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(54) Title: NOVEL PHARMACEUTICAL FORMULATIONS

(57) Abstract: The present invention relates to an oral pharmaceutical formulation, suitably tablets, suitably capsules, comprising an active pharmaceutical ingredient (API) and at least one micelle or liposome or microemulsion forming agent, wherein the active pharmaceutical ingredient is not 3'-[1-(3,4-dimethylphenyl)-1,5-dihydro-3-methyl-5-oxo-4H-pyrazol-4-ylidene]hydrazino]-2'-hydroxy-[1,1'-biphenyl]-3-carboxylic acid (INN name eltrombopag) or a pharmaceutically acceptable salt thereof. Particularly the API is not eltrombopag monoethanolamine or eltrombopag bis-(monoethanolamine).

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Novel Pharmaceutical Formulations

FIELD OF THE INVENTION

The present invention relates to an oral pharmaceutical formulation, suitably tablets, suitably capsules, comprising an active pharmaceutical ingredient (API) and at least one micelle or liposome or microemulsion forming agent, wherein the active pharmaceutical ingredient is not 3'-[(2Z)-[1-(3,4-dimethylphenyl)-1,5-dihydro-3-methyl-5-oxo-4H-pyrazol-4-ylidene]hydrazino]-2'-hydroxy-[1,1'-biphenyl]-3-carboxylic acid (INN name eltrombopag) or a pharmaceutically acceptable salt thereof. Particularly the API is not eltrombopag monoethanolamine or eltrombopag bis-(monoethanolamine).

BACKGROUND OF THE INVENTION

As reported in Promacta drug label, an open-label, randomized-Sequence, crossover trial was conducted to assess the effect of food on the bioavailability of eltrombopag. A standard high-fat breakfast significantly decreased plasma eltrombopag AUC_{0-∞} by approximately 59% and C_{max} by 65% and delayed T_{max} by 1 hour. The calcium content of this meal may have also contributed to this decrease in exposure. In a second trial, administration of a single 25-mg dose of eltrombopag for oral suspension to adults with a high-calcium, moderate-fat, moderate-calorie meal reduced plasma eltrombopag AUC_{0-∞} by 75% (90% CI: 71%, 88%) and C_{max} by 79% (90% CI: 76%, 82%). Administration of a single 25-mg dose of eltrombopag for oral suspension 2 hours after the high-calcium meal reduced plasma eltrombopag AUC_{0-∞} by 47% (90% CI: 40%, 53%) and C_{max} by 48% (90% CI: 40%, 54%). Administration of a single 25-mg dose of eltrombopag for oral suspension 2 hours before the high-calcium meal reduced plasma

eltrombopag AUC_{0-∞} by 20% (90% CI: 9%, 29%) and C_{max} by 14% (90% CI: 2%, 25%).

This is because eltrombopag chelates with coordinating metals, especially with calcium, and forms insoluble complex. As a result eltrombopag formulation has greatly reduced dissolution rate in the presence of calcium (example 8, WO/2008/136843). Hence eltrombopag has reduced bioavailability in the presence of food, especially calcium-rich food (negative food effect).

Generally, negative food effect is associated with BCS Class III drugs (high solubility and poor permeability; Reference: <https://cuvillier.de/de/shop/publications/6557>). For such poorly permeable drugs, absorptive transporter effects predominate and in presence of food the transporters are inhibited leading to negative food effect. However eltrombopag is low soluble/medium to high permeable. The mechanism for the negative food effect is predominantly related to the property of the drug that is, its tendency to chelate with polyvalent cations. It should be noted that the drop in bioavailability is seen only when the meals are fortified with high levels of calcium while there is negligible drop with low levels of calcium (Daphne D. Williams et. al. Clinical Therapeutics/Volume 31, Number 4, 2009).

Solubility enhancement with surfactant is generally used as means of mitigating positive food effect of poorly soluble drugs. For example the bioavailability of abiraterone increases with food. After a low fat meal, C_{max} and AUC are elevated 7- and 5-fold compared with the fasted state, whereas after a high fat meal there is a 17- and 10- fold elevation. WO2013/164473 teaches to mitigate food effect by including in the abiraterone formulation one or more lipid excipients and “The majority of these lipid excipients also have surfactant characteristics and many function to improve both the solubility and permeability of abiraterone”. Surfactant does not seem to be an effective means to address negative food effect issue, especially when the food effect is not due to low solubility of the compound, rather due to the complexing property of the compound with polyvalent cations, especially calcium in the meal.

Antibiotics like Tetracyclines have been well-known in literature for their interactions with Calcium and the reduction in bioavailability when co-administered with foods containing calcium. In its general statement concerning tetracyclines, the AHFS Drug Information notes that the absorption of tetracycline is reduced by 50% or more in the presence of food or milk, while the absorption of doxycycline may be reduced by up to 20% by food or milk. Meyer et al. examined the pharmacokinetics of doxycycline in nine healthy volunteers who received capsules containing 200 mg of doxycycline with 300 ml of milk or 300 ml of water. The authors of this study reported that simultaneous administration of milk diminished the peak plasma concentration of doxycycline by 24% and the absorption by 9%-35%. The authors concluded that, like other tetracyclines, doxycycline should not be administered together with milk. (Ref: Meyer, F.P., Specht, H., Quednow, B. et al. Influence of milk on the bioavailability of doxycycline — new aspects. *Infection* 17, 245–246 (1989). <https://doi.org/10.1007/BF01639529>).

Similar to tetracyclines, Quinolone e.g, Norfloxacin, ciprofloxacin, enoxacin, lomefloxacin, also show interaction with antacids containing aluminium and magnesium ions and milk products within gut. A study in 7 healthy subjects found that 300 mL of milk or yoghurt reduced the peak plasma levels of a single 500-mg dose of ciprofloxacin by 36% and 47%, respectively, and reduced its AUC by 33% and 36%, respectively (Neuvonen PJ, Kivistö KT, Lehto P. Interference of dairy products with the absorption of ciprofloxacin. *Clin Pharmacol Ther* (1991) 50, 498–50). In another study, 300 mL of milk reduced the AUC of ciprofloxacin 500 mg by about 30% (Hoogkamer JFW, Kleinbloesem CH. The effect of milk consumption on the pharmacokinetics of fleroxacin and ciprofloxacin in healthy volunteers. *Drugs* (1995) 49 (Suppl 2), 346–8). Accordingly, the FDA labeling information states restriction on the consumption of antacids, metal cations two hour before or 6 hour after ciprofloxacin dosing (https://www.accessdata.fda.gov/drugsatfda_docs/label/2016/019537s086lbl.pdf).

BRIEF DESCRIPTION OF THE DRAWINGS

Dissolution tests were performed according to Example 4 and some of the results were shown in the figures below.

Figure 1: Comparison of ETB115 DS, Promacta tablets and Vit E TPGS formulation (formulation 1 in 75mg) in the presence of 427mg or 450mg of calcium or control (absence of calcium), following the dissolution test described in EXAMPLE 4.

Figure 2A: Comparison of ETB115 DS in different drug load with Vit E TPGS formulation (20% vs. 24% in 75mg) in the presence of 427mg of calcium or control (absence of calcium)

Figure 2B: Comparison of ETB115 DS in lower drug load (6%) with Vit E TPGS formulation in the presence of 427mg of calcium or control (absence of calcium).

Figure 3A: Dissolution in presence of different concentration of Vit E TPGS in MOPS buffer calcium added 30 min before (Figure 3A) or 60 minutes after (Figure 3B) Promacta tablets in dissolution bowl.

Figure 4A: ETB115 DS with RH40 (formulation 9 in 50mg) in the presence of 427mg of calcium or control (absence of calcium)

Figure 4B: ETB115 DS with MEPC 7 (formulation 11) in the presence of 50 mg or 450mg of calcium or control (absence of calcium)

Figure 4C: ETB115 DS with Gellucire 48/16 in the presence of 427mg of calcium or control (absence of calcium)

Figure 4D: ETB115 DS with MEPC 3 (formulation 15) in the presence of calcium or control (absence of calcium)

Figure 5 Effect of surfactants with increasing value of HLB on the ETB115 dissolution in the absence (Figure 5A) and presence of 427mg calcium (Figure 5B)

Figure 6 Dissolution tests of lipid formulations of F2, F3 and F4

Figure 7 PAMPA test. ETB115 dissolution and permeation assay indicated as Fasted + High Calcium/Fasted flux ratio.

Figure 8 Comparison of doxycycline drug substance, doxycycline and Vit E TPGS formulation in the presence of 427mg calcium or control (absence of calcium), following the dissolution test described in EXAMPLE 10.

DESCRIPTION OF THE INVENTION

Surprisingly we have found that surfactant vitamin E TPGS can effectively mitigate the food effect on eltrombopag, i.e. mitigate the reduction of bioavailability in the presence of food, especially calcium rich food. We have found that vitamin E TPGS can effectively

- a) prevent eltrombopag from binding to polyvalent cations (e.g. calcium);
- b) prevent the formation of insoluble complex of eltrombopag and polyvalent cations (e.g. calcium);
- c) release soluble eltrombopag into the medium despite the presence of excessive amount of polyvalent cations (e.g. calcium), typically shown in an in vitro dissolution test;
- d) increase the solubilization of eltrombopag, typically shown in an in vitro dissolution test;
- e) prevent soluble eltrombopag from crashing out of the medium, typically despite the presence of excessive amount of polyvalent cations (e.g. calcium), typically shown in an in vitro dissolution test; and/or
- f) increase the permeation of eltrombopag through Flux in-vitro test PAMPA

Any one of the above effects or any combination thereof can be generally called the anti-calcium effect in this application.

While the above anti-calcium effect achieved with the invention was firstly demonstrated with eltrombopag, it is scientifically reasonable to expect that this finding is generally applicable to other APIs, having similar properties as eltrombopag (API of the Invention). Thus this invention would be useful, inter alia, in the field of pharmaceutical formulation, especially for APIs having food effect,

especially when the food effect is likely caused by the presence of polyvalent metals, such as calcium in the meal.

The present patent application carves out subject matter relating to eltrombopag, which was disclosed in the international patent application PCT/US2020/051593. Any mention of eltrombopag in the present patent application is for illustration purpose and any data relating to eltrombopag is to support the general inventive concept and to support the technical effects of the invention.

One property is that the API of the Invention is capable of chelating with polyvalent metal cations and form an API-metal complex. Polyvalent metals include but not limited to aluminum, calcium, copper, cobalt, gold, iron, bismuth, magnesium, manganese selenium and zinc. API of the Invention is capable of chelating with at least one polyvalent metal. Suitably the polyvalent metal is common in diet, including dietary supplement, such as calcium, iron, magnesium, manganese, copper, selenium and zinc. Suitably API of the Invention is capable of chelating with calcium.

One property of API is to have functional chemical groups in a configuration which results in formation of API-metal complex. Functional groups to which metal cation bind include but not limited to phenolic, ketone, catechol, carboxyl, amine, hydroxyimino and sulphhydryl groups. The functional groups have electron pairs which interact with the metal ion to form complex. API-metal complexation can include one or more functional group of one or two or multiple API molecules. Generally, a stable API-metal binding results in formation of five or six membered ring structure.

The term "API-metal complex" as used herein, is well understood by a skilled person in the field. Typically the binding affinity between the functional group and the metal is characterized by an association constant/formation constant/stability constant. It is an equilibrium constant and is a measure of strength of interaction between the API, i.e. ligand (L) and the metal ion (M). An API can form multiple complexes as ML, ML₂, ML₃ and likewise. Formation of each complex species is characterized by a constant also known as stepwise stability

constant e.g. K_1 , K_2 , K_3 and likewise. The overall complex formation constant K is equal to $K_1 * K_2 * K_3 \dots$ likewise. As would be apparent to a skilled person the measurement of formation constant is tedious and is affected by multiple factors i.e. metal ion, ionic charge, ligand/ion concentration, pH and other ions e.g. H^+ in the solution. Basis the measurement conditions the K value changes. Higher the value of stability constant, stronger is the complex. API that could form API-metal complex in the stomach, i.e. under acidic pH, e.g. $pH \leq 2$, is particularly included as API of the present Invention.

Alternatively, the protection of API under acidic condition could be extended to relatively basic conditions of the intestine, which is downstream to the stomach and is major site of drug absorption for most drugs.

The formed API-metal complex can either have reduced solubility/dissolution or reduced permeability or both, compared to the API alone, in the physiological conditions of absorption. Alternatively, the API-metal complex can have high solubility but reduced permeability in the physiological conditions of absorption. In general due to the API-metal chelating the bioavailability is reduced. Alternatively a medicament comprising API of the invention has negative food effect when taken together with food, especially calcium rich food, or vitamin/mineral supplement.

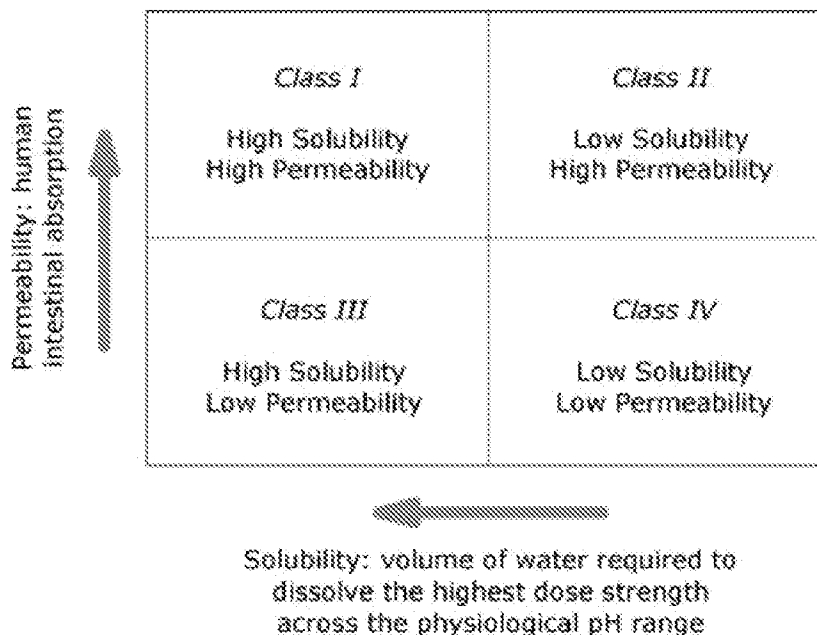
One additional property is that the API of the Invention has low solubility. The term "low solubility", as used herein and hereunder, is understood as below 100mg/ml, even more below 50mg/ml, even more below 25mg/ml, even more below 10mg/ml, even more below 2mg/ml in an aqueous medium. Suitably the API has a solubility at least above 0.05mg/ml. Suitably the solubility is tested under neutral pH condition, typically in aqueous medium.

One additional or alternative property is that the API has low solubility tested in acidic pH condition, typically in aqueous medium.

One additional or alternative property is that the API has high permeability.

One additional or alternative property is that the API is a class I, II, III or IV compound. Typically the API of the Invention is a class II compound or a class IV compound.

The API could belong to BCS class I, II, III or IV basis the API solubility and permeability. BCS classes of the API's can be defined as, Class I: High Solubility – High Permeability Class II: Low Solubility – High Permeability Class III: High Solubility – Low Permeability Class IV: Low Solubility – Low Permeability Reference (references: Waiver of In Vivo Bioavailability and Bioequivalence Studies for Immediate-Release Solid Oral Dosage Forms Based on a Biopharmaceutics Classification System Guidance for Industry (U.S. Department of Health and Human Services Food and Drug Administration Center for Drug Evaluation and Research (CDER)), December 2017, Biopharmaceutics) Here, a drug substance is considered highly soluble when the highest strength is soluble in 250 mL or less of aqueous media within the pH range of 1 - 6.8 at $37 \pm 1^\circ\text{C}$. High permeability refers to a condition when the systemic BA or the extent of absorption in humans is determined to be 85 percent or more of an administered dose based on a mass balance determination (along with evidence showing stability of the drug in the GI tract) or in comparison to an intravenous reference dose.



