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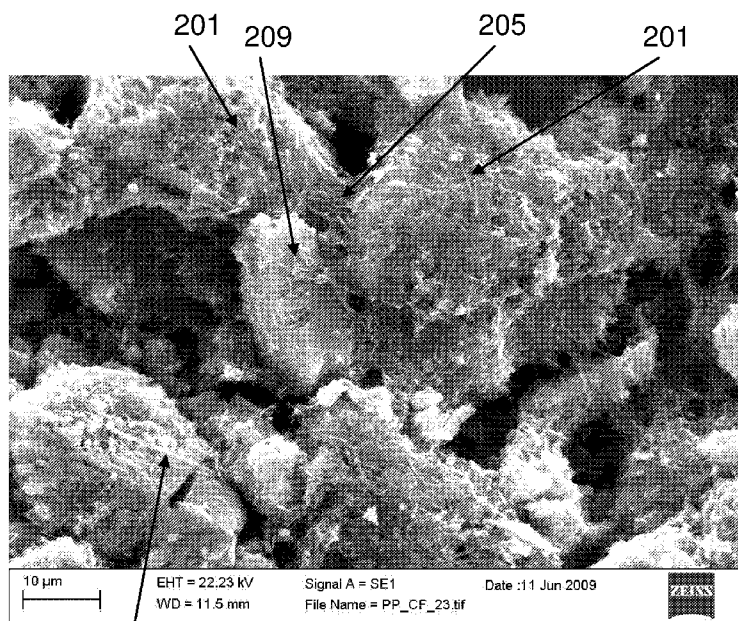


Figure 1

(57) Abstract: A composition comprising at least 50 weight % of a first particulate electroactive material and 3-15 weight % of a carbon additives mixture comprising elongate carbon nanostructures and carbon black, wherein: the elongate carbon nanostructures comprise at least a first elongate carbon nanostructure material and a different second elongate carbon nanostructure material; and the elongate carbon nanostructures : carbon black weight ratio is in the range 3 : 1 to 20 : 1.

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## COMPOSITION OF SI/C ELECTRO ACTIVE MATERIAL

**Field of the Invention**

The present invention relates to compositions comprising particles of an electroactive material and additives, and use of said compositions in devices including fuel cells and rechargeable metal ion batteries.

**Background of the Invention**

Rechargeable metal-ion batteries, for example lithium ion batteries, are extensively used in portable electronic devices such as mobile telephones and laptops, and are finding increasing application in electric or hybrid electric vehicles.

Rechargeable metal ion batteries have an anode layer; a cathode layer capable of releasing and re-inserting metal ions; and an electrolyte between the anode and cathode layers. When the battery cell is fully charged, metal ions have been transported from the metal-ion-containing cathode layer via the electrolyte into the anode layer. In the case of a graphite-based anode layer of a lithium ion battery, the lithium reacts with the graphite to create the compound  $\text{Li}_x\text{C}_6$  ( $0 \leq x \leq 1$ ). The graphite, being the electrochemically active material in the composite anode layer, has a maximum capacity of 372 mAh/g.

The use of a silicon-based active anode material, which may have a higher capacity than graphite, is also known.

WO2009/010758 discloses the etching of silicon powder in order to make silicon material for use in lithium ion batteries.

Xiao et al, J. Electrochem. Soc., Volume 157, Issue 10, pp. A1047-A1051 (2010), "Stabilization of Silicon Anode for Li-ion Batteries" discloses an anode comprising silicon particles and Ketjenblack carbon.

Lestriez et al, Electrochemical and Solid-State Letters, Vol.12, Issue 4, pp. A76-A80 (2009) "Hierarchical and Resilient Conductive Network of Bridged Carbon Nanotubes and Nanofibers for High-Energy Si Negative Electrodes" discloses a composite electrode containing multiwall carbon nanotubes and vapour-grown nanofibres.

US 2011/163274 discloses an electrode composite of silicon, a carbon nanotube and a carbon nanofibre.

It is an object of the invention to provide an anode composition for a metal ion battery that is capable of maintaining a high capacity.

It is a further objection of the invention to provide a composition for forming an anode of a metal ion battery from a slurry.

### **Summary of the Invention**

In a first aspect, the invention provides a composition comprising at least 50 weight % of a first particulate electroactive material and at least 3-15 weight % of carbon additives comprising elongate carbon nanostructures and carbon black, wherein:

the elongate carbon nanostructures comprise at least a first elongate carbon nanostructure material and a second elongate carbon nanostructure material; and

the elongate carbon nanostructures : carbon black weight ratio is in the range 3 : 1 to 20 : 1.

Optionally, the first particulate electroactive material is a silicon-comprising material.

Optionally, the composition comprises up to 80 wt %, optionally 60-80 weight %, of the first particulate electroactive material.

Optionally, the first particulate electroactive material comprises particles having a particle core and electroactive pillars extending from the particle core.

Optionally, the pillars of the silicon-comprising particles are silicon pillars.

Optionally, the core of the silicon-comprising particles comprises silicon.

Optionally, the silicon-comprising particles consist essentially of n- or p-doped silicon and wherein the pillars are integral with the core.

Optionally, the first elongate nanostructure has a mean average diameter of at least 100 nm.

Optionally, the second elongate carbon nanostructure material has a mean average diameter of no more than 90 nm, optionally a mean average diameter in the range of 40-90 nm.

Optionally, the first elongate nanostructure : second elongate nanostructure weight ratio is in the range 2.5 : 1 to 20 : 1.

Optionally, the at least first and second elongate carbon nanostructures each independently has an aspect ratio of at least 50.

Optionally, the first and second carbon elongate nanostructure materials are each independently selected from carbon nanotubes and carbon nanofibres.

Optionally, the first carbon elongate nanostructure material is a nanofibre and the second elongate carbon nanostructure material is a nanotube.

Optionally, the at least first and second elongate carbon nanostructure materials are provided in a total amount in the range of 0.1-15 weight % of the composition.

Optionally, one or more of the elongate carbon nanostructure materials has a functionalised surface, optionally a surface functionalised with a nitrogen-containing group or an oxygen containing group.

Optionally, the composition further comprises graphite.

Optionally, the graphite is provided in the composition in an amount of 1-30 wt %, optionally 1-20 wt %, optionally 1-15 wt %, optionally 1-12 wt%, optionally 1-6 wt %.

Optionally, the composition comprises 3-12 weight % of the carbon additives mixture.

Optionally, the crystallite length  $L_c$  of the graphite is optionally at least 50 nm, optionally at least 100 nm.

Optionally, the carbon black is provided in an amount of at least 0.5 weight % of the composition, and optionally less than 10 wt % of the composition, optionally less than 4 wt % of the composition.

In a second aspect, the invention provides a metal-ion battery comprising an anode, a cathode and an electrolyte between the anode and the cathode wherein the anode comprises a composition according to the first aspect.

In a third aspect, the invention provides a slurry comprising a composition according to the first aspect and at least one solvent.

In a fourth aspect, the invention provides a method of forming a metal-ion battery according to the second aspect, the method comprising the step of forming an anode by depositing a slurry according to the third aspect onto a conductive material and evaporating the at least one solvent.

In a fifth aspect, the invention provides a composition comprising a particulate electroactive material; a first elongate carbon nanostructure material; and a second elongate carbon nanostructure material, wherein the first elongate carbon nanostructure material has a mean average diameter of at least 100 nm; the second elongate carbon nanostructure material has a mean average diameter of less than 90 nm, and wherein the first particulate electroactive material comprises particles having a particle core and pillars of the electroactive material extending from the particle core.

The first particulate electroactive material, first elongate carbon nanostructure and second elongate nanostructure of the composition of the fifth aspect may be selected from materials described herein, including as described in the first aspect, and may be provided in weight percentages and / or ratios as described herein, including as described in the first aspect. The composition of the fifth aspect may comprise one or more further components, for example a binder, carbon black and / or graphite, and the one or more further components may be provided in weight percentages and / or ratios as described herein, including as described in the first aspect.

In a sixth aspect, the invention provides a composition comprising a first particulate electroactive material; a first elongate carbon nanostructure material; and a second elongate carbon nanostructure material, wherein the first elongate carbon nanostructure material has a mean average diameter of at least 100 nm; the second elongate carbon nanostructure material has a mean average diameter in the range of 40-90 nm.

The first particulate electroactive material, first elongate carbon nanostructure and second elongate nanostructure of the composition of the sixth aspect may be selected from materials described with reference to the first aspect, and may be provided in weight percentages and / or ratios as described herein, including as described in the first aspect. The composition of the sixth aspect may comprise one or more further components, for example a binder, carbon black and / or graphite as described herein, including as described in the first aspect, and the one or more further components may be provided in weight percentages and / or ratios as described herein, including as described in the first aspect.

In a seventh aspect, the invention provides a composition comprising a binder; a first particulate electroactive material and carbon additives comprising a first elongate carbon nanostructure material; a second elongate carbon nanostructure material; graphite; and carbon black.

The first particulate electroactive material, first elongate carbon nanostructure, second elongate nanostructure, binder, carbon black and graphite of the composition of the seventh aspect may be selected from materials described anywhere herein, including as described with reference to the first aspect, and may be provided in weight percentages and / or ratios as described herein, including as described in the first aspect.

In an eighth aspect, the invention provides composition comprising a binder; a first particulate electroactive material; carbon nanostructure additives comprising a first elongate carbon nanostructure material, a second elongate carbon nanostructure material, and optionally carbon black wherein the carbon nanostructure additives form no more than 10 weight % of the composition.

The binder, first particulate electroactive material, first elongate carbon nanostructure, second elongate nanostructure and carbon black of the composition of the eighth aspect may be selected from materials described herein, including as described in the first aspect, and may be provided in weight percentages and / or ratios as described herein, including as described in the first aspect, with the proviso that the carbon nanostructure additives form no more than 10 weight % of the composition. The composition of the

eighth aspect may comprise one or more further components, for example graphite as described herein, including as described in the first aspect, and the one or more further components may be provided in weight percentages and / or ratios as described herein, including as described in the first aspect.

In a ninth aspect, the invention provides a composition comprising a binder; a first particulate electroactive material; and at least one elongate carbon nanostructure material wherein the elongate carbon nanostructure material has a functionalised surface.

Optionally according to the ninth aspect, the composition comprises at least two elongate carbon nanostructure materials. One of the elongate carbon nanostructure materials of the ninth aspect may be a nanotube, and another may be a nanofibre.

The binder, first particulate electroactive material and at least one or at least two elongate carbon nanostructure of the composition of the ninth aspect may be selected from materials described herein, including as described in the first aspect, and may be provided in weight percentages and / or ratios as described herein, including as described in the first aspect. The composition of the ninth aspect may comprise one or more further components, for example carbon black and / or graphite, as described herein, including as described in the first aspect, and the one or more further components may be provided in weight percentages and / or ratios as described herein, including as described in the first aspect.

The compositions of the fifth to ninth aspects may be provided in a slurry, and may be used in forming an energy storage device, for example a rechargeable metal ion battery as described herein, including as described in the first aspect.

Weight percentages of components of a composition described herein are the weight percentages of those components in a porous or non-porous solid composition containing all components of the composition. In the case of a slurry containing a composition, it will be understood that the weight of the one or more solvents of the slurry does not form part of the composition weight as described herein.

### **Description of the Drawings**



The invention will now be described in more detail with reference to the drawings, in which:

Figure 1 illustrates schematically a metal ion battery according to an embodiment of the invention;

Figure 2 illustrates schematically a composite electrode according to an embodiment of the invention;

Figure 3A illustrates schematically a process of forming a pillared particle by an etching process;

Figure 3B illustrates schematically a process of forming a pillared particle by growing pillars on a core;

Figure 4A is a scanning electron microscope image of a composition according to an embodiment of the invention;

Figure 4B is a magnification of a region of the image of Figure 4A;

Figure 4C is a magnification of a region of the image of Figure 4B;

Figure 5 illustrates variation of specific discharge capacity as a function of the product of the cycle number and electrode capacity density in  $\text{mAh/cm}^{-2}$  for cells according to embodiments of the invention and comparative devices;

Figure 6A illustrates variation of electrode capacity density with cycle number for cells according to embodiments of the invention;

Figure 6B illustrates variation of end charge voltage with cycle number for the cells of Figure 6A;

Figure 7A illustrates variation of electrode capacity density with cycle number for cells according to embodiments of the invention;

Figure 7B illustrates variation of end charge voltage with cycle number for the cells of Figure 7A; and

Figures 8-10 illustrate variation of electrode capacity density in mAh/cm<sup>-2</sup> with cycle number for cells according to further embodiments of the invention.

### **Detailed Description of the Invention**

The structure of a rechargeable metal ion battery cell is shown in Fig. 1, which is not drawn to any scale. The battery cell includes a single cell but may also include more than one cell. The battery is preferably a lithium ion battery, but may be a battery of another metal ion, for example sodium ion and magnesium ion.

The battery cell comprises a current collector for the anode 10, for example copper, and a current collector for the cathode 12, for example aluminium, which are both externally connectable to a load or to a recharging source as appropriate. A composite anode layer containing active silicon particles 14 overlays the current collector 10 and a lithium containing metal oxide-based composite cathode layer 16 overlays the current collector 12 (for the avoidance of any doubt, the terms “anode” and “cathode” as used herein are used in the sense that the battery is placed across a load – in this sense the negative electrode is referred to as the anode and the positive electrode is referred to as the cathode. “Active material” or “electroactive material” as used herein means a material which is able to insert into its structure, and release therefrom, metal ions such as lithium, sodium, potassium, calcium or magnesium during the respective charging phase and discharging phase of a battery. Preferably the material is able to insert and release lithium. Preferred active materials include materials having silicon surface at a surface thereof, for example silicon particles or a composite of a material having a non-silicon core and a surface that is partly or wholly a silicon surface.)

