

(12) INTERNATIONAL APPLICATION PUBLISHED UNDER THE PATENT COOPERATION TREATY (PCT)

(19) World Intellectual Property Organization  
International Bureau



(43) International Publication Date  
30 May 2003 (30.05.2003)

PCT

(10) International Publication Number  
**WO 03/044028 A1**

- (51) International Patent Classification<sup>7</sup>: C07F 7/18, 7/14 45750 (US). **FILIPKOWSKI, Michelle, A.**; 729 Second Street, Marietta, OH 45750 (US). **HIMMELDIRK, Rodica, S.**; 4 Oak View Circle, Vincent, OH 45784 (US).
- (21) International Application Number: PCT/US02/32826
- (22) International Filing Date: 15 October 2002 (15.10.2002) (74) Agent: **DILWORTH, Michael, P.**; Crompton Corporation, 199 Benson Road, Middlebury, CT 06749 (US).
- (25) Filing Language: English
- (26) Publication Language: English (81) Designated States (*national*): BR, CA, CN, JP, KR.
- (30) Priority Data: 10/004,156 15 November 2001 (15.11.2001) US (84) Designated States (*regional*): European patent (AT, BE, BG, CH, CY, CZ, DE, DK, EE, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE, SK, TR).
- (71) Applicant: **CROMPTON CORPORATION** [US/US]; 199 Benson Road, Middlebury, CT 06749 (US). **Published:**  
— with international search report
- (72) Inventors: **WESTMEYER, Mark, D.**; 107 Cornerstone Drive, Marietta, OH 45740 (US). **HALE, Melinda, B.**; 531 Riverview Drive, Belmont, WV 26134 (US). **CHIL-DRESS, Shawn, R.**; 107 Briargate Drive, Marietta, OH
- For two-letter codes and other abbreviations, refer to the "Guidance Notes on Codes and Abbreviations" appearing at the beginning of each regular issue of the PCT Gazette.*



**WO 03/044028 A1**

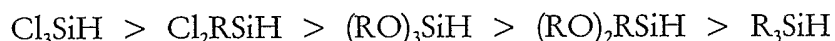
(54) Title: PROMOTED HYDROSILATION REACTIONS

(57) Abstract: A process of preparing compounds containing silicon-carbon bonds by the hydrosilation of a hydridoalkoxysilyl reactant with an olefinic reactant comprises operating the process in the presence of a platinum catalyst and a reaction promoter comprising a weakly nucleophilic amine of the formula  $NZ^1Z^2Z^3$  wherein  $Z^1$  is an aryl, alkaryl, or aralkyl group of  $C_6$  to  $C_{20}$  carbon atoms, or an organosilyl substituent,  $SiR_3$ , wherein R is an alkyl of  $C_1$  to  $C_{20}$  or an aryl of  $C_6$  to  $C_{10}$ ,  $Z^2$  is hydrogen, alkyl of  $C_1$  to  $C_{20}$ , an aryl, alkaryl, or aralkyl group of  $C_6$  to  $C_{20}$  carbon atoms, or  $SiR_3$ , wherein R is as previously defined;  $Z^3$  is the same as  $Z^1$  or  $Z^2$ , and, optionally, two of  $Z^1$ ,  $Z^2$  and  $Z^3$  taken together with the nitrogen atom form an aromatic heterocyclic ring.

## PROMOTED HYDROSILATION REACTIONS

Background Of The Invention

In the production of silicon compositions, transition metal catalysts have long been  
5 known to promote the hydrosilation reaction. Each transition metal-catalyzed  
hydrosilation reaction differs dramatically such that it is difficult to predict which  
transition metal will efficiently catalyze the hydrosilation reaction of a specific  
hydrosilyl reactant with a particular unsaturated reactant. For example, the effect of  
substituents on the silicon atom on adduct yield obtained in the platinum (Pt)-catalyzed  
10 reactions with 1-alkenes is in the following order of activity (R = Et):



The general trend for Pt-catalyzed hydrosilation reactions is that chlorosilanes are more  
15 reactive than alkoxy silanes (*Comprehensive Handbook on Hydrosilylation*; B. Marciniec,  
Ed.; Pergamon Press, New York, 1992; Ch.4; J. L. Speier *Adv. Organomet. Chem.* 1979, 17,  
407; E. Lukevics *Russ. Chem. Rev.* 1977, 46, 197). However, if one evaluates a different  
transition metal or olefin, the above trend may be different. For example, in the  
hydrosilation reaction of heptene with rhodium (Rh), the above trend is the reversed. Due  
20 to the relative importance of the Pt-catalyzed hydrosilation reaction in commercial  
production of organofunctional silanes, a process that improved both the reactivity and  
selectivity of alkoxy silanes in the hydrosilation relative to that seen with chlorosilanes  
would be valuable.

A number of patents in the art have disclosed that various promoters can increase the rates and/or selectivities of hydrosilation reactions. In terms of chemical structures or properties, the various types of promoters differ dramatically, such that it is not possible to predict which chemical structures or properties are important for promotion, or even  
5 which hydrosilation reactions may be promoted, since promotion will also depend on the chemical structures and properties of each of the hydridosilyl reactant, the unsaturated reactant, and the hydrosilation catalyst. For example, the reaction of trichlorosilane with allyl chloride is promoted by weak amines such as phenothiazine (V. T. Chuang U.S. Pat. No. 3,925,434), while the reaction of methyldichlorosilane with allyl chloride requires a  
10 more basic tertiary amine such as tributylamine (Ger. Patent 1,156,073; C. Hu et al. *Fenzi Cuihua*, 1988, 2, 38-43; see Chem. Abstr. 1989, 111, 78085m). Both of those reactions can be promoted with a second hydridosilane (U.S. 4,614,812) through a different promotion mechanism. Alkali metal carbonates or bicarbonates promote hydrosilations of allylic amines with hydridoalkoxysilanes (U. S. 4,481,364). Other hydrosilation reactions are  
15 promoted by phosphines, oxygen gas (D. L. Kleyer et al. U.S. Pat. No. 5,359,111), oxygen-containing organics including aldehydes, unsaturated ketones (R. Reitmeier et al. U.S. Pat. No. 5,663,400, H. M. Bank et al. U.S. Pat. No. 5,623,083), tertiary alcohols and silylated derivatives thereof, and propargylic alcohols and silylated derivatives thereof (H. M. Bank et al. U.S. Pat. No. 5,756,795), inorganic or organic salts including sodium alkoxides and  
20 compounds of tin and cobalt, and other organic compounds, including alcohols, diols, ethers and esters. Carboxylic acids, along with ketones, and esters thereof, appear to promote platinum-catalyzed hydrosilation reactions between hydridoalkoxysilanes and allylamine (USSR 415,268). The use of acetic acid in promoting hydrosilations involving trimethoxysilane has been coincidental with the use of vinylcyclohexene oxide as the

olefin, since acetic acid was discovered to be an impurity derived from early processes to make that epoxyolefin using peracetic acid (U. S. 2,687,406), as well as allyl glycidyl ether (*J. Am. Chem. Soc.* 1959, 81, 3350).

