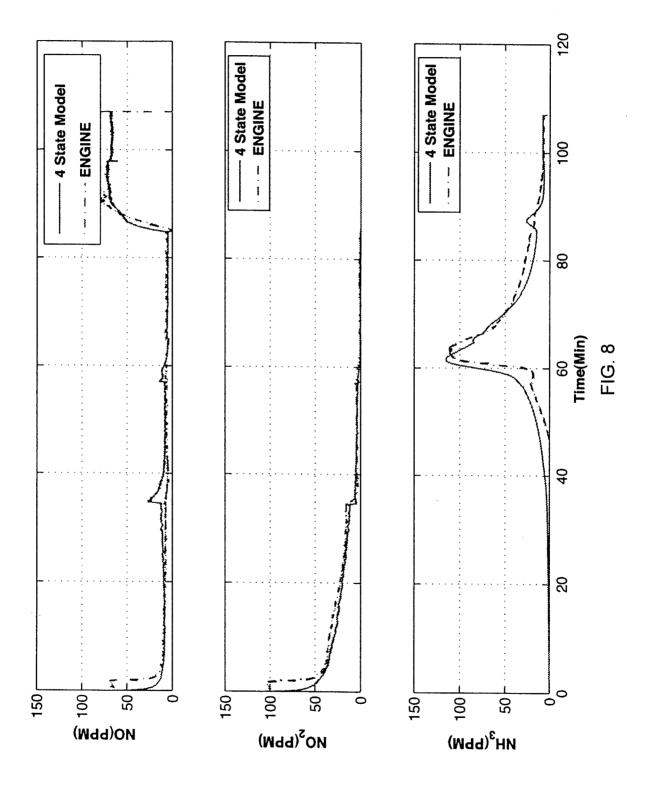
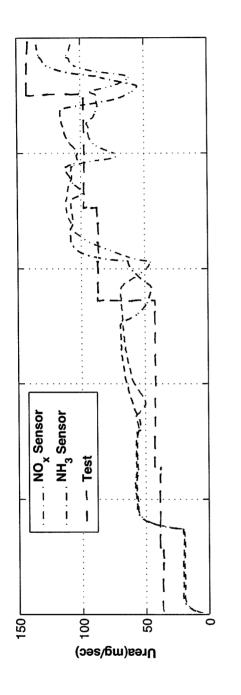


FIG. 5









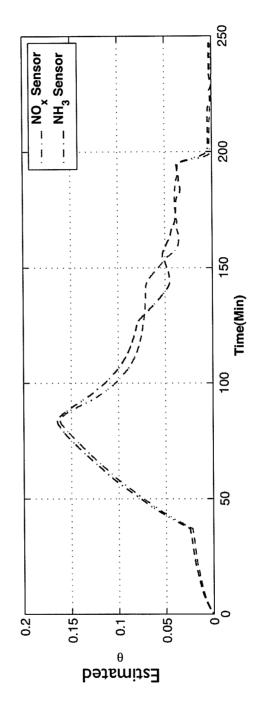


FIG. 9

#### INTERNATIONAL SEARCH REPORT

International application No. PCT/US2008/084980

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A. CLASSIFICATION OF SUBJECT MATTER IPC(8) - F01N 3/00 (2009.01) USPC - 60/286			
According to International Patent Classification (IPC) or to both national classification and IPC			
B. FIELDS SEARCHED			
Minimum documentation searched (classification system followed by classification symbols) IPC(8) - F01N 3/00, 3/08, 3/20 (2009.01) USPC - 60/282, 285, 286, 287, 295			
Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched			
Electronic data base consulted during the international search (name of data base and, where practicable, search terms used)			
PatBase			
C. DOCUMENTS CONSIDERED TO BE RELEVANT			
Category*	Citation of document, with indication, where ap	propriate, of the relevant passages	Relevant to claim No.
X	US 2002/0148220 A1 (PATCHETT et al) 17 October 2	002 (17.10.2002) entire document	1-19
Α	US 2007/0137181 A1 (UPADHYAY et al) 21 June 200	7 (21.06.2007) entire document	1-19
Α	US 2004/0098968 A1 (VAN NIEUWSTADT et al) 27 May 2004 (27.05.2004) entire document		1-19
Α	US 2003/0213234 A1 (FUNK et al) 20 November 2003 (20.11.2003) entire document		1-19
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Further documents are listed in the continuation of Box C.			
"A" docume	categories of cited documents: nt defining the general state of the art which is not considered	"T" later document published after the interned attended and not in conflict with the applic	ation but cited to understand
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16 January 2009 .		<b>29</b> JAN 2009	
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