The cathode 12 comprises a material capable of releasing and reabsorbing lithium ions for example a lithium-based metal oxide or phosphate,  $\text{LiCoO}_2$ ,  $\text{LiNi}_{0.8}\text{Co}_{0.15}\text{Al}_{0.05}\text{O}_2$ ,  $\text{LiMn}_x\text{Ni}_x\text{Co}_{1-2x}\text{O}_2$  or  $\text{LiFePO}_4$ .

A liquid electrolyte may be provided between the anode and the cathode. In the example of Figure 1, a porous plastic spacer or separator 20 is provided between the anode layer 14 and the lithium containing cathode layer 16, and a liquid electrolyte material is dispersed within the porous plastic spacer or separator 20, the composite anode layer 14

and the composite cathode layer 16. The porous plastic spacer or separator 20 may be replaced by a polymer electrolyte material and in such cases the polymer electrolyte material is present within both the composite anode layer 14 and the composite cathode layer 16. The polymer electrolyte material can be a solid polymer electrolyte or a gel-type polymer electrolyte.

When the battery cell is fully charged, lithium has been transported from the lithium containing metal oxide cathode layer 16 via the electrolyte into the anode layer 14.

A composition according to an embodiment of the invention comprises silicon-comprising particles, a binder and one or more additives. Each additive is preferably a conductive material. Each additive may or may not be an active material.

The silicon-comprising particles may be structured particles. One form of structured particles are particles having a core, which may or may not comprise silicon, with silicon-comprising pillars extending from the core. Another form of structured particles is porous silicon, in particular macroporous silicon, as described in more detail below.

Additives may be selected from: a first elongate carbon nanostructure; one or more further elongate carbon nanostructures; carbon black particles including acetylene black and ketjen black particles; and a material containing graphite or graphene particles. Each elongate carbon nanostructure is preferably selected from a nanotube and a nanofibre. A “nanostructure” material as used herein may mean a material comprising particles having at least one dimension less than 1 micron, preferably less than 500nm, more preferably less than 200nm.

With reference to Figure 2, which is not drawn to any scale, a composition according to an embodiment of the invention comprises silicon-comprising particles 201, a first elongate nanostructure 203, a second elongate nanostructure 205, carbon black particles 207, graphite particles 209 and binder 211. The silicon-comprising particles 201 illustrated in Figure 2 are pillared particles having a core with pillars extending from the core, however the silicon-comprising particles may or may not carry pillars.

The second elongate nanostructure material may become entangled with the pillars of the pillared silicon particles, and each nanostructure may wrap around some or all of the perimeter of one or more of the pillared silicon particle cores, and so may extend electronic conductivity beyond the pillared particle surface and / or lower barrier to conduction between the pillared particle surface and other conductive species, including the binder and other additives of the anode. The second elongate nanostructure may also be entangled with other components of the composition, for example graphite (if present).

The pillars, or other structural elements, of the silicon-comprising particles 201 may provide anchors for the nanofibres or nanotubes of the second elongate nanostructure material 205.

The larger diameter of the first elongate nanostructure material 203 may make it more rigid than the second elongate nanostructure material 205. The first elongate nanostructure material 203 may provide conduction paths within the composition that extend along the length of each nanostructure. These conduction paths may form the framework or support for conductive bridges between silicon-comprising particles 201 and between the silicon-comprising particles 201 and other components in the composite such as graphite particles 209.

Compositions of the invention may include only two different elongate nanostructure materials, for example as illustrated in Figure 2, or may include three or more different elongate nanostructure materials.

#### Silicon-comprising particles

The silicon-comprising particles may be structured particles. Structured particles include particles having a core and pillars extending from the core, and particles having pores on the particle surface or pores throughout the particle volume. A surface of a macroporous particle may have a substantially continuous network of the particle material at a surface of the particle with spaces, voids or channels within the material that may have dimensions of at least 50nm. Such voids may be present throughout the particle volume or may be restricted to regions of the particle. A particle may have regions of pillars and regions of pores. The pillars themselves may be microporous or mesoporous.

The silicon-comprising particles in compositions of the invention may consist essentially of n- or p-doped silicon or may contain one or more further materials. For example, in the case of pillared particles the particle may be selected from one of the following:

- a particle having a silicon core with pillars extending from and integral with the silicon core
- a particle having a non-silicon core of a conductive material, for example a graphite core, with pillars extending from the core; and
- a particle having a non-silicon core of a conductive material, for example a graphite core, coated with a silicon shell and having silicon pillars extending from and integral with the silicon shell.

The pillars may be core-shell structures, the inner core being of a different material to the outer shell material and where the core and/or shell contains silicon. In the case where the core and pillars are of different materials, the core may or may not be an electroactive material.

Figure 3A illustrates a first method of forming pillared particles wherein a starting material is etched to form a pillared particle wherein a starting material 301 is exposed to an etching formulation for selective etching at the surface of the starting material to produce a pillared particle 303 having a core 305 and pillars 307.

It will be appreciated that the volume of the particle core of the pillared particle formed by this method is smaller than the volume of the starting material, and the surface of the core is integral with the pillars. The size of the pillared particle may be the same as or less than the size of the starting material.

A suitable process for etching a material having silicon at its surface is metal-assisted chemical etching (alternatively called galvanic exchange etching or galvanic etching) which comprises treatment of the starting material with hydrogen fluoride, a source of metal ions, for example silver or copper, which electrolessly deposit onto the surface of the silicon and an oxidant, for example a source of nitrate ions. More detail on suitable etching processes can be found in, for example, Huang et al., *Adv. Mater.* 23, pp 285-308 (2011).

The etching process may comprise two steps, including a step in which metal is formed on the silicon surface of the starting material and an etching step. The presence of an ion that may be reduced is required for the etching step. Exemplary cations suitable for this purpose include nitrates of silver, iron (III), alkali metals and ammonium. The formation of pillars is thought to be as a result of etching selectively taking place in the areas underlying the electrolessly deposited metal.

The metal deposition and etching steps may take place in a single solution or may take place in two separate solutions.

Metal used in the etching process may be recovered from the reaction mixture for re-use, particularly if it is an expensive metal such as silver.

Exemplary etching processes suitable for forming pillared particles are disclosed in WO 2009/010758 and in WO 2010/040985.

Other etching processes that may be employed include reactive ion etching, and other chemical or electrochemical etching techniques, optionally using lithography to define the pillar array.

If the pillared particle comprises a first material at its core centre with a shell formed from a second material, for example carbon coated with silicon, then this particle may be formed by etching of silicon-coated carbon to a depth of less than the thickness of the silicon shell in order to form a pillared particle with a composite carbon / silicon core.

Etching may be to a depth of less than 2-10 microns, optionally at least 0.5 microns, to form pillars having a height of up to 10 microns. The pillars may have any shape. For example, the pillars may be branched or unbranched; substantially straight or bent; and of a substantially constant thickness or tapering.

The pillars may be formed on or attached to a particle core using methods such as growing, adhering or fusing pillars onto a core or growing pillars out of a core. Figure 3B illustrates a second method of forming pillared particles wherein pillars 307, preferably silicon pillars, for example silicon nanowires, are grown on or attached to a starting material 301 such as a silicon or carbon (e.g. graphite or graphene) starting material. The volume of the particle core 305 of the resultant pillared particle 303 may

be substantially the same as the volume of the starting material 301. In other words, the surface of the starting material may provide the surface of the particle core 305 from which the pillars 307 extend.

Exemplary methods for growing pillars include chemical vapour deposition (CVD) and fluidised bed reactors utilising the vapour-liquid-solid (VLS) method. The VLS method comprises the steps of forming a liquid alloy droplet on the starting material surface where a wire is to be grown followed by introduction in vapour form of the substance to form a pillar, which diffuses into the liquid. Supersaturation and nucleation at the liquid/solid interface leads to axial crystal growth. The catalyst material used to form the liquid alloy droplet may for example include Au, Ni or Sn.

Nanowires may be grown on one or more surfaces of a starting material.

Pillars may also be produced on the surface of the starting material using thermal plasma or laser ablation techniques.

The pillars may also be formed by nanowire growth out of the starting material using methods such as a solid-liquid-solid growth technique. In one example silicon or silicon-based starting material granules are coated with catalyst particles (e.g. Ni) and heated so that a liquid alloy droplet forms on the surface whilst a vapour is introduced containing another element. The vapour induces condensation of a product containing the starting material and the other element from the vapour, producing growth of a nanowire out of the starting material. The process is stopped before all of the starting material is subsumed into nanowires to produce a pillared particle. In this method the core of the pillared particle will be smaller than the starting material.

Silicon pillars grown on or out of starting materials may be grown as undoped silicon or they may be doped by introducing a dopant during the nanowire growth or during a post-growth processing step.

The pillars are spaced apart on the surface of the core. In one arrangement, substantially all pillars may be spaced apart. In another arrangement, some of the pillars may be clustered together.

The starting material for the particle core is preferably in particulate form, for example a powder, and the particles of the starting material may have any shape. For example, the starting material particles may be cuboid, cuboidal, substantially spherical or spheroid or flake-like in shape. The particle surfaces may be smooth, rough or angular and the particles may be multi-faceted or have a single continuously curved surface. The particles may be porous or non-porous.

A cuboid, multifaceted, flake-like, substantially spherical or spheroid starting material may be obtained by grinding a precursor material, for example doped or undoped silicon as described below, and then sieving or classifying the ground precursor material. Exemplary grinding methods include power grinding, jet milling or ball milling. Depending on the size, shape and form of the precursor material, different milling processes can produce particles of different size, shape and surface smoothness. Flake-like particles may also be made by breaking up / grinding flat sheets of the precursor material. The starting materials may alternatively be made by various deposition, thermal plasma or laser ablation techniques by depositing a film or particulate layer onto a substrate and by removing the film or particulate layer from the substrate and grinding it into smaller particles as necessary.

The starting material may comprise particles of substantially the same size. Alternatively, the starting material may have a distribution of particle sizes. In either case, sieves and/or classifiers may be used to remove some or all starting materials having maximum or minimum sizes outside desired size limits.

In the case where pillared particles are formed by etching a material comprising silicon, the starting material may be undoped silicon or doped silicon of either the p- or n-type or a mixture, such as silicon doped with germanium, phosphorous, aluminium, silver, boron and/or zinc. It is preferred that the silicon has some doping since it improves the conductivity of the silicon during the etching process as compared to undoped silicon. The starting material is optionally p-doped silicon having  $10^{19}$  to  $10^{20}$  carriers/cc.

Silicon granules used to form the pillared particles may have a silicon-purity of 90.00% or over by mass, for example 95.0% to 99.99%, optionally 98% to 99.98%.



The starting material may be relatively high purity silicon wafers used in the semiconductor industry formed into granules. Alternatively, the granules may be relatively low purity metallurgical grade silicon, which is available commercially and which may have a silicon purity of at least 98%; metallurgical grade silicon is particularly suitable because of the relatively low cost and the relatively high density of defects (compared to silicon wafers used in the semiconductor industry). This leads to a low resistance and hence high conductivity, which is advantageous when the pillar particles or fibres are used as anode material in rechargeable cells. Impurities present in metallurgical grade silicon may include Iron, Aluminium, Nickel, Boron, Calcium, Copper, Titanium, and Vanadium, oxygen, carbon, manganese and phosphorus. Certain impurities such as Al, C, Cu, P and B can further improve the conductivity of the starting material by providing doping elements. Such silicon may be ground and graded as discussed above. An example of such silicon is “Silgrain™” from Elkem of Norway, which can be ground and sieved (if necessary) to produce silicon granules, that may be cuboidal and / or spheroidal.

The granules used for etching may be crystalline, for example mono- or poly-crystalline with a crystallite size equal to or greater than the required pillar height. The polycrystalline granules may comprise any number of crystals, for example two or more.

Where the pillared particles are made by a growth of silicon pillars as described above, the starting material may comprise an electroactive material, and may comprise metal or carbon based particles. Carbon based starting materials may comprise soft carbon, hard carbon, natural and synthetic graphite, graphite oxide, fluorinated graphite, fluorine-intercalated graphite, graphene.

Graphene based starting materials may comprise particles comprising a plurality of stacked graphene nanosheets (GNS) and/or oxidised graphene nanosheets (ox-GNS), sometimes called Graphite Nano Platelets (GNP) or alternatively nano Graphene Platelets (NGP). NGP (or GNP) may have thicknesses of at least a few nanometres (e.g. at least 2nm) and larger dimensions of up to 100µm, preferably less than 40µm. Materials comprising a plurality of stacked graphene sheets are graphite materials. Methods of making graphene based particles include exfoliation techniques (physical, chemical or

mechanical), unzipping of MWCNT or CNT, epitaxial growth by CVD and the reduction of sugars.

The core of the silicon-comprising particle illustrated in Figure 3 is substantially spherical, however the particle core may have any shape, including substantially spherical, spheroidal (oblate and prolate), and irregular or regular multifaceted shapes (including substantially cube and cuboidal shapes). The particle core surfaces from which the pillars extend may be smooth, rough or angular and may be multi-faceted or have a single continuously curved surface. The particle core may be porous or non-porous. A cuboidal core may be in the form of a flake, having a thickness that is substantially smaller than its length or width such that the core has only two major surfaces.