Hydrosilation promotion effects are narrowly specific, and an effective promoter  
5 may work for a single hydrosilation reaction between a specific hydridosilane and a specific olefin. In addition to increasing reaction rates, yields, or selectivities, a promoter may act by preventing undesirable side reactions, which reduce yields/selectivities, such as undesired polymerization or formation of less desirable isomeric products. For example, added methanol is disclosed as being effective in reducing the undesired beta-isomer  
10 content in reaction products from platinum-catalyzed hydrosilations between trimethoxysilane and the epoxyolefins, i.e., vinylcyclohexene monoepoxide and allyl glycidyl ether (H. Takai et al. US Patent No. 4,966,981).

The use of amines in the hydrosilation of hydridosilane and acrylonitrile has been reported extensively, particularly tertiary amines in the presence of copper (Cu) (B. A.  
15 Bluestein U.S. Pat. No. 2,971,970, 1961; Z. V. Belyakova et al. translation from *Zhurnal Obshchei Khimii* 1964, 34, 1480-1484; A. Rajkumar et al. *Organometallics* 1989, 8, 549-550; H. M. Bank U.S. Pat. No. 5,283,348, and U.S. Pat. No. 5,103,033). U.S. Pat. No. 4,292,434 (T. Lindner et al.) describes the preparation of an amine-platinum catalyst and its use in the hydrosilation reaction. K. R. Mehta et al. in U.S. Pat. No. 5,191,103 reported  
20 the use of sterically hindered amines, phosphines or their equivalent salts in the presence of a platinum catalyst to promote the hydrosilation reaction.

In addition to promoting the hydrosilation reaction, amines have been reported to be inhibitors for the hydrosilation reaction. For example G. Janik et al. in U.S. Pat. No. 4,584,361 reported that amines inhibited polyorganosiloxane compositions at temperatures

below 40°C, but not at 135°C. Also R. P. Eckberg et al. reported the use of tertiary amines in the presence of both Rh and Pt catalysts to inhibit epoxy-polymerization in the production of epoxysilicones.

The hydrosilation reactions of many olefins, particularly amino-functional olefins,  
5 are either too slow or do not occur. For those olefins that do undergo hydrosilation, formation of the undesired  $\beta$ -isomer is a competing side reaction. The type of silane employed also impacts the rate of reaction. Typically, sluggish hydrosilation reactions result in an increase of the competing side-reactions, e.g., olefin isomerization or polymerization. Accordingly, a process which improves the reactivity and selectivity of  
10 the transition metal-catalyzed hydrosilation reactions of olefins continues to be a commercially desirable objective.

#### SUMMARY OF THE INVENTION

In accordance with the invention, a process is provided which comprises reacting  
(a) hydridoalkoxysilane with (b) olefin in the presence of (c) platinum catalyst and (d) a  
15 weakly nucleophilic amine of the formula  $NZ^1Z^2Z^3$ , wherein  $Z^1$  is an aryl, alkaryl, or aralkyl group of  $C_6$  to  $C_{20}$  carbon atoms, or an organosilyl group of the formula  $SiR_3$ , wherein R is alkyl of  $C_1$  to  $C_{20}$  or aryl of  $C_6$  to  $C_{10}$ ;  $Z^2$  is hydrogen, alkyl of  $C_1$  to  $C_{20}$ , an aryl, alkaryl, or aralkyl group of  $C_6$  to  $C_{20}$ , or  $SiR_3$ , wherein R is as previously defined;  $Z^3$  is the same as  $Z^1$  or  $Z^2$ ; and optionally two of  $Z^1$ ,  $Z^2$  and  $Z^3$  taken together with the  
20 nitrogen atom form an aromatic heterocyclic ring. The process of the invention exhibits improved yields and selectivities with respect to the desired reaction products.

DETAILED DESCRIPTION OF THE INVENTION

This invention provides a process for improving the yields and rates of the hydrosilation of alkoxyhydridosilane under relatively mild conditions using a weakly nucleophilic amine in the presence of a hydrosilation catalyst.

## 5 AMINES

Weakly nucleophilic amines containing substituents capable of  $\pi$ -interaction with the amine's lone pair of electrons such as aromatic or silicon-substituents can be employed in the practice of this invention. Thus, weakly nucleophilic amine promoters possess the general formulae  $NZ^1Z^2Z^3$  wherein  $Z^1$  is an aryl, alkaryl, or aralkyl group of six to twenty  
10 carbon atoms, or an organosilyl group of the formula  $SiR_3$ , wherein R is an alkyl of  $C_1$  to  $C_{20}$ , preferably  $C_1$  to  $C_4$ , or aryl of  $C_6$  to  $C_{10}$ ;  $Z^2$  is hydrogen, alkyl of  $C_1$  to  $C_{20}$ , preferably  $C_1$  to  $C_4$ , an aryl, alkaryl, or aralkyl group of  $C_6$  to  $C_{20}$ , or  $SiR_3$  wherein R is as previously defined; and  $Z^3$  is the same as  $Z^1$  or  $Z^2$ . Optionally, two of  $Z^1$ ,  $Z^2$  and  $Z^3$  taken together may form an aromatic heterocyclic ring including the nitrogen atom. Weakly nucleophilic  
15 amines include, but are not limited to, aniline, hexamethyldisilazane, phenothiazine, aminonaphthalene, benzylamine, pyridine and their corresponding derivatives. Aniline, benzylamine, and hexamethyldisilazane are the preferred amines with this invention, the choice depending on the hydridosilane and olefin reactants.

## 20 HYDRIDOSILANES

Promotable hydridosilanes in general can be represented by the formula  $R_nX_{3-n}SiH$ , wherein R is a branched or linear alkyl group of 1 to 18 carbon atoms, a cyclic alkyl group of four to eight carbon atoms or an aryl, alkaryl, or aralkyl group of six to twelve carbon atoms, optionally containing halogen, oxygen or nitrogen substituents with the proviso

that such substituents do not interfere with either hydrosilation or promotion, and X is an alkoxy group, selected from -OR, wherein R is as defined above, and n is 0, 1, or 2. The hydridosilanes may be alkoxysilanes selected from the group of trimethoxysilane, triethoxysilane, tri-n-propoxysilane, and triisopropoxysilane. Trimethoxysilane and triethoxysilane are preferred. Other hydridoalkoxysilanes include alkylalkoxysilanes such as methyldimethoxysilane, methyldiethoxysilane, dimethylmethoxysilane, and dimethylethoxysilane.

## OLEFINS

The olefins which can be employed in accordance with the invention are aliphatically unsaturated molecules, which may have certain functional substituents thereon. The term "olefins" utilized herein is being used in its broadest sense and therefore shall be understood to include alkenes, vinyl group-containing and allyl group-containing compounds. Terminal alkenes can be advantageously employed, such as the 1-alkenes, including ethylene, propylene, butene, pentene, hexene, octene, hexadecene, octadecene, trivinylcyclohexene, and the 2-alkyl-1-alkenes, such as 2-methylpropene, 2-methylbutene, diisobutylene, as well as non-terminal alkenes such as tertiary amylene and 2-butene. The 1-alkenes are preferred. Other suitable olefins include the epoxy olefins, such as vinylcyclohexene monoxide, allyl glycidyl ether, and allylic olefins, including, but not limited to allyl esters, allyl polyethers, and allylic tertiary amines, as well as their methallyl derivatives. Other olefins include the amino olefins, such as *N*-allylaniline, *N,N*-dimethallylamine, and *N*-ethylmethallylamine. Vinyl group-containing compounds include the vinyl esters and ethers, vinylsilanes, acrylates and methacrylates.