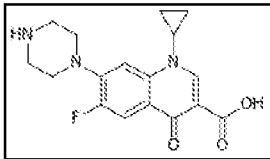
Reference for the image - The AAPS Journal, Vol. 11, No. 4, December 2009 (# 2009) DOI: 10.1208/s12248-009-9144-x

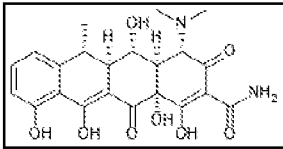
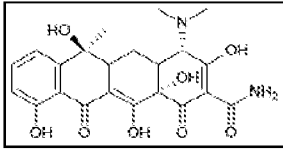
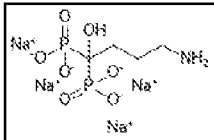
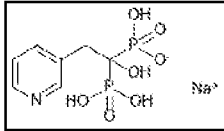
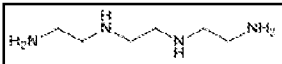
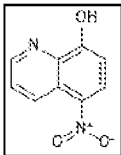
One additional or alternative property is that API could have Dose number of ≥ 1 . Where the dose number is the ratio of the Dose to the amount of the compound that can be dissolved in the dose Volume (default is 250 mL in human) of fluid at the pH of lowest solubility between pH 1.0 and pH 6.8. If the Dose Number is equal to or greater than 1.0 it means dose is not completely soluble, and if the value is below 1.0 it indicates dose is completely soluble. Alternatively, the API could have Dose number below 1.

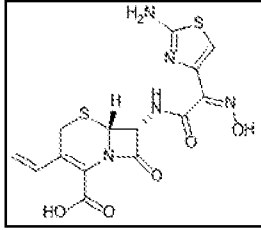
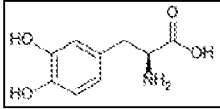
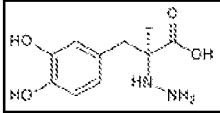
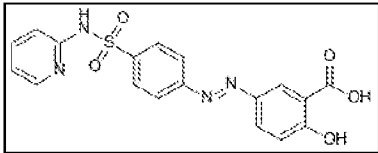
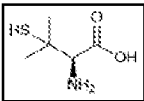
APIs have the chelating property alone or in combination with any one or more of the other properties are envisaged to benefit from the inclusion of Vitamin E TPGS to overcome the food effect due to the presence of polyvalent metal cations, especially calcium in the meal.

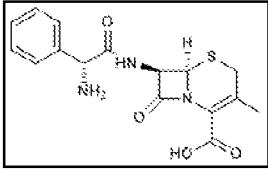
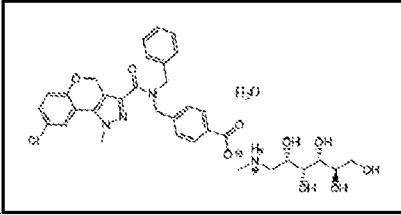
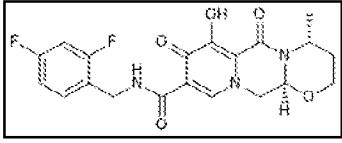
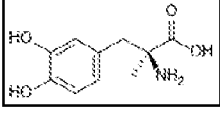
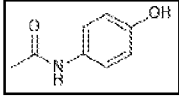
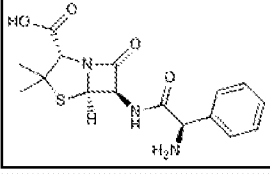
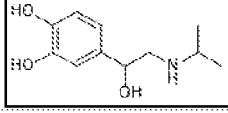
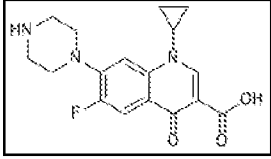
Suitably API of the Invention includes but not limited to the compounds listed in Table 1.

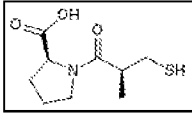
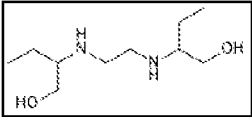
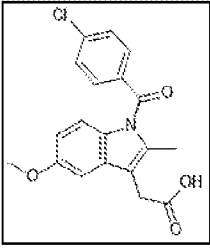
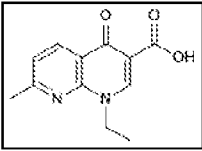
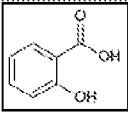
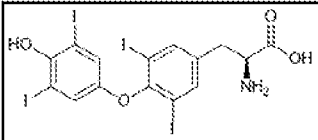
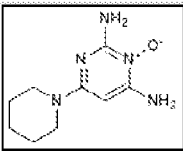
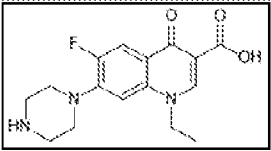
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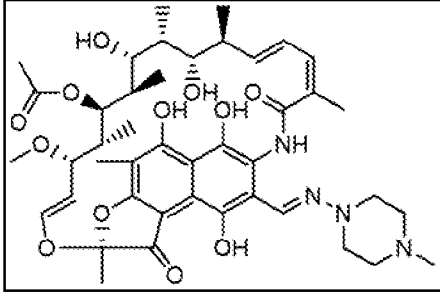
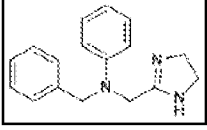
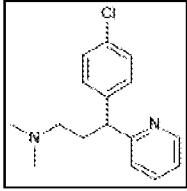
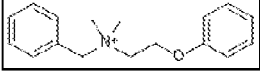
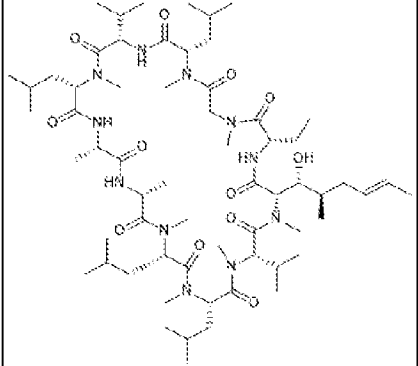
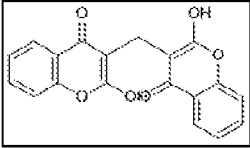
Name of the drug substance	Structure
Ciprofloxacin	 <p>The chemical structure of Ciprofloxacin is shown within a rectangular frame. It features a central quinolone ring system. At position 7, there is a piperazine ring. At position 8, there is a cyclopropyl ring. At position 6, there is a fluorine atom. At position 4, there is a carboxylic acid group (-COOH). The quinolone ring has a carbonyl group at position 2 and another at position 3.</p>

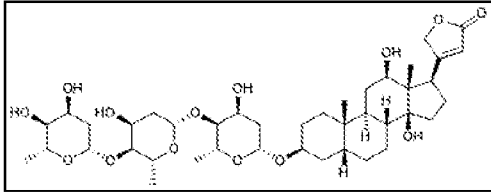
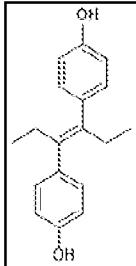
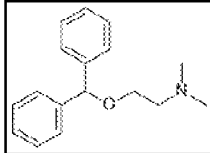
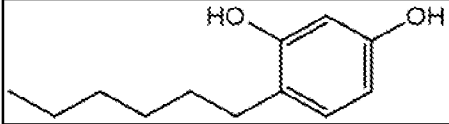
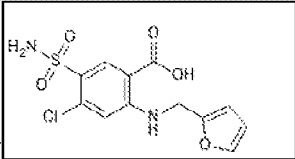
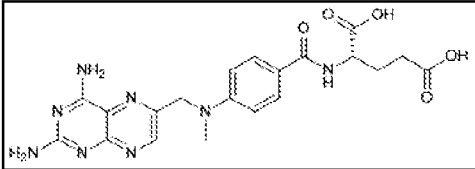
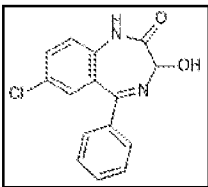
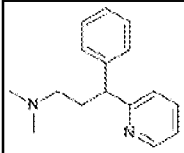
<p>Doxycycline</p>	 <p>The image shows the chemical structure of Doxycycline, a tetracycline antibiotic. It features a tetracyclic core with a dimethylamino group at C4, a hydroxyl group at C5, and a dimethylamino group at C12. The C11 position is substituted with a hydroxyl group and a dimethylamino group. The C10 position is substituted with a hydroxyl group and a dimethylamino group. The C7 position is substituted with a hydroxyl group and a dimethylamino group. The C6 position is substituted with a hydroxyl group and a dimethylamino group. The C5 position is substituted with a hydroxyl group and a dimethylamino group. The C4 position is substituted with a hydroxyl group and a dimethylamino group. The C12 position is substituted with a hydroxyl group and a dimethylamino group.</p>
<p>Tetracycline</p>	 <p>The image shows the chemical structure of Tetracycline, a tetracycline antibiotic. It features a tetracyclic core with a dimethylamino group at C4, a hydroxyl group at C5, and a dimethylamino group at C12. The C11 position is substituted with a hydroxyl group and a dimethylamino group. The C10 position is substituted with a hydroxyl group and a dimethylamino group. The C7 position is substituted with a hydroxyl group and a dimethylamino group. The C6 position is substituted with a hydroxyl group and a dimethylamino group. The C5 position is substituted with a hydroxyl group and a dimethylamino group. The C4 position is substituted with a hydroxyl group and a dimethylamino group. The C12 position is substituted with a hydroxyl group and a dimethylamino group.</p>
<p>Alendronate Sodium</p>	 <p>The image shows the chemical structure of Alendronate Sodium, a bisphosphonate. It consists of a central carbon atom bonded to two phosphonate groups and a propylamine group. The phosphonate groups are shown as Na^+ O^- $\text{P}(\text{OH})_2$ and Na^+ O^- $\text{P}(\text{OH})_2$. The propylamine group is $\text{CH}_2\text{CH}_2\text{CH}_2\text{NH}_2$.</p>
<p>Risedronate sodium</p>	 <p>The image shows the chemical structure of Risedronate sodium, a bisphosphonate. It consists of a central carbon atom bonded to two phosphonate groups and a pyridine ring. The phosphonate groups are shown as Na^+ O^- $\text{P}(\text{OH})_2$ and Na^+ O^- $\text{P}(\text{OH})_2$. The pyridine ring is attached to the central carbon atom.</p>
<p>Trientin</p>	 <p>The image shows the chemical structure of Trientin, a triazine derivative. It consists of a central carbon atom bonded to three nitrogen atoms, each of which is further bonded to a propylamine group ($\text{H}_2\text{N}-\text{CH}_2-\text{CH}_2-\text{CH}_2-\text{NH}_2$).</p>
<p>Nitroxoline</p>	 <p>The image shows the chemical structure of Nitroxoline, a pyridine derivative. It consists of a pyridine ring with a hydroxyl group at the 2-position and a nitro group at the 4-position.</p>

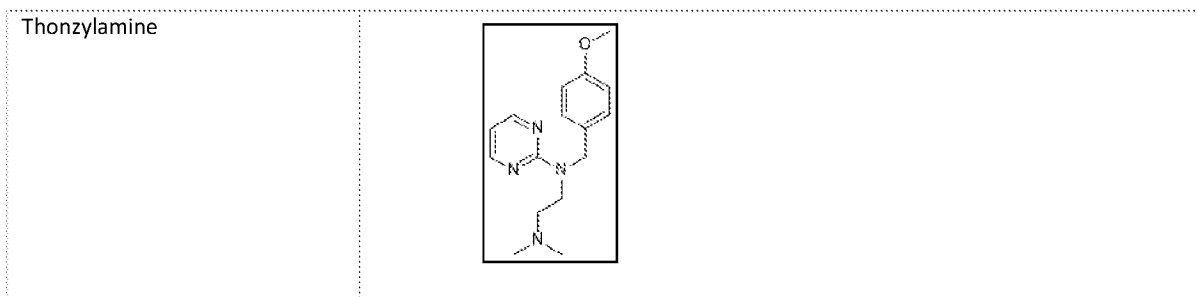
<p>Cefdinir</p>	 <p>The image shows the chemical structure of Cefdinir, a third-generation cephalosporin antibiotic. It features a central six-membered beta-lactam ring fused to a five-membered thiazolidine ring. A methyl group is attached to the thiazolidine ring, and a carboxylic acid group is attached to the beta-lactam ring. A side chain is attached to the 3-position of the beta-lactam ring, consisting of a methylene group, a methylene group, and a 5-amino-1,3,4-thiazolidine-2-ylidene group.</p>
<p>Levodopa</p>	 <p>The image shows the chemical structure of Levodopa (L-DOPA), a catecholamine neurotransmitter. It consists of a benzene ring with two hydroxyl groups at the 3 and 4 positions (catechol group). A propyl chain is attached to the benzene ring, with an amino group at the 2-position and a carboxylic acid group at the 1-position.</p>
<p>Carbidopa</p>	 <p>The image shows the chemical structure of Carbidopa (2S,3S)-2-(3,4-dihydroxyphenyl)propan-1-amine hydrochloride. It is a dihydroxyphenylethylamine derivative. The benzene ring has hydroxyl groups at the 3 and 4 positions. A propyl chain is attached to the benzene ring, with an amino group at the 2-position and a carboxylic acid group at the 1-position.</p>
<p>Sulfasalazine</p>	 <p>The image shows the chemical structure of Sulfasalazine, an anti-inflammatory drug. It consists of a pyridine ring connected via a sulfonamide group to a benzene ring. This benzene ring is further connected via an azo group to another benzene ring, which has a carboxylic acid group and a hydroxyl group at the 4 and 5 positions, respectively.</p>
<p>Penicillamine</p>	 <p>The image shows the chemical structure of Penicillamine, a chelating agent. It consists of a central carbon atom bonded to a methyl group, a propyl group, an amino group, and a carboxylic acid group.</p>

<p>Cefalexin</p>	 <p>The image shows the chemical structure of Cefalexin, a first-generation cephalosporin. It features a six-membered beta-lactam ring fused to a five-membered dihydrothiazolidine ring. A phenylacetamido group is attached to the 7-position of the beta-lactam ring, and a methyl group is attached to the 6-position of the dihydrothiazolidine ring. A carboxylic acid group is attached to the 4-position of the dihydrothiazolidine ring.</p>
<p>LMB763</p>	 <p>The image shows the chemical structure of LMB763, a nucleoside analog. It consists of a pyrimidine ring system with a chlorine atom at the 5-position and a methyl group at the 6-position. The 2-position is substituted with a side chain containing a carbonyl group, a methylene group, and a chiral center with a hydroxyl group. This side chain is further linked to a ribose sugar moiety.</p>
<p>Dolutegravir</p>	 <p>The image shows the chemical structure of Dolutegravir, a third-generation integrase strand transfer inhibitor. It features a central pyrimidine ring with a hydroxyl group at the 2-position and a methyl group at the 4-position. The 6-position is substituted with a side chain containing a carbonyl group, a methylene group, and a chiral center with a hydroxyl group. This side chain is further linked to a piperazine ring system.</p>
<p>Methyldopa</p>	 <p>The image shows the chemical structure of Methyldopa, a sympathomimetic amine. It consists of a benzene ring with two hydroxyl groups at the 3 and 4 positions. The 1-position is substituted with a side chain containing a methylene group and a chiral center with an amino group and a carboxylic acid group.</p>
<p>Paracetamol</p>	 <p>The image shows the chemical structure of Paracetamol, an analgesic and antipyretic. It consists of a benzene ring with a hydroxyl group at the 4-position and an acetamido group at the 1-position.</p>
<p>Ampicillin</p>	 <p>The image shows the chemical structure of Ampicillin, a penicillin antibiotic. It features a four-membered beta-lactam ring fused to a five-membered thiazolidine ring. A side chain containing a carbonyl group, a methylene group, and a chiral center with a hydroxyl group is attached to the 6-position of the beta-lactam ring. The 2-position of the thiazolidine ring is substituted with a methyl group and a methylamino group.</p>
<p>Isoprenaline</p>	 <p>The image shows the chemical structure of Isoprenaline, a beta-adrenergic agonist. It consists of a benzene ring with two hydroxyl groups at the 3 and 4 positions. The 1-position is substituted with a side chain containing a methylene group, a hydroxyl group, and a chiral center with an isopropylamino group.</p>
<p>Ciprofloxacin</p>	 <p>The image shows the chemical structure of Ciprofloxacin, a fluoroquinolone antibiotic. It features a central quinolone ring system with a fluorine atom at the 6-position and a carboxylic acid group at the 4-position. The 7-position is substituted with a side chain containing a piperazine ring and a cyclopropyl group.</p>

Captopril	 <chem>CC1(C)CCN1C(=O)[C@H](C)C(S)C</chem>
Ethambutol	 <chem>CC(O)CNCCNCC(O)C</chem>
Indomethacin	 <chem>CC1=C(C(=O)C2=CC=C(C=C2)Cl)N3C(=O)C=C(C(=O)O)C=C3OC1</chem>
Nalidixic	 <chem>CCN1C(=O)C=C2C(=O)N=C(C)N=C2C1C(=O)O</chem>
Salicylic Acid	 <chem>O=C(O)c1ccccc1O</chem>
Thyroxine	 <chem>NC(C(=O)O)Cc1ccc(I)c(O)c1Oc2ccc(I)c(O)c2I</chem>
Minoxidil	 <chem>NC1=NC(=NC(=N1)N)N2CCCCN2</chem>
Norfloxacin	 <chem>CCN1C(=O)C=C2C(=O)N=C(C)N=C2C(=O)O</chem>

