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(54) Title: GROWTH MEDIA COMPOSITIONS

(57) Abstract: The present disclosure is directed towards aqueous compositions including agricultural water, and a dispersion of acrylic copolymer; growth media compositions including a growth media and solids from the aqueous composition; and methods including combining a dispersion of acrylic copolymer and agricultural water to form an aqueous composition, and applying the aqueous composition to a growth media.

GROWTH MEDIA COMPOSITIONS

Field of Disclosure

[0001] Embodiments of the present disclosure are directed towards growth media compositions and methods of forming growth media compositions, more specifically, embodiments are directed towards growth media compositions including a growth media and solids from a dispersion of acrylic copolymer.

Background

[0002] Growth media may be utilized for plants to germinate, grow, and/or remain viable. There is continued focus in the industry on developing new and improved materials and/or methods that may be utilized with growth media.

Summary

[0003] The present disclosure provides aqueous compositions including agricultural water, and a dispersion of acrylic copolymer.

[0004] The present disclosure provides growth media compositions including a growth media and solids from the aqueous composition.

[0005] The present disclosure provides methods including combining a dispersion of acrylic copolymer and agricultural water to form an aqueous composition, and applying the aqueous composition to a growth media.

[0006] The above summary of the present disclosure is not intended to describe each disclosed embodiment or every implementation of the present disclosure. The description that follows more particularly exemplifies illustrative embodiments. In several places throughout the application, guidance is provided through lists of examples, which examples can be used in various combinations. In each instance, the recited list serves only as a representative group and should not be interpreted as an exclusive list.

Detailed Description

[0007] Growth media compositions and methods of forming growth media compositions are disclosed herein. As used herein "growth media compositions" are

compositions including a growth media and solids from a dispersion of acrylic copolymer.

[0008] Embodiments of the present disclosure can provide an advantageously improved, e.g., increased, water retention, as compared to other compositions. Embodiments of the present disclosure can provide that the dispersion of acrylic copolymer may be contacted, e.g. combined and/or applied to, with a growth media. For example, the dispersion of acrylic copolymer can be combined with agricultural water to form an aqueous composition that can be applied to the growth media. Once applied to the growth media, solids from the dispersion of acrylic copolymer may then help to retain water and/or nutrients in the growth media. Retaining water and/or nutrients in the growth media may help plants germinate, grow, and/or remain viable.

[0009] Embodiments of the present disclosure may advantageously provide an improved, e.g., reduced, growth media hardness, as compared to growth media not including solids from a dispersion of acrylic copolymer. A reduced growth media hardness may provide more favorable conditions for plant root growth, for instance.

[0010] Embodiments of the present disclosure may advantageously provide an improved, e.g., reduced, growth media surface crusting, as compared to growth media not including solids from a dispersion of acrylic copolymer. A reduced surface crusting may provide more favorable conditions for plant germination, growth, and/or viability. Also, a reduced surface crusting may provide for improved growth media maintenance, e.g., tilling.

[0011] Embodiments of the present disclosure may advantageously provide an improved, e.g., increased, holding of growth media fines, as compared to growth media not including solids from a dispersion of acrylic copolymer. Increasing the holding of growth media fines, which may be referred to as fixing growth media particles, may help to reduce erosion.

[0012] The dispersions of acrylic copolymer include a copolymer. Embodiments of the present disclosure provide that the copolymer can be a linear copolymer, a branched copolymer, a crosslinked copolymer, or a combination thereof. Embodiments of the present disclosure provide that the copolymer can be a random copolymer. The copolymer includes structural units of monomers. As used herein, the term "structural

units" refers to the remnant of the indicated monomer; thus a structural unit of ethyl acrylate is illustrated:

where the dotted lines represent points of attachment to the copolymer.

[0013] The copolymer includes structural units derived from an acrylic acid monomer and an acrylic ester monomer. The copolymer can include structural units derived from a macromolecular group. As used herein "acrylic" refers to ethylenically unsaturated, e.g., acrylic includes methacrylic.

[0014] As mentioned, the copolymer includes structural units derived from an acrylic acid monomer. Examples of the acrylic acid monomer include, but are not limited to, itaconic acid, fumaric acid, crotonic acid, acrylic acid, methacrylic acid, maleic acid, acryloxypropionic acid, citraconic acid, and combinations thereof. In one embodiment, a mono-ethylenically unsaturated carboxylic acid monomer may be utilized.

[0015] The copolymer can include from 30 weight percent to 75 weight percent of structural units derived from the acrylic acid monomer, based upon a total weight of the copolymer. All individual values and subranges from 30 weight percent to 75 weight percent are included; for example, the structural units derived from the acrylic acid monomer can be from a lower limit of 30, 32, or 35 weight percent to an upper limit of 75, 70, or 65 weight percent based upon the total weight of the copolymer.

[0016] The copolymer includes structural units derived from an acrylic ester monomer. Examples of the acrylic ester monomer include, but are not limited to, esters of 2-propenoic acid, 2-methylpropenoic acid, and combinations thereof. Embodiments of the present disclosure provide that C₁-C₈ alkyl acrylates and/or C₁-C₈ akyl methacrylates, such as methyl acrylate, ethyl acrylate, butyl acrylate, 2-ethylhexyl acrylate, methyl methacrylate, ethyl methacrylate, butyl methacrylate, and combinations thereof may be utilized.

[0017] The copolymer can include from 25 weight percent to 70 weight percent of structural units derived from the acrylic ester monomer, based upon a total weight of the copolymer. All individual values and subranges from 25 weight percent to 70 weight

percent are included; for example, the structural units derived from the acrylic ester monomer can be from a lower limit of 25, 28, or 30 weight percent to an upper limit of 70, 60, or 50 weight percent based upon the total weight of the copolymer.

[0018] The copolymer can include structural units derived from a vinylically unsaturated monomer. Examples of vinylically unsaturated monomers include, but are not limited to vinyl acetate, styrene, methylstyrene, diisobutylene, vinylpyrrolidone, vinylcaprolactame, and combinations thereof.

[0019] When the vinylically unsaturated monomer is utilized, the copolymer can include up to 20 weight percent of structural units derived from the vinylically unsaturated monomer, based upon the total weight of the copolymer. For instance, the structural units derived from the vinylically unsaturated monomer can be from a lower limit of 0.5, 1.0, or 3.0 weight percent to an upper limit of 20, 17, or 15 weight percent based upon the total weight of the copolymer.

[0020] The copolymer can include structural units, e.g., as a side chain, derived from a macromolecular group. The macromolecular group may be an alkoxylated ester and/or a urethane, for instance. The macromolecular group can include structural units of an ethylenically unsaturated alkyleneoxide macromonomer modified with an alkyl and/or aralkyl group.

[0021] Some embodiments of the present disclosure provide that the macromolecular group may be represented by either of the following structures:

$$[0022] \hspace{1cm} R^a \text{-} (OR^b)_x \text{-} (OR^c)_y \text{-} OC(O) \text{-} CR^d \hspace{--0.08cm} = \hspace{--0.08cm} CH_2$$

or

$$R^{a}$$
- $(OR^{b})_{x}$ - $(OR^{c})_{y}$ - $OC(O)$ - NH - $C(CH_{3})_{2}$ - (Ph) - CR^{d} = CH_{2}

where R^a is a group, such as a C_8 - C_{24} alkyl or aralkyl group; x + y is 10 to 100; R^b and R^c are each independently CH_2CH_2 , $CH_2CH(CH_3)$, or $CH_2CH_2CH_2$; R^d is H or C_1 - C_6 -alkyl; and Ph is a phenylene group. In one or more embodiments, x + y is 10 to 50. In one or more embodiments, x + y is 10 to 100. In one or more embodiments, x + y is 3 to 100. In one or more embodiments, R^b and R^c are both CH_2CH_2 . In one or more embodiments, R^d is methyl.