## CATALYST

Catalysts include those which contain platinum and that function as either homogeneous or heterogeneous hydrosilation catalysts. Typical catalysts include chloroplatinic acid and various solutions thereof, including solutions wherein the  
5 chloroplatinic acid has been chemically modified, chloroplatinate salts and their solutions, vinylsiloxane complexes containing platinum and solutions thereof (Karstedt catalyst), olefin and diolefin complexes of platinum and solutions thereof, and platinum deposited as the metal on various substrates, including carbon, alumina, silica, organically modified silicas, or base metals. Platinum complexes containing strongly bound ligands such as  
10 phosphines, acetylacetonate groups, or amines, may be promotable with the proviso that such ligands must not interfere with either the hydrosilation or the promotion. The catalyst should be used at a level of 0.5 to 100 ppm based on total charge, preferably 5 to 50 ppm, most preferably 5 to 15 ppm.

## 15 PROCESS

Promotion by amines is not subject to limitations regarding equipment, relative to size or type of material of construction. A wide variety of laboratory or commercial scale equipment currently capable of running hydrosilation reactions may be used. The hydrosilation process may be run in a batch, semi-batch or continuous mode.

20 Reaction conditions are also not narrowly critical with regard to temperature, pressure, or the absence or presence of inert solvents. Conditions currently in use for various hydrosilation reactions can be used for the promoted hydrosilations. It is possible that effective promotion will be accompanied by the added advantages of lowering reaction temperature, and catalyst concentration, or both. Preferred reaction conditions include a



temperature from about ambient temperature up to about 150°C with 60 to 120°C being most preferred. Generally, the process is carried out at a pressure of about 0.2 to 2.0 atmospheres (0.02-0.2 MPa), with ambient pressure being preferred, but operation at higher or lower pressures may be performed to maintain higher or lower reaction temperatures  
5 dependent on the volatilities of the respective reactants.

The residence time within the reactor is not critical but should be sufficient to achieve a satisfactory degree of conversion to the hydrosilated product, i.e., > 80%, within acceptable limits given the volume of the equipment and the desired rate of production.

Typical acceptable residence times are on the order of 0.5 to 4 hours.

10 Preferably the olefin should be present at a molar excess of 5-20%, though a stoichiometric equivalence or a molar excess of the silane may be used. The use of promoters of the instant invention can allow the use of lower molar excesses of olefins due to reduction of the competing olefin isomerization side-reaction.

The amine may be present at the start of the hydrosilation, or may be added during the  
15 reaction if it is not proceeding well (Caution – amine should not be added to incomplete reactions wherein significant quantities of both hydridosilyl reactant and olefinic reactant have accumulated, a rapid exothermic reaction may occur). The amines can be used at a concentration of 25 to 20,000 ppm (wt/wt); however, the preferred amine concentration is dependent on the olefin-silane system. The best mode of practice is to introduce the amine  
20 with the olefin and not with the alkoxysilane; although the amine can be introduced with the silane.

Promotion by amines is effective for those hydrosilation products which can be purified, as by distillation, and thusly separated from the amines, which may be lower or

higher boilers which will be stripped or remain in the distillation residue, and can be isolated for disposal, or reused to promote a subsequent batch of product.

### EXAMPLES

5           The following illustrative and comparative examples are intended to describe the present invention in more detail; however, they are not meant to limit the scope of the specification and claims. All parts and percentages presented in the examples below are by weight unless otherwise specified. The abbreviations g, mL, VCMX, AGE, TVC, CPA,  $\text{Pt}_2(\text{M}^*\text{M}^*)_3$  solution, Si-H, AcOH, MeOH, EtOH, and GC stand for gram, milliliter, 4-  
10 vinylcyclohexene monoxide, allyl glycidyl ether, a mixture of three structural isomers of trivinylcyclohexane, solution consisting of 10% (wt/wt) of hexachloroplatinic acid in ethanol, a solution of 12% (wt/wt) of tris(tetramethyldivinylsiloxane)diplatinum (0), any silicon hydride-containing species, acetic acid, methanol, ethanol, and gas chromatography, respectively. An internal standard was used in order to determine the percent of uneluted  
15 heavies for the GC analysis for those examples with VCMX. Uneluted heavies are defined as all components that did not elute under the GC conditions employed for the specific analysis. An \* indicates no internal standard was used.

### EXAMPLES OF THE INVENTION

#### **General procedure for the reaction of an olefin and an alkoxy silane.**

20           1.) **In the presence of an excess of olefin.** A typical reaction was conducted by treating 1.05 to 1.30 molar equivalents (vs. an alkoxy silane) of an olefin at room temperature, an amine and platinum precatalyst or a precatalyst solution. This solution was warmed. At 90 °C, the solution was treated with 1.00 molar equivalent of an alkoxy silane. The addition of the alkoxy silane resulted in an exothermic reaction. The

temperature of the solution was maintained between 90-100 °C throughout the silane addition. After the alkoxysilane addition was completed, the solution's temperature was maintained at 90 °C for one hour. After this time, solution was allowed to cool to room temperature. An aliquot of the crude reaction was analyzed by GC.

5           2.) In the presence of an excess of alkoxysilane. A typical reaction was conducted by treating 1.05 to 1.30 molar equivalents an alkoxysilane (versus the desired olefin) at room temperature with an amine and platinum precatalyst or a precatalyst solution. The amine can be dissolved either in the alkoxysilane or the olefin. This solution was warmed to ~80 °C (refluxing (MeO)<sub>3</sub>SiH), and treated with 1.00 molar equivalent of the desired  
10 olefin (or olefin-amine solution). The addition of olefin (or olefin-amine solution) resulted in an exothermic reaction. The solution's temperature was maintained between 90-100 °C throughout the olefin's addition. After the olefin addition was completed, the solution's temperature was maintained at 90 °C for one hour. After this time, the solution was allowed to cool to room temperature. An aliquot of This solution was analyzed by GC.

15

#### Comparative Example 1

At room temperature, 20.00 g of neat 1-octene was treated with 0.010 ml of CPA and warmed. At 90°C, the 1-octene solution was treated with 19.10 g of Cl<sub>3</sub>SiH. After the Cl<sub>3</sub>SiH addition was completed, the solution was maintained at 90°C for one hour. This  
20 solution was analyzed by GC.

Cl <sub>3</sub> SiH	Cl <sub>4</sub> Si	1-Octene	Octene isomers	Product
6.32	0.42	0.1	2.20	89.11

## Comparative Example 2

At room temperature, 20.00 g of neat 1-octene was treated with 0.010 ml of CPA and warmed. At 90°C, the 1-octene solution was treated with 18.00 g of (MeO)<sub>3</sub>SiH.