<p>Rifampicine</p>	 <p>The image shows the chemical structure of Rifampicine, a complex polycyclic molecule with multiple hydroxyl groups, a piperazine ring, and a decalin core.</p>
<p>Antazoline</p>	 <p>The image shows the chemical structure of Antazoline, featuring a piperazine ring connected to a benzyl group and a 2-phenylethyl group.</p>
<p>Chlorpheniramine</p>	 <p>The image shows the chemical structure of Chlorpheniramine, consisting of a piperazine ring linked to a 2-chlorophenyl group and a 2-pyridyl group.</p>
<p>Bephenium</p>	 <p>The image shows the chemical structure of Bephenium, which is a quaternary ammonium salt with a benzyl group and a 2-(benzyloxy)ethyl group attached to the nitrogen atom.</p>
<p>Cyclosporine</p>	 <p>The image shows the chemical structure of Cyclosporine, a large cyclic peptide consisting of 11 amino acid residues linked by amide bonds.</p>
<p>Dicumarol</p>	 <p>The image shows the chemical structure of Dicumarol, a dimeric coumarin compound with two coumarin units linked at the 3-position.</p>

<p>Digoxin</p>	 <p>The image shows the chemical structure of Digoxin, a cardiac glycoside. It consists of a digitoxigenin aglycone core with three digitoxose sugar units attached to the 3, 6, and 12 positions of the aglycone.</p>
<p>Diethylstilbestrol</p>	 <p>The image shows the chemical structure of Diethylstilbestrol, a synthetic estrogen. It features a central stilbene core with two ethyl groups on the double bond and two hydroxyl groups on the phenyl rings.</p>
<p>Diphenhydramine</p>	 <p>The image shows the chemical structure of Diphenhydramine, an antihistamine. It consists of a central carbon atom bonded to two phenyl rings, a dimethylaminoethyl group, and a hydroxyl group.</p>
<p>Hexylresorcinol</p>	 <p>The image shows the chemical structure of Hexylresorcinol, a preservative. It consists of a benzene ring with two hydroxyl groups at the 1 and 3 positions and a hexyl chain at the 4 position.</p>
<p>Furoseamide</p>	 <p>The image shows the chemical structure of Furoseamide, a diuretic. It features a benzimidazole core with a chlorine atom, a hydroxyl group, and a furfuryl group.</p>
<p>Methotrexate</p>	 <p>The image shows the chemical structure of Methotrexate, an antifolate drug. It consists of a pteridine ring system with an amino group, a methyl group, and a methoxy group, and a glutamic acid side chain.</p>
<p>Oxazepam</p>	 <p>The image shows the chemical structure of Oxazepam, a benzodiazepine. It features a seven-membered oxazepine ring fused to a benzene ring, with a chlorine atom and a phenyl group.</p>
<p>Pheniramine</p>	 <p>The image shows the chemical structure of Pheniramine, an antihistamine. It consists of a central carbon atom bonded to a phenyl ring, a piperazine ring, and a hydroxyl group.</p>



Thus the present invention relates to a pharmaceutical composition (pharmaceutical composition of the invention), preferably in an oral dosage form, comprising an API (API of the invention) or a pharmaceutically acceptable salt thereof and Vitamin E TPGS.

Vitamin E TPGS is chemically D-alpha-tocopheryl polyethylene glycol 1000 succinate. It is an amphipathic and hydrophilic surface active agent. It typically has a low critical micelle concentration about 0.02% w/w. It is traditionally being used as emulsifier, solubilizer, absorption enhancer and vehicle for lipid based drug delivery formulations. [Vitamin E TPGS as a molecular biomaterial for drug delivery <https://doi.org/10.1016/j.biomaterials.2012.03.046>].

Compound is in the free form (acidic or basic), solvate form or in a pharmaceutically acceptable salt form. The term “weight of the API” as used in the context of the present invention refers to the weight of the free form, unless specified otherwise.

In one embodiment the pharmaceutical composition of the invention comprises an API of the Invention, solvate, or a pharmaceutically acceptable salt thereof and vitamin E TPGS, wherein the weight of the API is not more than 80%, not more than 60%, suitably not more than 40%, suitably not more than 30%, suitably not more than 25%, suitably not more than 20% of the total weight of the API and vitamin E TPGS. In one embodiment the weight of API is not more than 30% of the total weight of API and vitamin E TPGS.

By way of example, formulation 1 in Example 1 consists of 95.6mg of eltrombopag bis-monoethanolamide, corresponding to 75mg of eltrombopag, and 382mg of vitamin E TPGS, then the weight of eltrombopag of the total weight of eltrombopag and vitamin E TPGS is 16.4% ($75/(75+382)$).

In one embodiment the weight of API is at least 2%, suitably at least 4%, suitably at least 5%, suitably at least 10% of the total weight of API and vitamin E TPGS. In one embodiment the weight of API is at least 4% of the total weight of API and vitamin E TPGS.

In one embodiment the pharmaceutical composition comprises API, solvate or a pharmaceutically acceptable salt thereof and vitamin E TPGS, wherein the weight of API is from 2% to 50%, suitably from 4% to 30%, suitably from 5% to 25%, suitably from 5 to 20% of the total weight of API and vitamin E TPGS.

In one embodiment vitamin E TPGS is the only surfactant in the pharmaceutical composition of the invention.

In one embodiment the pharmaceutical composition of the invention, preferably in an oral dosage form, consists essentially of or consists of API, solvate or a pharmaceutically acceptable salt thereof and vitamin E TPGS.

Besides vitamin E TPGS, additional surfactants and/or lipids could be added to the pharmaceutical composition of the invention. Typically one or two additional surfactant could be added. Alternatively or additionally one or two lipid could be added. Typically one additional surfactant could be added. Alternatively or additionally one lipid could be added. In one embodiment the pharmaceutical composition of the invention comprises API, solvate or a pharmaceutically acceptable salt thereof, vitamin E TPGS, Span 80, miglyol 812N, Labrasol.

In one embodiment the pharmaceutical composition of the invention comprises API, solvate or a pharmaceutically acceptable salt thereof, vitamin E TPGS and at least one more pharmaceutically acceptable excipients.

In one embodiment the at least one more pharmaceutically acceptable excipients include an anti-oxidant. In one embodiment the weight of the anti-oxidant is not more than 10%, suitably not more than 7%, suitably not more than 5%, suitably not more than 3%, suitably not more than 1%, of the total weight of the pharmaceutical composition. Preferably there is only one anti-oxidant in the composition. In one embodiment the anti-oxidant is selected from the list consisting of Vitamin E, Butylhydroxytoluol (BHT), Butylhydroxyanisol (BHA), Propyl

gallate, ascorbyl palmitate, ascorbic acid, EDTA and sodium metabisulfite or a mixture thereof, suitably 2 of the anti-oxidants from the list, preferably only one anti-oxidants from the list. In one embodiment the anti-oxidant is vitamin E. In one embodiment vitamin E is not more than 15%, suitably not more than 7%, suitably not more than 5% of the total weight of the pharmaceutical composition. In one embodiment vitamin E is 2-15%, suitably 2-10%, suitably 5% of the total weight of the pharmaceutical composition. In one embodiment the anti-oxidant is BHT. In one embodiment BHT is not more than 3%, suitably not more than 1%, suitably not more than 0.5%, suitably not more than 0.2%, suitably not more than 0.1 % of the total weight of the pharmaceutical composition. In one embodiment the anti-oxidant is BHA. In one embodiment BHA is not more 3%, suitably not more than 1%, suitably not more than 0.5%, suitably not more than 0.2% of the total weight of the pharmaceutical composition. In one embodiment the anti-oxidant is Propyl gallate. In one embodiment Propyl gallate is not more than 3%, suitably not more than 1%, suitably not more than 0.5% of the total weight of the pharmaceutical composition. In one embodiment the anti-oxidant is EDTA. In one embodiment weight of EDTA is not more than 10%%, suitably not more than 5%, suitably not more than 2% of the total weight of the pharmaceutical composition. In one embodiment EDTA is 1-5%, suitably 1-3%, suitably 2% of the total weight of the pharmaceutical composition. In one embodiment EDTA is in the form of disodium salt and the weight of EDTA is calculated based on the weight of EDTA disodium.

In one embodiment the pharmaceutical composition consists essentially of or consists of an API, a solvate or a pharmaceutically acceptable salt thereof, vitamin E TPGS and only one anti-oxidant.

There exists trace amount of vitamin E (about 1.5%) in vitamin E TPGS of GMP standard. This amount of vitamin E is generally regarded as impurities unless otherwise specified in this application.

In one preferred embodiment the anti-oxidant is EDTA.

In one embodiment the pharmaceutical composition comprises an API or a pharmaceutically acceptable salt thereof, vitamin E TPGS and EDTA. In one

embodiment the pharmaceutical composition consists essentially of or consists of an API, vitamin E TPGS and EDTA.

Without wishing to be bound by the theory, the effect of vitamin E TPGS could be attributed to partial solubilization of the API in Vitamin E TPGS, which upon contact with aqueous media forms micelles and thus minimizes the interaction of polyvalent cations, e.g. calcium, with the drug.

Eltrombopag, even in the form of bis-monoethanolamine salt, has low solubility in water as well as in a number of liquid/semisolid surfactants (data not shown). However it has been found that other surfactants/lipids also exhibit anti-calcium effect. Without wishing to be bound by the theory, this anti-calcium effect could be plausibly attributed to solubilization or partial solubilization of eltrombopag in such surfactants/lipids containing formulation, which upon contact with aqueous media forms colloids or vesicles, such as micelles or liposomes or microemulsion, and thus minimizes the interaction of polyvalent cations, e.g. calcium, with the drug.

Thus the present invention relates to a pharmaceutical composition (pharmaceutical composition of the invention), preferably in an oral dosage form, comprising an API or a pharmaceutically acceptable salt thereof and at least one colloid or vesicle forming agent. The term “vesicle” or “colloid”, as used here, can be broadly understood as spherical or non-spherical structures formed by amphiphilic molecules in an aqueous medium. The term “at least one colloid or vesicle forming agent” includes at least one micelle or liposome or microemulsion forming agent. Thus, the pharmaceutical composition of the invention comprises eltrombopag or a pharmaceutically acceptable salt thereof and at least one micelle or liposome or microemulsion forming agent. A micelle forming agent, liposome or microemulsion forming agent are not mutually exclusive. Some agents can form either micelle or liposome or microemulsion depending on the process or on the presence of other components in the medium.

Micelle is generally understood as a spherical structure with diameter about 1 to about 50 nm, about 1 to about 30 nm, about 1- about 20nm, about 1- about 10nm, formed with a single layer of amphiphilic molecules with the hydrophilic head directing towards the aqueous phase outside and the lipophilic tails forming a lipophilic compartment at the inside. Liposome is generally a bigger spherical structure with diameter about 30 to about 10000 nm, formed by one or more lipid bilayer(s) surrounding an aqueous inner core. Suitably the vesicle of the present invention is in the size range of about 1- about 5000 nm, about 1- about 3000 nm, about 1 –about 1000 nm, about 5-about 500 nm, about 5-about 100nm, about 5-about 30nm. The term colloidal-particle is understood as particles in the size range of about 5 to 10000nm which could be non-spherical and could be single layer or more layers. These colloidal-particles could comprise of polymers alone or in combination of surfactants and lipids. Microemulsion is generally understood as dispersed droplets in the size range of about 10 to about 10000 nm, oily droplets stabilized by a surfactant layer.

The term “at least one micelle or liposome forming agent” as used here, refers to an amphiphilic molecule that is capable of forming micelles or liposomes in an aqueous medium. Typically the concentration of the at least one micelle or liposome forming agent comprised by the pharmaceutical composition of the present invention is above the critical micelle concentration (CMC) or above the critical liposome concentration (CLC), suitably CMC or CLC at $37\pm 0.5^{\circ}\text{C}$ in an aqueous medium or in water, upon release from the composition into the medium. Suitably the aqueous medium is gastric fluid or gastric fluid simulates. Suitably the aqueous medium is small intestine fluid or small intestine fluid simulates. Typically the micelles or liposomes or microemulsions formed by the at least one micelle or liposome or microemulsion forming agent of the invention is capable of preventing or partially prevention the interaction of the API and the polyvalent cations, e.g. calcium, present in the medium.

The in vitro dissolution test as described in EXAMPLE 4 is an easy and effective method of screening suitable micelle or liposome or microemulsion forming agents for the purpose of the present invention. Vitamin E TPGS was

selected through this dissolution test as effective in mitigating the calcium effect on eltrombopag. Such effect was further confirmed in the Macroflux tests (PAMPA assays EXAMPLE 8).

A micelle/liposome/microemulsion forming agent for the purpose of the present invention should be pharmaceutically acceptable. Regulatory health authorities provide guidance of pharmaceutically acceptable excipients (e.g. <https://www.accessdata.fda.gov/scripts/cder/ig/index.cfm>). Furthermore the minimum concentration needed for a particular micelle forming agent for the purpose of the invention, e.g. above CMC in aqueous medium, should not exceed its maximum amount as set forth by the regulatory health authorities. For example 764 mg of Vitamin E TPGS is the maximum daily amount allowed for children older than one 1 year.

In one embodiment the present invention relates to a pharmaceutical composition (pharmaceutical composition of the invention), preferably in an oral dosage form, comprises an API or a pharmaceutically acceptable salt thereof and phospholipids. Preferably the phospholipids are pharmaceutically acceptable.

Phospholipids are surface-active, amphiphilic molecules, which comprise a polar head group and a lipophilic tail. The diacyl-phospholipids (DA-PL) comprise a glycerol backbone, which is esterified in positions 1 and 2 with fatty acids and in position 3 with phosphate, whereas phospholipids with one fatty acid tail are called “monoacyl-phospholipids” (MA-PL) or “lyso-phospholipids”. In typical membrane phospholipids, the phosphate group is further esterified with an additional alcohol, for instance in phosphatidylcholine (PC) with choline, in phosphatidylethanolamine (PE) with ethanolamine, and in phosphatidylglycerol (PG) with glycerol. Depending upon the structure of the polar region and pH of the medium, PE and PC are zwitterionic and have a neutral charge at pH values of about 7, whereas PG is negatively charged. The most common phospholipid is PC, and PC is the main component of lecithin.

Lecithin is described, e.g., in the United States Pharmacopoeia (USP) as a “complex mixture of acetone-insoluble phosphatides, which consists chiefly of PC,

PE, phosphatidylserine, and phosphatidylinositol, combined with various amounts of other substances such as triglycerides, fatty acids, and carbohydrates, as separated from the crude vegetable oil source. It contains not less than 50% of acetone-insoluble matter.” Normally, lecithin grades containing more than 80% PC do not comply anymore with the phamacopoeial definition and are called arbitrarily PC, whereas grades containing less than 80% PC can be arbitrarily called lecithin.

As understood by a skilled person, phospholipids, normally extracted from natural sources, is a mixture in which DA-PL is the predominant species over MA-PL. DA-PL is further a mixture with different phosphatidyl derivatives and different length and saturation of the fatty acids. Here below the two tables indicate the composition of lecithin obtained from different natural sources. Although phospholipids can be chemically synthesized, it is cheaper and environmental friendlier to obtain phospholipids from natural sources.

Table 2. Phospholipid composition of vegetable de-oiled lecithins, as derived from corresponding product specifications (%)

Phospholipid	Lecithin		
	Soybean	Sunflower seed	Rapeseed
PC	20-22	20-26	23-31
PE	16-22	4-10	9-15
PI	13-16	15-19	15-18
PA	5-10	2-5	5-10
LPC	<3	<3	<3

LPC: lyso-phosphotidyl choline

Table 3. Fatty acid composition of typical batches of vegetable deoiled lecithins (area %)

Fatty acid	Lecithin		
	Soybean	Sunflower seed	Rapeseed
C14:0	0.1	0.1	0.1
C16:0	21	16	10
C18:0	4.7	5.3	0.8
C18:1	9.9	21	49
C18:2	57	54	31
C18:3	5.0	0.2	4.4
C20:0	0.1	0.3	0.1
C22:0	0.4	1.5	0.1

In one embodiment the phospholipids is diacyl-phospholipids.

In one embodiment the phospholipids is lecithin.

The company Lipoid (<https://www.lipoid.com/en/node/10>) produces a big variety of phospholipids products suitable for the present invention, which includes but not limited to Lipoid 16:1/18-1, Lipoid , DMPG NA, Lipoid P 75, Lipoid S 80, Lipoid S, Lipoid R, Lipoid E and Lipoid E PG/DSPG.