The aspect ratio of a pillared particle core having dimensions of length L, width W and thickness T is a ratio of the length L to thickness T ( $L : T$ ) or width W to thickness T ( $W : T$ ) of the core, wherein the thickness T is taken to be the smallest of the 3 dimensions of the particle core. The aspect ratio is 1:1 in the case of a perfectly spherical core. Prolate or oblate spheroid, cuboidal or irregular shaped cores preferably have an aspect ratio of at least 1.2:1, more preferably at least 1.5:1 and most preferably at least 2:1. Flake like cores may have an aspect ratio of at least 3:1.

In the case of a substantially spherical core, pillars may be provided on one or both hemispheres of the core. In the case of a multifaceted core, pillars may be provided on one or more (including all) surfaces of the core. For example, in the case of a flake core the pillars may be provided on only one of the major surfaces of the flake or on both major surfaces.

The core material may be selected to be a relatively high conductivity material, for example a material with higher conductivity than the pillars, and at least one surface of the core material may remain uncovered with pillars. The at least one exposed surface of the conductive core material may provide higher conductivity of a composite anode layer comprising the pillared particles as compared to a particle in which all surfaces are covered with pillars.

The silicon particles may have at least one smallest dimension less than one micron. Preferably the smallest dimension is less than 500nm, more preferably less than 300nm. The smallest dimension may be more than 0.5 nm. The smallest dimension of a particle is defined as the size of the smallest dimension of an element of the particle such as the diameter for a rod, fibre or wire, the smallest diameter of a cuboid or spheroid or the smallest average thickness for a ribbon, flake or sheet where the particle may consist of the rod, fibre, wire, cuboid, spheroid, ribbon, flake or sheet itself or may comprise the rod, fibre, wire, cuboid, spheroid, ribbon, flake or sheet as a structural element of the particle.

Preferably the particles have a largest dimension that is no more than 100 $\mu$ m, more preferably, no more than 50 $\mu$ m and especially no more than 30 $\mu$ m.

Particle sizes may be measured using optical methods, for example scanning electron microscopy.

Preferably at least 20%, more preferably at least 50% of the silicon particles have smallest dimensions in the ranges defined herein. Particle size distribution may be measured using laser diffraction methods, for example using a MasterSizer<sup>RTM</sup> as described in more detail below, or optical digital imaging methods.

#### Elongate carbon nanostructure materials

A composition of the invention includes at least two elongate carbon nanostructure materials. A first elongate carbon nanostructure material may have a diameter (or smallest dimension) that is larger than that of the second elongate carbon nanostructure. The second nanostructure material may have a higher surface area per unit mass than the first nanostructure material. The first elongate nanostructure material may have a large enough diameter so that the nanostructure is relatively straight and rigid whereas the second elongate nanostructure may have a small enough diameter such that it can be flexible and curved or bent within the composite. Preferably the diameter (or smallest dimension) of the first elongate carbon nanostructure is at least 100nm. Preferably the diameter (or smallest dimension) of the second elongate carbon nanostructure is less than 100nm, more preferably less than 90 nm, more preferably less than 80nm. Preferably,

both the average thickness and average width of each of the first and second elongate carbon nanostructures is less than 500 nm.

Each of the elongate carbon nanostructure materials may have a large aspect ratio, the aspect ratio being the ratio of the largest and smallest dimensions of the material.

Preferably, the aspect ratio of the first elongate carbon nanostructure is in the range of about 40 to 180. Preferably the aspect ratio of the second carbon nanostructure is in the range of 200 to 500.

Elongate nanostructures may be selected from nanofibres and / or nanotubes and thin ribbons.

Nanotubes may be single-walled or multi-walled. Preferably, carbon nanotubes used in compositions of the invention are multi-walled. Walls of the nanotubes may be of graphene sheets.

Nanofibres may be solid carbon fibres or may have a narrow hollow core, and may be formed from stacked graphene sheets. An example of a suitable nanofibre material is VGCF<sup>RTM</sup> supplied by Showa Denko KK.

Optionally, the elongate nanostructures have a mean average length in the range of 3-50 $\mu$ m. Preferably the length of the first elongate nanostructure material is in the range 5-30 $\mu$ m.

Preferably the surface area of each elongate nanostructure material is no more than 100m<sup>2</sup>/g and at least 1m<sup>2</sup>/g.

The first elongate nanostructure may be a nanofibre having a surface area in the range of 10-20 m<sup>2</sup>/g

The second elongate nanostructure may be a nanotube have a surface area in the range of 40-80 m<sup>2</sup>/g.

The carbon nanostructures may be functionalised to improve adhesion or connection to other components in the composition, especially the silicon-comprising particles. For example carbon nanotubes can be functionalised with oxygen-containing groups, for example COOH, OH, CO and nitrogen containing groups, for example NH<sub>2</sub>. The second elongate nanostructure may be a carbon nanotube functionalised with COOH groups

which may promote connectivity to the surface of silicon-comprising particles or other electroactive particles.

A composition including a binder, silicon-comprising particles, two or more different elongate carbon nanostructure materials and any further additives may include each of the elongate nanostructure materials in an amount in the range of 0.25 – 20 weight %, optionally 0.25-10 wt % of the composition. The total amount of the two or more different elongate nanostructure materials in the composition may be in the range of 2-25 weight percent, optionally 3 – 13 weight percent.

### Carbon black

The composition may comprise carbon black, which may be characterised as a highly conducting particulate carbon, quasigraphitic in nature, composed of aggregates having a complex configuration (including but not limited to chain-like agglomerates) and of colloidal dimensions. Carbon black is typically made via the thermal decomposition and partial combustion of hydrocarbons. Various types of carbon black are available, including acetylene blacks. Examples of commercial products include Ketjen Black<sup>RTM</sup> EC600JD or EC300J supplied by AkzoNobel, Vulcan<sup>RTM</sup> XC72R manufactured by Cabot Corp, TokaBlack<sup>RTM</sup> 5500, 4500, 4400 or 4300 manufactured by Tokai Carbon Co., LTD. and DenkaBlack<sup>RTM</sup> FX-35 or HS-100 manufactured by Denki Kagaku Kogyo Kabushiki Kaisha. The composition may comprise a single type of carbon black or a blend of one or more types of carbon black. The carbon black particles may have dimensions in the range of 10-100nm and a surface area in excess of 50m<sup>2</sup>/g.

A composition including a binder, silicon-comprising particles, a first elongate carbon nanostructure and a second elongate carbon nanostructure, carbon black additive(s) and any further additives may include carbon black (of a single type or a blend of a plurality of types) in an amount of at least 0.25 weight % of the composition, and optionally less than 10 wt % of the composition. Preferably, the carbon black is present in an amount in the range 0.5 wt% to 4 wt% of the total composition. Ketjen Black EC600JD with an average particle size of 20-40nm and a surface area of >1000 m<sup>2</sup>/g is particularly preferred as an additive.

## Graphite

The composition may contain graphite particles, for example graphite flakes,, in addition to the elongated carbon nanostructures and carbon black particles described above .

Optionally the graphite is synthetic graphite.

The crystallite length  $L_c$  of the graphite particles is optionally at least 50 nm, optionally at least 100 nm. Graphite with a higher crystallite length  $L_c$  may be preferable as this may provide higher conductivity, and higher overall conductivity of the composite.

Suitable commercial products of graphite particles may include Timrex<sup>RTM</sup> SFG6, SFG10, SFG15, KS4 or KS6 manufactured by Timcal Ltd.

Graphite present in an anode of a metal ion battery may function as an active material. Active graphite may provide for a larger number of charge / discharge cycles without significant loss of capacity than active silicon, whereas silicon may provide for a higher capacity than graphite. Accordingly, an electrode composition having both silicon-containing active particles and a graphite active material may provide a metal ion battery, for example lithium ion battery, with the advantages of both high capacity and a large number of charge / discharge cycles. Depending on the type of graphite material and the charge/discharge conditions, the graphite additive in a silicon based composition may not be fully lithiated during charging and may have a negligible or zero contribution to the electrode capacity above that of the silicon based material. It may be used primarily to improve the overall conductivity of the composition.

Graphite present in the composition may also improve coating properties of a slurry of the composition as compared to a composition in which graphite is absent.

Graphite particles may be provided as a powder having a  $D_{50}$  size as measured using laser diffractometry of less than 50 microns, optionally less than 25 microns.

$D_n$  as used herein (for example,  $D_{50}$  or  $D_{90}$ ) means that at least  $n\%$  of the volume of the material is formed from particles have a measured spherical equivalent volume diameter equal to or less than the identified diameter.

Flake-like graphite particles may have a length, height and thickness wherein both length and width of the particles are each independently on average at least 5 times, optionally at least 10 times, the thickness of the particles. Average thickness of graphite flakes may be in the range of about 75-300 nm. Average dimensions may be measured from an SEM image of a sample of the particles.

A composition including a binder, silicon-comprising particles, graphite and any further additives may include graphite in an amount in the range of 2-30 wt %, optionally 2-15 wt %.

The present inventors have surprisingly found that the performance of a metal-ion battery having a composite anode containing both silicon-comprising particles and graphite particles may be affected by the size ratio of the silicon-comprising particles to the graphite particles.

Graphite as described herein may be graphene-based particles comprising a plurality of stacked graphene sheets. Graphene-based particles may comprise a plurality of stacked graphene nanosheets (GNS) and/or oxidised graphene nanosheets (ox-GNS), sometimes called Graphite Nano Platelets (GNP) or alternatively nano Graphene Platelets (NGP). NGP (or GNP) may have thicknesses of at least a few nanometres (e.g. at least 2nm) and larger dimensions of up to 100µm, preferably less than 40µm. Methods of making graphene-based particles include exfoliation techniques (physical, chemical or mechanical), unzipping of MWCNT or CNT, epitaxial growth by CVD and the reduction of sugars.

### Binder

The binder may be provided to provide cohesion of the particles and, in the case of use in a metal ion battery, for adhesion of the composition to an anode current collector.

The binder material may be a polymeric material, for example polyimide, polyacrylic acid (PAA) and alkali metal salts thereof, polyvinylalcohol (PVA), polyvinylidene fluoride (PVDF) and sodium carboxymethylcellulose (Na-CMC) or rubber based binders such as SBR. Mixtures of different binder materials may be used.

The binder may be provided in an amount in the range of 5-30 wt % of the composition.

### Composition

The silicon particles and the carbon additives and any other additives may each be provided in the form of a powder or slurry for ease of mixing and blending. For example a slurry can be made by mixing the silicon particles or carbon additives with an appropriate amount of aqueous (e.g. water) and / or non-aqueous (e.g. NMP) solvent. A slurry of a composition comprising the silicon particles, carbon additives and any other additives may be made by mixing all elements together with a solvent or alternatively may be made by first making more than one slurry, each slurry comprising one or more the individual elements of the composition in a solvent and then combining the separate slurries together to create a slurry containing all elements of the composition. The solvents of the separate starting slurries may be the same or may be different, as long as they are miscible when combined. A binder material with or without a solvent may also be added and blended to the composition or slurry. The resulting slurry may be deposited onto a substrate and dried to remove the solvent to form a composition for the electrode of a metal-ion battery.

The inventors have recognised that if a metal ion battery comprising a negative electrode comprising a silicon-comprising electroactive material is to cycle with a high capacity (for example, in excess of 500mAh per gram of active material) for in excess of 100-300 charge/discharge cycles, then the electrode composite structure should be uniformly porous and electronically well connected and designed to accommodate the volume changes of the electroactive material during cycling without mechanical or electronic disconnection of the active material from the composite structure.

In order to achieve this, the components within the composite may have moderate values of surface area per unit mass. A high surface area may provide higher reactivity of the active material or improved conductivity from the additives, however if the surface area of the components is too high, excessive formation of a solid-electrolyte interphase (SEI) layer may increase metal ion loss, cause reduction cycle life and cause reduction in porosity. In addition, an excessive surface area of the additives will require a higher



content of binder in the composition to effectively bind the components of the composite together and to adhere it to the current collector – which may reduce the overall volumetric capacity and make it difficult to provide an appropriate level of porosity in the composition.

When the composition is mixed with a solvent to form a slurry for depositing the composition onto a current collector, the mix of components with different shapes and varying volumes is preferably such that slurry comprise a uniform mixture with all components equally dispersed and of sufficiently low viscosity to enable thin, uniform coatings to be prepared.

The inventors have discovered that a negative electrode with a composition having the following properties may provide improved cycling performance as described above:

- (a) At least 50 wt% active material and no more than 80 wt%, the active material preferably comprising structured silicon particles
- (b) Binder in the range of 5-30wt%, preferably 10-20wt%.
- (c) First elongate carbon nanostructure material comprising nanostructures with a smallest dimension of more than 100nm in the amount of 0.25 to 20wt%, preferably 3-7wt%
- (d) Second elongate carbon nanostructure material comprising nanostructures with a smallest dimension of less than 100nm, preferably in the range 30-80nm, in the amount of 0.25 to 20wt%, more preferably 2-8wt%.
- (e) Carbon black in the range 0.25 to 10wt%, preferably 0.5 to 4 wt%.
- (f) Graphite particles and/or other additives, fillers and spacers in the range 2-30wt%
- (g) A porosity of at least 10-80%, preferably 20-60%.

wherein the total percentage of the above components adds up to 100 wt%. Preferably the total amount of the first and second elongate carbon nanostructures (c and d) in the

composition is in the range 2-25 wt%, especially 3-13wt%. Preferably the ratio of the mass of the first elongate carbon nanostructure material to the mass of the second elongate carbon nanostructure material is no more than 5:1, most preferably the ratio is in the range 0.1:1 to 5:1 and especially 0.5:1 to 2:1.