[0023] Some embodiments of the present disclosure provide that the macromolecular group may be represented by either of the following structures:

[0024] R^{a} -(OCH₂CH₂)_(z)-OC(O)-C(CH₃)=CH₂

or

 $R^{a}-(OCH_{2}CH_{2})_{(z)}-OC(O)-NH-C(CH_{3})_{2}-(Ph)-C(CH_{3})=CH_{2}$

where R^a is a C₈-C₂₄ alkyl or aralkyl group; z is 3 to 50; and Ph is a phenylene group.

As used herein "macromolecular" refers to a group having a molecular weight from 200 daltons to 7,000 daltons. All individual values and subranges from 200 daltons to 7,000 daltons are included; for example, the macromolecular group may have a molecular weight from a lower limit of 200, 225, or 250 daltons to an upper limit of 7,000, 6,750, or 6,500 daltons.

[0026] Some embodiments of the present disclosure provide that the macromolecular group, e.g., a portion thereof, such as Ra-O discussed above, may be hydrophobic. As used herein "hydrophobic" refers to the macromolecular group having a cLogP value from 3.5 to 20.0. The cLogP value may be determined using ChemBioDraw Ultra 13.0, available from PerkinElmer, which uses a chemical fragment algorithm to assess the octanol/water partition coefficient of a molecule and/or fragment based upon its constituent parts. All individual values and subranges from 3.5 to 20.0 are included; for example, the macromolecular group may have a cLogP value from a lower limit of 3.5, 3.7, or 4.0 to an upper limit of 20.0, 19.0, or 18.0.

The copolymer can include from 0.1 weight percent to 15 weight percent of structural units derived from the macromolecular group, based upon a total weight of the copolymer. All individual values and subranges from 0.1 weight percent to 15 weight percent are included; for example, the structural units derived from the macromolecular group can be from a lower limit of 0.1, 0.5, or 1 weight percent to an upper limit of 15, 10, or 5 weight percent based upon the total weight of the copolymer.

Some embodiments of the present disclosure provide that a chain transfer agent may be utilized in forming the dispersion of acrylic copolymer. Examples of the chain transfer agent include, but are not limited to, hydroxyethyl mercaptan, β-mercaptopropionic acid, C₄-C₂₂-alkylmercaptans, such as *n*-dodecylmercaptan, and combinations thereof. The chain transfer agent may be utilized to help control the molecular weight of the copolymer. The chain transfer agent may be utilized from 0.05 weight percent to 5 weight percent based upon a total weight of the acrylic acid monomer

and the acrylic ester monomer utilized to form the copolymer. All individual values and subranges from 0.05 weight percent to 5 weight percent are included; for example, the chain transfer agent may be utilized from a lower limit of 0.05, 0.1, or 0.5 weight percent to an upper limit of 5, 4.5, or 4 weight percent based upon the total weight of the acrylic acid monomer and the acrylic ester monomer utilized to form the copolymer.

[0029] Embodiments of the present disclosure provide that the copolymer has a weight average molecular weight (M_w) from 10,000 to 3,000,000 Daltons. All individual values and subranges from 10,000 to 3,000,000 Daltons are included; for example, the copolymer may have a M_w from a lower limit of 10,000, 25,000, or 50,000 to an upper limit of 3,000,000, 2,500,000, or 2,000,000 Daltons.

[0030] Embodiments of the present disclosure provide that the dispersion of the copolymer has an average particle diameter from 30 nanometers to 10 microns. All individual values and subranges from 30 nanometers to 10 microns are included; for example, the copolymer may have an average particle diameter from a lower limit of 30, 40, or 50 nanometers to an upper limit 10, 5, or 2 microns.

[0031] Some embodiments of the present disclosure provide that a cross-linking agent may be utilized in forming the dispersion of acrylic copolymer. The cross-linking agent can be a diethylenically unsaturated compound. Examples of the cross-linking agent include, but are not limited to, divinyl benzene, allyl methacrylate, diallyl phthalate, trimethylol propane triacrylate, 1,6-hexanediol diacrylate, ethylene glycol diacrylate, ethylene glycol diacrylate, and combinations thereof. Various amounts of the cross-linking agent may be utilized for different applications.

[0032] Embodiments of the present disclosure provide that the dispersion of acrylic copolymer can have a solids content from 10 weight percent to 65 weight percent, based upon a total weight of the dispersion of acrylic copolymer. All individual values and subranges from 10 weight percent to 65 weight percent are included; for example, the dispersion of acrylic copolymer can have a solids content from a lower limit of 10, 12, or 15 weight percent to an upper limit of 65, 55, or 45 weight percent based upon the total weight of the dispersion of acrylic copolymer. One or more embodiments provide that the dispersion of acrylic copolymer is an aqueous dispersion.

[0033] Embodiments of the present disclosure provide that the dispersion of acrylic copolymer can have a pH of 6 or lower. For instance, the dispersion of acrylic copolymer can have a pH from a lower limit of 1.0, 1.5, or 2.0 to an upper limit of 6.0, 5.7, or 5.5.

The dispersion of acrylic copolymer may be obtained commercially or prepared, e.g. by emulsion polymerization. Examples of commercially available dispersions of acrylic copolymers include those under the tradenames RHOPLEXTM, ACRYSOLTM, and ACULYNTM available from The Dow Chemical Company. Specific examples of commercially available dispersions of acrylic copolymers, where the acrylic copolymer does not include structural units derived from a macromolecular group, are RHOPLEXTM ASE-108NP, RHOPLEXTM ASE-60, and ACRYSOLTM ASE-95NP. Specific examples of commercially available dispersions of acrylic copolymers, where the acrylic copolymer does include structural units derived from a macromolecular group, are ACULYNTM 28, ACRYSOLTM DR-110, ACRYSOLTM TT-615, and ACRYSOLTM DR-72.

[0035] The dispersions of acrylic copolymer can be prepared from the above-described monomers by conventional emulsion polymerization, for instance. The emulsion polymerization can be performed at a pH of 5.0 or less using free-radical producing initiators, which may be utilized in an amount from 0.01 percent to 5 percent based on a total weight of the monomers, for instance.

[0036] Free-radical producing initiators include, but are not limited to peroxygen compounds, 2,2'-azobisisobutyronitrile, and high energy radiation sources. Examples of peroxygen compounds include inorganic persulfate compounds such as ammonium persulfate, potassium persulfate, sodium persulfate; peroxides such as hydrogen peroxide; organic hydroperoxides, for example, cumene hydroperoxide, and t-butyl hydroperoxide; organic peroxides, for example, benzoyl peroxide, acetyl peroxide, lauroyl peroxide, peracetic acid, and perbenzoic acid, which may be activated by a water-soluble reducing agent such as ferrous compound or sodium bisulfite.

