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THIXOTROPIC CLAY AQUEOUS SUSPENSIONS
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- (71) Applicant(s)
COLGATE PALMOLIVE COMPANY
- (72) Inventor(s)
JULIEN DRAPIER; CHANTAL GALLANT; DANIEL VAN DE GAER; GEORGES CHAZARD
- (74) Attorney or Agent
F B RICE & CO , 28A Montague Street, BALMAIN NSW 2041
- (56) Prior Art Documents
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GB 2116199
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- (57) Claim

1. An aqueous thixotropic automatic dishwasher composition comprising approximately by weight:
- (a) 5 to 35% alkali metal tripolyphosphate;
 - (b) 2.5 to 20% sodium silicate;
 - (c) 0 to ~~9.9~~⁹% alkali metal carbonate;
 - (d) 0.1 to 5% chlorine bleach stable, water-dispersible organic non-soap detergent active material;
 - (e) 0 to 5% chlorine bleach stable foam depressant;
 - (f) chlorine bleach compound in an amount to provide 0.2 to 4% of available chlorine;
 - (g) thixotropic clay thickener in an amount of 0.1 to 3% sufficient to provide the composition with a thixotropy index of 2 to 10;
 - (h) 0 to 8% of sodium hydroxide;
 - (i) a long chain fatty acid having from 8 to 24 carbon atoms as a physical stabilizer in an amount of 0.02 to 0.4% effective to increase the apparent viscosity and to increase the physical stability of the composition; and
 - (j) balance water.

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Name of Applicant :

 COLGATE-PALMOLIVE COMPANY

Address of Applicant :

 300 Park Avenue, New York, New York,
 10022, United States of America

Actual Inventor/s :

 Julien Drapier ; Chantal Gallant ;
 Daniel Van De Gaer ; Georges Chazard

Address for Service :

 F.B. RICE & CO.,
 Patent Attorneys,
 28A Montague Street,
 BALMAIN 2041.

Complete Specification for the invention entitled:

 Thixotropic Clay Aqueous Suspensions Containing Long Chain
 Saturated Fatty Acid Stabilizers

The following statement is a full description of this invention
including the best method of performing it known to us/

The present invention relates to thixotropic clay aqueous suspension with improved physical stability. More specifically the invention relates to the use of long chain fatty acids as physical stabilizers for thixotropic clay aqueous suspensions.

The present invention specifically relates to automatic dishwashing detergent compositions having thixotropic properties, improved chemical and physical stability, and with increased apparent viscosity, and which are readily dispersible in the washing medium to provide effective cleaning of dishware, glassware, china and the like.

Commercially available household-machine dishwasher detergents provided in power form have several disadvantages, e.g. non-uniform composition; costly operations necessary in their manufacture; tendency to cake in storage at high humidities, resulting in the formation of lumps which are difficult to disperse; dustiness, a source of particular irritation to users who suffer allergies; and tendency to cake in the dishwasher machine dispenser.

Recent research and development activity has focused on the gel or "thixotropic" form of such compositions, e.g. scouring cleansers and automatic-dishwasher products characterized as thixotropic pastes. Dishwasher products so provided are primarily objectionable in that they are insufficiently viscous to remain "anchored" in the dispenser cup of the dishwasher, and moreover yield spotty residues on dishware, glassware, china and the like. Ideally, thixotropic cleansing compositions should be highly viscous in a quiescent state, Bingham plastic in nature, and have relatively high yield values. When subjected to shear stresses, however, such as being shaken in a container or squeezed through an orifice, they should quickly fluidize and, upon cessation of the applied shear

stress, quickly revert to the high viscosity/Bingham plastic state. Stability is likewise of primary importance, i.e. there should be no significant evidence of phase separation or leaking after long standing.

5 The U.S. Patent Application Serial No. 744,754 filed 14 June, 1985, which is assigned to the applicant's assignee, and corresponds to Australian Patent No. 588881,



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5 The U.S. Patent Application Serial No. 744,754 filed
~~June 14, 1985, which is assigned to applicant's assignee,~~
is directed to thixotropic clay aqueous suspension
dishwashing detergent compositions containing metal salts
of long chain fatty acids, such as aluminum stearate as
10 physical stabilizing agents. The application Serial No.
744,754 compositions show improvement in the physical
stability of the detergent composition and improvement
against phase separation over those clay containing
compositions that do not contain the aluminum stearate.

15 The provision of automatic-dishwasher compositions in
gel form having the aforescribed properties, other than
for the improvements described in the above mentioned
application Serial No. 744,754, has thus far proven
problematical, particularly as regards compositions for
20 use in home dishwasher machines. For effective use, it is
generally recommended that the automatic dishwashing
detergent, hereinafter also designated ADD, contain (1)
sodium tripolyphosphate (NaTPP) to soften or tie up
hard-water minerals and to emulsify and/or peptize soil;
25 (2) sodium silicate to supply the alkalinity necessary for
effective detergency and to provide protection for fine
china glaze and pattern; (3) sodium carbonate, generally
considered to be optional, to enhance alkalinity; (4) a
chlorine-releasing agent to aid in the elimination of soil
30 specks which lead to water spotting; and (5)
defoamer/surfactant to reduce foam, thereby enhancing
machine efficiency and supplying requisite detergency.
See, for example, SDA Detergents in Depth, "Formulations
Aspects of Machine Dishwashing," Thomas Oberle (1974).
35 Cleansers approximating to the aforescribed compositions



are mostly liquids or powders. Combining such ingredients in a gel form effective for home-machine use has proved difficult. Generally, such compositions omit hypochlorite bleach, since it tends to react with other chemically active ingredients, particularly surfactant. Thus, U.S. Patent No. 4,115,308 discloses thixotropic automatic dishwasher pastes containing a suspending agent, e.g. CMC, synthetic clays or the like; inorganic salts including silicates, phosphates and polyphosphates; a small amount of surfactant and a suds depressor. Bleach is not disclosed. U.S. Patent No. 4,147,650 is somewhat similar, optionally including Cl-(hypochlorite) bleach but no organic surfactant or foam depressant. The product is described, moreover, as a detergent slurry with no apparent thixotropic properties.