Preferably the composition comprises structured silicon particles as described above. The inventors have discovered that all three carbon components c, d and e, within the weight amounts described above may produce a negative electrode with excellent cyclability. Without wishing to be bound by theory, it is believed that by using elongate carbon nanostructures such as MWCNT with diameters in the range 30-80nm and in the amounts described above, the MWCNT can become entangled with the structural features of the silicon structured particles to form short range conductive networks without excessive filling of the voids or spaces between the said structural features that are necessary to provide space for silicon expansion and access of electrolyte. The larger diameter, rigid first elongate carbon nanostructures, such as VGCF, provide conductive bridges for longer range electronic connections and help to provide a strong mechanical framework within the composition to withstand the volume expansion and contraction of the active material during cycling. It is believed that the highly dispersed carbon black may provide sufficient conductivity in the remaining locations within the composition. However if an excessive amount of any of the carbon additives is used then the effectiveness of the binder may be reduced and the uniformity of the composition may be reduced.

The composition may be formed by mixing the components of the composition, a solvent and optionally one or more of a surfactant, dispersant or porogen, and stirring the mixture. Two or more of the components may first be mixed together in a solvent before being added to the other components for a final mixing stage. The composition may then be deposited on a substrate and dried so that the solvent is evaporated to form a porous composite film.

## **Examples**

### Materials

Compositions were prepared with components selected from the following materials:

Pillared silicon particles formed by etching starting silicon particles available as “Silgrain™” from Elkem of Norway, wherein the starting silicon particles have a  $D_{50}$  particle size of 11.5-12.5 microns, or 24.5-25.5 microns as measured using a Mastersizer™ particle size analyzer available from Malvern Instruments Ltd. It will be understood that the resultant pillared particle may have a  $D_{50}$  that is smaller than that of the starting material, for example up to 2 or 4 microns smaller respectively.

VGCF carbon nanofibres available from Showa Denko, having an average diameter of 150 nm, an average length of 10-20 microns and a surface area of 13 m<sup>2</sup>/g.

Multiwalled carbon nanotubes from CheapTubes Inc having an average diameter of 50-80 nm, an average length of 15-20 microns and a surface area of 55-75 m<sup>2</sup>/g (hereinafter “MWCNT”).

Carbon black material available from AzkoNobel as Ketjenblack® EC600-JD having a surface area of 1400 m<sup>2</sup>/g and an average particle size of 20-40nm.

Carbon black material available from Denka as Denka black having a surface area of 69 m<sup>2</sup>/g and an average particle size of 35 nm.

Graphite available as TIMCAL TIMREX ® KS4, KS6, SFG6 and SFG10 having  $D_{10}$ ,  $D_{50}$  and  $D_{90}$  values (measured using a MasterSizer particle size analyser) and BET values as given in Table 2..

A sodium polyacrylate binder, hereinafter referred to as “NaPAA” was formed by partially neutralising commercially available polyacrylic PAA450K using sodium carbonate or sodium hydroxide to a 70% degree of neutralisation. A distribution of the particle sizes of a powder of starting material particles used to form pillared particles may be measured by laser diffraction, in which the particles being measured are typically assumed to be spherical, and in which particle size is expressed as a spherical equivalent volume diameter, for example using the Mastersizer™ particle size analyzer available from Malvern Instruments Ltd. A spherical equivalent volume diameter is the diameter of a sphere with the same volume as that of the particle being measured. If all particles in

the powder being measured have the same density then the spherical equivalent volume diameter is equal to the spherical equivalent mass diameter which is the diameter of a sphere that has the same mass as the mass of the particle being measured. For measurement the powder is typically dispersed in a medium with a refractive index that is different to the refractive index of the powder material. A suitable dispersant for powders of the present invention is water. For a powder with different size dimensions such a particle size analyser provides a spherical equivalent volume diameter distribution curve.

Figure 4A is a SEM image of a composition containing each of the aforementioned components following formation of a slurry of the composition and deposition of the composition onto a copper current collector and evaporation of the slurry solvent to form an anode layer.

The second elongate nanostructures 205, which in this case are multiwalled carbon nanotubes, are entangled with the silicon-comprising particles 201, which in this case are pillared silicon particles. The first elongate nanostructures 203, in this case a nanofibre, provides conductivity over a relatively long range, as shown for the annotated nanofibre 203 bridging two silicon particles.

The nanotubes provide medium range conductivity. Referring to Figures 4B and 4C, it can be seen that nanotubes 205 form a bridge extending across two silicon particles 201. The nanotubes and nanoparticles also provide for improved conductivity between the silicon particles and graphite flakes 209 of the composition.

#### General Device Process 1

Swagelok<sup>TM</sup>-style test cells were constructed using an anode comprising a composition comprising silicon pillared particles as the active material deposited with a coat weight of 13.5-15.5 grams of silicon per m<sup>2</sup> onto a 10µm thick copper foil, an NCA cathode (Li<sub>1+x</sub>Ni<sub>0.8</sub>Co<sub>0.15</sub>Al<sub>0.05</sub>O<sub>2</sub>) on an aluminium foil and a Tonen separator between the two electrodes. The electrodes and separator were wetted with an electrolyte solution of 1M LiPF<sub>6</sub> in EC/EMC containing VC (vinylene carbonate, 3wt%, FEC (fluoroethylene carbonate, 10wt%) and CO<sub>2</sub> (0.2wt%) as additives. The capacity of the NCA cathode was

3 times higher than the capacity of ppSi in the composite electrode that was designed to operate at 1200 mAh/g. The silicon pillared particles were prepared using metal-assisted etching of metallurgical grade silicon particles (with a silicon purity of 99.7-99.95wt%), to form irregular shaped pillars of lengths 1.5-2.5 $\mu$ m and thicknesses of 40-150nm such that the average mass of the pillars was 20-40% of the total silicon mass. The cells were cycled in such a way that the ppSi was charged to 1200mAh/g and discharged to a cut-off voltage of 2.5V. The cycling rate was C/2 for both charge and discharge. The electrode area was 1.13cm<sup>2</sup>.

#### Device Examples 1-2

Compositions of the following materials in the following weight ratios were prepared:

70 wt % pillared silicon particles (pillared particle D<sub>50</sub> = 11.1 microns)

14 wt % binder NaPAA

4 wt % graphite SFG6

12 wt % made up of elongate nanostructures, VGCF and EC600, as shown in Table 1.

Example (cell number)	Nanotube MWCNT (wt %)	VGCF (wt %)	Carbon black EC600 (wt %)	First, second and third cycle efficiencies (%)
1 (NG415)	5	5	2	36, 70, 100
2 (NG598)	8	1	3	73, 100, 79
Comparative Example 1 (NG469)	0	11	1	79, 76, 100
Comparative Example 2	11	0	1	73, 99, 80

(NG629)				
Comparative Example 3 (NG418)	7	1	4	72, 72, 100

Table 1

These compositions were used to prepare lithium-ion cells according to the General Device Process. The devices had first, second and third cycle efficiencies as shown in Table 2. The nth cycle efficiency is the ratio of the discharge capacity to the preceding charge capacity and provides an indication of the amount of lithium lost or retained within the anode or other cell components during the nth charge-discharge cycle, for example due to formation of the SEI layer.

With reference to Figure 5, normalised capacity starts to decrease at a lower cycle number for Comparative Example 1, which does not contain VGCF, and Comparative Example 2, which does not contain MWCNT, than Examples 1 and 2 that contain both MWCNT and VGCF.

Comparative Example 3 shows decrease in capacity before Example 1 or 2. Without wishing to be bound by any theory, it is believed that the high level of carbon black in this Example, may result in a high level of absorption of the binder due to the high surface area of the carbon black. Preferably, the weight ratio given by the combined mass of the elongate carbon nanostructures to the mass of the carbon black particles is in the range 3:1 to 20:1.

#### Device Examples 3-6

Compositions of the following materials in the following weight ratios were prepared:

70 wt % pillared silicon particles (pillared particle  $D_{10}= 11\mu\text{m}$ ,  $D_{50} = 21$  microns,  $D_{90}=39\mu\text{m}$ )

12 wt % binder NaPAA

6 wt % graphite

12 wt % made up of VGCF: multi-wall carbon nanotubes: EC600: Ketjenblack® EC600-JD: Denka black in the ratio of 4:1:1:2. Graphite was varied as shown in Table 2.

Example	Graphite Type	Graphite D <sub>10</sub> (microns)	Graphite D <sub>50</sub> (microns)	Graphite D <sub>90</sub> (microns)	Graphite BET surface area (m <sup>2</sup> /g)	First, second and third cycle efficiencies (%)
3	KS4	1.2	2.4	4.7	26	57, 85, 100
4	KS6	1.6	3.4	6.5	20	63, 87, 100
5	SFG6	1.7	3.5	6.5	17	63, 87, 100
6	SFG10	2.8	6.6	12.8	12.5	64, 86, 100

Table 2

The similarities in efficiencies for different sizes of graphite indicate that graphite size has little or no effect on first and subsequent cycle efficiencies.

These compositions were used to prepare lithium-ion cells according to the General Device Process. The devices had first, second and third cycle efficiencies as shown in Table 2.

Figure 6A depicts the evolution of the capacity density of Examples 3-6, and Figure 6B shows the evolution of the end of charge voltage for these cells with cycle number. The end of charge voltage was limited to 4.3V.

Figure 6B shows that cell resistance increases fastest for the anode of Example 5, containing SFG6. In particular, the cell resistance of Example 6 increases faster than for Example 3, containing KS4.

Example 6 delivers the highest capacity density over 350 cycles.

Device Examples 7-9

Devices were prepared as described with reference to Device Examples 3-6 except that the pillared silicon particles had a  $D_{50}$  size of 11  $\mu\text{m}$ , a  $D_{90}$  size of 20 $\mu\text{m}$  and a  $D_{10}$  size of 6 $\mu\text{m}$  and the graphite was varied as shown in Table 3.

Example	Graphite	Graphite $D_{50}$ (microns)	Graphite $D_{90}$ (microns)	Composition coat weight (g-Si/m <sup>2</sup> )	Graphite BET surface area (m <sup>2</sup> /g)	First, second and third cycle efficiencies (%)
7	KS4	2.4	4.7	13.8	26	69, 80, 100
8	KS4	2.4	4.7	13.6	26	81, 81, 100
9	SFG6	3.5	6.5	14	17	77, 81, 100

Table 3

The measured cycling efficiencies in Table 3 indicate that the performance improvement in Example 9 is not simply down to the lower surface area of SFG6 leading to less SEI layer being formed in the first few cycles.

In contrast to the relative performance described above with reference to Examples 3 and 5, Figure 7A shows that Example 9 containing SFG6 with pillared silicon particles having a  $D_{50}$  of 11 microns maintains its capacity for a larger number of cycles than Examples 7 or 8 containing KS4, and Figure 7B shows that cell resistance increases fastest for the anode of Examples 7 and 8, indicating a relationship between silicon particle size and graphite size for optimum performance. Preferably, the silicon : graphite  $D_{50}$  ratio is at least 2:1, optionally it is no more than 4.5:1, optionally no more than 4 : 1. Most preferably it is in the range 3:1 to 4:1.

General Device Process 2



Swagelok<sup>TM</sup>-style test cells were constructed using an anode comprising a composition comprising 70 wt % silicon pillared particles as the active material deposited with a coat weight of 30 g/m<sup>2</sup> with 5% variance of silicon onto a 10µm thick copper foil, an NCA cathode (Li<sub>1+x</sub>Ni<sub>0.8</sub>Co<sub>0.15</sub>Al<sub>0.05</sub>O<sub>2</sub>) on an aluminium foil and a Tonen separator between the two electrodes. The electrodes and separator were wetted with an electrolyte solution of 1M of LiPF<sub>6</sub> and LiBOB in a weight ratio of 95:5, in a solvent comprising FEC (fluoroethylene carbonate) and EMC with a volume ratio of 1:1 and 3wt% of VC (vinylene carbonate) as an additive. The silicon pillared particles were prepared using metal-assisted etching of metallurgical grade silicon particles (with a silicon purity of 99.7-99.95wt%). The pillared silicon particles had a D<sub>50</sub> of 11.5µm and BET=19.9m<sup>2</sup>/g. The anode composition also comprised 14 wt% binder, wherein the binder was 70% neutralised Na-PAA with a molecular weight of 450k and various amounts of graphite additive SFG6, as shown in Table 4.

Unless otherwise specified, the ratio of the carbon additives in the conductive carbon additives mixture (Cmix) is CB:MWCNT:VGCF = 2:5:5, wherein CB is a carbon black material EC600. The amount of Cmix was varied as shown in Table 4.

The formulations of the anode composition of Device Examples 11, 12 and Comparative Examples 4-8 are shown in Table 4. These compositions were used to prepare lithium-ion cells according to the General Device Process 2.

Example (cell number)	Anode Formulation Si:Binder:Graphite:Cmix	Anode Conductivity (S/cm)	Anode Peel Strength (mN/cm)	First Cycle Loss (%)
Comparative Example 4	70:14:14:2	0.15	112	14.6
Example 11	70:14:4:12	1.31	196	17.7
Example 12	70:14:1:15	3.05	138	18.7
Comparative Example	70:14:0:16	2.00	216	18.7

5				
Comparative Example 6	70:14:12:4 where Cmix is VGCF:MWCNT in 1:1 ratio, no carbon black	0.20	60	15.7
Comparative Example 7	70:14:16:0	0.04	62	17.3
Comparative Example 8	70:14:4:12 where Cmix is VGCF:MWCNT in 1:1 ratio, no carbon black	0.85	170	15.7

Table 4

The cells were charged to 1000mAh/g at constant current and constant voltage. The first cycle was at C/25 between 4.2 to 3V. The subsequent cycles were at C/3 between 4.1 and 3 V.

The losses from the first cycle were measured using different sections of the anode coatings in half test cells vs lithium metal foil, in order to exclude losses due to the cathode. The devices prepared according to General Device Process 2 had anode conductivity, anode peel strength and a first cycle loss as shown in Table 4.