Synthetic DA-PLs, such as Lipoid PC, Lipoid PE, Lipoid PG, Lipoid PA, Lipoid PS can also be purchased from Lipoid catalog.

The Company ALC (<http://www.americanlecithin.com/aboutphos.html>) also offers a variety of phospholipids.

In one embodiment the phospholipids is negative charged. It has been observed that pharmaceutical composition of the present invention comprising negative charged phospholipids exerts stronger anti-calcium effect than neutral charged phospholipids at pH values of about 7. Without wishing to be bound by the theory, negatively charged phospholipids can have the anti-calcium effect by additionally capturing positively charged calcium. Thus in one embodiment the phospholipids is negatively charged. In one embodiment the phospholipids is lecithin. In one embodiment the phospholipids is phosphatidylglycerol.

In aqueous medium diacyl-phospholipids normally form liposomes. The participation of surfactant in the formation of vesicles increases the curvature, which results in liposomes with smaller diameter or results in thermodynamically stable micelles. The addition of surfactant, such as monoacyl-phospholipids or bile salt, reduces the viscosity of the formulation.

Thus in one preferred embodiment, the pharmaceutical composition comprises eltrombopag or a pharmaceutically acceptable salt thereof, diacyl-phospholipids and at least one surfactant, preferably one surfactant. In one embodiment the at least one surfactant is monoacyl-phospholipids. In one embodiment the at least one surfactant is bile salt. In one embodiment the at least one surfactant is vitamin E TPGS.

In one embodiment the pharmaceutical composition comprises diacyl-phospholipids and monoacyl-phospholipids. Typically the molar ratio between monoacyl-phospholipids and diacyl-phospholipids is from at least about 1:20 to up to about 1:4, to up to about 1:3, to up to about 1:2, to up to about 1:1.

In one embodiment the pharmaceutical composition comprises diacyl-phospholipids and at least one bile salt, preferably one bile salt. Typically the molar ratio between diacyl-phospholipids and the bile salt is from about 3:1 to about 1:3, from about 2:1 to about 1:2 and more typically about 1:1.

In one embodiment the pharmaceutical composition comprises diacyl-phospholipids, monoacyl-phospholipids and at least one bile salt, preferably one bile salt.

Suitable bile salts include, but not limited to, sodium cholate, sodium deoxycholate, sodium chenodeoxycholate, sodium lithocholate, sodium ursodeoxycholate, sodium hyodeoxycholate, glycine conjugated sodium glycocholate, sodium glycodeoxycholate, sodium glycochenodeoxycholate, sodium glycoursodeoxycholate, taurine conjugated sodium taurocholate, sodium taurodeoxycholate, sodium taurochenodeoxycholate.

In one embodiment the bile salt is selected from a group consisting of sodium taurocholate, sodium taurodeoxycholate, sodium taurochenodeoxycholate,

sodium glycocholate, sodium glycodeoxycholate and sodium glycochenodeoxycholate.

In one embodiment the bile salt is selected from a group consisting of sodium cholate, sodium deoxycholate, sodium glycocholate, sodium taurocholate, and sodium taurodeoxycholate.

In one embodiment the bile salt is sodium taurocholate.

In one embodiment the bile salt is sodium glycocholate.

Alternatively bile salts exist in abundance in gastric intestinal tract, which could interact with the DA-PL released from the composition to form micelles or small sized liposomes even if the composition itself does not comprise bile salt.

In one embodiment the phospholipids is monoacyl-phospholipids. Lipoid LPC 80 contains 70%-80% of monoacyl-phospholipids, while the rest is mainly diacyl-phospholipids. In one embodiment the pharmaceutical composition comprises phospholipids, wherein phospholipids is predominantly lyso-phospholipids. Used in this context, the term “predominantly” is understood that the molar ratio between monoacyl-phospholipids and diacyl-phospholipids is from at least about 1:1 to up to about 2:1, to up to about 3:1, to up to about 4:1, to up to about 5:1.

In one embodiment, the pharmaceutical composition of the invention, comprises eltrombopag or a pharmaceutically acceptable salt thereof, phospholipids and at least one co-solvent. Co-solvent is miscible with water and can increase the solubilization of the drug. Preferably the co-solvent is pharmaceutically acceptable. The minimum concentration needed for the co-solvent for the purpose of the invention should not exceed its maximum amount regulated by the regulatory health authorities. Generally the weight of co-solvent does not exceed 20%, does not exceed 15%, does not exceed 10%, does not exceed 5% of the total weight of the pharmaceutical composition.

Commonly used co-solvent includes but not limited to PEG300, propylene glycol.

In one embodiment the co-solvent is PEG 300.

In one embodiment, the pharmaceutical composition of the invention, comprises an API or a pharmaceutically acceptable salt thereof, phospholipids and at least one viscosity-lowering agent. Viscosity-lowering agent includes but not limited to glycerol. Generally the weight of co-solvent does not exceed 15%, does not exceed 10%, does not exceed 5% of the total weight of the pharmaceutical composition.

In one embodiment, in the pharmaceutical composition of the invention comprising phospholipids, the weight of the API, calculated in its free form, is not more than 40%, typically and preferably not more than 30%, not more than 20%, not more than 15% of the total weight of the pharmaceutical composition. In one embodiment the weight of the API, calculated in its free form, is not more than 20% of the total weight of the pharmaceutical composition. In one embodiment the weight of the API, calculated in its free form, is between about 3% to about 40%, between about 3% to about 30%, between about 5%- about 25%, preferably between about 5%- about 20% of the total weight of the pharmaceutical composition.

In one embodiment the weight of phospholipids is at least at least 35%, at least 50%, at least 60%, at least 70% of the total weight of the pharmaceutical composition.

In one embodiment the weight of phospholipids is not more than 90% of the total weight of the pharmaceutical composition.

In one embodiment the weight of diacyl-phospholipids is not more than 80%, not more than 60%, not more than 50% of the total weight of the pharmaceutical composition.

In one embodiment the weight of diacyl-phospholipids is between about 35% to about 85%, between about 50% to about 75% of the total weight of the pharmaceutical composition.

In one embodiment, the pharmaceutical composition of the invention, comprises, consists essentially of or consists of

- a) About 4-20%w/w API or a pharmaceutically acceptable salt thereof, calculated based on the API free form; and
- b) About 45% w/w to about 94% to 95% w/w of phospholipids.

In one embodiment, the pharmaceutical composition of the invention, comprises, consists essentially of or consists of

- c) About 4-20%w/w API or a pharmaceutically acceptable salt thereof, calculated based on the API free form;
- d) About 45% w/w to about 85% w/w of phospholipids; and
- e) About 10% to 45% bile salt.

In one embodiment, the pharmaceutical composition of the invention, comprises, consists essentially of or consists of an API or a pharmaceutically acceptable salt thereof, phospholipids, preferably diacyl- phospholipids, and a bile salt. In one embodiment, the pharmaceutical composition of the invention, comprises, consists essentially of or consists of an API or a pharmaceutically acceptable salt thereof, phospholipids, preferably diacyl- phospholipids, a bile salt and a viscosity-lowering agent. In one embodiment the viscosity-lowering agent is glycerol. In one embodiment, the pharmaceutical composition of the invention, comprises, consists essentially of or consists of

- f) About 4-20%w/w API or a pharmaceutically acceptable salt thereof, calculated based on the free form of the API;
- g) About 40-80% w/w phospholipids, , preferably diacyl- phospholipids;
- h) About 10-40% w/w bile salt; and
- i) About 0-10% w/w viscosity-lowering agent.

In one embodiment the pharmaceutical composition of the invention, comprises, consists essentially of or consists of an API or a pharmaceutically acceptable salt thereof, phospholipids, preferably predominately lyso-phospholipids. In one embodiment, the pharmaceutical composition of the invention, comprises, consists essentially of or consists of an API or a pharmaceutically acceptable salt thereof, phospholipids, preferably predominately lyso- phospholipids and a co-solvent. Preferably the co-solvent is PEG 300.

In one embodiment the pharmaceutical composition of the invention, comprises, consists essentially of or consists of

- a) About 5-20% w/w API or a pharmaceutically acceptable salt thereof, based on the free form of the API;
- b) About 60% to 85% w/w phospholipids, calculated based on the total weight of phospholipids, wherein phospholipids comprises predominately lyso- phospholipids;
- c) About 0%-10% w/w co-solvent, preferably the co-solvent is PEG 300.

In one aspect the present invention relates to a method of manufacturing the pharmaceutical composition comprising the steps of mixing the API or a pharmaceutically acceptable salt thereof and phospholipids in a solution (Solution). Thus the resulting pharmaceutical composition is in the liquid form, which can be administered as oral solutions, as concentrated solutions to be filled in capsules or dosed by pipetting a small volume in a drink like water or juice. Organic solvent in the Solution can be evaporated to result in solid or semi-solid cake. Such cake can be directly formulated into tablets or filled into capsules. Optionally such case can be re-hydrated with water to arrive at a solution, that can be filled into capsules. The capsule is preferably sealed by banding.

Alternatively the pharmaceutical formulation of the present invention comprising phospholipids is in solid form. Typically the Solution can be dried in the presence of a sugar (Van Hoogevest, European Journal of Pharmaceutical Sciences, Vol 108, Page 1-12, 2017). Alternatively, in order to convert unsaturated mono- and diacyl-phospholipids into powders organic (ethanolic) solvent solutions of the phospholipids can be mixed as explained in WO2003063835A with absorbing porous carriers like Neuselin Grade 2 (Fuji Chemicals) and subsequently the solvent is removed under vacuum. Furthermore CA2352178 teaches a method of mixing the phospholipids containing solution with polymers, dry and grind to obtain a free flowing powder.

The present invention relates to a pharmaceutical composition (pharmaceutical composition of the invention), preferably in an oral dosage form,

comprising an API or a pharmaceutically acceptable salt thereof and at least one surfactant.

Suitably the term “at least one surfactant” refers to one, two, three or four surfactants present in the pharmaceutical composition of the present invention. Suitably the term “at least one surfactant” refers to one, two or three surfactants present in the pharmaceutical composition of the present invention. Suitably the term “at least one surfactant” refers to one or two surfactants present in the pharmaceutical composition of the present invention. Suitably the term “at least one surfactant” refers to only one surfactant present in the pharmaceutical composition of the present invention.

An empirical system known as hydrophilic /lipophilic balance (HLB) is commonly used to categorize amphiphilic surfactants by the degree of affinity for the oily phase compared to aqueous phases in a formulation.

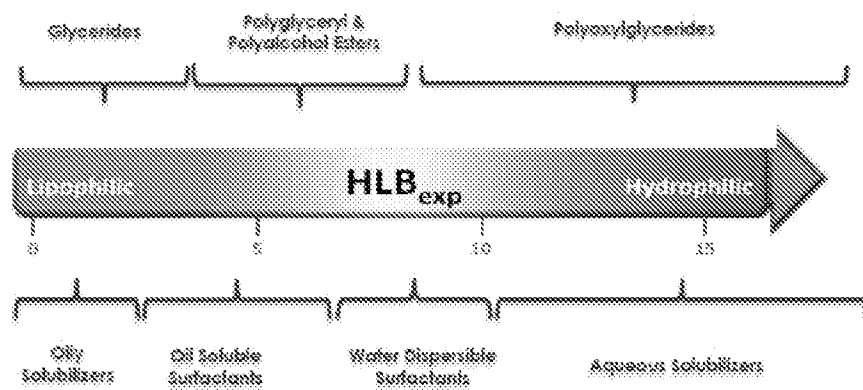


Figure 2: Amphiphilicity and functionality expressed on the HLB scale

Reference : Source of above figure Gattefosse

The HLB of a surfactant can be determined by analytical methods and referred to as practical HLB. Alternatively the HLB value can be obtained theoretically. For nonionic surfactants the common way to calculate HLB is called Griffin method (Griffin, William C. (1954), "Calculation of HLB Values of Non-Ionic Surfactants" (PDF), Journal of the Society of Cosmetic Chemists, 5 (4): 249, archived from the original (PDF) on 2014-08-12, retrieved 2013-05-25). Practically surfactants providers usually provides information on HLB value. HLB value can

vary to certain degree, for example ± 3 , ± 2 or ± 1 between different providers or between different batches of the same provider largely due to varying degree of polymerization, e.g. the number of PEG repeats. HLB values of commonly used and/or commercially available surfactants are collected from general literature, including product catalogs and are presented in Table 4.

According to the Gattefosse diagram above, the surfactant suitable for the present invention is typically in the range of water dispersible surfactant, preferably in the range of aqueous solubilizers. Typically the API is at least partially solubilized by the aid of the at least one surfactant.

The working surfactant examples suggest that surfactant suitable for the present invention is likely more towards the hydrophilic end of the diagram above. Thus in one embodiment the at least one surfactant has a HLB value above 7, about 8, above 9, preferably above 10, more preferably above 11. The HLB value should be viewed with certain degree of flexibility due to the variation range of ± 3 , ± 2 or ± 1 in practice. In one embodiment the at least one surfactant has a HLB value below 20, below 18, preferably below 17, more preferably below 16. In one embodiment the at least one surfactant has a HLB value in the range of 9-20, preferably 10-19, preferably 10-18, preferably 11-17, more preferably 12-16.

The surfactant suitable for the present invention is preferably a polyethoxylated / polyethylene glycol / PEG fatty acid ester derivative, such as PEG 40 hydrogenated castor oil (Cremophor RH 40), PEG 35 castor oil (Cremophor EL), PEG 32 monostearate (Gelucire 48/16), PEG 15 hydroxystearate (Solutol HS 15), or Vitamin E TPGS (d- α -tocopheryl PEG 1000 succinate) or mixtures thereof. The respective HLB of about 14-16, 12-14, 12, 14-16, or 13 of the above molecules is related to but not exclusively dependent on the number of ethylene oxide repeat units in the PEG chain.

The below table contains commonly used surfactants with indication of their properties and suitability for the present invention.

Table 4 List of Surfactants and Lipids:

Ref	Class/type	Surfactant	HLB	IIG level	Application	Pegylation (yes/no)	useable for ETB
Chemical book	Non-ionic	Polyethylene Glycol Monocetyl Ether	15.7		Cosmetic	yes	likely
	Non-ionic	POLYETHYLENE GLYCOL MONOOLEYL ETHER	16.9		Cosmetic-Shampoo	yes	likely
	Non-ionic	Glycerol tristearate	5.8	225mg	Oral formulation	No	
	Non-ionic	Sorbitan monopalmitate	6.7	2%/w/w	Topical Formulation	No	
	Non-ionic	TRIOLEIN			Injectable preparation	No	
	Non-ionic	Span 20	8.6	83.9mg	Oral formulation	yes	
	Non-ionic	Span 60	4.7	3.5mg	Oral formulation	yes	
	Non-ionic	Span 80	4.3	153.9mg	Oral formulation	yes	
	Non-ionic	Tween 80	15	418mg	Oral formulation	yes	likely
	Non-ionic	Tween 85	11			yes	likely
	Non-ionic	Tween 60	15	20mg/1ml	Oral emulsion	yes	likely
	Non-ionic	Polysorbate 20	16.7	56.25mg	Oral formulation	yes	likely
	Non-ionic	Polyoxyethylene stearate	18.8	25mg/5ml	Oral Concentrate	yes	yes
	Non-ionic	Glyceryl Monooleate				no	
	Non-ionic	SORBITAN TRIOLEATE	1.8	1.5mg/5ml	Powder for suspension	no	
	Non-ionic	Polyoxyethylene lauryl ether	water soluble	5.22%W/W	Topical Formulation	yes	likely
	Non-ionic	Propyleneglycol alginate		250mg	Powder for suspension	no	
	Non-ionic	GLYCEROL MONOHYDROXYSTEARATE				no	
	Non-ionic	Fatty acids, lanolin, isopropyl esters	10		Cosmetic	no	
	Non-ionic	Poly(ethylene glycol) distearate			Topical formulation	Yes	likely
	Non-ionic	MYRISTYL MYRISTATE	8.5		Cosmetic		
	Non-ionic	SUCROSE DISTEARATE					
	Non-ionic	SORBITAN SESQUIOLEATE		2.5%W/W	Topical formulation		
	Non-ionic	SORBITAN TRISTEARATE	2.1	0.5%W/W	Topical Formulation		
	Non-ionic	glycerine monostearate				no	
	Non-ionic	Fatty alcohol polyoxyethylene ether N=3				YES	likely
	Non-ionic	castor oil polyoxyethylene (90) ether				YES	likely
	Non-ionic	MONOMYRISTIN	11.5			No	
	Non-ionic	alkyl polyglucoside				No	
	Non-ionic	TRIDECETH-4			cosmetic		
	Non-ionic	MONOCAPRYLIN	6			no	
	Non-ionic	TRILAURIN		Soluble in water		no	
	Non-ionic	DILAURIN	7			no	
	Non-ionic	MONOLAURIN				no	
	Non-ionic	C ⁸ ~C ⁹ alkyl phenyl polyoxyethylene (15) ether				yes	likely
	Non-ionic	C ¹² ~C ¹⁸ fatty alcohol polyoxyethylene (35) ether				yes	likely