[0037] The emulsion polymerization can include one or more emulsifiers. Examples of emulsifiers include, but are not limited to anionic surfactants, nonionic surfactants, amphoteric surfactants, and zwitterionic surfactants. Examples of anionic

emulsifiers include alkali metal alkyl aryl sulfonates, alkali metal alkyl sulfates, and sulfonated alkyl esters, among others. Specific examples of emulsifiers are sodium dodecylbenzenesulfonate, sodium disecondary-butylnaphthalene sulfonate, sodium lauryl sulfate, disodium dodecyldiphenyl ether disulfonate, disodium noctadecylsulfosuccinamate and sodium dioctylsulfosuccinate. Examples of nonionic emulsifiers include structures based on polyethylene oxide or oligosaccharides hydrophilic heads, among others. Various amounts or emulsifier may be utilized for different applications.

[0038] Optionally, other ingredients known for emulsion polymerizations may be utilized, such as chelating agents, buffering agents, inorganic salts, and pH adjusting agents, among others. Various amounts other ingredients may be utilized for different applications.

[0039] As mentioned, the dispersion of acrylic copolymer, e.g., solids from a dispersion of acrylic copolymer, can be combined with agricultural water to form an aqueous composition that can be applied to the growth media. As used herein "agricultural water" refers to water that is utilized for plant growth, e.g., plants that provide food and/or fiber. Combining the dispersion of acrylic copolymer and the agricultural water may be performed in any suitable manner, e.g., the dispersion of acrylic copolymer and the agricultural water may be added to a vessel and agitated therein.

[0040] The dispersion of acrylic copolymer and the agricultural water may be combined such that a concentration of solids from the dispersion of acrylic copolymer is from 50 parts per million to 100,000 parts per million based upon a total weight of the aqueous composition. All individual values and subranges from 50 parts per million to 100,000 are included; for example, the dispersion of acrylic copolymer and agricultural water may be combined such that the concentration of solids from the dispersion of acrylic copolymer is from a lower limit of 50, 100, 250, 500, 1,000, 2,000, 5,000, or 10,000 parts per million to an upper limit of 100,000, 75,000, 50,000, 35,000, 25,000, 20,000, or 15,000 parts per million based upon the total weight of the aqueous composition.

[0041] Embodiments of the present disclosure provide that the acrylic copolymer can be insoluble in the aqueous composition, e.g., at natural pH. While not wishing to be bound to theory, it is believed that the acrylic copolymer is insoluble in the aqueous composition due to a pH of the aqueous composition. As used herein "insoluble in the aqueous composition" indicates that the acrylic copolymer does not appreciably swell with water, e.g., the acrylic copolymer does not form a gel with the water and/or dissolve. Because the acrylic copolymer is insoluble in the aqueous composition, the aqueous composition advantageously can be stored and/or applied to the growth media at a suitable viscosity, in contrast to some other polymeric materials.

[0042] The aqueous composition can have a pH from 1.0 to 6.0. All individual values and subranges from 1.0 to 6.0 are included; for example, the aqueous composition can have a pH from a lower limit of 1.0, 1.5, or 2.0 to an upper limit of 6.0, 5.7, or 5.5.

The aqueous compositions can include an adjunct material. Examples of adjunct materials include, but are not limited to fertilizers, insecticides, herbicides, and/or stabilizers, among others. Various amounts of adjunct material may be utilized for different application. The aqueous compositions can have one or more characteristics such as salinity, electrical conductivity, alkalinity, iron content, and/or pH, among others making the aqueous composition suitable for utilization with a growth media for plants that provide food and/or fiber.

As mentioned, the aqueous composition can be applied to a growth media, e.g., to form a growth media composition. The growth media compositions disclosed herein include a growth media. As used herein, "growth media" refers to material that is utilized for plant growth, e.g., plants that provide food and/or fiber. The growth media may be an agricultural soil, for instance. The growth media may include sand, silt, clay, and/or organic material, for example. Various amounts of sand, silt, clay, and/or organic material may be utilized for different applications.

[0045] The growth media compositions can have a concentration of solids from the dispersion of acrylic copolymer that is from 10 parts per million to 50,000 parts per million based upon a total weight of the growth media. All individual values and subranges from 10 parts per million to 50,000 parts per million are included; for example, the growth media composition can include from a lower limit or 10, 20, 25, 50, 100, or

200 parts per million to an upper limit of 50,000, 45,500, 40,000, 35,500, 25,000, 15,000, or 5,000 parts per million of solids from the dispersion of acrylic copolymer based upon the total weight of the growth media.

[0046] The growth media compositions can include an adjunct material. Examples of adjunct materials include, but are not limited to fertilizers, insecticides, herbicides, pH adjusting agents, and/or stabilizers, among others. Various amounts of adjunct material may be utilized for different applications.

[0047] While not wishing to be bound to theory, it is believed that one or more components of the growth media compositions may neutralize, e.g., increase the pH, components of the dispersion of acrylic copolymer that are applied to the growth media. As the pH is increased the acrylic copolymer can advantageously become water soluble, e.g., the acrylic copolymer swells with water. Once applied to the growth media, solids from the dispersion of acrylic copolymer can help to retain water and/or nutrients in the growth media; provide an improved growth media hardness; provide an improved growth media surface crusting; and/or provide an improved holding of growth media fines. In contrast to other polymeric materials that are applied to agricultural soils, the acrylic copolymer, as discussed herein, has a water insoluble state and a water soluble state, which are believed to be pH dependent.

The growth media composition can have a pH from 6.0 to 11.0. All individual values and subranges from 6.0 to 11.0 are included; for example, the growth media composition can have a pH from a lower limit of 6.0, 6.5, or 7.0 to an upper limit of 11.0, 10.0, 9.0, or 8.0. One or more embodiments of the present disclosure provide that growth media composition can have a pH as low as 5.5.

[0049] Some embodiments of the present disclosure are directed toward methods. The methods disclosed herein can include combining a dispersion of acrylic copolymer and agricultural water to form an aqueous composition and applying the aqueous composition to a growth media. Applying the aqueous composition to the growth media can form a growth media composition, i.e. the growth media compositions disclosed herein. While not wishing to be bound to theory, because the acrylic copolymers, as discussed herein, are in the water insoluble state in the aqueous composition, the acrylic copolymers will not appreciably swell. This lack of swelling can allow the aqueous

composition to have little or no significant viscosity increase, e.g., the viscosity of the aqueous compositions may remain essentially constant. Because there is no significant viscosity increase, the aqueous compositions may be applied to the growth media by processes, e.g., spraying, that could not be effectively utilized with some other polymers that swell in water.

[0050] Some embodiments of the present disclosure provide that applying the aqueous composition to a growth media can be a surface treatment. Examples of surface treatments include, but are not limited to, spraying, sprinkling, and/or pouring the aqueous composition on to the growth media. For instance, the aqueous composition may be sprayed on to a surface of the growth media, such as spraying an agricultural field.

[0051] Some embodiments of the present disclosure provide that applying the aqueous composition to a growth media can be a bulk treatment. Examples of a bulk treatments include, but are not limited to, mixing the aqueous composition with the growth media, and injecting of the aqueous composition beneath the surface of the growth media.

[0052] The methods disclosed herein may be directed toward increasing water retention, reducing growth media hardness, reducing growth media surface crusting, and/or increasing holding of growth media fines, among others.

EXAMPLES

[0053] Growth media (sandy loam M-SL-PF; 63 wt% sand, 18 wt% silt, and 19 wt% clay; obtained from AGVISE Laboratories). The growth media had an organic matter of approximately 2.8 to 3.2 wt%, a cation exchange capacity (CEC) between 20.9 and 24.7, a field capacity between 22.7 and 30.1, and pH between 6.6 and 8.0.