U.S. Patent No. 3,985,668 describes abrasive scouring cleaners of gel-like consistency containing (1) suspending agent, preferably the Smectite and attapulgite types of clay; (2) abrasive, e.g. silica sand or perlite; and (3) filler comprising light density powdered polymers, expanded perlite and the like, which has a bouyancy and thus stabilizing effect on the composition in addition to serving as a bulking agent, thereby replacing water otherwise available for undesired supernatant layer formation due to leaking and phase destabilization. The foregoing are the essential ingredients. Optional ingredients include hypochlorite bleach, bleach stable surfactant and buffer, e.g. silicates, carbonates, and monophosphates. Builders, such as NaTPP, can be included as further optional ingredients to supply or supplement building function not provided by the buffer, the amount of such builder not exceeding 5% of the total composition, according to the patent. Maintenance of the desired (greater than) pH 10 levels is achieved by the buffer/builder components. High pH is said to minimize

decomposition of chlorine bleach and undesired interaction between surfactant and bleach. Foam killer is not disclosed.

In U.K. Patent Application GB 2,116,199A and GB
5 2,140,450A, both of which are assigned to
Colgate-Palmolive, liquid ADD compositions are disclosed
which have properties desirably characterizing
thixotropic, gel-type structure and which include each of
the various ingredients necessary for effective detergency
10 with an automatic dishwasher. The normally gel-like
aqueous automatic dishwasher detergent composition having
thixotropic properties includes the following ingredients,
on a weight basis:

- (a) 5 to 35% alkali metal tripolyphosphate;
- 15 (b) 2.5 to 20% sodium silicate;
- (c) 0 to 9% alkali metal carbonate;
- (d) 0.1 to 5% chlorine bleach stable, water dispersible
organic detergent active material;
- (e) 0 to 5% chlorine bleach stable foam depressant;
- 20 (f) chlorine bleach compound in an amount to provide
about 0.2 to 4% of available chlorine;
- (g) thixotropic thickener in an amount sufficient to
provide the composition with thixotropy index of
about 2.5 to 10;
- 25 (h) sodium hydroxide, as necessary, to adjust pH; and
(i) balance water.

ADD compositions so formulated are low-foaming; are
readily soluble in the washing medium and most effective
at pH values best conducive to improved cleaning
30 performance, viz, pH 10.5-14. The compositions are
normally of gel consistency, i.e. a highly viscous, opaque
jelly-like material having Bingham plastic character and
thus relatively high yield values. Accordingly, a
definite shear force is necessary to initiate or increase
35 flow, such as would obtain within the agitated dispenser

cup of an energized automatic dishwasher. Under such conditions, the composition is quickly fluidized and easily dispersed. When the shear force is discontinued, the fluid composition quickly reverts to a high viscosity, the fluid composition quickly reverts to a high viscosity, Bingham plastic state closely approximating its prior consistency.

U.S. Patent 4,511,487 dated April 16, 1985 describes a low-foaming detergent paste for dishwashers. The patented thixotropic cleaning agent has a viscosity of at least 30 pa.s at 29°C as determined with a rotational viscometer at a spindle speed of 5 revolutions per minute. The composition is based on a mixture of finely divided hydrated sodium metasilicate, an active chlorine compound and a thickening agent which is a foliated silicate of the hectorite type. Small amount of nonionic tensides and alkali metal carbonates and/or hydroxides may be used.

The formation of organoclays by the interaction of clays (such as bentonite and hectorite) with organic compounds such as quaternary ammonium salts, has also been described (W.S. Mardis, JAOCs, Vol. 61, No. 2, p.382 (1984)).

While these previously disclosed liquid ADD formulations are not subject or are subject to a lesser degree to one or more of the above described deficiencies, it has been found that further improvements in physical stability at lower costs are desired to increase the shelf-life of the product and thereby enhance consumer acceptance.

At the same time it would be highly desirable to increase the physical stability of other clay based thixotropic liquid formulations, such as scouring cleansers; dental pastes, "liquid" soaps, and the like.

Accordingly, it is an objective of the invention to provide anti-settling additives for thixotropic clay

aqueous suspensions.

It is another object of the invention to provide liquid ADD compositions having thixotropic properties with improved physical stability and rheological properties at
5 lower costs by using fatty acids in place of the more expensive metal salts of fatty acids.

It is still another object of the invention to provide thixotropic liquid ADD compositions having reduced levels of thixotropic thickener without adversely
10 effecting the generally high viscosities at low shear rates and lower viscosities at high shear rates which are characteristic of the desired thixotropic properties.