	Non-ionic	C ⁿ (8 ~ 9 ⁿ) alkyl phenyl polyoxyethylene (8) ether				yes	likely
	Non-ionic	alkyl phenyl polyoxyethylene ether				yes	likely
	Non-ionic	octyl phenyl polyoxyethylene (30) ether				yes	likely
	Non-ionic	dibenzyl biphenyl polyoxyethylene ether				yes	likely
	Non-ionic	nonyl phenyl polyoxyethylene (9) ether				yes	likely
	Non-ionic	octyl phenyl polyoxyethylene (3) ether				yes	likely
	Non-ionic	castor oil polyoxyethylene (30) ether				yes	likely
	Non-ionic	polyoxyethylene (10) castor oil ether				yes	likely
	Non-ionic	DECYL OLEATE	Insoluble in water			no	
	Non-ionic	Trimethylolpropane t	2			no	
	Non-ionic	SUCROSE COCOATE	3			no	
	Non-ionic	CETYL LACTATE	Insoluble in water		Cosmetic		
	Non-ionic	Sucrose stearate	Slightly soluble in water	44.5mg	Oral formulation		
	Non-ionic	Isooctyl palmitate			cosmetic		
	Non-ionic	Pentaerythryl tetrastearate					
	Non-ionic	Isopropyl myristate			Topical formulation		
	Non-ionic	Isooctadecanoic acid, ester with 1,2,3-propanetriol					
	Non-ionic	ethylene glycol monostearate				yes	
	Non-ionic	HEXAETHYLENE GLYCOL MONOOCTYL ETHER				yes	
	Non-ionic	glycerine monolaurate				no	
	Non-ionic	1-Glyceryl caprate				no	
	polymer	Hydroxypropyl methyl cellulose	Soluble in water	670mg	Oral formulation	no	No
	polymer	Hydroxyethyl Cellulose		400mg	Oral Formulation	no	
	polymer	Soluplus® Polyvinyl caprolactam-polyvinyl acetate- polyethylene glycol graft copolymer.		NA		yes	No
Gattefosse brochure	Non-ionic	Gelucire 48/16 HLB 12 (PEG32 stearate or Polyoxyyl stearate)	12			yes	yes
Gattefosse brochure	Non-ionic	Labrasol ALF HLB 12	12				yes
Gattefosse brochure	Non-ionic	Gelucire 44/14 HLB 11	11	3 mg/ 218 mg basis RDS			Very likely
Gattefosse brochure	Non-ionic	Gelucire 50/13 HLB 11	11				Very likely
Gattefosse brochure	Non-ionic	Labrafil M 1944 CS HLB 9 9					
Gattefosse brochure	Non-ionic	Labrafil M 2125 CS HLB 9 9					
Gattefosse brochure	Non-ionic	Labrafil M 2130 CS HLB 9 9					

Gattefosse brochure	Non-ionic	Glyceryl mono stearate HLB 11	3.8				
	Non-ionic	Pluronic® L-31 Non-ionic 1100 1.0-7.0	1 to 7	NA		yes	
	Non-ionic	Pluronic® L-35 Non-ionic 1900 18.0-23.0	18 to 23			yes	
	Non-ionic	Pluronic® L-61 Non-ionic 2000 1.0-7.0	1 to7			yes	
	Non-ionic	Pluronic® L-81 Non-ionic 2800 1.0-7.0	1 to7			yes	
	Non-ionic	Pluronic® L-64 Non-ionic 2900 12.0-18.0	12 to 18			yes	
	Non-ionic	Pluronic® L-121 Non-ionic 4400 / Poloxamer 401	1 to7			yes	
	Non-ionic	Pluronic® P-123 Non-ionic 5800 /Poloxamer 403	7 to 9	NA		yes	
	Non-ionic	Pluronic® F-68 Non-ionic 8400 / Poloxamer 188	>24	100		yes	No
	Non-ionic	Pluronic® F-108 Non-ionic 14600 / Poloxamer 338	>24	NA		yes	
https://patents.google.com/patent/KR101841756B1/en?q=non-ionic+surfactant+drug&ocq=non-ionic+surfactant+drug	Non-ionic	Brij®30 Non-ionic Polyoxyethylene (4) lauryl ether 9.7	9.7	NA		yes	likely
https://patents.google.com/patent/KR101841756B1/en?q=non-ionic+surfactant+drug&ocq=non-ionic+surfactant+drug	Non-ionic	Brij®35 Non-ionic Polyoxyethylene (23) lauryl ether 16.9	16.9	NA		yes	likely
https://patents.google.com/patent/KR101841756B1/en?q=non-ionic+surfactant+drug&ocq=non-ionic+surfactant+drug	Non-ionic	Brij®52 Non-ionic Polyoxyethylene (2) cetyl ether 5.3	5.3	NA		yes	
https://patents.google.com/patent/KR101841756B1/en?q=non-ionic+surfactant+drug&ocq=non-ionic+surfactant+drug	Non-ionic	Brij®56 Non-ionic Polyoxyethylene (10) Cetyl ether 12.9	12.9	NA		yes	likely
https://patents.google.com/patent/KR101841756B1/en?q=non-ionic+surfactant+drug&ocq=non-ionic+surfactant+drug	Non-ionic	Brij®58 Non-ionic Polyoxyethylene (20) Cetyl ether 15.7	15.7	NA		yes	likely
https://patents.google.com/patent/KR101841756B1/en?q=non-ionic+surfactant+drug&ocq=non-ionic+surfactant+drug	Non-ionic	Brij®72 Non-ionic Polyoxyethylene (2) stearyl ether 4.9	4.9	NA		yes	

https://patents.google.com/patent/KR101841756B1/en?q=non-ionic+surfactant+drug&pg=non-ionic+surfactant+drug	Non-ionic	Brij®76 Non-ionic Polyoxyethylene (10) stearyl ether 12.4	12.4	NA		yes	likely
https://patents.google.com/patent/KR101841756B1/en?q=non-ionic+surfactant+drug&pg=non-ionic+surfactant+drug	Non-ionic	Brij®78 Non-ionic Polyoxyethylene (20) stearyl ether 15.3	15.3	NA		yes	likely
https://patents.google.com/patent/KR101841756B1/en?q=non-ionic+surfactant+drug&pg=non-ionic+surfactant+drug	Non-ionic	Brij®92V Non-ionic Polyoxyethylene (2) oleyl ether 4.9	4.9	NA		yes	
https://patents.google.com/patent/KR101841756B1/en?q=non-ionic+surfactant+drug&pg=non-ionic+surfactant+drug	Non-ionic	Brij®93 Non-ionic Polyoxyethylene (2) oleyl ether 4	4	NA		yes	
https://patents.google.com/patent/KR101841756B1/en?q=non-ionic+surfactant+drug&pg=non-ionic+surfactant+drug	Non-ionic	Brij®96V Non-ionic Polyethylene glycol oleyl ether 12.4	4	NA		yes	
https://patents.google.com/patent/KR101841756B1/en?q=non-ionic+surfactant+drug&pg=non-ionic+surfactant+drug	Non-ionic	Brij®97 Non-ionic Polyoxyethylene (10) oleyl ether 12	12	NA		yes	likely
https://patents.google.com/patent/KR101841756B1/en?q=non-ionic+surfactant+drug&pg=non-ionic+surfactant+drug	Non-ionic	Brij®98 Non-ionic Polyoxyethylene (20) oleyl ether 15.3	15.3	NA		yes	likely
https://patents.google.com/patent/KR101841756B1/en?q=non-ionic+surfactant+drug&pg=non-ionic+surfactant+drug	Non-ionic	Brij®700 Non-ionic Polyoxyethylene (100) stearyl ether 18	18	NA		yes	likely

https://patents.google.com/patent/KR101841756B1/en?q=non-ionic+surfactant+drug&qq=non-ionic+surfactant+drug	Non-ionic	Span®20 Non-ionic Sorbitan monolaurate 8.6	8.6	83.9		No	
https://patents.google.com/patent/KR101841756B1/en?q=non-ionic+surfactant+drug&qq=non-ionic+surfactant+drug	Non-ionic	Span®40 Non-ionic Sorbitan monopalmitate 6.7	6.7	NA		No	
https://patents.google.com/patent/KR101841756B1/en?q=non-ionic+surfactant+drug&qq=non-ionic+surfactant+drug	Non-ionic	Span®60 Non-ionic Sorbitan monostearate 4.7	4.7	62.5		No	
https://patents.google.com/patent/KR101841756B1/en?q=non-ionic+surfactant+drug&qq=non-ionic+surfactant+drug	Non-ionic	Span®65 Non-ionic Sorbitan tristearate 2.1	2.1	NA		No	
https://patents.google.com/patent/KR101841756B1/en?q=non-ionic+surfactant+drug&qq=non-ionic+surfactant+drug	Non-ionic	Span®80 Non-ionic Sorbitan monooleate 4.3	4.3	153.9		No	
https://patents.google.com/patent/KR101841756B1/en?q=non-ionic+surfactant+drug&qq=non-ionic+surfactant+drug	Non-ionic	Span®85 Non-ionic Sorbitan trioleate 1.8	1.8	1.5mg/5ml		No	
https://patents.google.com/patent/KR101841756B1/en?q=non-ionic+surfactant+drug&qq=non-ionic+surfactant+drug	Non-ionic	Myrj®45 Non-ionic Polyoxyethylene monostearate	10.8	NA		yes	likely
https://patents.google.com/patent/KR101841756B1/en?q=non-ionic+surfactant+drug&qq=non-ionic+surfactant+drug	Non-ionic	Myrj®49 Non-ionic Polyoxyethylene monostearate	15	NA		yes	likely

https://patents.google.com/patent/KR101841756B1/en?q=non-ionic+surfactant+drug&qq=non-ionic+surfactant+drug	Non-ionic	Myrj®52 Non-ionic Polyoxyethylene monostearate	16.9	NA		yes	likely
https://patents.google.com/patent/KR101841756B1/en?q=non-ionic+surfactant+drug&qq=non-ionic+surfactant+drug	Non-ionic	Myrj®53 Non-ionic Polyoxyethylene monostearate	17.9	NA		yes	likely
https://patents.google.com/patent/KR101841756B1/en?q=non-ionic+surfactant+drug&qq=non-ionic+surfactant+drug	Non-ionic	Kolliphor EL/ELP – Macroglycerol ricinoleate		599.4 mg		yes	yes
https://patents.google.com/patent/KR101841756B1/en?q=non-ionic+surfactant+drug&qq=non-ionic+surfactant+drug	Non-ionic	Polyoxyethylene (20) sorbitan monolaurate – Tween 20 HLB 16.7	16.7	56.25		yes	likely
https://patents.google.com/patent/KR101841756B1/en?q=non-ionic+surfactant+drug&qq=non-ionic+surfactant+drug	Non-ionic	Polyoxyethylene (20) sorbitan monopalmitate – Tween 40 HLB 15.6	15.6	0.05 mg/ml		yes	likely
https://patents.google.com/patent/KR101841756B1/en?q=non-ionic+surfactant+drug&qq=non-ionic+surfactant+drug	Non-ionic	Polyoxyethylene (20) sorbitan monostearate – Tween 60 HLB 15	15	25mg/5ml		yes	likely
https://patents.google.com/patent/KR101841756B1/en?q=non-ionic+surfactant+drug&qq=non-ionic+surfactant+drug	Non-ionic	Polyoxyethylene (20) sorbitan mono-oleate – Tween 80 HLB 15	15	418.37		yes	likely
Handbook of excipient for HLB	Non-ionic	Polyoxyethylene (20) sorbitan tristearate – Tween 65	10.5	NA		yes	likely
https://patents.google.com/patent/KR101841756B1/en?q=non-ionic+surfactant+drug&qq=non-ionic+surfactant+drug	Non-ionic	Polyoxyethylene (20) sorbitan tri-oleate – Tween 85 HLB 11	11	NA		yes	likely
..	Non-ionic	(Polyoxyethylene-(20)-sorbitanmonoisostearate) - Tween 120	14.9			yes	likely

	Non-ionic	Vitamin E TPGS (d- α -tocopheryl PEG 1000 succinate), CAS-No. 9002-96-4	ca. 13			Yes	yes
	Non-ionic	Polyoxyl 40 hydrogenated castor oil (Cremophor or Kolliphor RH 40)	14 to 16	450 mg		Yes	yes
	Non-ionic	Polyoxyl 35 castor oil (Cremophor or Kolliphor EL and ELP)	12 to 14	599.4 mg		Yes	yes
	Non-ionic	PEG 15 Hydroxystearate (Solutol HS 15)	14 to 16	NA		Yes	Very likely
	Non-ionic	Polyoxyl 60 hydrogenated castor oil (Cremophor RH 60)	15 to 17	NA		Yes	likely
	Non-ionic	PEG-25 Hydrogenated Castor Oil	10.8			Yes	likely
	Non-ionic	Polyoxyl 5 castor oil (PEG-5 castor oil) HLB 3.7	3.7			Yes	
	Non-ionic	Polyoxyl 9 castor oil (PEG-9 castor oil)	<10.8 (PEG 25 Hydrogenated castor oil)			Yes	likely
	Non-ionic	Polyoxyl 15 castor oil (PEG-15 castor oil)	<10.8 (PEG 25 Hydrogenated castor oil)			Yes	likely
Excipient handbook	Non-ionic	Polyoxyl 6 cetostearyl ether	10 to 12		Topical	Yes	likely
Excipient handbook	Non-ionic	Polyoxyl 20 cetostearyl ether	15		Topical	Yes	likely
Excipient handbook	Non-ionic	Polyoxyl 25 cetostearyl ether	15 to 17		Topical	Yes	likely
Excipient handbook	Non-ionic	Polyoxyl 9 Lauryl ether,	13.6		Topical	Yes	likely
Excipient handbook	Non-ionic	Polyoxyl 10 oleyl ether, Brij 96	12.4		Topical	Yes	likely
Excipient handbook	Non-ionic	Polyoxyl 20 oleyl ether, Brij 98	15.9		Topical	Yes	likely
Excipient handbook	Non-ionic	Polyoxyl 21 steryl ether, Brij 721	15.5		Topical	Yes	likely
Excipient handbook	Non-ionic	Polyoxyl 100 steryl ether	18.8		Topical	Yes	likely

In one embodiment the at least one surfactant is an ionic surfactant.

In one embodiment the at least one surfactant is an anionic surfactant.

For example, bile salts are anionic surfactants. Commonly and preferred used bile salts are described in the earlier section of this application.

In one embodiment the at least one surfactant is a nonionic surfactant.

In one embodiment the at least one surfactant is vitamin E TPGS.

In one embodiment the at least one surfactant is a polyoxyethylene castor oil derivative, which includes but not limited to polyoxyl 5 castor oil (PEG-5 castor oil), polyoxyl 9 castor oil (PEG-9 castor oil), polyoxyl 15 castor oil (PEG-15 castor oil), polyoxyl 35 castor oil (Cremophor EL, or PEG-35 castor oil), polyoxyl 40 hydrogenated castor oil (Cremophor RH 40 or PEG-40 hydrogenated castor oil), Polyoxyl 60 hydrogenated castor oil (Cremophor RH 60 or PEG-60 hydrogenated castor oil).

In one embodiment the at least one surfactant is a polyoxyethylene alkyl ether, which includes but not limited to polyethylene glycol monoacetyl ether, polyethylene glycol monolauryl ether, polyethylene glycol monooleyl ether, polyethylene glycol monostearyl ether.

In one embodiment polyoxyethylene alkyl ether is selected from a group consisting of Polyoxyl 20 cetostearyl ether, Polyoxyl 10 cetyl ether, Polyoxyl 20 cetyl ether, Polyoxyl 23 laur lauryl ether Polyoxyl 23 lauryl ether Polyoxyl 23 lauryl ether, Polyoxyl 10 oleyl ether, Polyoxyl 20 oleyl ether, Polyoxyl 10 stearyl ether and Polyoxyl 21 stearyl ether.