#### 1. Effect of carbon black.

With reference to Figure 8, Example 11 comprising 12 wt% Cmix which contains carbon black, shows an improved conductivity and cycle life compared to Comparative Example 8, which has the same anode formulation but where the presence of the carbon black is substituted with an equal amount of MWCNT and VGCF.

Comparative Example 6, which also has zero carbon black and differs from Comparative Example 8 in that the respective amounts of graphite and elongate carbon material are swapped round, also shows an inferior performance compared to Example 11 where carbon black is present.

## 2. Weight range of conductive carbon additives

Increasing the amount of conductive additives may increase the first cycle lithium loss (because of the high surface areas of these additives) but may also improve the peel strength, i.e. adhesion of the composite anode to the current collector.

Decreasing the amount of conductive carbon may decrease the electrical conductivity of the composite, leading to higher running losses as active particles become isolated during cycling.

With reference to Figure 9, Comparative Example 4 comprising only 2 wt% Cmix, shows an inferior performance compared to Examples 11 and 12 comprising Cmix of 12 wt% and 15 wt% respectively.

Figure 10 shows an inferior performance with decreased capacity of Comparative Example 5 comprising 16 wt% Cmix compared to Example 11 comprising Cmix of 12 wt%.

The data have been normalised to the value of the discharge capacity on the second cycle.

SEM images of comparative examples in which one or more of the elongate carbon nanostructures and carbon black are absent were found to be more porous and more loosely connected than devices according to embodiments of the invention.

## 3. Effect of conductive carbon

Figure 10 shows that when the amount of Cmix (Comparative Example 7) is set to zero, the cycling performance is worse compared to that of the cell Example 11 comprising 4 wt% graphite and 12 wt% Cmix.

The amount of graphite is preferably in the range of 1-12 wt% (when Si is 70 wt%). Larger amounts of graphite may reduce peel strength.

The invention has been described with reference to electroactive silicon as the first particulate electroactive material, however it will be understood that the invention may be

applied to other electroactive materials that have a bulk volume expansion of more than 10% when fully lithiated or is capable of having a specific capacity of greater than 300mAh/g, or may be a metal or semi-metal that can reversibly form an alloy with lithium. Other exemplary electroactive materials are tin; aluminium; electroactive compounds including oxides, nitrides, fluorides, carbides and hydrides, for example compounds of tin, aluminium and silicon; and alloys thereof.

The invention has been described with reference to rechargeable lithium ion batteries, however it will be understood that the invention may be applied to metal ion batteries of metal ions other than lithium, and moreover that the invention may be applied to other energy storage devices, for example fuel cells.

Although the present invention has been described in terms of specific exemplary embodiments, it will be appreciated that various modifications, alterations and/or combinations of features disclosed herein will be apparent to those skilled in the art without departing from the scope of the invention as set forth in the following claims.

## Claims

1. A composition comprising at least 50 weight % of a first particulate electroactive material and 3-15 weight % of a carbon additives mixture comprising elongate carbon nanostructures and carbon black, wherein:  
  
the elongate carbon nanostructures comprise at least a first elongate carbon nanostructure material and a different second elongate carbon nanostructure material; and  
  
the elongate carbon nanostructures : carbon black weight ratio is in the range 3 : 1 to 20 : 1.
2. A composition according to claim 1 wherein the first particulate electroactive material is a silicon-comprising material.
3. A composition according to any preceding claim wherein the first particulate electroactive material comprises particles having a particle core and electroactive pillars extending from the particle core.
4. A composition according to claims 2 and 3 wherein the pillars of the silicon-comprising particles are silicon pillars.
5. A composition according to claim 4 wherein the core of the silicon-comprising particles comprises silicon.
6. A composition according to any of claims 2-5 wherein the silicon-comprising particles consist essentially of n- or p-doped silicon and wherein the pillars are integral with the core.
7. A composition according to any preceding claim wherein the first elongate nanostructure has a mean average diameter of at least 100 nm.
8. A composition according to any preceding claim wherein the second elongate carbon nanostructure material has a mean average diameter of no more than 90 nm, optionally a mean average diameter in the range of 40-90 nm.

9. A composition according to any preceding claim wherein the first elongate nanostructure : second elongate nanostructure weight ratio is in the range 2.5 : 1 to 20 : 1.
10. A composition according to any preceding claim wherein the at least first and second elongate carbon nanostructures each independently has an aspect ratio of at least 50.
11. A composition according to any preceding claim wherein the first and second carbon elongate nanostructure materials are each independently selected from carbon nanotubes and carbon nanofibres.
12. A composition according to claim 11 wherein the first carbon elongate nanostructure material is a nanofibre and the second elongate carbon nanostructure material is a nanotube.
13. A composition according to any preceding claim wherein the at least first and second elongate carbon nanostructure materials are provided in a total amount in the range of 0.1-15 weight % of the composition.
14. A composition according to any preceding claim wherein one or more of the elongate carbon nanostructure materials has a functionalised surface, optionally a surface functionalised with a nitrogen-containing group or an oxygen containing group.
15. A composition according to any preceding claim wherein the composition further comprises graphite.
16. A composition according to claim 15 wherein the graphite is provided in the composition in an amount of 1-30 wt %, optionally 1-20 wt %, optionally 1-15 wt %, optionally 1-12 wt%, optionally 1-6 wt %,.
17. A composition according to any preceding claim wherein composition comprises 3-12 weight % of the carbon additives mixture.

18. A composition according to claim 16 or 17 wherein the crystallite length  $L_c$  of the graphite is optionally at least 50 nm, optionally at least 100 nm.
19. A composition according to any preceding claim wherein the carbon black is provided in an amount of at least 0.5 weight % of the composition, and optionally less than 10 wt % of the composition, optionally less than 4 wt % of the composition.
20. A metal-ion battery comprising an anode, a cathode and an electrolyte between the anode and the cathode wherein the anode comprises a composition according to any preceding claim.
21. A slurry comprising a composition according to any of claims 1-19 and at least one solvent.
22. A method of forming a metal-ion battery according to claim 20, the method comprising the step of forming an anode by depositing a slurry according to claim 21 onto a conductive material and evaporating the at least one solvent.
23. A composition comprising a first particulate electroactive material; a first elongate carbon nanostructure material; and a second elongate carbon nanostructure material, wherein the first elongate carbon nanostructure material has a mean average diameter of at least 100 nm; the second elongate carbon nanostructure material has a mean average diameter of less than 90 nm, and wherein the first particulate electroactive material comprises particles having a particle core and pillars of the electroactive material extending from the particle core.
24. A composition comprising a first particulate electroactive material; a first elongate carbon nanostructure material; and a second elongate carbon nanostructure material, wherein the first elongate carbon nanostructure material has a mean average diameter of at least 100 nm; and the second elongate carbon nanostructure material has a mean average diameter in the range of 40-90 nm.

25. A composition comprising a binder; a first particulate electroactive material and carbon additives comprising a first elongate carbon nanostructure material; a second elongate carbon nanostructure material; graphite; and carbon black.
26. A composition comprising a binder; a first particulate electroactive material; carbon nanostructure additives comprising a first elongate carbon nanostructure material, a second elongate carbon nanostructure material, and optionally carbon black wherein the carbon nanostructure additives form no more than 10 weight % of the composition.
27. A composition comprising a binder; a first particulate electroactive material; and at least one elongate carbon nanostructure material wherein the elongate carbon nanostructure material has a functionalised surface.
28. A composition according to claim 27 wherein the composition comprises at least two elongate carbon nanostructure materials.