[0054] Calcium stock (0.04 M) was formed by mixing calcium chloride (2.21 grams; certified ACS; obtained from Acros Organics) and deionized water (497.7 grams). Magnesium stock (0.04 M) was formed by mixing magnesium chloride hexahydrate crystals (4.05 grams; certified ACS; obtained from Fischer Chemical) and deionized water (495.7 grams). Hard water was formed by mixing the calcium stock (68.5 grams), the magnesium stock (17.0 grams), and MILLI-Q® water (910.6 grams).

[0055] Polypropylene cartridges (60 mL; obtained from Applied Separations, catalog #2446) were each cut to a height of 54 mm. Polypropylene centrifuge tubes (50 mL; obtained from Fisher Scientific, catalog #06-443-18) were each cut at the preprinted 27.5 mL mark.

[0056] The respective cut cartridges were weighed and growth media was added to each respective cut cartridge; the growth media was gently pressed down in each of the cut cartridges to a height of approximately 28.5 mm.

[0057] The cut centrifuge tubes were placed into a holder. The cut cartridges containing the growth media were respectively placed on top of the cut centrifuge tubes.

Comparative Example A, for cycle 1, was prepared as follows. Hard water was added to a cartridge including the growth media to form Comparative Example A. Comparative Example A was weighed to determine an initial weight. Thereafter, Comparative Example A was centrifuged at 1200 rpm for 4 hours and weighed again. Thereafter, Comparative Example A was centrifuged at 4200 rpm for 3 hours and weighed again. Cycle 2 followed cycle 1, where MILLI-Q® water (9 mL) was added to the resultant cartridge and contents from cycle 1. Thereafter, the cycle 2 cartridge and contents were centrifuged and weighed as in cycle 1. Cycle 3 followed cycle 2, where MILLI-Q® water (9 mL) was added to the resultant cartridge and contents from cycle 2. Thereafter, the cycle 3 cartridge and contents were centrifuged and weighed as in cycle 2. Following the weighing the samples after Cycle 3, they were dried for approximately 24 hours at 105 °C, and then weighed again to determine the dry weight. Four replicates were prepared for each sample. Weights of any additive, cartridges, growth media, and hard water are reported in Table 1.

Example 1, an aqueous composition including agricultural water and a dispersion of acrylic copolymer, was prepared as follows. Hard water, e.g., agricultural water, and ACULYNTM 28 (dispersion of acrylic copolymer; 20 weight percent solids; obtained from The Dow Chemical Company) were combined in a container to form Example 1. Examples 2-7, aqueous compositions including agricultural water and a dispersion of acrylic copolymer, were prepared as Example 1 with the change that RHOPLEXTM ASE-108NP (dispersion of acrylic copolymer; 18 weight percent solids; obtained from The Dow Chemical Company); RHOPLEXTM ASE-60 (dispersion of

acrylic copolymer; 28 weight percent solids; obtained from The Dow Chemical Company); ACRYSOLTM ASE-95NP (dispersion of acrylic copolymer; 18 weight percent solids; obtained from The Dow Chemical Company); ACRYSOLTM DR-110 (dispersion of acrylic copolymer; 30 weight percent solids; obtained from The Dow Chemical Company); ACRYSOLTM TT-615 (dispersion of acrylic copolymer; 30 weight percent solids; obtained from The Dow Chemical Company); and ACRYSOLTM DR-72 (dispersion of acrylic copolymer; 30 weight percent solids; obtained from The Dow Chemical Company) were respectively used in place of ACULYNTM 28. Solids from the dispersion of acrylic copolymer based upon a total weight of the respective Example and pH are reported in Table 1.

Table 1

	Dispersion of acrylic copolymer (grams)	PPM of solids from the dispersion of acrylic copolymer based upon a total weight of the Example	pН
Example 1	0.0300 (ACULYN TM 28)	5,000	4.0
Example 2	0.0300 (RHOPLEX TM ASE-108NP)	5,000	4.7
Example 3	0.0301 (RHOPLEX TM ASE-60)	5,000	5.0
Example 4	0.0300 (ACRYSOL TM ASE-95NP)	5,000	4.3
Example 5	0.0300 (ACRYSOL TM DR-110)	5,000	5.0
Example 6	0.0300 (ACRYSOL TM TT-615)	5,000	4.5
Example 7	0.0301 (ACRYSOL TM DR-72)	5,000	5.0

[0060] Example 8, a composition including the growth media and solids from a dispersion of acrylic copolymer, was prepared as follows for cycle 1. Example 1 was added to a cartridge including the growth media to form Example 8. Example 8 was weighed to determine an initial weight. Thereafter, Example 8 was centrifuged at 1200 rpm for 4 hours. Thereafter, Example 8 was centrifuged at 4200 rpm for 3 hours. Cycle 2 followed cycle 1, where MILLI-Q® water (9 mL) was added to the resultant cartridge and contents from cycle 1. Thereafter, the cycle 2 cartridge and contents were centrifuged and weighed as in cycle 1. Cycle 3 followed cycle 2, where MILLI-Q® water (9 mL) was added to the resultant cartridge and contents from cycle 2. Thereafter, the cycle 3 cartridge and contents were centrifuged and weighed as in cycle 2. Afterwards the samples were dried for approximately for 24 hours at 150 °C and then weighed again to determine the weight of the solids, which was used to determine the weight of the water left in the cartridge after each cycle. Four replicates were performed for each sample, which are labeled as Run 1-4. Weights of any additive, cartridges, growth media, and hard water are reported in Table 2.

[0061] Examples 9-14 were prepared as Example 8, with the change that Examples 2-7 were respectively used in place of Example 1. Weights of any additive, cartridges, growth media, and hard water are reported in Table 2. The pH of each of Examples 8-14 and Comparative Example A were in the range from 7.0 to 7.5.

Table 2

	Dispersion of acrylic copolymer (grams)	Cartridge (grams)	Growth media (grams)	Hard Water (grams)
Comparative Example A Replicate 1 Cycle 1	-	9.7959	12.9722	6.0024
Comparative Example A Replicate 2 Cycle 1	-	9.7498	12.9687	6.0018
Comparative Example A Replicate 3 Cycle 1	-	8.0295	12.9684	6.0003
Comparative Example A Replicate 4 Cycle 1	-	9.8770	12.9655	6.0048