In one embodiment the at least one surfactant is a PEG stearate, e.g. PEG 15 Hydroxystearate (Solutol HS 15, polyethylene glycol (PEG)-15-hydroxystearate) or PEG 32 stearate (Gelucire 48/16, Polyethylene glycol monostearate, Polyoxyl stearate).

In one embodiment the at least one surfactant is a polyoxyethylene sorbitan fatty acid ester, which includes but not limited to Tween 80 (Polysorbate 80, Polyoxyethylene (20) sorbitan monooleate).

In one embodiment the at least one surfactant is selected from the list consisting of all the surfactants labelled as likely or very likely in Table 4.

In one embodiment the at least one surfactant is selected from the list consisting of Vitamin E TPGS, PEG 40 hydrogenated castor oil (Cremophor RH 40) or Kolliphor RH40), PEG 32 monostearate (Gelucire 48/16), Gelucire 44/14, Gelucire 50/13, labrasol, and PEG 35 castor oil (Cremophor EL), PEG 15

hydroxystearate (Solutol HS 15), Tween 80 or a mixture of any 3 or any 2 of the surfactants from the list.

Derived from hydrogenated castor oil and ethylene oxide, Kolliphor® RH40 is used as a non-ionic oil-in-water solubilizer and emulsifying agent. BASF catalog contains product information of Kolliphor® RH40.

Labrasol® (Synonym: CAPRYLOCAPROYL MACROGOL-8 / POLYOXYL-8 GLYCERIDES Caprylocaproyl polyoxylglycerides PEG-8 Caprylic/Capric Glycerides (FDA IIG)) is a non-ionic water dispersible surfactant composed of well-characterised polyethylene glycol (PEG) esters, a small glyceride fraction and free PEG. Self-emulsify forming a fine dispersion (SMEDDS). Ferromet catalog contains product information of Labrasol®.

Gelucire® 48/16 is a polyethylene glycol monostearate (type I) NF and consists of PEG-32 (MW 1500) esters of palmitic (C16) and stearic (C18) acids.

In one embodiment one of the at least one surfactant is Kolliphor RH40.

In one embodiment one of the at least one surfactant is Gelucire® 48/16.

In one embodiment the pharmaceutical composition of the invention comprises an API or a pharmaceutically acceptable salt thereof and at least one surfactant, wherein the weight of the API, based on its free form, is not more than 80%, not more than 60%, suitably not more than 40%, suitably not more than 30%, suitably not more than 25%, suitably not more than 20% of the total weight of the API and the at least one surfactant. In one embodiment the weight of the API is not more than 30% of the total weight of the API and the at least one surfactant. For the sake of clarity, if there are more than one surfactant in the composition, the weight of the at least one surfactant is the total weight of all the surfactants. The weight of API is based on the weight of the free form of the API.

In one embodiment the weight of the API is at least 2%, suitably at least 5%, suitably at least 10% of the total weight of the API and the at least one surfactant. In one embodiment the weight of the API is at least 5% of the total weight of the API and the at least one surfactant.

In one embodiment the pharmaceutical composition of the invention comprises an API or a pharmaceutically acceptable salt thereof and at least one surfactant, wherein the weight of the API is from 2% to 50%, suitably from 5% to 40%, suitably from 5% to 30%, suitably from 5% to 25%, suitably from 10 to 20%, of the total weight of the API and the at least one surfactant. In one embodiment the composition, the weight of API is from 10% to 20% of the total weight of the API and the at least one surfactant.

In one embodiment the pharmaceutical composition further comprises at least one more pharmaceutically acceptable excipients.

In one embodiment the at least one more pharmaceutically acceptable excipients include an anti-oxidant. In one embodiment the weight of the anti-oxidant is not more than 10%, suitably not more than 7%, suitably not more than 5%, suitably not more than 3%, suitably not more than 1%, of the total weight of the pharmaceutical composition. In one embodiment the anti-oxidant is selected from a list consisting of Vitamin E, Butylhydroxytoluol (BHT), Butylhydroxyanisol (BHA), Propyl gallate, ascorbyl palmitate, ascorbic acid, EDTA and sodium metabisulfite or a mixture thereof, suitably 2 of the anti-oxidants from the list

In one embodiment the pharmaceutical composition consists essentially of or consists of an API or a pharmaceutically acceptable salt thereof and one surfactant and one anti-oxidant. In one embodiment the anti-oxidant is not more than 10%, suitably not more than 7%, suitably not more than 5%, of the total weight of the pharmaceutical composition.

In one embodiment the at least one anti-oxidant is EDTA.

In one embodiment the pharmaceutical composition of the present invention, preferably in an oral dosage form, comprises an API or a pharmaceutically acceptable salt thereof, Kolliphor RH40 and EDTA. In one embodiment the pharmaceutical composition of the present invention, preferably in an oral dosage form, consists of an API or a pharmaceutically acceptable salt thereof, Kolliphor RH40 and EDTA. In one embodiment the weight of the API is not more than 30% of the total weight of the API and Kolliphor RH40.

Suitably, the at least one more pharmaceutically acceptable excipients include a diluent (also known as filler or bulking agent) and/or a binder and/or a lubricant and/or a disintegrant. Those skilled in the art will recognize that a given material may provide one or more functions in the tablet formulation, although the material is usually included for a primary function.

Diluents provide bulk, for example, in order to make the tablet a practical size for processing. Diluents may also aid processing, for example, by providing improved physical properties such as flow, compressibility, and tablet hardness. Because of the relatively high percentage of diluent and the amount of direct contact between the diluent and the active compound in the typical pharmaceutical formulation, the interaction of the diluent with the active compound is of particular concern to the formulator. Examples of diluents suitable for general use include: water-soluble fillers and water-insoluble fillers, such as calcium phosphate (e.g., di and tri basic, hydrated or anhydrous), calcium sulfate, calcium carbonate, magnesium carbonate, kaolin, spray dried or anhydrous lactose, cellulose (e.g., microcrystalline cellulose, powdered cellulose), pregelatinized starch, starch, lactitol, mannitol, sorbitol, maltodextrin, powdered sugar, compressible sugar, sucrose, dextrose, and inositol. The diluents that do not contain coordinating metals and diluents that are non-reducing sugars are suitable for tablets of the current invention. Suitable diluents for use in this invention include microcrystalline cellulose, powdered cellulose, pregelatinized starch, starch, lactitol, mannitol, sorbitol, and maltodextrin. Unsuitable diluents include calcium phosphate (e.g., di and tri basic, hydrated or anhydrous), calcium sulfate, calcium carbonate, magnesium carbonate, kaolin, and spray dried or anhydrous lactose. In one embodiment of the present invention, the diluent is composed of one or both of Mannitol and microcrystalline cellulose.

Binders impart cohesive properties to the powdered material. Examples of binders suitable for use in the present invention include: starch (e.g., paste, pregelatinized, mucilage), gelatin, sugars (e.g., sucrose, glucose, dextrose, molasses, lactose, dextrin, xylitol, sorbitol), polymethacrylates, natural and synthetic gums (e.g., acacia, alginic acids and salts thereof such as sodium alginate, gum tragacanth,

Irish moss extract, panwar gum, ghatti gum, guar gum, zein), cellulose derivatives [such as carboxymethyl cellulose and salts thereof, methyl cellulose (MC), hydroxypropyl methyl cellulose (HPMC), hydroxypropyl cellulose (HPC), hydroxyethyl cellulose (HEC) and ethyl cellulose (EC)], polyvinylpyrrolidone, Veegum, larch arabogalactan, polyethylene glycol, waxes, water, alcohol, magnesium aluminum silicate, and bentonites. In one embodiment of the present invention, the binder comprises polyvinylpyrrolidone (PVP).

Lubricants are generally used to enhance processing, for example, to prevent adhesion of the formulation material to manufacturing equipment, reduce interparticle friction, improve rate of flow of the formulation, and/or assist ejection of the formulations from the manufacturing equipment. Examples of lubricants suitable for use in the present invention include: talc, stearates (e.g., magnesium stearate, calcium stearate, zinc stearate, palmitostearate), stearic acid, hydrogenated vegetable oils, glyceryl behenate, polyethylene glycol, ethylene oxide polymers (e.g., CARBOWAXes), liquid paraffin, sodium lauryl sulfate, magnesium lauryl sulfate, sodium oleate, sodium stearyl fumarate, DL-leucine, and silica derivatives (e.g., colloidal silicon dioxide, colloidal silica, pyrogenic silica, and hydrated sodium silicoaluminate). In one embodiment of the present invention, the lubricant comprises magnesium stearate.

Disintegrants are employed to facilitate breakup or disintegration of the formulation after administration. Examples of disintegrants suitable for use in the present invention include: starches, celluloses, gums, crosslinked polymers, and effervescent agents, such as corn starch, potato starch, pregelatinized starch, modified corn starch, croscarmellose sodium, crospovidone, sodium starch glycolate, Veegum HV, methyl cellulose, microcrystalline cellulose, cellulose, modified cellulose gum (e.g., Ac-Di-Sol R), agar, bentonite, montmorillonite clay, natural sponge, cation exchange resins, ion exchange resins (e.g., polyacrin potassium), alginic acid and alginates, guar gum, citrus pulp, carboxymethylcellulose and salts thereof such as sodium lauryl sulfate, magnesium aluminum silicate, hydrous aluminum silicate, sodium bicarbonate in admixture

with an acidulant such as tartaric acid or citric acid. In one embodiment of the present invention, the disintegrant is sodium starch glycolate.

In the foregoing embodiments, the diluent is suitably a combination of mannitol and microcrystalline cellulose, the non-reducing sugar is suitably mannitol, the binder is suitably polyvinylpyrrolidone, the lubricant is suitably magnesium stearate, and the disintegrant is suitably sodium starch glycolate.

Oral dosage form is meant to be taken orally, typically as instructed by the medicine manufacturer. Common oral dosage form includes but not limited to solid dosage form such as tablets, capsules, pellets, lozenges, granules and powders and liquid dosage form such as syrup. In one embodiment the oral dosage form is tablet. In one embodiment the oral dosage form is to be taken into the mouth directly. In one embodiment the oral dosage form is firstly to be suspended/dissolved/dispersed/mixed prior to oral administration. For example a dispersible tablet is firstly dispersed in sufficient liquid, such water/juice prior to oral administration. In one embodiment oral dosage form is a tablet. In one embodiment the tablet is a dispersible tablet. In one embodiment the tablet is to be taken directly into the mouth. In one embodiment the oral dosage form is granules.

In one embodiment the pharmaceutical composition of the invention is in the form of capsule. In one embodiment the pharmaceutical composition of the invention is in the form of soft capsule. In one embodiment the pharmaceutical composition of the invention is in the form of hard capsule. Capsule can be gelatin or non-gelatin based. An example of non-gelatin based capsule is hypromellose (HPMC) based capsule. In one embodiment the hard capsule is HPMC based capsule.

In one embodiment, the pharmaceutical composition of the invention is semi-solid. The composition is prepared by melting the at least one surfactant and incorporating the compound and filling the molten mass into capsules which upon cooling forms a semi-solid in the capsule.

The oral dosage form, suitable tablets, capsules or granules, suitable capsules, containing the pharmaceutical composition of the present invention

typically comprise at least about 5mg or at least about 10mg of the API, suitably at least about 10mg of the API per dosage form. The oral dosage form, suitable tablets, capsules or granules, suitably capsules, containing the pharmaceutical formulation of the present invention typically comprise at most about 200mg, 150mg, 100mg or 75mg, suitably at most about 75mg of the API per dosage form. In one embodiment the pharmaceutical composition comprises about 10mg to about 100mg of the API, or about 10mg to 75mg, or about 5mg to 75mg of the API per dosage form.

In another preferred embodiment, due to the increased bioavailability of the API in the pharmaceutical composition of the present invention, the oral dosage form comprises less amount of the API but is bioequivalent to the corresponding formulation not employing the present invention.

In one embodiment, the pharmaceutical composition of the invention is free from polyvalent metals. In one embodiment, the pharmaceutical composition of the invention is free from polyvalent metals, apart from magnesium stearate.

In one aspect the invention relates to a process for making the pharmaceutical composition of the invention. For example, homogenization, extrusion, spray granulation, spray layering, spray congealing can be employed.

In one embodiment, the present invention relates to the process for preparing the pharmaceutical composition comprising the steps of homogenizing an API or a pharmaceutically acceptable salt thereof and the at least one surfactant.

In one embodiment the present invention relates to the process A for preparing the pharmaceutical composition comprising the steps of:

- a) Melting the at least one surfactant, preferably by heating, preferably by heating above its melting temperature;
- b) Adding the API or a pharmaceutically acceptable salt thereof to the molten mass;
- c) Homogenizing the mixture of b); and
- d) Formulating mixture c) into a pharmaceutical dosage form, preferably in oral dosage form.

In one embodiment the present invention relates to the process B for preparing the pharmaceutical composition comprising the steps of:

- a) Adding the API or a pharmaceutically acceptable salt thereof to the at least one surfactant;
- b) Heating the mixture a) to melt; preferably heating to the temperature above the melting temperature of the at least one surfactant;
- c) Homogenizing the mixture of b); and
- d) Formulating mixture c) into a pharmaceutical dosage form, preferably in oral dosage form.

In one embodiment the process of A or B comprises a step of adding at least one more pharmaceutically acceptable excipients before or after any one of the steps of a) to c). In one embodiment the step of adding at least one more pharmaceutically acceptable excipients is performed after step c). The at least one more excipients includes, but not limited to one or more of fillers, binders, disintegrants and/or anti-oxidant. Preferably the mixture of c) is mixed with additional filler, binder, disintegrant, lubricant and/or anti-oxidant before formulating into a pharmaceutical dosage form, preferably in oral dosage form, preferably in tablet or in capsule.

Suitably oxidative stress should be avoided during the manufacturing process.

In one embodiment the present invention relates to the process A1 for preparing the pharmaceutical composition comprising the steps of:

- a) Melting vitamin E TPGS, preferably by heating, preferably by heating above its melting temperature;
- b) Optionally adding an anti-oxidant, e.g. EDTA to the molten mass and mixing thoroughly;
- c) Adding the API, solvate or a pharmaceutically acceptable salt thereof, to the molten mass;
- d) Homogenizing the mixture of c);
- e) Formulating mixture d) into a capsule, suitably a HPMC capsule; and
- f) Optionally seal the capsule by banding.

In one embodiment, the oral dosage form is a soft or hard gel capsule. The capsules may be prepared according to methods known in the art, suitably filling the pharmaceutical composition of the present invention into the soft or hard capsule, e.g. a standard two-piece hard gelatin capsule. Suitable capsules contain the pharmaceutical composition of the invention in liquid, semi-solid or granules form.

Suitably the pharmaceutical composition of the present invention is in the form of hard capsule. Suitably the shell of the hard capsule is hypromellose (HPMC) based. Suitably the body of the cap of the shell is sealed together by banding. Banding capsule can be performed by applying a small amount of a water/ethanol mixture at the cap and body interface followed by a gentle warming to fuse the two capsule parts together or by capsule banding process where a thin layer of gelatin or HPMC is placed over the edge of the capsule cap and body. Banding provides additional advantage by delaying contact of the drug with gastric fluid.

The pharmaceutical composition of the invention is capable of mitigating the food effect on the API, i.e. mitigating the reduction of bioavailability in the presence of food, especially calcium-rich food.

In one embodiment the pharmaceutical composition of the invention, preferably in an oral dosage form, comprising an API or a pharmaceutically acceptable salt thereof and at least one surfactant, wherein more than 40%, more than 50%, preferably more than 55%, more preferably more than 60%, more preferably more than 65%, more preferably more than 70%, more preferably more than 80%, more preferably more than 90% of the API is released in a dissolution test performed in the presence of excessive amount of calcium, typically measured at a definite time point after the addition of the drug. Typically the dissolution test is carried out as exemplified in Example 4. Typically the dissolution is measured at least 20 minutes, at least 30 minutes, at least 45 minutes, at least 60 minutes, or at the time point when the dissolution reaches plateau, suitably at 30 minutes, after the addition of eltrombopag into the test solution.

Normally the API forms complex with calcium. Thus the term “excessive amount of calcium” refers to the molar ratio of calcium over the API is suitably

higher than 20, suitably higher than 40, suitably higher than 60, suitably higher than 80. Suitably the term “excessive amount of calcium” refers to the molar ratio of calcium over the API is between 40 to 80, suitably between 40 to 60.

In one embodiment, the amount of the API released from the pharmaceutical composition of the invention is not reduced by more than 50%, suitably not reduced by more than 40%, suitably not reduced by more than 30%, suitably not reduced by more than 25%, suitably not reduced by more than 20% in the dissolution test carried out in the presence of excessive calcium when compared to that in the absence of calcium, while other conditions are kept identical, typically measured at a definite time point after the addition of the drug. Suitably the dissolution tests are carried out substantially according to Example 4, or according to Example 4A. Suitably the amount of released API is measured at 30 minutes after the addition of the pharmaceutical composition of the invention to the test medium.