Figure 1

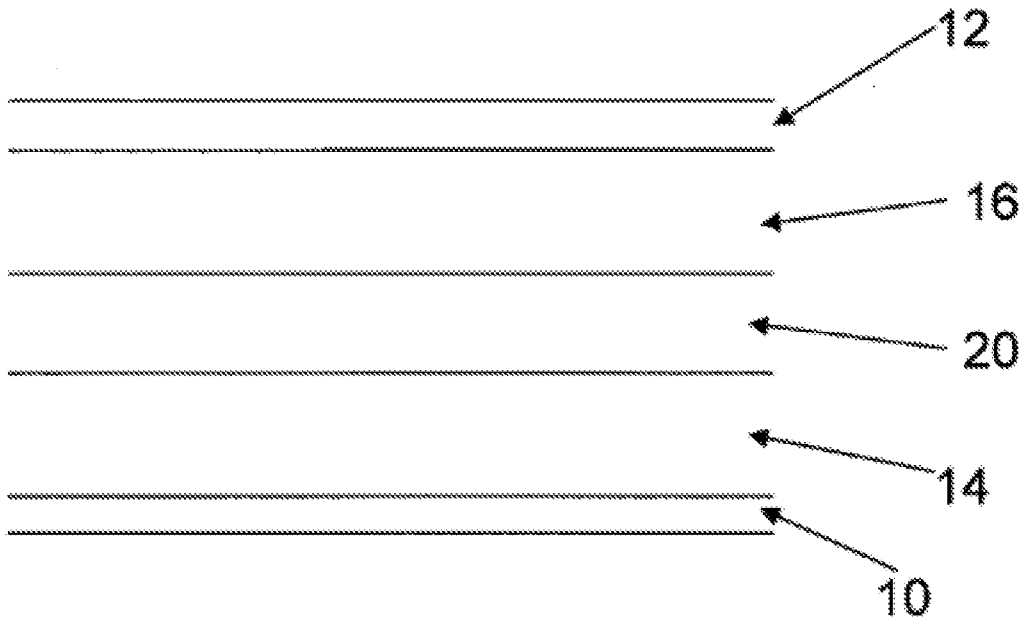


Figure 2

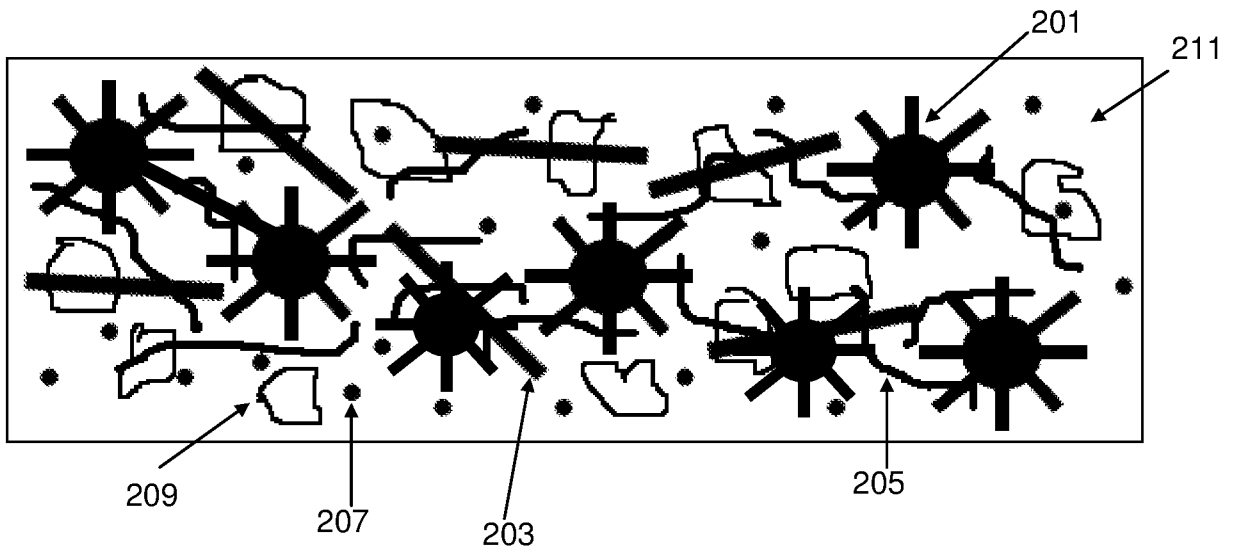


Figure 3A

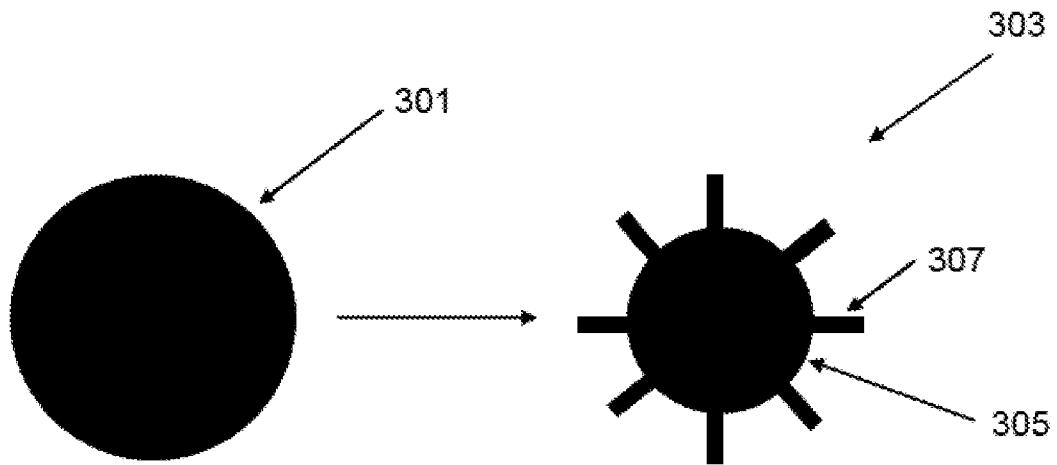


Figure 3B

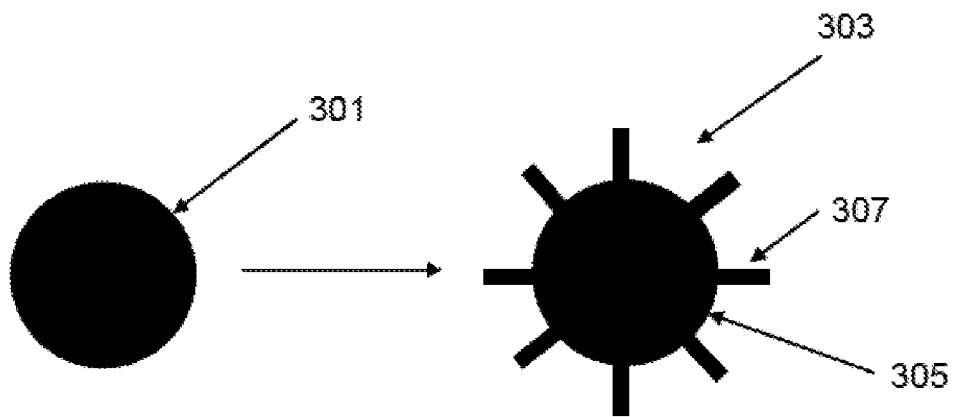


Figure 4A

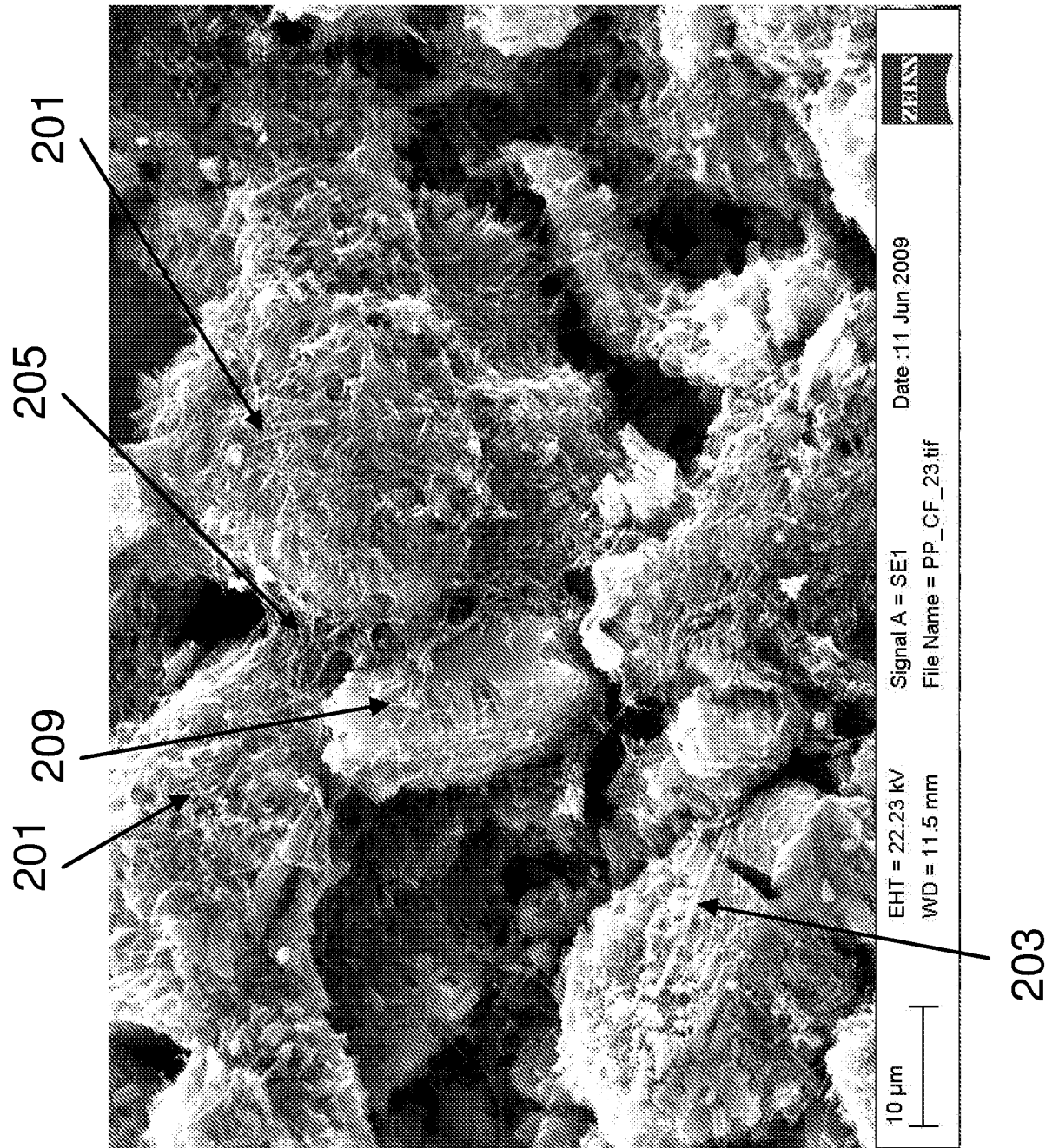


Figure 4B

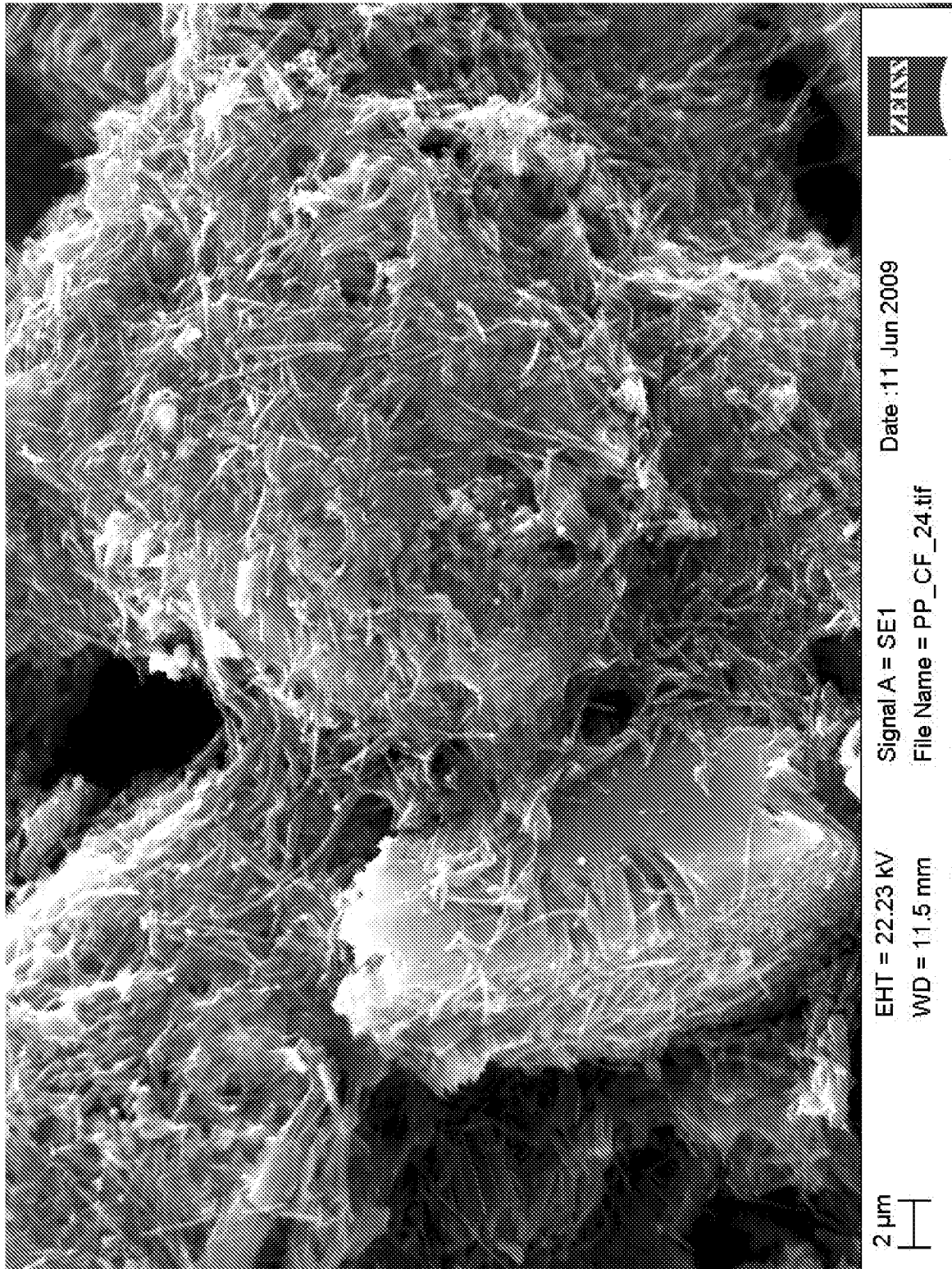


Figure 4C

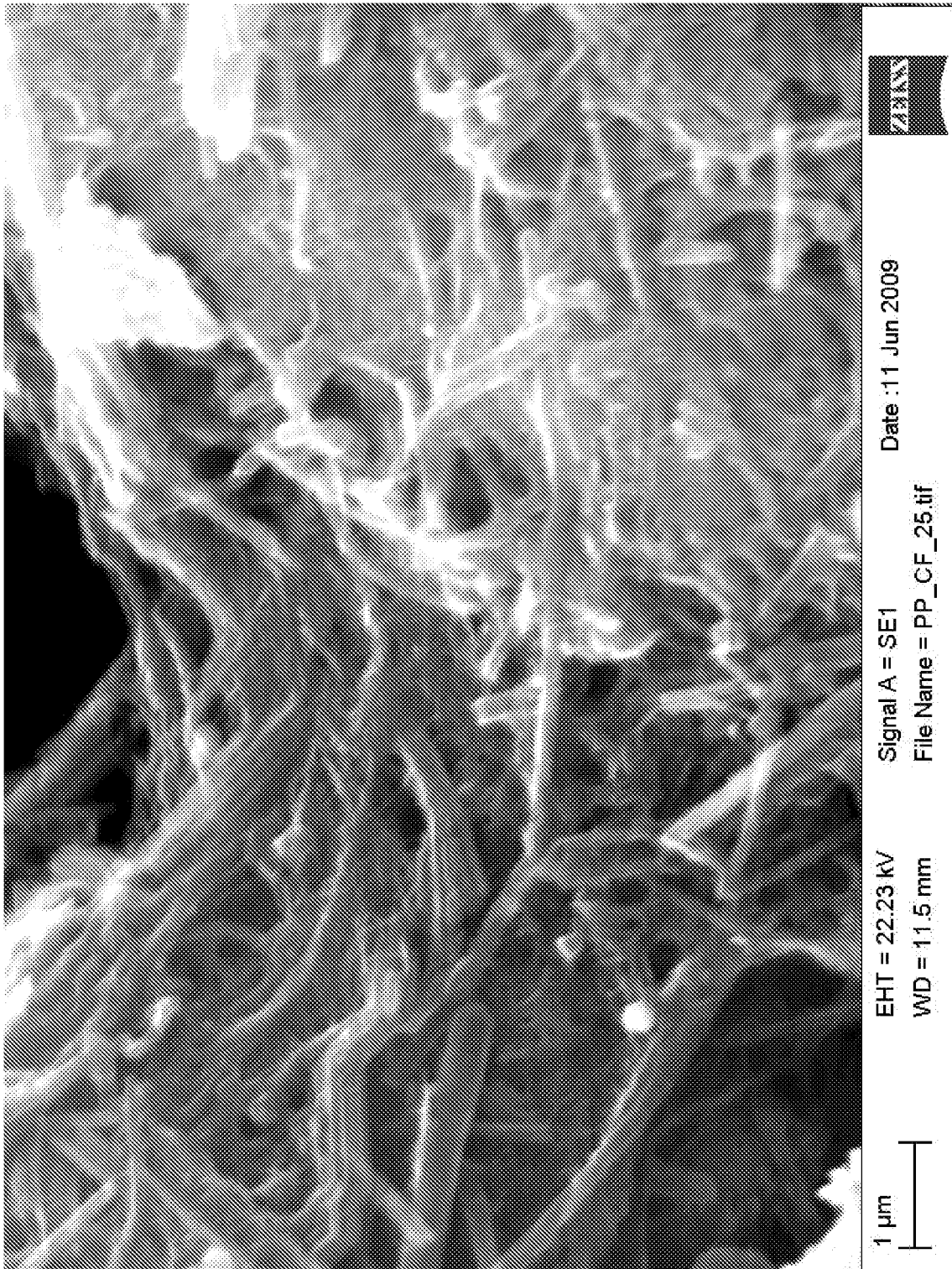


Figure 5

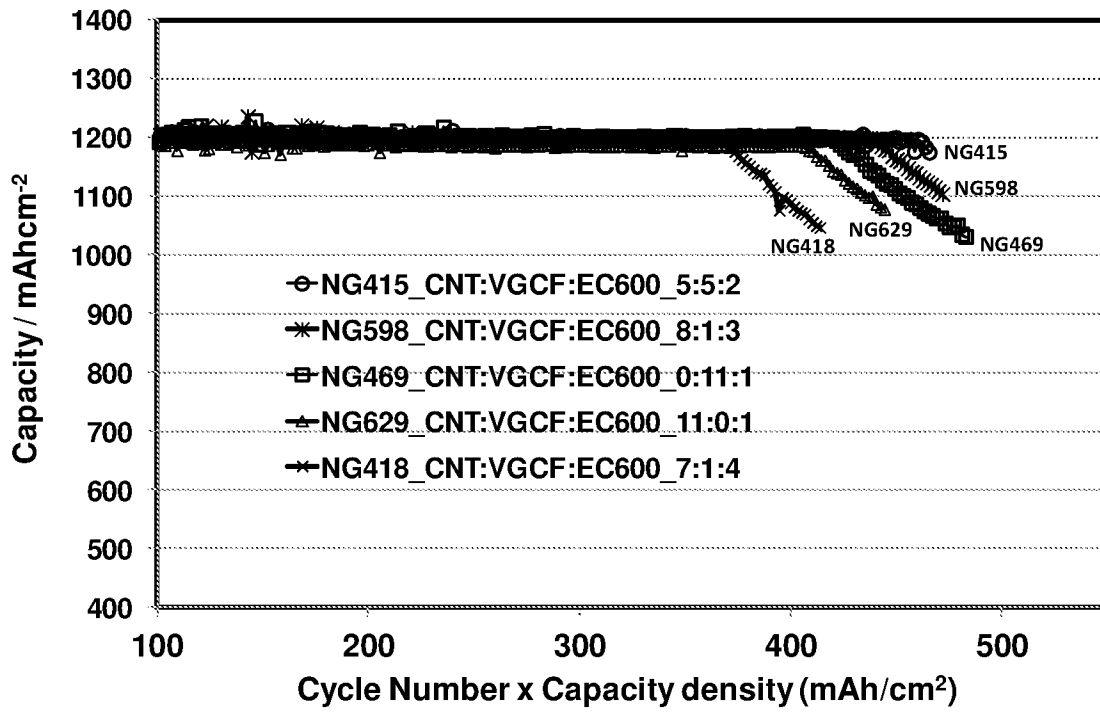


Figure 6A

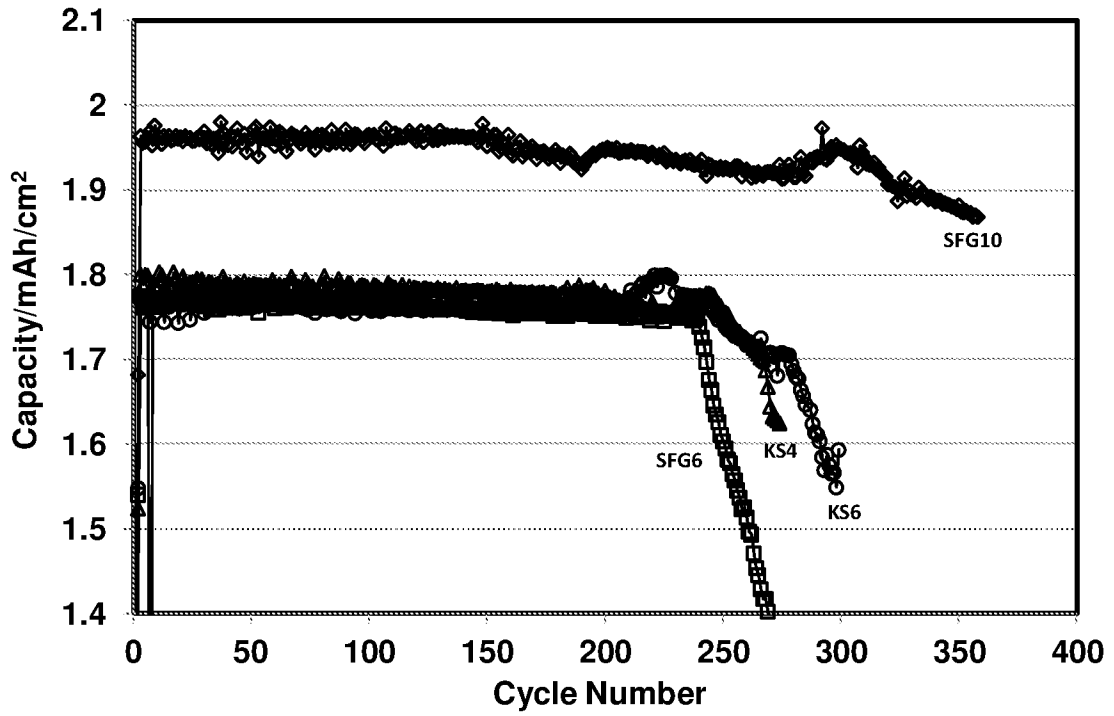


Figure 6B

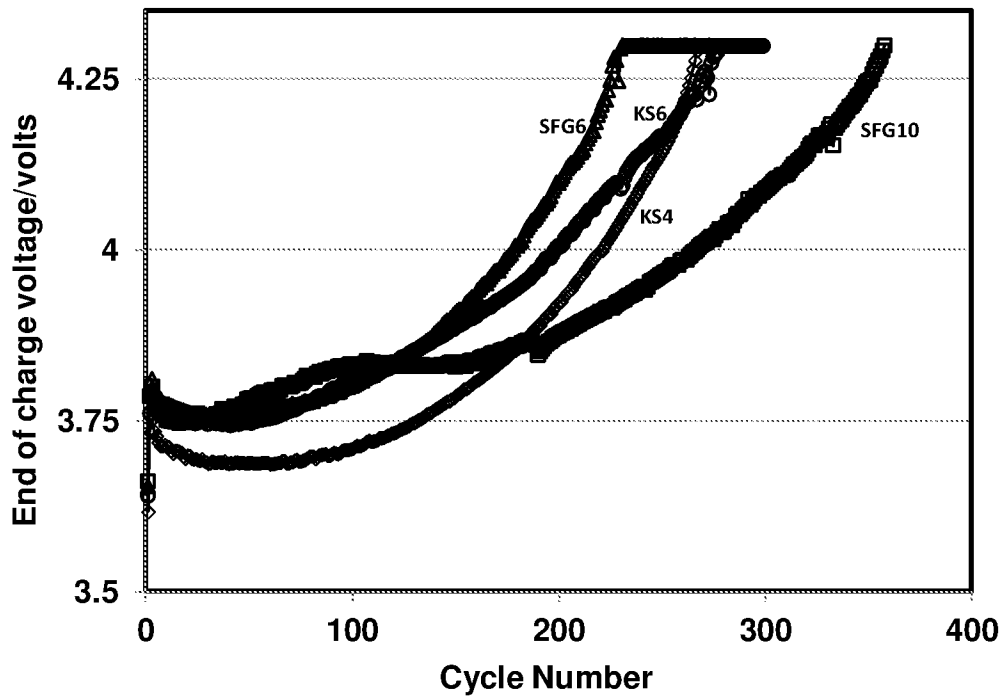


Figure 7A

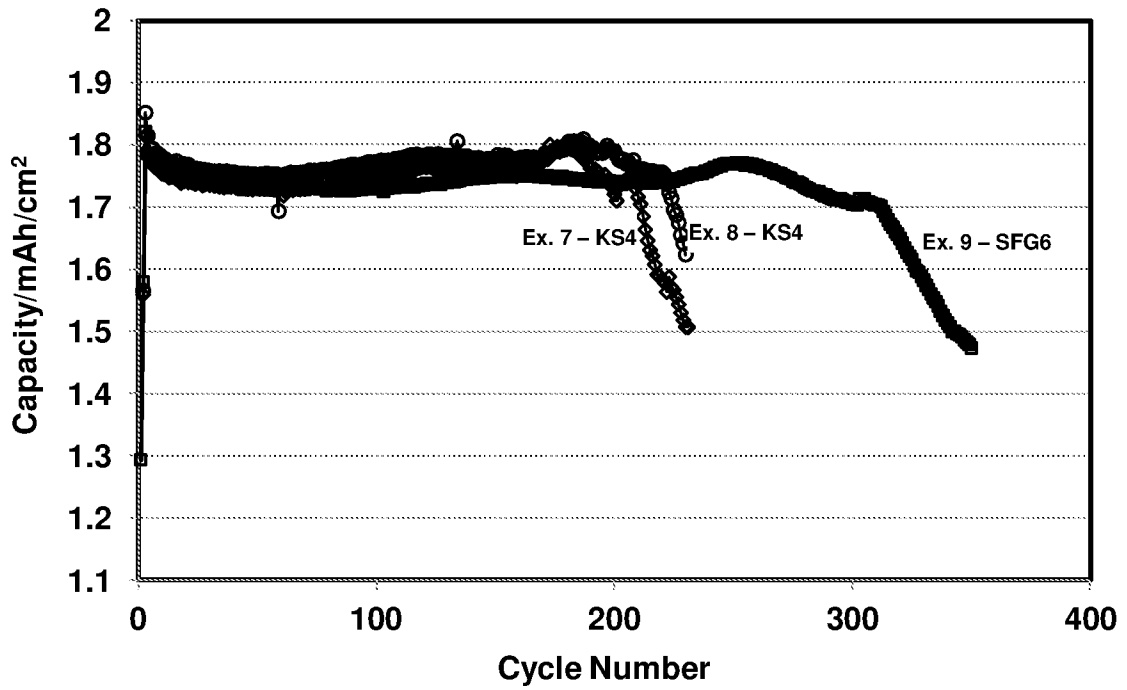


Figure 7B

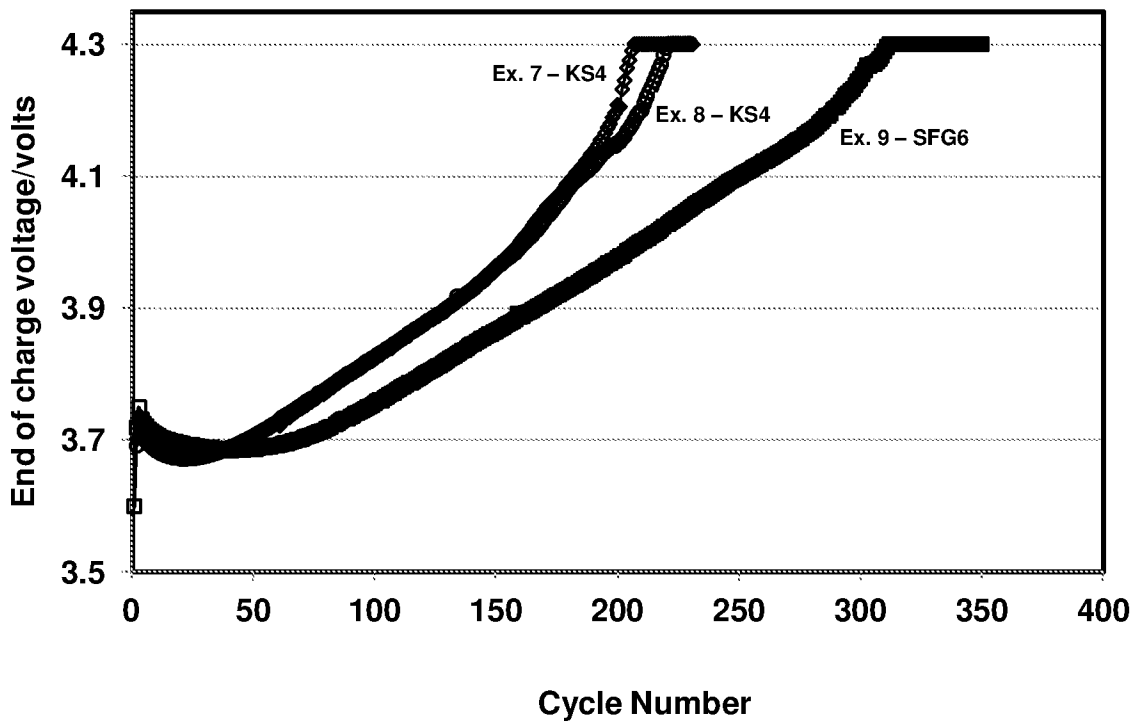




Figure 8

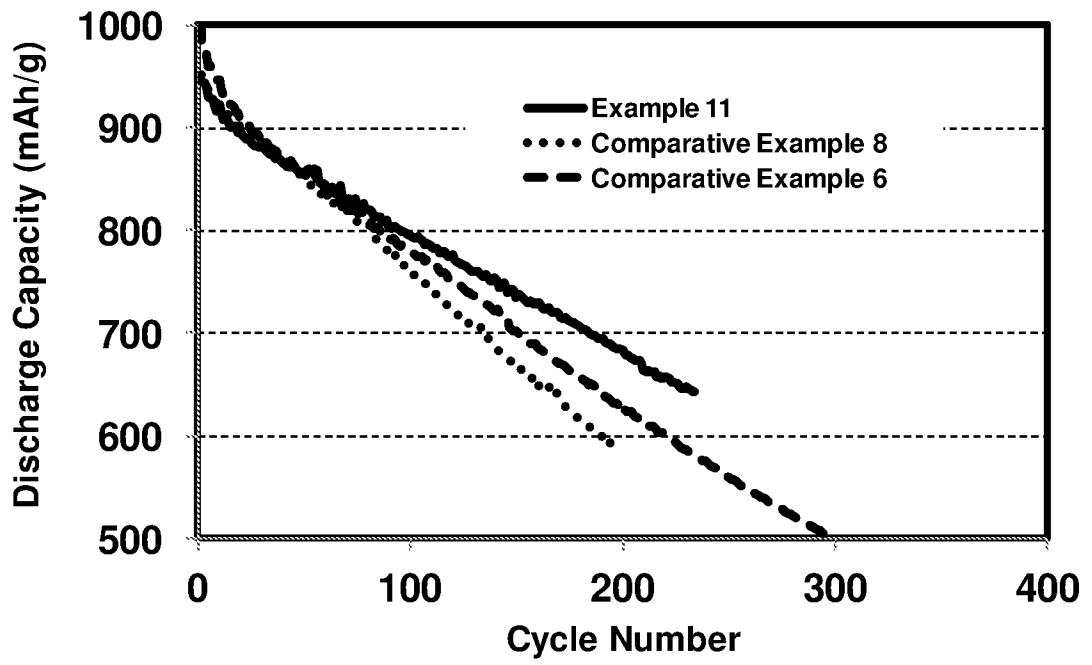


Figure 9

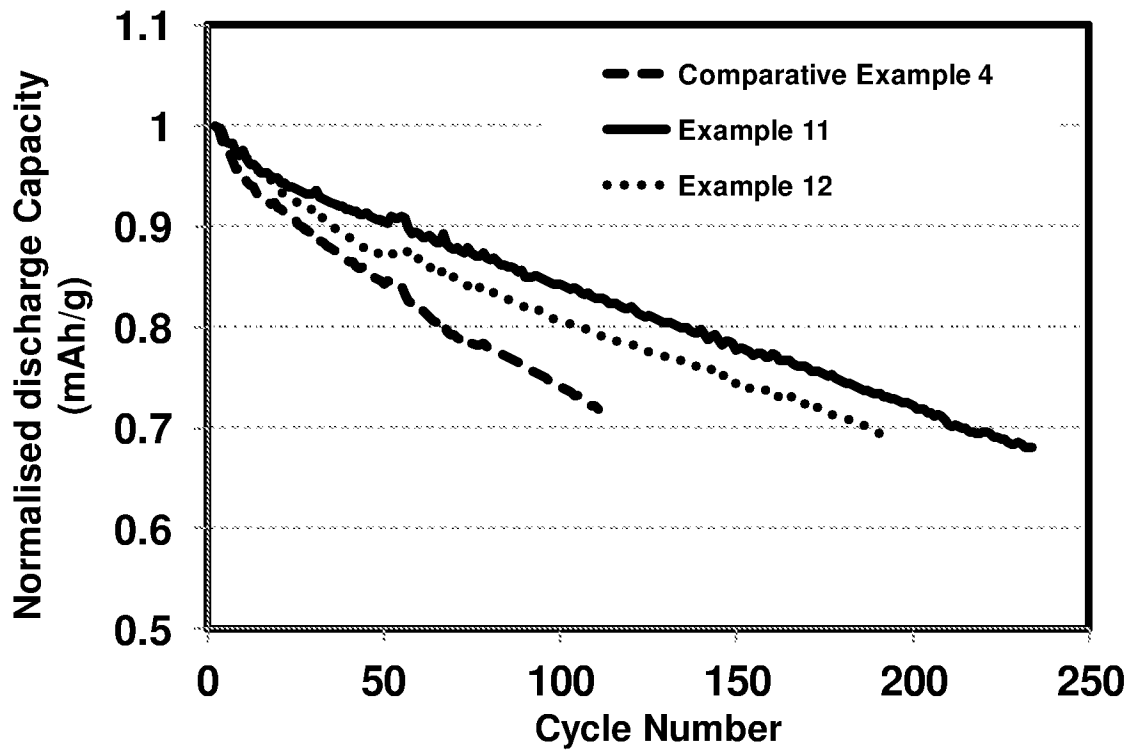
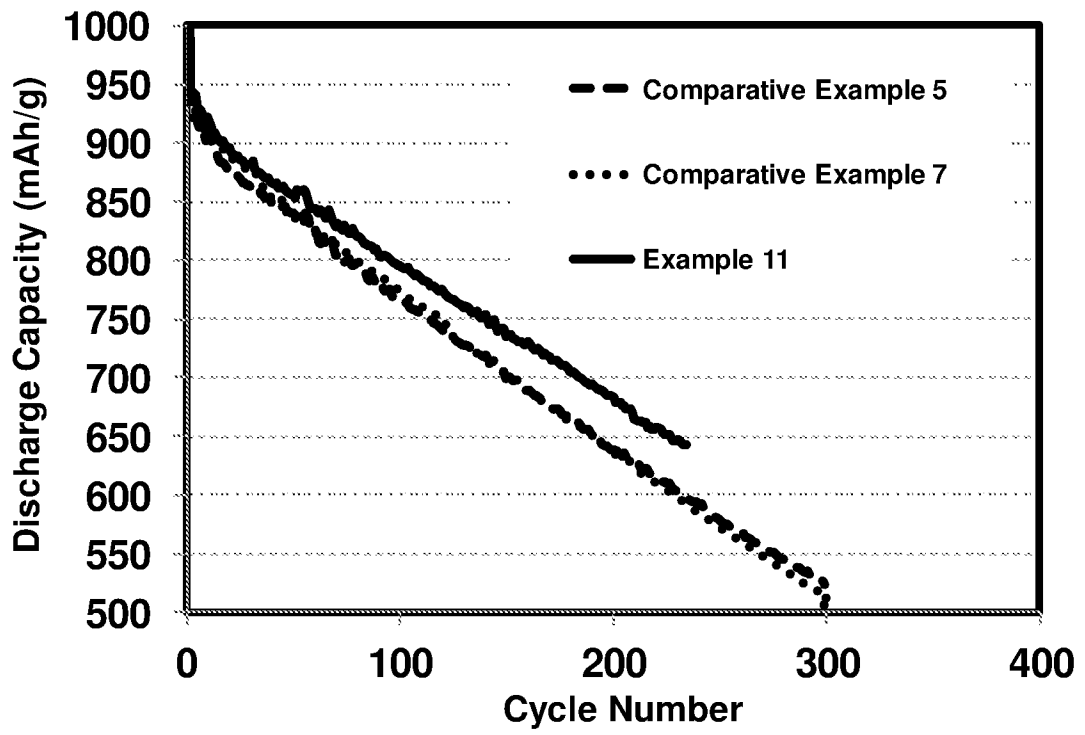


Figure 10



INTERNATIONAL SEARCH REPORT

International application No  
PCT/GB2013/050189

A. CLASSIFICATION OF SUBJECT MATTER  
INV. H01M4/134  
ADD.  
According to International Patent Classification (IPC) or to both national classification and IPC

B. FIELDS SEARCHED  
Minimum documentation searched (classification system followed by classification symbols)  
H01M

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Electronic data base consulted during the international search (name of data base and, where practicable, search terms used)  
EPO-Internal, WPI Data

C. DOCUMENTS CONSIDERED TO BE RELEVANT

Category*	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
X,P	WO 2012/093224 A1 (COMMISSARIAT ENERGIE ATOMIQUE [FR]; RENAULT SA [FR]; JOUANNEAU-SI LARB) 12 July 2012 (2012-07-12) the whole document * see p.5, l.31 - p.12, l.6; claims *	1-28
X,P	WO 2012/028858 A1 (NEXEON LTD [GB]; GREEN MINO [GB]) 8 March 2012 (2012-03-08) the whole document * see claims *	1-28
X	US 2010/143798 A1 (ZHAMU ARUNA [US] ET AL) 10 June 2010 (2010-06-10) the whole document * see [0070], [0071]; Figs. "a, 2B;claims *	1-28
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Further documents are listed in the continuation of Box C.

See patent family annex.

\* Special categories of cited documents :

<p>"A" document defining the general state of the art which is not considered to be of particular relevance</p> <p>"E" earlier application or patent but published on or after the international filing date</p> <p>"L" document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified)</p> <p>"O" document referring to an oral disclosure, use, exhibition or other means</p> <p>"P" document published prior to the international filing date but later than the priority date claimed</p>	<p>"T" later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention</p> <p>"X" document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step when the document is taken alone</p> <p>"Y" document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art</p> <p>"&amp;" document member of the same patent family</p>