More broadly, it is an object of this invention to improve the stability of aqueous thixotropic clay based
15 compositions, especially liquid automatic dishwasher detergent pastes or gels, by incorporating in the clay aqueous suspension a minor amount of a fatty acid effective to increase the apparent viscosity of the formulation and to inhibit the settling of the suspended
20 particles and to prevent phase separation.

These and other objects of the invention which will become more readily understood from the following detailed description of the invention and preferred embodiments thereof are achieved by incorporating in a normally
25 gel-like aqueous liquid composition a small but effective amount of a physical stabilizer which is a long chain fatty acid. More particularly, according to a preferred and specific embodiment of the invention, there is provided a normally gel-like automatic dishwasher
30 detergent composition in which is incorporated an amount of a long chain fatty acid which is effective to increase the apparent viscosity of the formulation and to inhibit settling of the suspended particles, such as thixotropic agent.

35 In accordance with this particular aspect, the

present invention provides a normally gel-like aqueous automatic dishwasher detergent composition having thixotropic properties which include, on a weight basis:

- (a) 5 to 35% alkali metal tripolyphosphate;
- 5 (b) 2.5 to 20% sodium silicate;
- (c) 0 to 9% alkali metal carbonate;
- (d) 0.1 to 5% chlorine bleach stable, water dispersible organic detergent active material;
- (e) 0 to 5% chlorine bleach stable foam depressant;
- 10 (f) chlorine bleach compound in an amount to provide about 0.2 to 4% of available chlorine;
- (g) thixotropic thickener in an amount sufficient to provide the composition with thixotropy index of about 2.5 to 10; and
- 15 (h) 0 to 8% sodium hydroxide;
- (i) a long chain fatty acid in an amount effective to increase apparent viscosity and the the physical stability of the composition; and
- (j) balance water.

20 Also related to this specific aspect, the invention provides a method for cleaning dishware in an automatic dishwashing machine with an aqueous wash bath containing an effective amount of the liquid automatic dishwasher detergent (LADD) composition as described above.

25 According to this aspect of the invention, the LADD composition can be readily poured into the dispensing cup of the automatic dishwashing machine and will, within just a few seconds, promptly thicken to its normal gel-like or pasty state to remain securely within the dispensing cup
30 until shear forces are again applied thereto, such as by the water spray from the dishwashing machine.

Generally, LADD effectiveness is directly related to (a) available chlorine levels; (b) alkalinity; (c) solubility in washing medium; and (d) foam inhibition.

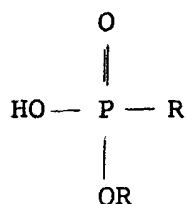
35 It is preferred herein that the pH of the LADD composition

be at least about 9.5, more preferably from about 10.5 to 14.0 and most preferably at least about 11.5. The presence of carbonate is also often needed herein, since it acts as a buffer helping to maintain the desired pH level. Excess carbonate is to be avoided, however, since it may cause the formation of needle-like crystals of carbonate, thereby impairing the stability, as well as impairing the dispensibility of the product from, for example, squeeze tube bottles. Caustic soda (NaOH) serves the further function of neutralising the phosphoric or phosphonic acid ester foam depressant when present. About 0.5 to 6 wt% of NaOH and about 2 to 9 wt% of sodium carbonate in the LADD composition are typical, although it should be noted that sufficient alkalinity may be provided by the NaTPP and sodium silicate.

The NaTPP employed in the LADD composition in a range of about 8 to 35 wt%, preferably about 20 to 30 wt%, should preferably be free of heavy metal which tends to decompose or inactivate the preferred sodium hypochlorite and other chlorine bleach compounds. The NaTPP may be anhydrous or hydrated, including the stable hexahydrate with a degree of hydration of 6 corresponding to about 18% by weight of water or more. Especially preferred LADD compositions are obtained, for example, when using a 0.5:1 to 2:1 weight ratio of anhydrous to hexahydrated NaTPP, values of about 1:1 being particularly preferred.

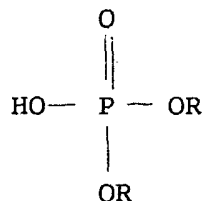
Foam inhibition is important to increase dishwasher machine efficiency and minimize destabilizing effects which might occur due to the presence of excess foam within the washer during use. Foam may be sufficiently reduced by suitable selection of the type and/or amount of detergent active material, the main foam-producing component. The degree of foam is also somewhat dependent on the hardness of the wash water in the machine whereby suitable adjustment of the proportions of NaTPP which has

a water softening effect may aid in providing the desired degree of foam inhibition. However, there may optimally be included a chlorine bleach stable foam depressant or inhibitor where a low foam LADD is desired. Particularly effective are the alkyl phosphonic acid esters of the formula



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available for example from BASF-Wyandotte (PCUK-PAE), and especially the alkyl acid phosphate esters of the formula



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available, for example, from Hooker (SAP) and Knapsack (LPKn-158), in which one or both R groups in each type of ester may represent independently a C₁₂₋₂₀ alkyl group. Mixtures of the two types, or any other chlorine bleach stable types, or mixtures of mono- and di-esters of the same type, may be employed. Especially preferred is a mixture of mono- and di-C₁₆₋₁₈ alkyl acid phosphate esters such as monostearyl/distearyl acid phosphates 1.2/1 (Knapsack) or 4/1 (UGINE KULHPLAN). When employed, proportions of 0.1 to 5 wt%, preferably about 0.1 to 0.5 wt%, of foam depressant in the composition is typical, the weight ratio of detergent active component (d) to foam depressant (e) generally ranging from about 10:1 to 1:1 and preferably about 5:1 to 1:1. Other defoamers which may be used include, for example, the known silicones. In addition, it is an advantageous feature of this invention that many of the stabilizing long chain fatty acids, such

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as stearic acid and behenic acid also act as foam killers.