In one embodiment the pharmaceutical composition of the invention, preferably in an oral dosage form, comprising an API or a pharmaceutically acceptable salt thereof and at least one surfactant, e.g. vitamin E TPGS, wherein plasma API AUC_{0-∞} is not reduced by more than 40%, preferably not reduced by more than 35%, 30%, 25%, 20%, 15%, 10% when the pharmaceutical composition is taken with a high-calcium, moderate-fat, moderate-calorie meal. A standard high-calcium, moderate-fat, moderate-calorie meal contains about 372 calories ± 20% , about 9 g ± 10 % fat, and about 448 mg ± 10% calcium. Preferably a standard high-calcium, moderate-fat, moderate-calorie meal contains about 372 calories, about 9 g fat, and about 448 mg calcium. In one preferred embodiment plasma the API AUC_{0-∞} is not reduced by more than 20% when the pharmaceutical composition is taken with a high-calcium, moderate-fat, moderate-calorie meal.

In one embodiment the pharmaceutical composition of the invention, preferably in an oral dosage form, comprising an API or a pharmaceutically acceptable salt thereof and at least one surfactant, e.g. vitamin E TPGS, wherein plasma API C_{max} is not reduced by more than 40%, preferably not reduced by more than 35%, 30%, 25%, 20%, 15%, 10% when the pharmaceutical composition is

taken with a high-calcium, moderate-fat, moderate-calorie meal. In one preferred embodiment plasma API C_{max} is not reduced by more than 20% when the pharmaceutical composition is taken with a high-calcium, moderate-fat, moderate-calorie meal.

In one embodiment the pharmaceutical composition of the invention, preferably in an oral dosage form, comprising an API or a pharmaceutically acceptable salt thereof and at least one surfactant, e.g. vitamin E TPGS, wherein plasma API AUC_{0-∞} taken with a high-calcium, moderate-fat, moderate-calorie meal is within about 80% and about 125%, suitably within about 80% and about 100%, suitable within about 80% and about 90% of the AUC_{0-∞} taken without a meal, e.g. on an empty stomach. In one embodiment the pharmaceutical composition of the invention, preferably in an oral dosage form, comprising an API or a pharmaceutically acceptable salt thereof and at least one surfactant, e.g. vitamin E TPGS, wherein plasma API C_{max} taken with a high-calcium, moderate-fat, moderate-calorie meal is within about 80% and about 125%, suitably within about 80% and about 100%, suitable within about 80% and about 90% of the C_{max} taken without a meal, e.g. on an empty stomach.

As used herein, the term “about” in relation to a numerical value x means, for example, +/-10%, suitably +/-5%, +/-2%.

In one embodiment there is no requirement to take the pharmaceutical composition “on an empty stomach (e.g. 1 hour before or 2 hours after a meal)” and/or “at least 2 hours before or 4 hours after other medications (e.g., antacids), calcium-rich foods and/or ” in the drug label of the medicine containing the pharmaceutical composition of the invention.

Without further elaboration, it is believed that one skilled in the art can, using the preceding description, utilize the present invention to its fullest extent. The following Examples, therefore, are to be construed as merely illustrative and not a limitation of the scope of the present invention.

EXAMPLES**EXAMPLE 1**

Capsules comprising different amount of vit E TPGS

Capsules comprising eltrombopag olamine and Vitamin E TPGS shown in Table 5 were prepared.

Table 5

Capsule strength (mg)	Composition (formulation 1)	mg/capsule	mg/capsule	%w/w
75	ETB115	95.6*	1.912	20.0
	Vit E TPGS	382	7.640	80.0
	Total	477.6	9.552	100.0

*Free acid equivalent

* As used in the tables of the Examples, %w/w indicates each components' weight percentage of the total composition. For example ETB115 95.6mg is the 20% of eltrombopag bis-olamine of a total of 477.6 mg of the composition (drug load).

Hard gel capsules were prepared as follows: VitE TPGS is first melted at 60-70C in a suitable container depending on the batch weight. Then compound is added and continuously mixed with a homogenizer. Aliquots equivalent to the fill weight are filled into the body of the capsules and allowed to cool to room temp. In case of hard capsules, the body is closed with the cap. The body and cap of this two-piece hard shell capsule is securely sealed. The sealing can be done either by spraying

a small amount of a water/ethanol mixture at the cap and body interface followed by a gentle warming to fuse the two capsule parts together or by capsule banding process where a thin layer of gelatin or HPMC is placed over the edge of the capsule cap and body. In both cases, specialized automated machines can be employed.

Example 2

Capsules comprising eltrombopag, Vit E TPGS and various anti-oxidants

Table 6

Capsule strength (mg)	Composition (formulation 7)	mg/capsule	%w/w
75	ETB115	95.6*	20.0
	VitE TPGS	381.45	79.9
	BHA	0.55	0.12
	Total	477.6	100.0

Table 7

Components	Composition per unit [%]	Composition per unit 25 mg [mg/unit] (formulation 14A)	Composition per unit 50 mg [mg/unit] (formulation 14B)	Composition per unit 75 mg [mg/unit] (formulation 14C)
Eltrombopag Olamine	20.0	31.9	63.8	95.7
(Vitamin E TPGS)	77.9	124.0	248.0	372.0
Edetate Disodium	2.1	3.3	6.7	10.0
Hypromellose capsule	-	1 capsule	1 capsule	1 capsule
Total	100.0	159.2	318.5	477.7

Capsule formulations containing Eltrombopag, vit E TPGS and an anti-oxidant were manufactured in a similar manner as described in example 1. The anti-oxidant was added after vit E TPGS was melted and it is further mixed by stirring. Drug was added then to the mixture. Whenever possible, it is preferred to avoid/reduce oxidative stress during the manufacture process and to minimize exposure to water during storage.

Example 3

Capsule formulations comprising eltrombopag and various surfactants

Table 8

Capsule strength (mg)	Composition (formulation 9)	mg/capsule	%w/w
75	ETB115	95.6	25.8
	Kolliphor Rh 40	275.0	74.2
	Total	370.6	100.0
50	ETB115	63.8	18.8
	Kolliphor Rh 40	275.0	81.2
	Total	338.8	100.0

Table 9

Capsule strength (mg)	Composition (formulation 10)	mg/capsule	%w/w
75	ETB115	95.6	25.8
	Vit E TPGS	137.5	37.1
	Kolliphor Rh 40	137.5	37.1
	Total	370.6	100.0
50	ETB115	63.8	18.8
	Vit E TPGS	137.5	40.6
	Kolliphor Rh 40	137.5	40.6
	Total	338.8	100.0

Capsule formulations containing eltrombopag and various surfactants were manufactured in a similar manner as described in example 1.

Table 10 (formulation 11)

ETB115-ORA-0024-004 (MEPC 7)		
Composition [^]	Gm	% w/w
VitETPGS	15	18.75%
Span 80	15	18.75%
Miglyol 812N	5	6.25%
Labrasol	25	31.25%
Ethanol	20	25%
	80	100%

[^] Placebo MEPC was prepared first followed by addition of ETB115 at 2%w/v (20mg/mL)

Table 11 (formulation 15)

MEPC 3	%
---------------	----------

Kolliphor RH 40	45
Maisine CC	27
Propylene glycol	18
Ethanol	10
Total	100

50mg ETB115 is suspended in 1ml of the above concentrate. (2 x size 0 capsules each with 0.5ml of the formulation = equivalent to 50mg ETB115 was tested for dissolution in MOPS + SIF buffer)

Example 4

Dissolution test

Reagents

MOPS Buffer
(3-morpholinopropane-1-sulfonic acid)

e.g. Sigma Aldrich, AR grade or equivalent

SIF powder
(Sodium taurocholate + Lecithin)

e.g. Bio-relevant, FaSSIF/FeSSIF/FassGF powder or equivalent

Dissolution Condition

Test Medium

MOPS+SIF Buffer

Dissolve 20.9 g of MOPS buffer in 1 liter of water and mixed well. Adjust the pH 6.8 ± 0.1 with Sodium hydroxide. Add 0.74 g of SIF powder and stir gently until mixed well

Test Medium for Reference Preparation

0.5% Tween 80 in MOPS buffer pH 6.8

Dissolve 20.9 g of MOPS buffer in 1 liter of water and mixed well. Adjust the pH 6.8 ± 0.1 with Sodium hydroxide. Add 5ml (5.2g) Tween-80 in 1 liter MOPS buffer and Mixed Well.

Speed of Rotation

100 \pm 3 rpm

Volume of test medium

900ml

Temperature

37.0 \pm 0.5°C

Test Procedure

Test solution (profile)

This test can be carried out by an automated dissolution test system. In dissolution profiles, Samples are required to collect at 30, 45, 60, 90 and 105 minutes (Infinity testing at 200rpm). Do not replace the medium removed. At each time point, withdraw 10 mL of the solution and immediately filter through a Pall Acrodisc PSF GxF/Glass 1 μ m Automated Certified syringe filters (recommended) or ROBY25/GF55 glass fiber filter 0.7 μ m. Collect the test solution into a container for UV measurement.

Calcium challenge procedure:

At 30 minutes prior to adding the dosage form, 139mg, 1185mg, 1249mg of calcium chloride, equivalent to 50 mg or 427 mg or 450 mg of calcium element, is added and allowed to dissolve completely. After 30 minutes, add the dosage form and consider this as time 0 for defining the dissolution profile.

Sampling time for profile generation may be adapted based on project need.

Reference solutions	This corresponds to 0.055 mg/mL as free acid of ETB115	
Wavelengths	Sample wavelength	424 nm

Dissolution test according to EXAMPLE 4 was carried out for formulation 1 in 75mg in comparison to Promacta. As shown in Figure 1, vitamin E TPGS has effectively maintained the dissolution rate in the presence of excessive amount of calcium.

The anti-calcium effect by other surfactants or mixture of surfactants is shown in Figure 4.

Data in Figure 4 were generated following EXAMPLE 4

- a. 0.5% Gelucire 48/16 (0.5% = 4.5grams) was added to 900ml MOPS + SIF buffer,
- b. after 40minutes - 427mg Ca (solution) was added to the above media.
- c. After 30minutes - the Promacta 75mg tablet was added to this media (containing Gelucire 48/16 and Ca) and tested for dissolution.
- d. Separate "Control - without the addition of calcium" dissolution was also done for comparison.

Similarly, other suitable surfactants can be selected as above. Following the teaching of this example, the effect of surfactant on the API of the Invention can be tested.

EXAMPLE 4A

Modifications of dissolution test

The above dissolution test was designed for eltrombopag. Minor modifications depending on the compound to be tested is recommended for the optimal results. For example each one of the below parameters or any combination thereof can be adjusted depending on the compound to be tested and the chelating metal to be tested.

Speed of rotation	75±3 rpm or 100 ± 4 rpm
Note	Selection of rotation speed to be adapted based on the Drug substance and Drug product properties and type of formulation to be evaluated. Other speed within the common general knowledge can also be chosen.
Test medium	pH 6.8 MOPS Buffer or pH 6.8 MOPS + SIF buffer
Note	Addition of SIF (Sodium Taurocholate and Lecithin) and amount of SIF in MOPS buffer to be adapted based on the drug substance properties. In principle the more soluble the drug, the less amount of SIF or no SIF is required if the drug has good solubility.
Evaluation	<i>Quantitative evaluation performed using suitable quantitation techniques like UV or HPLC method.</i> Determine the absorbance or peak area in the chromatogram of the test medium, reference and test solutions using a suitable UV-VIS spectrophotometer or HPLC techniques.
Note	Selection of suitable quantification techniques adapted based on the nature and properties of drug substance and drug product and also Metal ion selected for <i>In vitro</i> assessment of negative food effect.

EXAMPLE 5

Impact of vitamin E TPGS concentration and sequence of calcium addition on dissolution

To further understand effect of Vitamin E TPGS on drug release, dissolution of 75 mg Promacta® tablets was performed; (1) with different concentrations of Vitamin E TPGS and (2) sequence of calcium addition in MOPS buffer.

Dissolution in MOPS buffer to understand the effect of Vitamin E TPGS on % release from 75 mg Promacta® tablets

Table 12

Experiment no	Amount of calcium (addition time ¹)	Vitamin E TPGS concentration % w/v, (addition time ¹)
1	50mg elemental Calcium (-30min)	0.1, (-70 min)
2		0.3, (-70 min)
3		0.5, (-70 min)
4	50mg elemental Calcium (+60min)	0.1, (-70 min)
5		0.3, (-70 min)
6		0.45, (-70 min)

Results of the dissolution from group with calcium added 30 min before addition of Promacta® tablet showed Vitamin E TPGS concentration dependent release (Figure 3A). As Vitamin E TPGS is increased from 0.1 to 0.5% w/v in dissolution media, the % of drug release at 105 min increases from 50 to 82%.

On the contrary, when the calcium is added 60 min after addition of Promacta® tablets, the release is relatively unaffected by the concentration of Vitamin E TPGS at 0.3 and 0.45% w/v (Figure 3B).

From above results, the effect of Vitamin E TPGS can be summed up as 1) eltrombopag dissolution shows concentration dependence when calcium is present in the media from the beginning which is likely to be the situation in-vivo, 2) eltrombopag once solubilized in dissolution media in presence of Vitamin E TPGS, effect of calcium mediated drop in dissolution is mitigated. So, partially solubilized suspension of DS in Vitamin E TPGS could contribute to reduce calcium mediated food effect.

EXAMPLE 6

Effects of surfactants with different HLB value on drug dissolution

HLB solution preparation:

HLB-8 solution:

Mixed 32.5ml Span-80 and 17.5ml Tween-80.

HLB-10.7 solution:
Mixed 20ml Span-80 and 30ml Tween-80.

HLB-12.8 solution:
Mixed 10ml Span-80 and 40ml Tween-80.

Dissolution media Preparation:

0.1% solution of HLB-4.3 (Span-80) in MOPS:
Mixed 2ml of SPAN-80 with 2000ml of MOPS buffer. Mixed well.

0.1% solution of HLB-8 in MOPS:
Mixed 2ml of HLB-8 with 2000ml of MOPS buffer. Mixed well.

0.1% solution of HLB-10.7 in MOPS:
Mixed 2ml of HLB-10.7 solution with 2000ml of MOPS buffer. Mixed well.

0.1% solution of HLB-12.8 in MOPS:
Mixed 2ml of HLB-12.8 Solution with 2000ml of MOPS buffer. Mixed well.

0.1% solution of HLB-15.0(Tween-80) in MOPS:
Mixed 2ml of Tween-80 with 2000ml of MOPS buffer. Mixed well.

Control Preparation:

Dropped 1 Promacta tablet, kept in sinker in 900ml with respective dissolution media.
After each specified time point, sample was withdrawn automatically and filtered through ROBY 25/GF 55 Filter. Dissolution results are given in Fig. 5A.

With 427mg calcium Preparation:

Added 1185mg CaCl₂ 30 minutes before dropping 1 Promacta tablet. Dropped 1 tablet, kept in sinker in 900ml with respective dissolution media. After each specified time point, sample was withdrawn automatically and filtered through ROBY 25/GF 55 Filter. Dissolution results are given in Fig. 5B.

The results showed that with surfactants with higher HLB value resulted in higher solubilization of ETB115 and had stronger anti-calcium effect. Similar trend can be expected if ETB115 is replaced with another API of the Invention.

EXAMPLE 7

Compositions comprising phospholipids

Lipid based formulations were prepared with varying ratio of the components as per weights in Table 13. Lipoid E80 S(Cas No. 93685-90-6), glycocholic acid, glycerol and ETB115 were first dissolved in organic solvent in a round bottom flask to obtain a optically clear solution and then solvent was evaporated gradually, which results is solid gel like cake. The resulting solid gel cake readily disperses with water with brief use of sonication and heat. The resulting viscous fluid gel like formulation was filled into hard gelatin capsules by weight and allowed to cool to room temperature. The body is closed with the cap and used for dissolution studies.

Alternatively the resulting solid gel cake can be directly filled into capsules by further extrusion or can be hydrated with required amount of non-aqueous hydrophilic or lipophilic solvent for filling in soft gel capsules.

The lipid particles size tested upon dilution with water were found to be around 190nm in size with good uniformity as determined by Malvern dynamic light scattering technique. We anticipate that the true particle size of the formulations would be much smaller if tested undiluted.