	0.0500			
Example 8	0.0300	0.0541	12.0720	6.0020
Replicate 1	(ACULYN TM 28)	8.0541	12.9729	6.0020
Cycle 1	0.0201			
Example 8	0.0301	0.0607	12.0677	6.0177
Replicate 2	(ACULYN TM 28)	8.0605	12.9677	6.0155
Cycle 1				
Example 8	0.0300			
Replicate 3	(ACULYNTM 28)	9.7614	12.9715	6.0043
Cycle 1				
Example 8	0.0300			
Replicate 4	(ACULYNTM 28)	9.8509	12.9729	6.0025
Cycle 1				
Example 9	0.0300			
Replicate 1	(RHOPLEX TM	9.8667	12.9719	6.0006
Cycle 1	ASE-108NP)			
Example 9	0.0300			
Replicate 2	(RHOPLEX TM	7.9534	12.9736	6.0001
Cycle 1	ASE-108NP)			
Example 9	0.0300			
Replicate 3	(RHOPLEX TM	8.0694	12.9727	6.0018
Cycle 1	ASE-108NP)			
Example 9	0.0300			
Replicate 4	(RHOPLEX TM	9.8292	12.9691	6.0015
Cycle 1	ASE-108NP)	3.0232	12,5051	0,0010
Example 10	0.0301			
Replicate 1	(RHOPLEX TM	9.8355	12.9681	6.0118
Cycle 1	ASE-60)	7.0333	12.7001	0.0110
Example 10	0.0300			
Replicate 2	(RHOPLEX TM	9.7908	12.9685	5.9996
Cycle 1	ASE-60)	2.7700	12.7003	3.7770
Example 10	0.0300			
Replicate 3	(RHOPLEX TM	9.8512	12.9698	6.0043
Cycle 1	ASE-60)	7.0312	12.7070	0.0043
Example 10	0.0300			
Replicate 4	(RHOPLEX TM	8.0523	12.9731	5.9990
Cycle 1	ASE-60)	0.0525	12.9/31	3.3330
Example 11	0.0300			
Replicate 1	(ACRYSOL TM	9.7859	12.9722	6.0022
Cycle 1		9.7039	12.9/22	0.0022
	ASE-95NP)			
Example 11	0.0300 (ACRYSOL TM	0.6160	12.9697	6.0011
Replicate 2	`	9.6169	12.909/	0.0011
Cycle 1	ASE-95NP)			
Example 11	0.0300	0.0210	12.0664	6,000
Replicate 3	(ACRYSOL TM	8.0210	12.9664	6.0008
Cycle 1	ASE-95NP)			
Example 11	0.0300	0.0505	12.0070	60010
Replicate 4	(ACRYSOL TM	8.0795	12.9670	6.0019
Cycle 1	ASE-95NP)			
Example 12	0.0300	0.0511	10.0710	
Replicate 1	(ACRYSOL TM	9.8714	12.9718	6.0072
Cycle 1	DR-110)			
Example 12	0.0300			
Replicate 2	(ACRYSOL TM	9.5866	12.9675	6.0000
Cycle 1	DR-110)			
Example 12	0.0300	9.6362	12.9736	6.0023

Replicate 3	(ACRYSOL TM			
Cycle 1	DR-110)			
Example 12	0.0300			
Replicate 4	$(ACRYSOL^{TM})$	9.8608	12.9687	5.9997
Cycle 1	DR-110)			
Example 13	0.0300			
Replicate 1	$(ACRYSOL^{TM})$	8.1029	12.9675	6.0028
Cycle 1	TT-615)			
Example 13	0.0300			
Replicate 2	$(ACRYSOL^{TM})$	9.7021	12.9701	5.9998
Cycle 1	TT-615)			
Example 13	0.0300			
Replicate 3	$(ACRYSOL^{TM})$	9.8199	12.9678	6.0034
Cycle 1	TT-615)			
Example 13	0.0300			
Replicate 4	$(ACRYSOL^{TM})$	9.8399	12.9729	6.0057
Cycle 1	TT-615)			
Example 14	0.0301			
Replicate 1	$(ACRYSOL^{TM})$	9.8758	12.9726	6.0115
Cycle 1	DR-72)			
Example 14	0.0301			
Replicate 2	$(ACRYSOL^{TM})$	8.0207	12.9685	6.0133
Cycle 1	DR-72)			
Example 14	0.0300			
Replicate 3	$(ACRYSOL^{TM})$	9.7660	12.9740	6.0012
Cycle 1	DR-72)			
Example 14	0.0300			
Run 4	$(ACRYSOL^{TM})$	9.8003	12.9662	6.0027
Cycle 1	DR-72)			

[0062] For each replicate of each cycle performed, an amount of water retained following centrifuging was determined by weighing the sample after the centrifuge, oven drying, and weighing again. Alternatively, the amount of water can be determined by measuring the moisture of the soil with a moisture analyzer and subtracting the weight of the container, calculated dry soil weight and added additive weight from the post-centrifuge weight. For each replicate of each cycle performed for Examples 8-14 and Comparative Example A, an amount of water retained and the change in the percentage of water retained vs Comparative Example A was calculated. The results are reported in Table 3.

Table 3

	Cyc	cle 1	C	ycle 2	C	ycle 3
		Change in		Change in		Change in
	Water	percentage	Water	percentage	Water	percentage
1	retained	of water	retained	of water	retained	of water
	(grams)	retained vs	(grams)	retained vs	(grams)	retained vs
		Comparative		Comparative		Comparative

		Example A		Example A		Example A
Comparative		•		•		•
Example A	2.7921	-	2.8538	-	2.9066	-
Replicate 1						
Comparative						
Example A	2.7893	-	2.8554	-	2.9102	-
Replicate 2						
Comparative						
Example A	2.7794	_	2.8406	_	2.9052	_
Replicate 3	2.775		2.0.00		2.,, 002	
Comparative						
Example A	2.8091	_	2.8722	_	2.9246	_
Replicate 4	2.0071		2.0722		2.7210	
Example 8						
Replicate 1	3.0806	+10.2830	3.1006	+8.5491	3.1340	+7.6025
Example 8	3.0992	+10.9940	3.1131	+9.0310	3.1450	+8.0241
Replicate 2						
Example 8	3.1094	+11.3249	3.1379	+9.8657	3.1547	+8.3238
Replicate 3	3.1071	11,3219	3.1377	19,0027	3,10 17	10.5250
Example 8	3.1466	+12.6454	3.1796	+11.3144	3.2031	+9.9746
Replicate 4	3.1400	112.0434	3.1770	111.5144	3.2031	17.7740
Example 9	2 8080	±2.7522	2.9426	±2 0229	2.0627	±1.7272
Replicate 1	2.8980	+3.7523	2.9420	+3.0238	2.9627	+1.7272
Example 9	0.0070	. 2 2 7 0 0	2 02 12	. 2 7106	2 0 5 5 5	
Replicate 2	2.8872	+3.3509	2.9343	+2.7186	2.9577	+1.5411
Example 9						
Replicate 3	2.9036	+3.9471	2.9406	+2.9482	2.9694	+1.9517
Example 9						
Replicate 4	2.9237	+4.6971	2.9595	+3.6400	2.9799	+2.3419
Example 10						
Replicate 1	2.9275	+4.8409	2.9564	+3.5390	2.9761	+2.2189
Example 10						
	2.9263	+4.7918	2.9567	+3.5436	2.9734	+2.1203
Replicate 2						
Example 10	2.9193	+4.5336	2.9494	+3.2804	2.9665	+1.8759
Replicate 3						
Example 10	2.8860	+3.3129	2.9098	+1.8658	2.9415	+0.9898
Replicate 4		0.0123	2.7070	2,000		0,7070
Example 11	2.9339	+5.0366	2.9826	+4.4233	3.0067	+3.2370
Replicate 1	2.7337	15.0500	2.7020	1.1233	3.0007	13.2370
Example 11	2.9293	+4.8924	2.9708	+4.0305	2.9904	+2.6974
Replicate 2	2.7273	14.6724	2.7700	14.0303	2.5504	12.07/4
Example 11	2.0206	+4.9295	2.0790	14.2000	2 0012	12.0076
Replicate 3	2.9296	+4.9293	2.9780	+4.3088	3.0013	+3.0976
Example 11	2 0206	. 7 2 4 6 7	2 00 17	. 4 5202	2 0 1 4 7	. 2 5 1 5 0
Replicate 4	2.9386	+5.2465	2.9847	+4.5382	3.0145	+3.5459
Example 12						
Replicate 1	2.9734	+6.4543	3.0040	+5.1761	3.0255	+3.8860
Example 12						
Replicate 2	2.9727	+6.4632	3.0004	+5.0836	3.0203	+3.7406
Example 12						
	2.9094	+4.1490	2.9619	+3.6881	2.9905	+2.6704
Replicate 3						
Example 12	2.9151	+4.3903	2.9716	+4.0649	2.9991	+3.0025
Replicate 4						
Example 13	2.8937	+3.6335	2.9322	+2.6945	2.9652	+1.8476
Replicate 1						3.5170