Although any chlorine bleach compound may be employed in the compositions of this invention, such as dichloro-isocyanurate, dichloro-dimethyl hydantoin, or chlorinated TSP, alkali metal, e.g. potassium, lithium, magnesium and especially sodium, hypochlorite is preferred. The composition should contain sufficient chlorine bleach compound to provide about 0.2 to 4.0% by weight of available chlorine, as determined, for example, by acidification of 100 parts of the composition with excess of hydrochloric acid. A solution containing about 0.2 to 4.0% by weight of sodium hypochlorite contains or provides roughly the same percentage of available chlorine. About 0.8 to 1.6% by weight of available chlorine is especially preferred. For example, sodium hypochlorite (NaOCl) solution of from about 11 to about 14% available chlorine in amounts of about 3 to 20%, preferably about 7 to 12%, can be advantageously used.

The sodium silicate, which provides alkalinity and protection of hard surface, such as fine china glaze and pattern, is employed in an amount ranging from about 2.5 to 20 wt%, preferably about 5 to 15 wt%, in the composition. The sodium silicate is generally added in the form of an aqueous solution, preferably having an $\text{Na}_2\text{O}:\text{SiO}_2$ ratio of about 1:2 to 1:2.8.

Detergent active material useful herein must be stable in the presence of chlorine bleach, especially hypochlorite bleach, and those of the organic anionic, amine oxide, phosphine oxide, sulphoxide or betaine water dispersible surfactant types are preferred, the first mentioned anionics being most preferred. They are used in amounts ranging from about 0.1 to 5%, preferably about 0.3 to 2.0%. Particularly preferred surfactants herein are the linear or branched alkali metal mono- and/or di-(C_{8-14}) alkyl diphenyl oxide mono and/or disulphates

or disulfonates, commercially available for example as DOWFAX (Registered Trademark) 3B-2 and DOWFAX 21A-1. In addition, the surfactant should be compatible with the other ingredients of the composition. Other suitable

5 surfactants include the primary alkylsulphates, alkylsulphonates, alkylaryl-sulphonates and sec.-alkylsulphates. Examples include sodium C₁₀-C₁₈ alkylsulphates such as sodium dodecylsulphate and sodium tallow alcohol sulphate; sodium C₁₀-C₁₈

10 alkanesulphonates such as sodium hexadecyl-1-sulphonate and sodium C₁₂C₁₈ alkylbenzenesulphonates such as sodium dodecylbenzenesulphonates. The corresponding potassium salts may also be employed.

As other suitable surfactants or detergents, the

15 amine oxide surfactants are typically of the structure R₂R¹N-O, in which each R represents a lower alkyl group, for instance, methyl, and R¹ represents a long chain alkyl group having from 8 to 22 carbon atoms, for instance a lauryl, myristyl, palmityl or cetyl group.

20 Instead of an amine oxide, a corresponding surfactant phosphine oxide R₂R¹PO or sulfoxide RR¹SO can be employed. Betaine surfactants are typically of the structure R₂R¹N - R¹COO⁻, in which each R represents a lower alkylene group having from 1 to 5 carbon atoms.

25 Specific examples of these surfactants are lauryl-dimethylamine oxide, myristyldimethylamine oxide, the corresponding phosphine oxides and sulfoxides, and the corresponding betaines, including dodecyldimethylammonium acetate, tetradecyldiethylammonium

30 pentanoate, hexadecyldimethylammonium hexanoate and the like. For biodegradability, the alkyl groups in these surfactants should be linear, and such compounds are preferred.

Surfactants of the foregoing type, all well known in

35 the art, are described, for example, in U.S. Patents

3,985,669 and 4,271,030.

Thixotropic thickeners, i.e. thickeners or suspending agents which provide an aqueous medium with thixotropic properties, are known in the art and may be organic or
5 inorganic water soluble, water dispersible or colloid-forming, and monomeric or polymeric, and should of course be stable in these compositions, e.g. stable to high alkalinity and chlorine bleach compounds, such as sodium hypochlorite. Those especially preferred generally
10 comprise the inorganic, colloid-forming clays of smectite and/or attapulgite types. These materials were generally used in amounts of about 1.0 to 10, preferably 1.2 to 5 wt%, to confer the desired thixotropic properties and Bingham plastic character in the assignee's prior
15 disclosed LADD formulations of the aforementioned GB 2,116,199A and GB 2,140,450A. It is one of the advantages of the LADD formulations of the present invention that the desired thixotropic properties and Bingham plastic character can be obtained in the presence of the fatty
20 acid stabilizers with lesser amounts of the thixotropic thickeners. For example, amounts of the inorganic colloid-forming clays of the smectite and/or attapulgite types in the range of from about 0.1 to 3%, preferably 0.1 to 2.5%, especially 0.1 to 2%, are generally sufficient to
25 achieve the desired thixotropic properties and Bingham plastic character when used in combination with the physical stabilizer.