After the (MeO)<sub>3</sub>SiH addition was completed, the solution was maintained at 90°C for one hour. This solution was analyzed by GC.

(MeO) <sub>3</sub> SiH	(MeO) <sub>4</sub> Si	1-Octene	Octene isomers	Product
12.8	0.2	66.2	3.8	15.9

## 10 Comparative Example 3

At room temperature, 20.00 g of neat 1-octene was treated with 0.019 ml of acetic acid, 0.010 ml of CPA and warmed. At 90°C, the 1-octene solution was treated with 18.00 g of (MeO)<sub>3</sub>SiH. After the (MeO)<sub>3</sub>SiH addition was completed, the solution was maintained at 90°C for one hour. This solution was analyzed by GC.

15

(MeO) <sub>3</sub> SiH	(MeO) <sub>4</sub> Si	1-Octene	Octene isomers	Product
0.1	0.3	1.9	8.0	87.6

## Example 1

20 At room temperature, 20.00 g of neat 1-octene was treated with 0.020 ml of aniline, 0.010 ml of CPA and warmed. At 90°C, the 1-octene solution was treated with 18.00 g of (MeO)<sub>3</sub>SiH. After the (MeO)<sub>3</sub>SiH addition was completed, the solution was maintained at 90°C for one hour. This solution was analyzed by GC.

(MeO) <sub>3</sub> SiH	(MeO) <sub>4</sub> Si	1-Octene	Octene isomers	Product
0.1	0.3	8.4	5.5	83.8

For examples 2-8, all reactions were conducted using a 20% molar excess of 1-octene (98%  
 5 purity) versus (MeO)<sub>3</sub>SiH, an amine promoter and 10 ppm Pt as a solution of  
 chloroplatinic acid at 90 °C followed by one hour at 90 °C after the addition was  
 completed. All solutions were analyzed using gas chromatography. The GC data for  
 Examples 2-8 are summarized in Table 1.

10 **Table 1.** The effect of amines on the Pt-catalyzed, amine-promoted hydrosilation reaction of 1-octene.<sup>a</sup>

Examples	Amine Promoters	(MeO) <sub>3</sub> SiH	(MeO) <sub>4</sub> Si	1-Octene	Octene isomers <sup>b</sup>	Product
15	2 1000 ppm phenothiazine	8.8	0.4	54.8	6.5	28.4
	3 640 ppm NH[Si(CH <sub>3</sub> ) <sub>3</sub> ] <sub>2</sub>	6.2	0.9	41.4	5.2	44.9
	4 600 ppm diphenylamine	10.4	1.2	59.1	6.3	21.4
	5 800 ppm di(n-butyl)amine	16.7	1.4	78.6	0.4	1.7
	6 640 ppm NH(t-butyl)[Si(CH <sub>3</sub> ) <sub>3</sub> ]	12.8	1.1	75.6	1.3	8.6
	7 640 ppm N(CH <sub>3</sub> )[Si(CH <sub>3</sub> ) <sub>3</sub> ] <sub>2</sub>	8.4	1.6	40.6	7.6	40.1
20	8 640 ppm N[Si(CH <sub>3</sub> ) <sub>3</sub> ] <sub>3</sub>	10.5	1.2	46.4	9.7	28.1

<sup>a</sup>All reactions were conducted at 90°C using 20% mole excess of 1-octene (98% purity) and 10 ppm Pt

(CPA).

<sup>b</sup>This value is the sum of the three isomers of octene observed.

25

#### Example 9

At room temperature, 20.68 g of neat VCMX was treated with 0.020 g of aniline,  
 0.018 ml of CPA and warmed. At 90°C, the VCMX solution was treated with 18.43 g of  
 (MeO)<sub>3</sub>SiH. After the (MeO)<sub>3</sub>SiH addition was completed, the solution was maintained at  
 30 90°C for one hour. This solution was analyzed by GC.

(MeO) <sub>3</sub> SiH	(MeO) <sub>4</sub> Si	VCMX	VCMX Isomer	Product	Uneluted Heavies
0.1	1.9	1.6	3.8	85.3	3.5

For Examples 10-17, all reactions were conducted using a 10% molar excess of VCMX

5 (97% purity) versus (MeO)<sub>3</sub>SiH, an amine promoter and 10 ppm Pt as a solution of chloroplatinic acid at 90 °C followed by one hour at 90 °C after the addition was completed. All solutions were analyzed using gas chromatography. The GC data for Examples 10-17 are summarized in Table 2.

10 **Table 2.** GC data for the platinum-catalyzed, amine-promoted hydrosilation reaction of VCMX and TMS.<sup>a</sup>

Examples	Amine Promoters	(MeO) <sub>3</sub> SiH	(MeO) <sub>4</sub> Si	VCMX	VCMX	Product	Uneluted
					Isomer		Heavies
10	500 ppm aniline	0.1	1.9	1.6	3.8	85.3	3.5
15	11 500 ppm p-anisidine	0.1	1.0	4.6	0.9	88.4	1.2
	12 500 ppm 4-	0.1	1.3	1.3	5.1	87.6	0.8
	13 500 ppm benzyl amine	0.1	0.8	1.9	2.44	84.9	7.8
	14 640 ppm NH[Si(CH <sub>3</sub> ) <sub>3</sub> ] <sub>2</sub>	0.1	0.5	3.4	1.1	92.2	1.0
	15 500 ppm pyridine	46.5	0.3	50.6	1.4	0.1	0.1
20	16 500 ppm phenothiazine	29.1	0.7	30.4	1.1	21.9	14.5
	17 500 ppm triethylamine	31.8	6.8	39.7	0.1	15.3	0.9

<sup>a</sup>All reactions were conducted using a 10% mole excess of VCMX (97%) vs. (MeO)<sub>3</sub>SiH (99%) and 10 ppm Pt (CPA). No additional gelation inhibitor or promoter was used.

25

#### Comparative Example 4

At room temperature, 20.07 g of neat AGE was treated with 0.010 ml of CPA and warmed. At 90 °C, the AGE solution was treated with 18.0 g of (MeO)<sub>3</sub>SiH. After the (MeO)<sub>3</sub>SiH addition was completed, the solution was maintained at 90 °C for one hour.

30 This solution was analyzed by GC.

(MeO) <sub>3</sub> SiH	(MeO) <sub>4</sub> Si	AGE	AGE isomers	β-isomer	γ-isomer
5.7	0.8	13.4	7.4	0.8	63.2

## Example 18

5 At room temperature, 20.10 g of neat AGE was treated with 0.020 g of phenothiazine, 0.010 ml of CPA and warmed. At 90°C, the AGE solution was treated with 18.0 g of (MeO)<sub>3</sub>SiH. After the (MeO)<sub>3</sub>SiH addition was completed, the solution was maintained at 90°C for one hour. This solution was analyzed by GC.

10	(MeO) <sub>3</sub> SiH	(MeO) <sub>4</sub> Si	AGE	AGE isomers	β-isomer	γ-isomer
	5.4	0.8	13.4	7.4	0.8	63.2

## Example 19

15 At room temperature, 20.10 g of neat AGE was treated with 0.025 mL of NH[Si(CH<sub>3</sub>)<sub>3</sub>]<sub>2</sub>, 0.010 ml of CPA and warmed. At 90°C, the AGE solution was treated with 18.0 g of (MeO)<sub>3</sub>SiH. After the (MeO)<sub>3</sub>SiH addition was completed, the solution was maintained at 90°C for one hour. This solution was analyzed by GC.