Example 13 Replicate 2	2.9341	+5.0601	2.9787	+4.3030	3.0014	+3.0711
Example 13 Replicate 3	2.9437	+5.4234	2.9845	+4.5255	3.0055	+3.2310
Example 13 Replicate 4	2.9354	+5.0844	2.9613	+3.6717	2.9870	+2.5549
Example 14 Replicate 1	2.9358	+5.1019	2.9738	+4.1125	2.9994	+2.9837
Example 14 Replicate 2	2.8989	+3.8117	2.9281	+2.5431	2.9710	+2.0390
Example 14 Replicate 3	2.9746	+6.4784	3.0191	+5.6860	3.0427	+4.4581
Example 14 Replicate 4	2.9683	+6.3157	3.0187	+5.7345	3.0423	+4.5062

[0063] The data of Table 3 illustrates that for replicates 1-4 of cycles 1-3, Examples 8-14 advantageously had an improved, i.e. increased, water retention as compared to Comparative Example A.

[0064] Growth media hardness. Comparative Example B was prepared as follows. Growth media (50 g), described above, was added to cylindrical container that allowed for liquid drainage. Deionized water (20 mL) was added to the growth media; one week later, deionized water (100 mL) was added to the growth media, excess water drained from the container and collected in a collection container; after 24 hours more deionized water (100 mL) was added to the growth media, excess water drained from the container and collected in the collection container; after another 24 hours more deionized water (100 mL) was added to the growth media, excess water drained from the container and collected in the collection container; thereafter the growth media was dried at ambient conditions for 1 week. After 1 week, growth media hardness was measured with a soil pocket penetrometer obtained from Certified Material Testing Products. The results are reported in Table 4.

[0065] Comparative Example C was prepared as Comparative Example B with the change that RESERVOIR® DG (soil additive; obtained from HELENA®) was added (500 ppm based upon a total weight of the growth media and 20 mL of deionized water; the RESERVOIR® DG was added as a solid followed by the 20 mL of deionized water) to the growth media one week prior any addition of further deionized water to the growth media. The results are reported in Table 4.

[0066] Examples 15-16 were prepared as Comparative Example B with the change that ACULYNTM 28 (20 mL of ACULYNTM 28 and deionized water; Example 15: 500 ppm ACULYNTM 28 based upon a total weight of the ACULYNTM 28 and deionized water; Example 16: 1000 ppm ACULYNTM 28 based upon a total weight of the ACULYNTM 28 and deionized water) was utilized rather than the polyacrylamide. The results are reported in Table 4.

Table 4

	Additive	Growth media hardness (kg/cm²)
Example 15	ACULYN™ 28 (500 ppm)	0.75
Example 16	ACULYN™ 28 (1000 ppm)	0.75
Comparative Example B	Deionized water only	2.00
Comparative Example C	RESERVOIR® DG (500 ppm)	1.25

[0067] The data of Table 4 illustrates that each of Example 15 and Example 16 provide improved, i.e., reduced, growth media hardness as compared to both Comparative Example B and Comparative Example C.

[0068] Holding of growth media fines was determined by visual inspection of drainage collected in the respective collection containers. The results are reported in Table 5.

Table 5

	Additive	Visual Inspection of collected drainage
Example 15	ACULYN™ 28 (500 ppm)	Clear
Example 16	ACULYN™ 28 (1000 ppm)	Clear
Comparative Example B	Deionized water only	Yellowish, Cloudy

Comparative Example C	RESERVOIR® DG	Yellowish, Cloudy
	(500 ppm)	

[0069] The data of Table 5 illustrates that each of Example 15 and Example 16 provide improved, i.e., increased, holding of growth media fines as compared to both Comparative Example B and Comparative Example C. The cloudy appearance of collected drainage for Comparative Example B and Comparative Example C indicates that fines, at a concentration to be visually perceived, were present in the respective collected drainages. The clear appearance of collected drainage for Example 15 and Example 16 indicates that fines, at a concentration to be visually perceived, were not present in the respective collected drainages.

Claims

What is claimed:

An aqueous composition comprising:
 agricultural water; and
 solids from a dispersion of acrylic copolymer.

- 2. The aqueous composition of claim 1, wherein the acrylic copolymer includes structural units derived from an acrylic acid monomer and structural units derived from acrylic ester monomer.
- 3. The aqueous composition of claim 2, wherein the acrylic copolymer includes structural units derived from a macromolecular group.
- 4. The aqueous composition of any one of claims 1-3, wherein the dispersion of acrylic copolymer is from 10 weight percent to 55 weight percent solids based upon a total weight of the dispersion of acrylic copolymer.
- 5. The aqueous composition of any one of claims 1-4, wherein solids from the dispersion of acrylic copolymer is from 50 parts per million to 100,000 parts per million based upon a total weight of the aqueous composition.
- A growth media composition comprising:a growth media; andsolids from the aqueous composition of any one of claims 1-5.
- 7. The growth media composition of claim 6, wherein a concentration of solids from the dispersion of acrylic copolymer is from 10 parts per million to 50,000 parts per million based upon a total weight of the growth media.
- 8. A method comprising:

combining solids from a dispersion of acrylic copolymer and agricultural water to form an aqueous composition; and

applying the aqueous composition to a growth media.

- 9. The method of claim 8, wherein applying the aqueous composition to the growth media comprises a surface treatment.
- 10. The method of claim 8, wherein applying the aqueous composition to the growth media comprises a bulk treatment.

International application No PCT/US2018/034044

A. CLASSIFICATION OF SUBJECT MATTER INV. A01N25/04 A01N25/10

C09K17/22

A01G24/35

ADD.