Smectite clays include montmorillonite (bentonite), hectorite, attapulgite, smectite, saponite, and the like.
30 Montmorillonite clays are preferred and are available under tradenames such as Thixogel (Registered trademark) No. 1 and Gelwhite (Registered Trademark) GP, H, etc., from Georgia Kaolin Company; and ECCAGUM (Registered Trademark) GP, H, etc., from Luthern Clay Products.
35 Attapulgite clays include the materials commercially

available under the tradename Attagel (Registered Trademark), i.e. Attagel 40, Attagel 50 and Attagel 150 from Engelhard Minerals and Chemicals Corporation. Mixtures of smectite and attapulgite types in weight ratios of 4:1 to 1:5 are also useful herein. Thickening or suspending agents of the foregoing types are well known in the art, being described, for example, in U.S. Patent No. 3,985,668 referred to above. Abrasives or polishing agents should be avoided in the LADD compositions as they may mar the surface of fine dishware, crystal and the like.

The amount of water contained in these compositions should, of course, be neither so high as to produce unduly low viscosity and fluidity, nor so low as to produce unduly high viscosity and low flowability, thixotropic properties in either case being diminished or destroyed. Such amount is readily determined by routine experimentation in any particular instance, generally ranging from about 30 to 75 wt%, preferably about 35 to 65 wt%. The water should also be preferably deionized or softened.

So far, the description of the LADD product, except as otherwise noted, conforms to the compositions as disclosed in the aforementioned U.K. Patent Applications GB 2,116,199A and GB 2,140,450A, which are assigned to applicants' assignee.

The LADD products of the prior U.K. Patent Application GB 2,116,199A and GB 2,140,450 exhibit improved rheological properties as evaluated by testing product viscosity as a function of shear rate. The compositions exhibited higher viscosity at a low shear rate and lower viscosity at a high shear rate, the data indicating efficient fluidization and gellatin well within the shear rates extant within the standard dishwasher machine. In practical terms, this means improved pouring and processing characteristics as well as less leaking in

the machine dispenser-cup, compared to prior liquid or gel ADD products. For applied shear rates corresponding to 3 to 30 rpm, viscosities (Brookfield) correspondingly ranged from about 10,000 to 30,000 cps to about 3,000 to 7,000 5 cps, as measured at room temperature by means of an LVT Brookfield viscometer after 3 minutes using a No. 4 spindle. A shear rate of 7.4 sec^{-1} corresponds to a spindle rpm of about 3. An approximate ten-fold increase in shear rate produces about a 3- to 9-fold reduction in 10 viscosity. With prior ADD gels, the corresponding reduction in viscosity was only about two-fold. Moreover, with such compositions, the initial viscosity taken at about 3 rpm was only about 2,500 to 2,700 cps. The compositions of the assignee's prior invention thus 15 exhibit threshold fluidizations at lower shear rates and of significantly greater extent in terms of incremental increases in shear rate versus incremental decrease in viscosity. This property of the LADD products of the prior invention is summarized in terms of a thixotropic 20 index (TI) which is the ratio of the apparent viscosity at 3 rpm and at 30 rpm. The prior compositions have a TI of from 2 to 10. The LADD compositions tested exhibited substantial and quick return to prior quiescent state consistency when the shear force was discontinued.

25 The present invention is based upon the discovery that the physical stability, i.e. resistance to phase separation, settling, etc., of the U.K. Patent Applications GB 2,116,199A and GB 2,140,450 and the U.S. Patent Application Serial No. 744,754 liquid aqueous ADD 30 compositions can be significantly improved or not adversely affected while at the same time significantly increasing the apparent viscosity and improving the physical stability of the formulations and at lower cost, by adding to the composition a small but effective amount 35 of a long chain fatty acid.

As an example of the improvement in rheological properties, it has been found that the viscosities at low shear rates, e.g. at a spindle rpm of about 3, apparent viscosities may often be increased from two- to three-fold
5 with the incorporation of as little as 0.2% or less, e.g. 0.16%, of the fatty acid stabilizer. At the same time, the physical stability may be improved to such an extent that even after a long time, the compositions containing the fatty acid stabilizers do not undergo any visible
10 phase separation.

The preferred long chain fatty acids are the higher aliphatic fatty acids having from about 8 to about 24 carbon atoms, more preferably from about 10 to 24 carbon atoms, and especially preferably from about 12 to 22
15 carbon atoms, inclusive of the carbon atom of the carboxyl group of the fatty acid. The aliphatic radical may be saturated or unsaturated and may be straight or branched. Straight chain saturated fatty acids are preferred. Mixtures of fatty acids may be used, such as
20 those derived from natural sources, such as tallow fatty acid, coco fatty acid, soya fatty acid, etc., or from synthetic sources available from industrial manufacturing processes.

Thus, examples of the fatty acids which can be used
25 as stabilizers include, for example, decanoic acid, dodecanoic acid, palmitic acid, myristic acid, stearic acid, behenic acid, oleic acid, eicosanoic acid, tallow fatty acid, coco fatty acid, soya fatty acid, mixtures of these acids, etc. Behenic acid, stearic acid and mixed
30 fatty acids are preferred, with behenic acid being the most preferred.