20	(MeO) <sub>3</sub> SiH	(MeO) <sub>4</sub> Si	AGE	AGE isomers	β-isomer	γ-isomer
	5.4	1.5	1.0	10.0	1.0	77.1

For example 20-22, all reactions were conducted using a 20% molar excess of AGE (99% purity) versus (MeO)<sub>3</sub>SiH, an amine promoter and 10 ppm Pt as a solution of chloroplatinic acid at 90 °C followed by one hour at 90 °C after the addition was  
25 completed. All solutions were analyzed using gas chromatography. The GC data for comparative Example 4 and Examples 18-22 are summarized in Table 3.

**Table 3.** GC data for the hydrosilation reaction of trimethoxysilane and AGE.<sup>a</sup>

Examples	Amine Promoters	(MeO) <sub>3</sub> SiH	(MeO) <sub>4</sub> Si	AGE	AGE isomers	β-isomer	γ-isomer	
5	CE4	Control	5.7	0.8	13.4	7.4	0.8	63.2
	18	500 ppm phenothiazine	5.4	1.5	1.0	10.0	1.0	77.1
	19	640 ppm NH[Si(CH <sub>3</sub> ) <sub>3</sub> ] <sub>2</sub>	2.1	1.8	0.8	9.8	1.1	76.4
	20	500 ppm acetic acid	0.1	0.3	6.0	11.2	0.5	79.8
10	21	500 ppm pyridine	35.1	1.6	44.8	1.1	0.1	12.3
	22	500 ppm triethylamine	3.1	41.4	50.8	0.2	0.1	0.2

<sup>a</sup>All reactions were conducted at 90 °C using a 20% mole excess of AGE (99%) and 10 ppm Pt (CPA).

### 15 Comparative Example 5

At room temperature, 19.80 g of neat (MeO)<sub>3</sub>SiH was treated with 0.016 ml of Pt<sub>2</sub>(M\*M\*)<sub>3</sub> and warmed. At ~85 °C, the TMS solution was treated with 20.02 g of *N*-allylaniline. After the *N*-allylaniline addition was completed, the solution was maintained at 90°C for one hour. This solution was analyzed by GC.

20

(MeO) <sub>3</sub> SiH	(MeO) <sub>4</sub> Si	<i>N</i> -allylaniline	<i>N</i> -propylaniline	β-isomer	γ-isomer
4.2	2.3	2.0	1.7	23.6	59.4

### Example 23

25 At room temperature, 19.80 g of neat (MeO)<sub>3</sub>SiH was treated with 0.40 g of aniline, 0.016 ml of Pt<sub>2</sub>(M\*M\*)<sub>3</sub> and warmed. At ~85°C, the (MeO)<sub>3</sub>SiH solution was treated with 20.02 g of *N*-allylaniline. After the *N*-allylaniline addition was completed, the solution was maintained at 90°C for one hour. This solution was analyzed by GC.

30



(MeO) <sub>3</sub> SiH	(MeO) <sub>4</sub> Si	<i>N</i> -allylaniline	<i>N</i> - propylaniline	$\beta$ -isomer	$\gamma$ -isomer
5.4	1.1	1.1	0.8	2.1	82.1

## Example 24

5 At room temperature, 19.80 g of neat (MeO)<sub>3</sub>SiH was treated with 0.016 ml of Pt<sub>2</sub>(M\*M\*)<sub>3</sub> and warmed. At ~85°C, the (MeO)<sub>3</sub>SiH solution was treated with a solution consisting of 0.40 g of aniline dissolved in 20.02 g of *N*-allylaniline. After the *N*-allylaniline addition was completed, the solution was maintained at 90°C for one hour. This solution was analyzed by GC.

10

(MeO) <sub>3</sub> SiH	(MeO) <sub>4</sub> Si	<i>N</i> -allylaniline	<i>N</i> - propylaniline	$\beta$ -isomer	$\gamma$ -isomer
0.1	1.7	1.0	0.6	0.6	84.1

For examples 25-28, all reactions were conducted using a 10% molar excess of (MeO)<sub>3</sub>SiH versus *N*-allylaniline (97% purity), an amine promoter and 20 ppm Pt as a solution of [Pt<sub>2</sub>(M\*M\*)<sub>3</sub>] at 85 °C followed by one hour at 90 °C after the addition was completed. All solutions were analyzed using gas chromatography. The GC data for Comparative Example 5 and Examples 23-28 are summarized in Table 4.

20 **Table 4.** GC data for the hydrosilation reaction of *N*-allylaniline.<sup>a</sup>

Examples	Amine Promoters	(MeO) <sub>3</sub> SiH	(MeO) <sub>4</sub> Si	<i>N</i> - allylaniline	<i>N</i> - propylaniline	$\beta$ -isomer	$\gamma$ -isomer
CE5	Control	4.2	2.3	2.0	1.7	23.6	59.4
23	1.0% aniline in (MeO) <sub>3</sub> SiH	5.4	1.1	1.1	0.8	2.1	82.1
24	1.0% aniline in <i>N</i> -allylaniline	<0.1	1.7	1	0.6	0.6	84.1
25	1.0% <i>N,N</i> -dimethylaniline	2.2	0.6	7.0	1.3	16.7	57.6
26	1.0% pyridine	2.6	0.6	1.4	4.6	4.1	60.1
27	500 ppm benzylamine	8.7	0.3	82.6	<0.1	0.1	4.6

28    500 ppm phenothiazine                    6.0        0.5        34.5        1.7        6.4        39.8

<sup>a</sup>All reactions were conducted at 85°C a 10% mole excess of (MeO)<sub>3</sub>SiH versus *N*-allylaniline using 20 ppm Pt [Pt<sub>2</sub>(M<sup>3</sup>M<sup>3</sup>)<sub>3</sub>] and the amine was dissolved in the *N*-allylaniline.

5

### Comparative Example 6

At room temperature, 12.10 g of neat TVC was treated with 0.010 ml of CPA and warmed. At 90°C, the TVC solution was treated with 19.80 g of (MeO)<sub>3</sub>SiH. After the (MeO)<sub>3</sub>SiH addition was completed, the solution was maintained at 90°C for one hour.

10 This solution was analyzed by GC.

(MeO) <sub>3</sub> SiH	(MeO) <sub>4</sub> Si	TVC	Product (Si:TVC ratio)		
			1:1	2:1	3:1
19.0	0.6	34.1	39.5	6.7	0.1

15

### Example 29

At room temperature, 12.10 g of neat TVC was treated with 0.016 g of aniline, 0.010 ml of CPA and warmed. At 90°C, the TVC solution was treated with 19.8 g of (MeO)<sub>3</sub>SiH. After the (MeO)<sub>3</sub>SiH addition was completed, the solution was maintained at

20 90°C for one hour. This solution was analyzed by GC.