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Date of the actual completion of the international search  30 April 2013	Date of mailing of the international search report  22/05/2013
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Name and mailing address of the ISA/ European Patent Office, P.B. 5818 Patentlaan 2 NL - 2280 HV Rijswijk Tel. (+31-70) 340-2040, Fax: (+31-70) 340-3016	Authorized officer  Stellmach, Joachim
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## INTERNATIONAL SEARCH REPORT

International application No  
PCT/GB2013/050189

C(Continuation). DOCUMENTS CONSIDERED TO BE RELEVANT		
Category*	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
X	US 2002/164479 A1 (MATSUBARA KEIKO [JP] ET AL) 7 November 2002 (2002-11-07) the whole document * see [0005] - [0008]; [0018] - [0045]; claims *	1-28
X	----- LESTRIEZ B; DESAEVER S; DANET J; MOREAU P; PLEE D; GUYOMARD D: "Hierarchical and resilient conductive network of bridged carbon nanotubes and nanofibers for high-energy Si negative electrodes", ELECTROCHEMICAL AND SOLID-STATE LETTERS, vol. 12, no. 4, 1 September 2009 (2009-09-01), pages A76-A80, XP002696365, the whole document * see p.A76, right col., *	1-28
X	----- WO 2005/096414 A2 (DEGUSSA [DE]; PETRAT FRANK-MARTIN [DE]; HENNIGE VOLKER [DE]; ALBRECHT) 13 October 2005 (2005-10-13) the whole document * see p.5, 1.25 p.8, 1.25; claims *	1-28
Y	----- WO 2010/026332 A1 (ARKEMA FRANCE [FR]; CT NAT DE RECH SCIENT [FR]; PLEE DOMINIQUE [FR]; L) 11 March 2010 (2010-03-11) the whole document * see p.9, 1.9 - p.11, 1.26; exs. 1 -4: claims *	1-28
Y	----- WO 2009/033082 A2 (BOSTON SCIENT SCIMED INC [US]; CHEN JOHN JIANHUA [US]; HORN DANIEL J [ ]) 12 March 2009 (2009-03-12) the whole document * see p.5, 1.18 - p.9, 1.16; claims *	1-28
Y	----- WO 2009/010758 A2 (NEXEON LTD [GB]; GREEN MINO [GB]) 22 January 2009 (2009-01-22) cited in the application the whole document * see p.2, 1.4 - p.3, 1.3; claims *	1-28
Y	----- WO 2006/135375 A2 (UNIV CALIFORNIA [US]; JIN SUNGHO [US]; CHEN LI-HAN [US]; AUBUCHON JOSE) 21 December 2006 (2006-12-21) the whole document * see p.5, 1.3 - 15; claims *	1-28