According to International Patent Classification (IPC) or to both national classification and IPC

B. FIELDS SEARCHED

Minimum documentation searched (classification system followed by classification symbols)

A01N C09K A01G

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Electronic data base consulted during the international search (name of data base and, where practicable, search terms used)

EPO-Internal, CHEM ABS Data, WPI Data

C. DOCUM	C. DOCUMENTS CONSIDERED TO BE RELEVANT				
Category*	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.			
Υ	US 4 540 427 A (HELBLING GOTTFRIED [CH]) 10 September 1985 (1985-09-10) examples 1-3	1-10			
х	EP 0 101 253 A2 (ALLIED COLLOIDS LTD [GB]) 22 February 1984 (1984-02-22)	1,5-7			
Υ	page 4, line 10 - line 14 page 6, line 29 - line 32 examples 1-22	1-10			

Further documents are listed in the continuation of Box C.	X See patent family annex.
"A" document defining the general state of the art which is not considered to be of particular relevance "E" earlier application or patent but published on or after the international filing date "L" document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified) "O" document referring to an oral disclosure, use, exhibition or other means "P" document published prior to the international filing date but later than the priority date claimed	"T" later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention "X" document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step when the document is taken alone "Y" document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art "&" document member of the same patent family
Date of the actual completion of the international search	Date of mailing of the international search report
8 August 2018	21/08/2018
Name and mailing address of the ISA/	Authorized officer
European Patent Office, P.B. 5818 Patentlaan 2 NL - 2280 HV Rijswijk Tel. (+31-70) 340-2040, Fax: (+31-70) 340-3016	Habermann, Jörg

1

International application No
PCT/US2018/034044

	tion). DOCUMENTS CONSIDERED TO BE RELEVANT	
Category*	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
Υ	MICHAEL S. JOHNSON: "The effects of gel-forming polyacrylamides on moisture storage in sandy soils", JOURNAL OF THE SCIENCE OF FOOD AND AGRICULTURE, vol. 35, no. 11, 1 November 1984 (1984-11-01), pages 1196-1200, XP55497403, GB ISSN: 0022-5142, DOI: 10.1002/jsfa.2740351110 the whole document	1-10
X	RABAT NURUL EKMI ET AL: "Effect of Different Monomers on Water Retention Properties of Slow Release Fertilizer Hydrogel", PROCEDIA ENGINEERING, ELSEVIER, AMSTERDAM, NL, vol. 148, 12 July 2016 (2016-07-12), pages 201-207, XP029637788, ISSN: 1877-7058, DOI:	1,5-7
Υ	10.1016/J.PROENG.2016.06.573 abstract table 1 figure 3 paragraphs 3.2, 3.3, 4	1-10
X	SHAUKAT ALI SHAHID ET AL: "Improvement in the Water Retention Characteristics of Sandy Loam Soil Using a Newly Synthesized Poly(acrylamide-co-acrylic Acid)/AlZnFe204 Superabsorbent Hydrogel Nanocomposite Material", MOLECULES, vol. 17, no. 8, 3 August 2012 (2012-08-03), pages 9397-9412, XP55497400,	1,5-7
Υ	DOI: 10.3390/molecules17089397 figure 6 table 2 paragraph 4	1-10
X Y	GB 1 591 415 A (UNION CARBIDE CORP) 24 June 1981 (1981-06-24) page 1, line 29 - line 31 page 2, line 29 - line 31 page 14, line 1 - line 3 page 16, line 1 - line 3 examples 1-17 tables I, II	1,5-7 1-10

International application No
PCT/US2018/034044

WO 83/00498 A1 (UNILEVER PLC [GB]; UNILEVER NV [NL]) 17 February 1983 (1983-02-17) page 8, line 3 - line 15 claims 1, 7, 9, 10 EP 0 122 797 A1 (MITSUI PETROCHEMICAL IND [JP]) 24 October 1984 (1984-10-24) page 1, line 4 - line 14 examples 1-29 EP 1 247 444 A2 (DEGUSSA [DE]) 9 October 2002 (2002-10-09)	1,5-7 1-10 1 1-10
17 February 1983 (1983-02-17) page 8, line 3 - line 15 claims 1, 7, 9, 10 EP 0 122 797 A1 (MITSUI PETROCHEMICAL IND [JP]) 24 October 1984 (1984-10-24) page 1, line 4 - line 14 examples 1-29 EP 1 247 444 A2 (DEGUSSA [DE]) 9 October 2002 (2002-10-09)	1
[JP]) 24 October 1984 (1984-10-24) page 1, line 4 - line 14 examples 1-29 EP 1 247 444 A2 (DEGUSSA [DE]) 9 October 2002 (2002-10-09)	
page 1, line 4 - line 14 examples 1-29 EP 1 247 444 A2 (DEGUSSA [DE]) 9 October 2002 (2002-10-09)	1-10
9 October 2002 (2002-10-09)	
3 00000C1 E00E (E00E 10 03)	1,5-10
paragraphs [0021], [0029], [0031] 	1-10
WO 03/000621 A1 (PEPPMOELLER REINMAR [DE]; FABRITZ GERHARD [DE])	1,6
examples 1-5 claim 3 page 7, line 19 - line 29	1-10
WO 2006/026406 A2 (ABSORBENT TECHNOLOGIES INC [US]; D2 POLYMER TECHNOLOGIES INC	1,5-7
[05]; MCK) 9 March 2006 (2006-03-09) paragraphs [0061] - [0070] table 1 claims 1, 3, 6, 24	1-10
DATABASE WPI Week 201561 Thomson Scientific, London, GB; AN 2015-514813 XP002783643, & KR 101 547 306 B1 (JUNG H K)	1,5-7
abstract	1-10
DATABASE WPI Week 201640 Thomson Scientific, London, GB; AN 2016-26047V XP002783644, & CN 105 503 477 A (ZHONGSHAN LVLANG AUXILIARIES CO LTD) 20 April 2016 (2016-04-20)	1,5-8
abstract	1-10
	FABRITZ GERHARD [DE]) 3 January 2003 (2003-01-03) examples 1-5 claim 3 page 7, line 19 - line 29 WO 2006/026406 A2 (ABSORBENT TECHNOLOGIES INC [US]; D2 POLYMER TECHNOLOGIES INC [US]; MCK) 9 March 2006 (2006-03-09) paragraphs [0061] - [0070] table 1 claims 1, 3, 6, 24 DATABASE WPI Week 201561 Thomson Scientific, London, GB; AN 2015-514813 XP002783643, & KR 101 547 306 B1 (JUNG H K) 25 August 2015 (2015-08-25) abstract DATABASE WPI Week 201640 Thomson Scientific, London, GB; AN 2016-26047V XP002783644, & CN 105 503 477 A (ZHONGSHAN LVLANG AUXILIARIES CO LTD) 20 April 2016 (2016-04-20) abstract