(MeO) <sub>3</sub> SiH	(MeO) <sub>4</sub> Si	TVC	Product (Si:TVC ratio)		
			1:1	2:1	3:1
7.5	0.9	0.4	0.6	27.0	63.3

25

### Example 30

At room temperature, 12.10 g of neat TVC was treated with 0.032 g of aniline, 0.010 ml of CPA and warmed. At 90°C, the TVC solution was treated with 19.8 g of

(MeO)<sub>3</sub>SiH. After the (MeO)<sub>3</sub>SiH addition was completed, the solution was maintained at 90°C for two hours. This solution was analyzed by GC.

5	(MeO) <sub>3</sub> SiH	(MeO) <sub>4</sub> Si	TVC	Product (Si:TVC ratio)		
				1:1	2:1	3:1
	4.2	3.3	0.1	0.1	7.3	77.4

For examples 31-36, all reactions were conducted using a 10% molar excess of (MeO)<sub>3</sub>SiH versus TVC, an amine promoter and 10 ppm Pt as a solution of chloroplatinic acid at 90 °C followed by one hour at 90°C after the addition was completed. All solutions were analyzed using gas chromatography. The GC data for Examples 29-36 are summarized in Table 5.

Table 5. GC data for the hydrosilation reaction of TVC.<sup>a</sup>

15	Examples	Amine Promoters	(MeO) <sub>3</sub> SiH	(MeO) <sub>4</sub> Si	TVC	Product (Si:TVC ratio)		
						1:1	2:1	3:1
	CE6	Control	19.0	0.6	34.1	39.5	6.7	0.1
20	29	500 ppm aniline	7.5	0.9	0.4	0.6	25.3	63.3
	30	1000 ppm aniline	7.2	3.6	0.2	0.6	13.1	72.0
	31	500 ppm acetic acid	17.8	1.1	2.6	22.9	42.5	9.6
	32	500 ppm <i>N,N</i> -dimethylaniline	14.5	0.1	36.9	38.1	6.7	0.1
	33	500 ppm phenothiazine	16.1	0.1	59.4	24.6	0.1	0.1
25	34	500 ppm triethylamine	20.5	1.1	58.1	20.2	0.1	0.1
	35	500 ppm pyridine	29.5	1.1	50.7	15.1	1.2	0.1
	36	640 ppm NH[Si(CH <sub>3</sub> ) <sub>3</sub> ] <sub>2</sub>	12.3	1.3	0.5	14.8	51.5	18.6

<sup>a</sup>All reactions were conducted at 90 °C using a 10% mole excess of TMS versus TVC using 10 ppm Pt (CPA).

30

#### Comparative Example 7

At room temperature, 12.10 g of neat TVC was treated with 0.010 ml of CPA and warmed. At 90°C, the methyl-diethoxysilane solution was treated with 19.8 g of TVC

solution containing 0.020 ml of aniline. After the TVC addition was completed, the solution was maintained at 90°C for one hour. This solution was analyzed by GC.

	(EtO) <sub>2</sub> MeSiH	(EtO) <sub>3</sub> MeSi	TVC	Product (Si:TVC ratio)		
5				1:1	2:1	3:1
	7.4	2.6	0.1	0.1	5.4	80.1

For examples 37-38, all reactions were conducted using a 10% molar excess of Me(EtO)<sub>2</sub>SiH versus TVC, an amine promoter and 10 ppm Pt as a solution of chloroplatinic acid at 90°C followed by one hour at 90°C after the addition was completed. All solutions were analyzed using gas chromatography. The GC data for comparative Example 7 and Examples 37-38 are summarized in Table 6.

Table 6. GC data for the hydrosilation reaction of 1,2,4-trivinylcyclohexane (TVC).<sup>a</sup>

Examples	Amine Promoters	(EtO) <sub>2</sub> MeSiH	(EtO) <sub>3</sub> MeSi	TVC	Product (Si:TVC ratio)			
					1:1	2:1	3:1	
20	CE7	Control <sup>o</sup>	7.4	2.6	0.1	0.1	5.4	80.1
	37	1000 ppm aniline	6.7	2.6	3.2	0.1	0.1	87.0
	38	640 ppm NH[Si(CH <sub>3</sub> ) <sub>3</sub> ] <sub>2</sub> in	1.2	1.9	0.5	4.9	15.3	74.5
		TVC						

<sup>a</sup>All reactions were conducted at 90°C using a 10% mole excess of Me(EtO)<sub>2</sub>SiH using 10 ppm Pt (CPA).

#### Comparative Example 8

At room temperature, 6.50 g of neat TVC was treated with 0.007 ml of CPA and warmed. At 90°C, the TVC solution was treated with 24.48 g of (EtO)<sub>3</sub>SiH solution.

After the  $(\text{EtO})_3\text{SiH}$  addition was completed, the solution was maintained at  $90^\circ\text{C}$  for 2 hours. This solution was analyzed by GC.

	$(\text{EtO})_3\text{SiH}$	$(\text{EtO})_4\text{Si}$	TVC	Product (Si:TVC ratio)		
5				1:1	2:1	3:1
	46.8	1.5	2.9	25.6	16.5	0.5

#### Example 39

At room temperature, 6.50 g of neat TVC was treated with 0.007 ml of CPA, 0.031 mL of aniline and warmed. At  $90^\circ\text{C}$ , the  $(\text{EtO})_3\text{SiH}$  solution was treated with 24.48 g of  $(\text{EtO})_3\text{SiH}$ . After the  $(\text{EtO})_3\text{SiH}$  addition was completed, the solution was maintained at  $90^\circ\text{C}$  for 2 hours. This solution was analyzed by GC.

	$(\text{EtO})_3\text{SiH}$	$(\text{EtO})_4\text{Si}$	TVC	Product (Si:TVC ratio)		
15				1:1	2:1	3:1
	16.1	2.4	0.1	0.1	13.2	65.9

#### Comparative Example 9

At room temperature, 31.51 g of neat eugenol methyl ether was treated with 0.010 ml of CPA and warmed. At  $90^\circ\text{C}$ , the eugenol methyl ether solution was treated with 18.0 g of  $(\text{MeO})_3\text{SiH}$ . After the  $(\text{MeO})_3\text{SiH}$  addition was completed, the solution was maintained at  $90^\circ\text{C}$  for one hour. This solution was analyzed by GC.

	$(\text{MeO})_3\text{SiH}$	$(\text{MeO})_4\text{Si}$	Eugenol methyl ether	Eugenol methyl ether Isomer	Product
25	3.9	0.1	25.8	18.5	49.4

#### Comparative Example 10

At room temperature, 31.51 g of neat eugenol methyl ether was treated with 0.025 g of acetic acid, 0.010 ml of CPA and warmed. At 90°C, the eugenol methyl ether solution was treated with 18.0 g of (MeO)<sub>3</sub>SiH. After the (MeO)<sub>3</sub>SiH addition was completed, the solution was maintained at 90°C for one hour. This solution was analyzed by GC.