Naturally, for LADD compositions, as well as any other applications where the invention composition will or may come into contact with articles used for the handling,
35 storage or serving of food products or which otherwise may

come into contact with or be consumed by people or animals, the use of the fatty acids as the stabilizing agent are of particular advantage because of their known low toxicity. For this purpose, the stearic acid and
5 behenic acid are especially preferred as generally safe food additives. Another distinct advantage of the use of the fatty acids as stabilizers is their lower cost as compared to the fatty acid metal salts.

Many of these fatty acids are commercially
10 available. For example, the stearic acid and behenic acid are readily available.

Mixed fatty acids, such as the naturally occurring acids, e.g. coco acid, as well as mixed fatty acids resulting from the commercial manufacturing process are
15 also advantageously used as an inexpensive but effective source of long chain fatty acids.

The amount of the fatty acid stabilizers to achieve the desired enhancement of physical stability and apparent viscosity increase will depend on such factors as the
20 nature of the fatty acid, the nature and amount of the thixotropic agent, detergent active compound, inorganic salts, especially TPP, other LADD ingredients, as well as the anticipated storage and shipping conditions.

Generally, however, amounts of the fatty acid
25 stabilizing agents in the range of from about 0.02 to 1%, preferably from about 0.06 to 0.8%, especially preferably from about 0.08 to 0.4%, provide the increase in apparent viscosity and the long term stability and absence of phase separation upon standing or during transport at both low
30 and elevated temperatures as are required for a commercially acceptable product.

From the examples to be given below, it will be seen that, depending on the amounts, proportions and types of physical stabilizers and thixotropic agents, the addition
35 of the fatty acids not only increases physical stability

but also provides a simultaneous increase in apparent viscosity. Ratios of fatty acid to thixotropic agent in the range of from about 0.08 to 0.4 weight percent fatty acid and from about 1.3 to 2.5 weight percent thixotropic agent are usually sufficient to provide these simultaneous benefits and, therefore, the use of these ingredients in these ratios is most preferred.

According to one preferred method of making these compositions, one should dissolve or disperse first all the inorganic salts, i.e. carbonate (when employed), silicate and tripolyphosphate, in the aqueous medium. Thickening agent is added last. The foam depressor (when employed) is preliminarily provided as an aqueous dispersion, as is the thickening agent. The foam depressant dispersion, caustic soda (when employed) and inorganic salts are first mixed at elevated temperatures in aqueous solution (deionized water) and, thereafter, cooled, using agitation throughout. Bleach, surfactant, fatty acid stabilizer and thickener dispersion at room temperature are thereafter added to the cooled (25-35°C) solution. Excluding the chlorine bleach compound, total salt concentration (NaTpp, sodium silicate and carbonate) is generally about 20 to 50 weight percent, preferably about 30 to 40 weight percent in the composition.

Another highly preferred method for mixing the ingredients of the LADD formulations involves first forming a mixture of the water, foam suppressor (when employed), detergent, physical stabilizer (fatty acid) and thixotropic agent, e.g. clay. These ingredients are mixed together under high shear conditions, preferably starting at room temperature, to form a uniform dispersion. This this premixed portion, the remaining ingredients are introduced under low shear mixing conditions. For instance, the required amount of the premix is introduced into a low shear mixer and thereafter the remaining

ingredients are added, with mixing, either sequentially or simultaneously. Preferably, the ingredients are added sequentially, although it is not necessary to complete the addition of all of one ingredient before beginning to add
5 the next ingredient. Furthermore, one or more of the ingredients can be divided into portions and added at different times. Good results have been obtained by adding the remaining ingredients in the following sequence: sodium hydroxide, alkali metal carbonate,
10 sodium silicate, alkali metal tripolyphosphate (hydrated), alkali metal tripolyphosphate (anhydrous or up to 5% water), bleach (preferably, sodium hypochlorite) and sodium hydroxide.

Other conventional ingredients may be included in
15 these compositions in small amounts, generally less than about 3 weight percent, such as perfume, hydrotropic agents such as the sodium benzene, toluene, xylene and cumene sulphonates, preservatives, dyestuffs and pigments and the like, all of course being stable to chlorine
20 bleach compound and high alkalinity (properties of all the components). Especially preferred for colouring are the chlorinated phthalocyanines and polysulphides of aluminosilicate which provide, respectively, pleasing green and blue tints. TiO_2 may be employed for
25 whitening or neutralizing off-shades.

The liquid ADD compositions of this invention are readily employed in known manner for washing dishes, other kitchen utensils and the like in an automatic dishwasher, provided with a suitable detergent dispenser, in an
30 aqueous wash bath containing an effective amount of the composition.

While the invention has been particularly described in connection with its application to liquid automatic dishwasher detergents it will be readily understood by one
35 of ordinary skill in the art that the benefits which are

obtained by the addition of the long chain fatty acids, namely increased apparent viscosity and increased physical stability of the clay based thixotropic suspension, will apply equally well to other clay based thixotropic suspensions, such as the scouring paste formulations described in the aforementioned U.S. Patent 3,985,668.

The invention may be put into practice in various ways and a number of specific embodiments will be described to illustrate the invention with reference to the accompanying examples.

All amounts and proportions referred to herein are by weight of the composition unless otherwise indicated.

Example 1

In order to demonstrate the effect of the fatty acid stabilizer a liquid ADD formulation is prepared as follows.