International application No
PCT/US2018/034044

citation of document, with indication, where appropriate, of the relevant passages D T Davidson: "Soil Stabilisation with	Relevant to claim No.
D. T. Davidson: "Soil Stabilisation with	
Chemicals",	1,2,4-8, 10
in May 1962 (1962-05-18), pages 1-347, XP55497634, Ames, Iowa, U.S.A. Retrieved from the Internet: URL:http://publications.iowa.gov/22154/1/I ADOT_IHRB_Bulletin_22_Soil_Stabilization_C hemicals_1960.pdf [retrieved on 2018-08-06] page 152, line 25 - page 153, line 2 page 154; figure 9 page 173; table III	
<pre>.: "ACULYN(TM) 28 Rheology Modifier/Stabilizer",</pre>	1-4
1 January 2006 (2006-01-01), pages 1-11, XP55497602, Retrieved from the Internet: URL:https://www.dow.com/assets/attachments/business/pcare/aculyn/aculyn_28/tds/aculyn_28.pdf	
the whole document	1-10
DATABASE WPI Week 197703 Thomson Scientific, London, GB; AN 1977-04643Y XP002783645, & JP S51 139115 A (JAPAN EXLAN CO LTD) 30 November 1976 (1976-11-30) abstract	1-10
US 4 051 630 A (BISHOP RICHARD TIMOTHY) 4 October 1977 (1977-10-04) column 1, line 4 - line 9 column 1, line 18 - line 20 column 1, line 49 - line 58 examples 1, 4, 5	1,5-10
DATABASE WPI Week 200011 Thomson Scientific, London, GB; AN 2000-119403 XP002783646, & JP H11 323331 A (SHOWA HIGH POLYMER CO LTD) 26 November 1999 (1999-11-26) abstract	1-10
-/	
	XP55497634, Ames, Iowa, U.S.A. Retrieved from the Internet: URL:http://publications.iowa.gov/22154/1/I ADOT IHRB_Bulletin_22_Soil_Stabilization_C hemicals_1960.pdf [retrieved on 2018-08-06] page 152, line 25 - page 153, line 2 page 154; figure 9 page 173; table III .: "ACULYN(TM) 28 Rheology Modifier/Stabilizer", 1 January 2006 (2006-01-01), pages 1-11, XP55497602, Retrieved from the Internet: URL:https://www.dow.com/assets/attachments /business/pcare/aculyn/aculyn_28/tds/aculy n_28.pdf [retrieved on 2018-08-06] the whole document DATABASE WPI Week 197703 Thomson Scientific, London, GB; AN 1977-04643Y XP002783645, & JP S51 139115 A (JAPAN EXLAN CO LTD) 30 November 1976 (1976-11-30) abstract US 4 051 630 A (BISHOP RICHARD TIMOTHY) 4 October 1977 (1977-10-04) column 1, line 4 - line 9 column 1, line 49 - line 58 examples 1, 4, 5 DATABASE WPI Week 200011 Thomson Scientific, London, GB; AN 2000-119403 XP002783646, & JP H11 323331 A (SHOWA HIGH POLYMER CO LTD) 26 November 1999 (1999-11-26) abstract

International application No
PCT/US2018/034044

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	tion). DOCUMENTS CONSIDERED TO BE RELEVANT	T	
Category*	Citation of document, with indication, where appropriate, of the relevant passages	ļ!	Relevant to claim No.
X Y	US 4 227 911 A (LEONARD JAMES D ET AL) 14 October 1980 (1980-10-14) column 8. line 53 - line 69		1,2,4-10 1-10
•	column 8, line 53 - line 69 column 11, line 21 - line 27 examples 1, 4(G)-4(K) claims 1, 7		1 10
X	Dow: "ACRYSOL(TM) Rheology Modifiers",		1-4
	<pre>1 October 2013 (2013-10-01), pages 1-16, XP055498200, Switzerland Retrieved from the Internet: URL:http://msdssearch.dow.com/PublishedLit eratureDOWCOM/dh_08f8/0901b803808f8b0d.pdf ?filepath=coatings/pdfs/noreg/884-00076.pd f&fromPage=GetDoc [retrieved on 2018-08-08]</pre>		
Υ	"ASE Chemistry"; "HASE Chemistry"		1-10

Information on patent family members

International application No
PCT/US2018/034044

			,	.018/034044
Patent document cited in search report	Publication date	Patent family member(s)	/	Publication date
US 4540427 A	10-09-1985	CA 11341 CH 6351 DE 30654 EP 00233 ES 81057 FR 24624 GB 20547 IL 602 JP S56223 MA 188	26 A5 51 D1 47 A1 70 A1 53 A1 96 A 23 A 97 A1 59 A	08-12-1983 26-10-1982 15-03-1983 08-12-1983 04-02-1981 01-07-1981 13-02-1981 18-02-1981 30-09-1982 02-03-1981 01-04-1981 31-07-1981 10-09-1985 24-06-1981
EP 0101253 A	2 22-02-1984	CA 12218 DE 33726 DK 3741 EP 01012 ES 85024 FI 8329 GR 789 IL 694 NO 8329	35 D1 33 A 53 A2 64 A1 41 A 28 B 56 A 37 A 84 A	07-08-1986 19-05-1987 27-08-1987 18-02-1984 22-02-1984 01-01-1985 18-02-1984 02-10-1984 31-10-1986 20-02-1984 01-09-1983 17-12-1985
GB 1591415 A	24-06-1981	BE 8580 CA 11349 DE 27379 FR 23625 GB 15914	81 A 41 A1 86 A1 15 A 95 B 53 A 29 B2 88 A 89 A	13-11-1980 23-02-1978 02-11-1982 09-03-1978 24-03-1978 24-06-1981 30-10-1981 21-05-1985 08-04-1978 25-06-1986 15-08-1985 28-02-1978 26-07-1978
WO 8300498 A	1 17-02-1983	AU 5477 CA 12158 CA 12169 DE 32624 DE 32737 DK 1546 DK 1547 EP 00722 EP 00722 ES 83083 ES 83083 GB 21045 GB 21093 GR 762	97 A 27 D1 97 D1 83 A 83 A 13 A1 14 A1 46 A1 47 A1 32 A	13-06-1985 31-10-1985 30-12-1986 20-01-1987 28-03-1985 20-11-1986 07-04-1983 07-04-1983 16-02-1983 16-02-1983 16-08-1983 16-08-1983 09-03-1983 02-06-1983 04-08-1984

Information on patent family members

International application No
PCT/US2018/034044

						,	016/034044
	atent document d in search report		Publication date		Patent family member(s)		Publication date
				IE IN JP JP NZ NZ PH WO WO ZA	53134 156248 156787 S58501233 S58501234 201482 201483 19119 8300482 8300498 8205595	B A A A A A1 B	06-07-1988 08-06-1985 02-11-1985 28-07-1983 28-07-1983 14-12-1984 30-08-1985 06-01-1986 17-02-1983 17-02-1983 28-03-1984 28-03-1984
EP	0122797	A1	24-10-1984	DE EP	3464761 0122797		20-08-1987 24-10-1984
EP	1247444	A2	09-10-2002	DE EP US ZA	10116442 1247444 2002157141 200202504	A2 A1	10-10-2002 09-10-2002 24-10-2002 24-10-2002
WO	03000621	A1	03-01-2003	AT CN CY DE EP ES HK IL JP PT US US WO	478835 1633400 1110940 10130427 1399397 2296567 1079761 158841 4133807 2005500407 1399397 2004132869 2008114095 2010083719 03000621	A T1 A1 T1 A1 A B2 A E A1 A1	15-09-2010 29-06-2005 10-06-2015 27-03-2003 24-03-2004 01-05-2008 15-10-2010 05-07-2006 13-08-2008 06-01-2005 23-11-2010 08-07-2004 15-05-2008 08-04-2010 03-01-2003
WO	2006026406	A2	09-03-2006	AU CA CN EP JP KR US US US	2005280088 2576967 101432309 1781709 2008511726 20070051868 2006047068 2008113866 2009069185 2006026406 200508113	A1 A2 A A A1 A1 A1 A2	09-03-2006 09-03-2006 13-05-2009 09-05-2007 17-04-2008 18-05-2007 02-03-2006 15-05-2008 12-03-2009 09-03-2006 29-11-2006
KR	101547306	B1	25-08-2015	NONE			
CN	105503477	A	20-04-2016	NONE			
JP	S51139115	Α	30-11-1976	NONE			
US	4051630	А	04-10-1977	AU CA US	8760275 1076382 4051630	Α	23-06-1977 29-04-1980 04-10-1977

Information on patent family members

International application No
PCT/US2018/034044

					32018/034044
Patent document cited in search report		Publication date	P.	atent family nember(s)	Publication date
			ZM	16875 A1	21-02-1977
JP H11323331	Α	26-11-1999	NONE		
US 4227911	Α	14-10-1980	NONE		