5

(MeO) <sub>3</sub> SiH	(MeO) <sub>4</sub> Si	Eugenol methyl ether	Eugenol methyl ether Isomer	Product
0.2	0.1	25.8	10.6	47.5

#### Example 40

At room temperature, 31.51 g of neat eugenol methyl ether was treated with 0.025 g of benzylamine, 0.010 ml of CPA and warmed. At 90°C, the eugenol methyl ether solution was treated with 18.0 g of (MeO)<sub>3</sub>SiH. After the (MeO)<sub>3</sub>SiH addition was completed, the solution was maintained at 90°C for one hour. This solution was analyzed by GC.

15

(MeO) <sub>3</sub> SiH	(MeO) <sub>4</sub> Si	Eugenol methyl ether	Eugenol methyl ether isomers	Product
0.3	0.7	1.2	9.7	75.0

For examples 41-44, all reactions were conducted using a 20% molar excess of eugenol methyl ether versus (MeO)<sub>3</sub>SiH, an amine promoter and 10 ppm Pt as a solution of chloroplatinic acid at 90°C followed by one hour at 90°C after the addition was completed. All solutions were analyzed using gas chromatography. The GC data for Comparative Examples 9 and 10 and Examples 40-44 are summarized in Table 7.

Table 7. GC data for the hydrosilation reaction of eugenol methyl ether.<sup>a</sup>

Examples	Amine Promoters	(MeO) <sub>3</sub> SiH	(MeO) <sub>4</sub> Si	Eugenol Methyl Ether	isomers	Product	
5	CE9	Control	3.9	<0.1	25.8	18.5	49.4
	CE10	500 ppm acetic acid	0.2	0.1	25.8	10.6	47.5
	40	500 ppm benzyl amine	0.3	0.7	1.2	9.7	75.0
	41	500 ppm aniline	15.5	0.2	12.0	5.0	66.2
	42	500 ppm phenothiazine	5.2	0.2	41.7	12.3	38.7
10	43	500 ppm 2-methoxybenzylamine	2.2	1.8	30.7	6.0	58.0
	44	500 ppm triethylamine	4.9	<0.1	32.6	6.3	55.1

<sup>a</sup>All reactions were conducted at 90 °C using a 20% mole excess of eugenol methyl ether and 10 ppm Pt (CPA).

15

#### Comparative Example 11

At room temperature, 107.78 g of neat hexadecene was treated with 0.440 ml of CPA and warmed. At 90°C, the hexadecene solution was treated with 48.9 g of (MeO)<sub>3</sub>SiH. After the (MeO)<sub>3</sub>SiH addition was completed, the solution was maintained at 90°C for one hour. This solution was analyzed by GC.

(MeO) <sub>3</sub> SiH	(MeO) <sub>4</sub> Si	Hexadecene	Hexadecene isomers	Product
5.4	0.5	52.2	12.5	23.6

#### 25 Comparative Example 12

At room temperature, 36.6 g of neat hexadecene was treated with 0.112 g of acetic acid, 0.017 ml of CPA and warmed. At 90°C, the hexadecene solution was treated with 22.3 g of (MeO)<sub>3</sub>SiH. After the (MeO)<sub>3</sub>SiH addition was completed, the solution was maintained at 90°C for one hour. This solution was analyzed by GC.

30

(MeO) <sub>3</sub> SiH	(MeO) <sub>4</sub> Si	Hexadecene	Hexadecene isomers	Product
2.9	0.4	10.4	10.8	68.2

## Example 45

5 At room temperature, 36.7 g of neat hexadecene was treated with 0.081 g of aniline, 0.012 ml of Pt<sub>2</sub>(M\*M\*)<sub>3</sub> and warmed. At 90°C, the hexadecene solution was treated with 18.6 g of (MeO)<sub>3</sub>SiH. After the (MeO)<sub>3</sub>SiH addition was completed, the solution was maintained at 90°C for one hour. This solution was analyzed by GC.

10	(MeO) <sub>3</sub> SiH	(MeO) <sub>4</sub> Si	Hexadecene	Hexadecene isomers	Product
	0.2	0.5	0.7	3.9	86.28



## WHAT IS CLAIMED IS:

1. A process comprising reacting hydridoalkoxysilane with (b) olefin in the presence of (c) platinum catalyst and (d) a weakly nucleophilic amine of the formula  $NZ^1Z^2Z^3$  wherein  $Z^1$  is an aryl, alkaryl, or aralkyl group of  $C_6$  to  $C_{20}$  carbon atoms, or an organosilyl substituent of the formula  $SiR_3$ , wherein R is an alkyl of  $C_1$  to  $C_{20}$  or an aryl of  $C_6$  to  $C_{10}$ ,  $Z^2$  is hydrogen, alkyl of  $C_1$  to  $C_{20}$ , an aryl, alkaryl, or aralkyl group of  $C_6$  to  $C_{20}$  carbon atoms, or  $SiR_3$ , wherein R is as previously defined;  $Z^3$  is the same as  $Z^1$  or  $Z^2$ , and, optionally, two of  $Z^1$ ,  $Z^2$  and  $Z^3$  taken together with the nitrogen atom form an aromatic heterocyclic ring.
2. The process according to claim 1 wherein the hydridoalkoxysilane corresponds to the formula  $R_nX_{3-n}SiH$  wherein R is a branched or linear alkyl group of 1 to 18 carbon atoms, a cyclic alkyl group of four to eight carbon atoms or an aryl, alkaryl, or aralkyl group of six to twelve carbon atoms, optionally containing halogen, oxygen, or nitrogen substituents with the proviso that such substituents do not interfere with either hydrosilation or promotion, and X is  $-OR$ , wherein R is as defined above, and n is 0, 1 or 2.
3. The process according to claim 2 wherein n is 0 or 1, and X is selected from the group consisting of ethoxy and methoxy.
4. The process according to claim 2 wherein the hydridoalkoxysilane is selected from the group consisting of trimethoxysilane, triethoxysilane, tri-n-propoxysilane, triisopropoxysilane, methyldimethoxysilane, methyldiethoxysilane, dimethylmethoxysilane, and dimethylethoxysilane.
5. The process according to claim 2 wherein the hydridoalkoxysilane is selected from the group consisting of trimethoxysilane and triethoxysilane.

6. The process according to claim 1 wherein the olefin is selected from the group consisting of alkenes, vinyl-group containing compounds and allylic-group containing compounds.
7. The process according to claim 1 wherein the olefin is selected from the group consisting of ethylene, propylene, butene, pentene, hexene, octene, hexadecene, octadecene, trivinylcyclohexene, 2-methylpropene, 2-methylbutene, diisobutylene, tertiary amylene, 2-butene, vinylcyclohexene monoxide, allyl glycidyl ether, allyl esters, allyl polyethers, allylic tertiary amines, and methallyl derivatives thereof, *N*-allylaniline, *N,N*-dimethallylamine, *N*-ethylmethallylamine, vinyl esters and ethers, vinylsilanes, acrylates and methacrylates.
8. The process according to claim 1 wherein the olefin is selected from the group consisting of ethylene, propylene, butene, pentene, hexene, octene, hexadecene, octadecene, trivinylcyclohexene, 2-methylpropene, 2-methylbutene and diisobutylene.
9. The process according to claim 1 wherein the platinum catalyst is chloroplatinic acid.
10. The process according to claim 1 wherein the weakly nucleophilic amine is used at a level of 25 to 20,000 parts per million by weight of the combined weight of hydridoalkoxysilane and olefin.
11. The process according to claim 1 wherein the weakly nucleophilic amine is selected from the group consisting of aniline, hexamethyldisilazane, phenothiazine, aminonaphthalene, benzylamine, pyridine and derivatives thereof.
12. The process according to claim 1 wherein the olefin is an amino olefin.
13. The process according to claim 1 wherein the olefin is hexadecene.