	<u>Percent</u>
Deionized water	41.10
Caustic soda solution (50% NaOH)	2.20
20 Sodium carbonate, anhydrous	5.00
Sodium silicate, 47.5% solution of Na ₂ O:SiO ₂ ratio of 1:2.4	15.74
25 Sodium TPP (substantially anhydrous-i.e. 0-5%, especially 3%, moisture) (Thermphos NW)	12.00
30 Sodium TPP (hexahydrate) (Thermphos N hexa)	12.00

The mixture is cooled at 25-30°C and agitation maintained throughout, and the following ingredients at room temperature are added thereto:

	<u>Percent</u>
35 Sodium hypochlorite	

	solution (11% available chlorine)	9.00
	Monostearyl phosphate	0.16
5	DOWFAX 3B-2 (45% Na monodecyl/didecyl diphenyl oxide disulphonate-aqueous solution)	0.80
	Physical stabilizer (fatty acid or fatty acid salt)	X
	Pharmagel H	2.00

10 There are three formulations prepared in which X = 0%, X = 0.10% calcium stearate and X = 0.16% behenic acid.

The monostearyl phosphate foam depressant (when employed) and Dowfax 3B-2 detergent active compound fatty acid stabilizer are added to the mixture just before the Pharmagel H thickener.

15 The Run 1 is a control formulation which includes the monostearyl phosphate anti-foam agent, but which does not contain a fatty acid stabilizer.

20 The Run 2 is a control formulation of Run 1 to which has been added a calcium stearate stabilizing agent of application Serial No. 744,754.

25 The Run 3 is a formulation of the present invention in which behenic acid ($\text{CH}_3(\text{CH}_2)_{20}\text{COOH}$) is used as the stabilizing agent and the monostearyl phosphate anti-foam agent is optionally omitted.

Each of the resulting liquid ADD formulations as shown in the Table are measured for apparent viscosity at 3 and 30 rpm. The results obtained are also shown in Table.

30 From the data reported in the Table the following conclusions are reached:

The incorporation of 0.1% calcium stearate in a 2.0% Pharmagel H containing formula Run 2 (control) leads to an increase in the apparent viscosity Table, Run 1 (control).

35 The incorporation of 0.16% behenic acid in a 2%

Pharmagel H containing formula Run 3 (invention) leads to a significant increase in the apparent viscosity as compared to both the control Runs 1 and 2.

TABLE

5

RUN	FORMULATION	BROOK LVT VISCOSITY	
		(KCPS) 3RPM	(1) 30RPM
10 1 (control)	H ₂ O = 41.10% Monostearyl Phosphate = 0.16% Stabilizer = 0% Pharmagel H = 2.0%	18	4.9
15 2 (control)	H ₂ O = 41.0% Monostearyl Phosphate = 0.16% Ca Stearate = 0.1% Pharmagel H = 2.0%	24	3.8
20 3 (invention)	H ₂ O = 41.0% Monostearyl Phosphate = 0% Behenic Acid = 0.16% Pharmagel H = 2.0%	87	10.2

(1) Measured with spindle 4 after three minutes at 3 and 30 rpm on 24 hour old samples.

Example 2

25 The following gel-like thixotropic liquid ADD is prepared following the same general procedures as in

Example 1:

Ingredient	Amount (A.I.) Wt%
Sodium silicate (47.5% solution of Na ₂ O:SiO ₂ ratio of 1:2.4)	7.48
Monostearyl phosphate	0.16
Dowfax 3B-2	0.36
Thermphos NW	12.0
35 Thermphos N hexa	12.0

	Behenic Acid	0.1
	Sodium carbonate, anhydrous	5.0
5	Caustic soda solution (50% NaOH)	3.1
	Pharmagel H	1.5
	Sodium hypochlorite solution (11%)	1.0
	Water	balance
10	Minor amounts of perfume, color, etc. can also be added to formulation.	

Example 3

The following gel-like thixotropic liquid ADD is prepared following the same general procedures as in

15 Example 1:

	<u>Ingredient</u>	<u>Amount (A.I.) Wt%</u>
	Sodium silicate (47.5% solution of $\text{Na}_2\text{O}:\text{SiO}_2$ ratio of 1:2.4)	7.48
20	Monostearyl phosphate	0.16
	Dowfax 3B-2	0.36
	Thermphos NW	12.0
	Thermphos N hexa	12.0
	Stearic Acid	0.2
25	Sodium carbonate, anhydrous	5.0
	Caustic soda solution (50% NaOH)	3.1
	Pharmagel H	1.0
30	Sodium hypochlorite solution (11%)	1.0
	Water	balance

Minor amounts of perfume, color, etc. can also be added to formulation.