Y	----- WO 2006/097380 A1 (DEGUSSA [DE]; PETRAT FRANK-MARTIN [DE]; WIGGERS HARTMUT [DE]; REEKEN B) 21 September 2006 (2006-09-21) the whole document * see p.4, 1.5 - p.7, 1.20; claims *	1-28

## INTERNATIONAL SEARCH REPORT

Information on patent family members

International application No

PCT/GB2013/050189

Patent document cited in search report	Publication date	Patent family member(s)	Publication date
WO 2012093224 A1	12-07-2012	FR 2970245 A1 WO 2012093224 A1	13-07-2012 12-07-2012
WO 2012028858 A1	08-03-2012	CA 2810331 A1 GB 2483373 A TW 201228082 A WO 2012028858 A1	08-03-2012 07-03-2012 01-07-2012 08-03-2012
US 2010143798 A1	10-06-2010	NONE	
US 2002164479 A1	07-11-2002	CN 1374712 A US 2002164479 A1	16-10-2002 07-11-2002
WO 2005096414 A2	13-10-2005	AT 394806 T CN 101095251 A DE 102004016766 A1 EP 1730800 A2 JP 5096136 B2 JP 2007534118 A KR 20070004794 A US 2007281216 A1 WO 2005096414 A2	15-05-2008 26-12-2007 20-10-2005 13-12-2006 12-12-2012 22-11-2007 09-01-2007 06-12-2007 13-10-2005
WO 2010026332 A1	11-03-2010	CN 102197519 A EP 2351121 A1 FR 2935546 A1 JP 2012501515 A KR 20110063634 A US 2011163274 A1 WO 2010026332 A1	21-09-2011 03-08-2011 05-03-2010 19-01-2012 13-06-2011 07-07-2011 11-03-2010
WO 2009033082 A2	12-03-2009	EP 2197508 A2 JP 2010537793 A US 2009149948 A1 WO 2009033082 A2	23-06-2010 09-12-2010 11-06-2009 12-03-2009
WO 2009010758 A2	22-01-2009	AT 535033 T AT 548768 T CA 2693460 A1 CN 101790805 A EP 2183804 A2 EP 2194596 A2 EP 2204868 A2 EP 2533331 A2 EP 2533332 A2 HK 1138943 A1 HK 1140855 A1 JP 4834814 B2 JP 2010533637 A JP 2011046603 A JP 2011222522 A JP 2012254934 A KR 20100058486 A KR 20120120392 A RU 2010100662 A SG 185308 A1 TW 200917550 A US 2010178565 A1	15-12-2011 15-03-2012 22-01-2009 28-07-2010 12-05-2010 09-06-2010 07-07-2010 12-12-2012 12-12-2012 19-10-2012 06-07-2012 14-12-2011 28-10-2010 10-03-2011 04-11-2011 27-12-2012 03-06-2010 01-11-2012 27-08-2011 29-11-2012 16-04-2009 15-07-2010

# INTERNATIONAL SEARCH REPORT

Information on patent family members

International application No PCT/GB2013/050189
---

Patent document cited in search report	Publication date	Patent family member(s)	Publication date
		US 2011067228 A1	24-03-2011
		WO 2009010758 A2	22-01-2009
WO 2006135375	A2	21-12-2006	
		US 2007207318 A1	06-09-2007
		WO 2006135375 A2	21-12-2006
WO 2006097380	A1	21-09-2006	
		CN 101137765 A	05-03-2008
		DE 102005011940 A1	21-09-2006
		EP 1859073 A1	28-11-2007
		JP 2008532912 A	21-08-2008
		KR 20070111521 A	21-11-2007
		US 2008145761 A1	19-06-2008
		WO 2006097380 A1	21-09-2006