14. The process according to claim 1 wherein the reaction is conducted at a temperature of from about ambient temperature up to about 150°C and a pressure of about 0.2 to about 2.0 atmospheres.
15. The process according to claim 1 wherein a molar excess of olefin relative to the hydridoalkoxysilane is employed in the reaction.

## INTERNATIONAL SEARCH REPORT

International Application No

PCT/US 02/32826

A. CLASSIFICATION OF SUBJECT MATTER  
 IPC 7 C07F7/18 C07F7/14

According to International Patent Classification (IPC) or to both national classification and IPC

## B. FIELDS SEARCHED

Minimum documentation searched (classification system followed by classification symbols)  
 IPC 7 C07F

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Electronic data base consulted during the international search (name of data base and, where practical, search terms used)

CHEM ABS Data, EPO-Internal

## C. DOCUMENTS CONSIDERED TO BE RELEVANT

Category °	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
P,A	TANG, HUADONG ET AL: "A Novel Aryl Amide-Bridged Ladderlike Polymethylsiloxane Synthesized by a Amido H-Bonding Self-Assembled Template" JOURNAL OF THE AMERICAN CHEMICAL SOCIETY (8 AUGUST 2002), 124(35), 10482-10488 , XP001118909 ISSN: 0002-7863 page 10483, column 2 --- -/--	1-15

Further documents are listed in the continuation of box C.

Patent family members are listed in annex.

° Special categories of cited documents:

- \*A\* document defining the general state of the art which is not considered to be of particular relevance
- \*E\* earlier document but published on or after the international filing date
- \*L\* document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified)
- \*O\* document referring to an oral disclosure, use, exhibition or other means
- \*P\* document published prior to the international filing date but later than the priority date claimed

- \*T\* later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention
- \*X\* document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step when the document is taken alone
- \*Y\* document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art.
- \*&\* document member of the same patent family

Date of the actual completion of the international search

17 January 2003

Date of mailing of the international search report

29/01/2003

Name and mailing address of the ISA

European Patent Office, P.B. 5818 Patentlaan 2  
 NL - 2280 HV Rijswijk  
 Tel. (+31-70) 340-2040, Tx. 31 651 epo nl,  
 Fax: (+31-70) 340-3016

Authorized officer

Elliott, A

## INTERNATIONAL SEARCH REPORT

International Application No

PCT/US 02/32826

C.(Continuation) DOCUMENTS CONSIDERED TO BE RELEVANT		
Category °	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
P,A	LAVASTRE, OLIVIER ET AL: "500 micrometre diameter beads as single reactors to screen organometallic catalysts: discovery of a new supported catalyst for the hydrosilylation of ketones" NEW JOURNAL OF CHEMISTRY (7 JUNE 2002), 26(6), 745-749 , XP009003854 ISSN: 1144-0546 the whole document ---	1-15
P,A	DOHERTY, SIMON ET AL: "Iminophosphines: synthesis, formation of 2,3-dihydro-1H-benzo[1,3]azaphosphol-3-ium salts and N-(pyridin-2-yl)-2-diphenylphosphinylaniline, coordination chemistry and applications in platinum group catalyzed Suzuki coupling reactions and hydrosilylations" JOURNAL OF ORGANOMETALLIC CHEMISTRY (1 MAY 2002), 650(1-2), 231-248 , XP004351225 ISSN: 0022-328X page 245, paragraph 3.3 ---	1-15
P,A	FANG, PENG-FEI ET AL: "Synthesis of polysiloxane-supported fullerene platinum complex and its catalytic behavior" FENZI CUIHUA - JOURNAL OF MOLECULAR CATALYSIS (CHINA) (APRIL 2002), 16(2), 147-149 , XP001121850 ISSN: 1001-3555 abstract ---	1-15
X	US 3 925 434 A (CHUANG VINCENT T) 9 December 1975 (1975-12-09) cited in the application claims 1,3,6-11 ---	1-15
X	DE 11 56 073 B (GOLDSCHMIDT AG TH) 24 October 1963 (1963-10-24) cited in the application the whole document ---	1-15
	-/--	

## INTERNATIONAL SEARCH REPORT

International Application No  
PCT/US 02/32826

C.(Continuation) DOCUMENTS CONSIDERED TO BE RELEVANT		
Category *	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
Y	<p>CHUNYE H ET AL: "Studies on hydrosilylation of allyl chloride catalysed by chloroplatinic acid-amine systems"</p> <p>FENZI CUIHUA - JOURNAL OF MOLECULAR CATALYSIS, GAI KAN BIANJIBU, LANZHOU, CN, vol. 2, no. 1, March 1988 (1988-03), pages 38-43, XP009003505 ISSN: 1001-3555 cited in the application abstract</p>	1-15
X	<p>US 5 191 103 A (MEHTA KUNJ R ET AL) 2 March 1993 (1993-03-02) cited in the application the whole document</p>	1-15
X	<p>US 4 292 434 A (LINDNER TASSILO ET AL) 29 September 1981 (1981-09-29) cited in the application the whole document</p>	1-15
Y	<p>US 5 283 348 A (BANK HOWARD M) 1 February 1994 (1994-02-01) cited in the application the whole document</p>	1-15
A	<p>LUKEVICS E: "Latest research on the hydrosilylation reaction"</p> <p>RUSSIAN CHEMICAL REVIEWS, vol. 46, no. 3, 1977, pages 264-277, XP009003506 ISSN: 0036-021X cited in the application page 264, column 1</p>	1-15

## INTERNATIONAL SEARCH REPORT

Information on patent family members

International Application No

PCT/US 02/32826

Patent document cited in search report		Publication date	Patent family member(s)	Publication date
US 3925434	A	09-12-1975	BE 837763 A1	22-07-1976
			DE 2602171 A1	29-07-1976
			GB 1502221 A	22-02-1978
			JP 1086984 C	26-02-1982
			JP 51095023 A	20-08-1976
			JP 56029873 B	10-07-1981
DE 1156073	B	24-10-1963	NONE	
US 5191103	A	02-03-1993	AT 163935 T	15-03-1998
			DE 69224728 D1	16-04-1998
			DE 69224728 T2	02-07-1998
			EP 0550064 A2	07-07-1993
US 4292434	A	29-09-1981	DE 3000768 A1	16-07-1981
			DE 3166097 D1	25-10-1984
			EP 0032377 A2	22-07-1981
			JP 1324277 C	27-06-1986
			JP 56103187 A	18-08-1981
			JP 60052159 B	18-11-1985
US 5283348	A	01-02-1994	DE 69426173 D1	30-11-2000
			DE 69426173 T2	05-04-2001
			EP 0633265 A1	11-01-1995
			JP 7145177 A	06-06-1995