THE CLAIMS DEFINING THE INVENTION ARE AS FOLLOWS:-

1. An aqueous thixotropic automatic dishwasher composition comprising approximately by weight:
 - (a) 5 to 35% alkali metal tripolyphosphate;
 - (b) 2.5 to 20% sodium silicate;
 - (c) 0 to ^{9%}~~20%~~ alkali metal carbonate;
 - (d) 0.1 to 5% chlorine bleach stable, water-dispersible organic non-soap detergent active material;
 - (e) 0 to 5% chlorine bleach stable foam depressant;
 - (f) chlorine bleach compound in an amount to provide 0.2 to 4% of available chlorine;
 - (g) thixotropic clay thickener in an amount of 0.1 to 3% sufficient to provide the composition with a thixotropy index of 2 to 10;
 - (h) 0 to 8% of sodium hydroxide;
 - (i) a long chain fatty acid having from 8 to 24 carbon atoms as a physical stabilizer in an amount of 0.02 to 0.4% effective to increase the apparent viscosity and to increase the physical stability of the composition; and
 - (j) balance water.
2. The composition of claim 1, wherein the physical stabilizer (i) is an aliphatic fatty acid having 10 to 24 carbon atoms.
3. The composition of claim 2, wherein the acid has from 12 to 22 carbon atoms.
4. The composition of claim 1 wherein the physical stabilizer (i) is stearic acid.
5. The composition of claim 1 wherein the physical stabilizer (i) is behenic acid.
6. The composition of claim 1 wherein the physical stabilizer (i) is present in an amount of from 0.06 to 0.4%.
7. The composition of claim 1 wherein the physical stabilizer (i) is present in an amount of from 0.08 to 0.4%.



8. The composition of claim 1 wherein the thixotropic thickener (g) is an inorganic, colloid-forming clay.
9. The composition of claim 8 wherein the clay is a montmorillonite clay, an attapulgite clay, a hectorite clay or a smectite clay.
10. The composition of claim 8 wherein the amount of the clay thickener is in the range of from 0.1 to 2.5%.
11. The composition of claim 8 which contains from 0.06 to 0.4% of the physical stabilizer (i) and from 0.1 to 2% by weight of an inorganic, colloid-forming clay as the thixotropic thickener (g).
12. The composition of claim 8 which contains from 0.8 to 0.4% of the physical stabilizer (i) and 0.25% to 1.0% of an inorganic colloid-forming clay as the thixotropic thickener (g).
13. The composition of claim 1 in which the chlorine bleach compound (f) is sodium hypochlorite.
14. The composition of claim 1 which contains 0.1 to 0.5% of the foam depressant (e).
15. The composition of claim 14 in which the foam depressant is an alkyl acid phosphate ester or an alkyl phosphonic acid ester containing one or two C₁₂₋₂₀ alkyl groups, or a mixture thereof.
16. The composition of claim 1 having a pH of 10.5 to 13.5.
17. A method of cleaning soiled dishware in an automatic dishwashing machine which comprises contacting the soiled dishware in an automatic dishwashing machine in an aqueous washbath having dispersed therein an effective amount of the composition of claim 5.
18. An aqueous thixotropic automatic dishwasher composition which comprises by weight:
 - (a) 5 to 35% alkali metal tripolyphosphate;
 - (b) 2.5 to 20% sodium silicate;
 - (c) 0 to 9% alkali metal carbonate;



- (d) 0.1 to 5% chlorine bleach stable, water-dispersible organic non-soap detergent active material;
 - (e) 0 to 5% chlorine bleach stable foam depressant;
 - (f) chlorine bleach compound in an amount to provide 0.2 to 4% of available chlorine;
 - (g) thixotropic clay thickener in an amount of 0.1 to 2.5% sufficient to provide the composition with a thixotropy index of 2 to 10;
 - (h) 0 to 8% of sodium hydroxide;
 - (i) a long chain aliphatic fatty acid having from 10 to 24 carbon atoms as a physical stabilizer in an amount of 0.06 to 0.4% effective to increase the apparent viscosity and to increase the physical stability of the composition; and
 - (j) 35 to 65 weight percent water.
19. The aqueous thixotropic automatic dishwasher composition of claim 18 wherein the thixotropic clay thickener is in an amount of 0.1 to 2.0%.
20. The aqueous thixotropic automatic dishwasher composition of claim 18 wherein the long chain fatty acid stabilizer is in an amount of 0.08 to 0.4%.
21. The aqueous thixotropic automatic dishwasher composition of claim 20 wherein the long chain fatty acid stabilizer is in an amount of about 0.16%.
22. An aqueous thixotropic automatic dishwasher composition consisting essentially of by weight:
- (a) 5 to 35% alkali metal tripolyphosphate;
 - (b) 2.5 to 20% sodium silicate;
 - (c) 0 to 9% alkali metal carbonate;
 - (d) 0.1 to 5% chlorine bleach stable, water-dispersible organic non-soap detergent active material;
 - (e) 0 to 5% chlorine bleach stable foam depressant;
 - (f) chlorine bleach compound in an amount to provide 0.2 to 4% of available chlorine;

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- (g) thixotropic clay thickener in an amount of 0.1 to 2.0% sufficient to provide the composition with a thixotropy index of 2 to 10;
- (h) 0 to 8% of sodium hydroxide;
- (i) a long chain aliphatic fatty acid having from 12 to 22 carbon atoms as a physical stabilizer in an amount of 0.08 to 0.4% effective to increase the apparent viscosity and to increase the physical stability of the composition; and
- (j) 35 to 65 weight percent water.

23. A method for improving stability against phase separation of an aqueous thixotropic automatic dishwasher composition comprising a gel-like aqueous dispersion of at least one water-insoluble finely divided detergent builder material, said method comprising incorporating in the dispersion from 0.1 to 2.5 weight percent of clay thickening agent and from 0.08 to 0.4 weight percent of a long chain aliphatic C₁₂-C₂₂ fatty acid, whereby said composition has a viscosity at low shear conditions which is substantially higher than the viscosity at low shear conditions of the composition without the long chain fatty acid.

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COLGATE-PALMOLIVE COMPANY
Patent Attorneys for the
Applicant:

F.B. RICE & CO.