Surprisingly, we found that lipid: drug weight ratio of 5:1 or 9:1 have almost same degree of drug associated, it was 94% for 5:1 vs 104% for 9:1 ratio signifying almost majority of drug is associated with the lipid or micelle. Association was measured by filtration through 0.2 micron filter and concentration of the filtrate was assessed using Liquid chromatography. The addition of bile salt and glycerol improves the dispersion and hydration time. The increasing amount of bile salt decreases the mixed micelles particle size. Furthermore, upon dilution with GI simulated fluids, these particles were found be stable in both Fed (FeSSIF) and Fasted (FaSSIF) state simulated media as observed by no dramatic change in the particle size. In fact, in presence of Fed state media, the lipid based mixed micelle formulation dispersed more readily into micelle-based formulation. Formulation also showed increased solubility in both fast and fed simulated GI fluids which justifies that compound had less tendency to precipitate upon dilution and a

lipid dispersion is formed upon dilution (Table 15). Lastly, the improved solubility of more than 100x upon dilution with bio relevant media translated to enhanced dissolution in dissolution studies (following the protocol as described in EXAMPLE 4) when compared to control Promacta formulation. Mixed micelle formulation L-F2 showed >90% dissolution within 30min and showed that negative effect of calcium on dissolution is being mitigated by using lipid based mixed micelle formulation.

Lipid complex with hydrophilic co-solvent (F3 and F4):

To prepare lipid complex, capsules were prepared as per weights in Table 14. Lipoid P LPC (Cas No. 9008-30-4) or Lipoid E80 S, PEG 300 co-solvent and ETB115 were first dissolved in organic solvent (DCM/Methanol 1:1) in a round bottom flask and then solvent was evaporated which results in solid cake. This solid cake is then hydrated with required amount of distilled water which results into a highly viscous gel like formulation. Upon rehydration the fluid aliquots equivalent to the fill weight and filled into the body of the capsules and allowed to cool to room temperature. In case of hard capsules, the body is closed with the cap.

These lipid complex formulation dispersed with water with brief use of sonication and heat. The lipid particles formed after hydration with water are around 31nm in size (LPC lipid L-F4) and around 490nm (Lipoid E 80S L-F3) with good uniformity as determined by malvern dynamic light scattering technique. The composition L-F4 was selected based on the highest association, stability upon dilution with biorelevant media and also due to the small micelle like particle formation. The weight ratio of 7:1 was found to provide the highest level of association/entrapment of drug, close to 100%. Alternately, 5:1 ratio can also be utilized as it shows minimum impact on association (L-F4-B) Lipid complex formulation showed enhanced stability upon dilution in biorelevant media as evident by no major change in size (Table 15) and also provided better solubility. The enhanced solubility in biorelevant media also resulted in better dissolution profile (70-80% in 60 min) compared to control Promacta in the dissolution studies described in

EXAMPLE 4. The results are shown in Figure 6. Although, the dissolution of F3 was low, we believe that F3 formulation could be further optimized to improve solubility and stability as we had some technical processing challenges while scaling up the formulations.

In summary, it was demonstrated that lipid based formulation, mixed micelle or lipid complexes both can facilitate high association of the compound with lipids which in turn resulted in improved solubilization upon dilution in GI biorelevant media and provided close to complete dissolution of drug in presence of calcium where perhaps compound was shielded from interaction with calcium.

Examples L-F2

Table 13. Capsules comprising Lipid based Micelles with bile acid are shown below:

Composition	%w/w*	mg/capsules	%w/w	mg/capsule
		(formulation L-F2)		(formulation L-F2-B)
ETB115	5.5	95.6	14.2	95.6
Lipoid E80 S	49.6	860	71.2	478.0
Lipoid P LPC 80	NA	NA	NA	NA
Na Glycocholic acid	38.4	666.2	12.5	83.7
PEG 300	NA	NA	NA	NA
Glycerol	6.4	111.7	2.1	14.3
Physical appearance before hydration		Wine red, slightly viscous gel cake		Wine red, slightly viscous gel cake
Hydration Water in gms		1000		1195
Total	100	1733.5	100	671.59
% Association		104		94
Lipid:drug ratio wt		9		5

*Free acid equivalent

* As used in the tables of the Examples, %w/w indicates each components' weight percentage of the total composition. For example ETB115 95.6mg is the 5.5% of eltrombopag bis-olamine of a total of 1638 mg of the composition (drug load).

Examples L-F4 and L-F3

Table 14. Capsules containing Lipoid P LPC, Lipoid E 80 S Lipid complex formulations are shown below:

Composition	%w/w	mg/capsule	%w/w	mg/capsule	%w/w	mg/capsule	%w/w	mg/capsule
		(formulation L-F4)		(formulation L-F4-B)		(formulation L-F4-C)		(formulation L-F3)
ETB115	11.4	95.6	13.8	95.6	16.7	95.6	9.1	95.6
Lipoid E80 S	NA	NA	69.0	478.0	83.3	478.0	82.0	860.4
Lipoid P LPC 80	79.7	669.8	NA	NA	NA	NA	NA	NA
Na Glycocholic acid	NA							
PEG 300	8.9	74.56	17.2	119.5		0.0	8.9	93.2
Glycerol								
Physical appearance before hydration		Wine red, highly viscous cake		Wine red, highly viscous cake		Wine red, highly viscous cake		Wine red, highly viscous cake
Hydration Water in gms		1250		1195		1195.0		1623.0
Total	100	839.96	100	693.1	100.0	573.6	100.0	1049.2
% Association		98		100		42.0		101.2
Lipid:drug ratio wt		7		5		5		9

*Free acid equivalent

* As used in the tables of the Examples, %w/w indicates each components' weight percentage of the total composition. For example ETB115 95.6mg is the 11.4% of eltrombopag bis-olamine of a total of 744.36 mg of the composition (drug load).

Table 15. Particle size and solubility data for L-F2 mixed micelle and L-F4 lipid complex formulations

Formulation Type	Dilution Media	Particle size (nm)	Fraction solubility (mg/ml)
Formulation L-F2 mixed micelle	Water	186.7	na
	FaSSIF	241.4	5.35
	FeSSIF	46.14	6.41
Formulation L-F4 lipid complex	Water	31.07	na
	FaSSIF	22.11	4.59

	FeSSIF	25.89	4.88
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Similarly lipid formulations comprising API other eltrombopag can be prepared according to the teaching of this example.

EXAMPLE 8

PAMPA test

A study combining the experimental determination of dissolution and simultaneous flux through an artificial lipidic membrane was conducted on a 12.5, 25, 50 and 75 mg dose of Promacta and a 55 and 75 mg dose of formulation 1 (in Capsule) of the invention. In addition, a mixed micelle formulation (F2) was evaluated at a 37 mg dose.

FaSSIF media was prepared according to instruction from biorelevant.com. Additionally, each media are prepared containing 450 mg of elemental calcium, to mimic a high calcium meal and understand the impact of calcium and dissolution and resultant flux. This was achieved by adding calcium chloride to the media. The media are added, 900 mL in total, to a USP II apparatus, equipped with a paddle attachment for stirring. The dosage unit was introduced to each media. Dissolution analysis were conducted in each media: FaSSIF, V2; FaSSIF, V2 + Calcium. This described set-up is considered the donor compartment.

A receiver compartment was introduced into the USP II dissolution set-up. This consists of a minaturized USP II paddle set-up. The bottom of the vessel has a 0.45 um PVDF membrane. Concentrations of eltrombopag in the donor vessel and receiver vessel are monitored with fiber optic probes.

The capsule formulation containing about 80% vitamin E TPGS, as well as the F2 formulation has improved the Fasted + High Calcium/Fasted flux ratio significantly: Ratio of 0.9 for the 55 mg dose and for the F2 formulation (compared to 0.2 for Promacta @ 50 mg dose); Ratio of 0.7 for the 75 mg dose (compared to 0.3 for Promacta @ 75 mg dose).

PAMPA test can be used or modified if needed to test the anti-calcium effect on other APIs according to the teaching of the present invention.

EXAMPLE 9

Food effect study in health volunteers

The effect of food, high and low in calcium content, on the pharmacokinetics of an API of the Invention after administration of the pharmaceutical composition of the invention (e.g. capsule comprising (80% w/w vitamin E TPGS and 20% w/w of API) will be investigated. The treatment will consist of single oral doses administered in a fasted state, and in various fed conditions: high-fat high-calorie (HFHC) high-calcium meal, HFHC low-calcium meal, high-fat low-calorie (HFLC) low-calcium meal. Subjects will undergo 4 treatment periods with a washout of 7 to 10 days between 2 consecutive doses. The primary objective is to evaluate the effect of food, high or low in calcium, on the API of the Invention pharmacokinetics, including but not limited to the measure of AUC, Tmax and Cmax.

EXAMPLE 10

Doxycycline Vitamin E TPGS composition

Table 16

Composition	% w/w	Composition per unit (mg)
Doxycycline Monohydrate	21.7	105.85
Vitamin E TPGS	76.3	372.00
EDTA disodium salt	2.0	10.0
Hypromellose capsule size 1	NA	1 capsule
Total	100	487.85

Dissolution test was performed substantially similar to EXAMPLE 4 with the following changes:

Sample time is 10,15,30,45,60,75, 90min;

Paddle rotation speed was 75 rpm;

Test medium: pH 6.8 MOPS Buffer;

Calcium element concentration: 427mg calcium;

Wavelength : 274nm.

The calcium effect on Doxycycline is mitigated by the presence of Vitamin E TPGS as shown in Fig. 8.

WHAT IS CLAIMED IS:

1. A pharmaceutical composition in an oral dosage form comprising an API, a solvate or a pharmaceutically acceptable salt thereof, and vitamin E TPGS, wherein said API is not eltrombopag.
2. The pharmaceutical composition of claim 1, wherein the API is capable of chelating with polyvalent metal cation and forming API-polyvalent metal complex.
3. The pharmaceutical composition of claim 2, wherein the polyvalent metal is selected from the group consisting of iron, calcium, magnesium, aluminium, selenium and zinc.
4. The pharmaceutical composition of any one of the claims 1-3, wherein the weight of the API is not more than 40% of the total weight of API and vitamin E TPGS.
5. The pharmaceutical composition according to any one of the preceding claims, wherein the weight of the API from about 5% to about 30% of the total weight of the API and vitamin E TPGS.
6. The pharmaceutical composition of any one of the preceding claims further comprising at least one anti-oxidant, wherein preferably the anti-oxidant is selected from a list consisting of Vitamin E, Butylhydroxytoluol (BHT), Butylhydroxyanisol (BHA), Propyl gallate, ascorbyl palmitate, ascorbic acid, EDTA and sodium metabisulfite or a mixture thereof.
7. A pharmaceutical composition consisting essentially of or consisting of an API, a solvate or a pharmaceutically acceptable salt thereof, and vitamin E TPGS.

8. A pharmaceutical composition consisting essentially of or consisting of an API, or a pharmaceutically acceptable salt thereof, vitamin E TPGS and one anti-oxidant.
9. A pharmaceutical composition in an oral dosage form comprising an API or a pharmaceutically acceptable salt thereof and at least one micelle forming agent.
10. The pharmaceutical composition according to any one of the claims 14-16, wherein the weight of the API from 5% to 40% of the total weight of the API and the at least one micelle forming agent.
11. The pharmaceutical composition of claim 9 or 10, wherein the at least one micelle forming agent is a surfactant.
12. The pharmaceutical composition of any one of the claims 19, wherein the at least one surfactant is a non-ionic surfactant.
13. The pharmaceutical composition according to any one of the claims 14-20, wherein the at least one surfactant is selected from the list consisting of Vitamin E TPGS, PEG 40 hydrogenated castor oil (Cremophor RH 40 or Kolliphor RH40), PEG 15 hydroxystearate (Solutol HS 15), PEG 32 monostearate (Gelucire 48/16), Gelucire 44/14, Gelucire 50/13, labrasol, PEG 35 castor oil (Cremophor EL) and Polyoxyethylene (20) sorbitan monooleate (Polysorbate 80, Tween 80), or a mixture thereof.
14. The pharmaceutical composition according to any one of the claims 19-24, wherein the weight of the API is from 5% to 30% of the total weight of the API and the at least one surfactant.

15. The pharmaceutical composition of claim 9 or 10, wherein the at least one micelle forming agent is a phospholipid.
16. The pharmaceutical composition of claim 15, wherein the phospholipids is diacyl- phospholipids.
17. The pharmaceutical composition of claim 16, wherein the phospholipids is monoacyl- phospholipids.
18. The pharmaceutical composition of any one of the claims 15-17 further comprising at least one bile salt.
19. The pharmaceutical composition of claim 18 wherein the bile salt is sodium taurocholate or sodium glycocholate,
20. The pharmaceutical composition of any one of the claim 15 to 19, wherein the weight of the API, calculated in its free form, is between about 5% to about 30% of the total weight of the pharmaceutical composition.
21. The pharmaceutical composition according to any one of the preceding claims in the form of capsule.
22. The pharmaceutical composition according to any one of the preceding claims, wherein more than 40% of the API is released within 45 minutes in a dissolution test performed in the presence of excessive amount of calcium.
23. The pharmaceutical composition according to any one of the preceding claims, wherein plasma API AUC_{0-∞} taken with a high-calcium, moderate-fat, moderate-calorie meal is within about 80% and about 125% within of the AUC_{0-∞} taken on an empty stomach.

24. A process for preparing the pharmaceutical composition according to any one of the claims 1-8, comprising the steps of:
 - a) Melting vitamin E TPGS, preferably by heating above its melting temperature;
 - b) Optionally adding an anti-oxidant, e.g. EDTA to the molten mass and mixing thoroughly
 - c) Adding the API or a pharmaceutically acceptable salt thereof to the molten mass and stirring to mix thoroughly;
 - d) Filling mixture c) into a capsule, suitably a HPMC capsule; and
 - e) Optionally seal the capsule by banding.

25. The pharmaceutical composition of any one of the preceding claims, wherein the API is doxycycline.

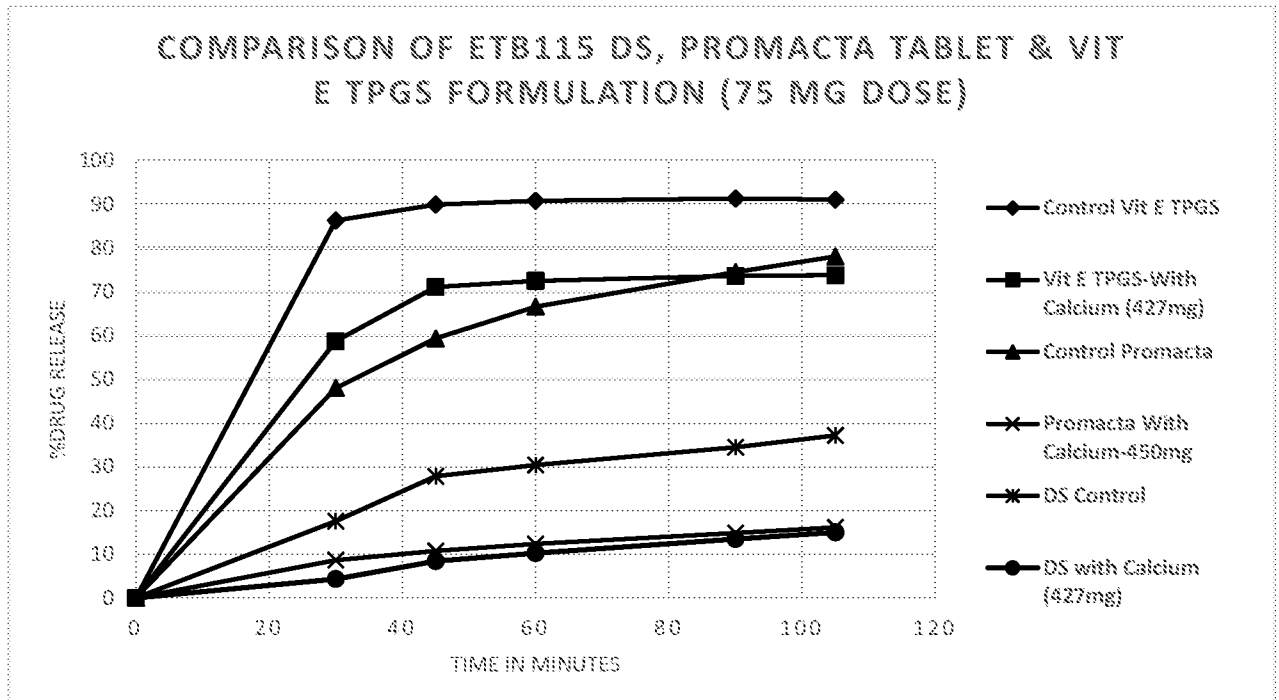


Fig. 1

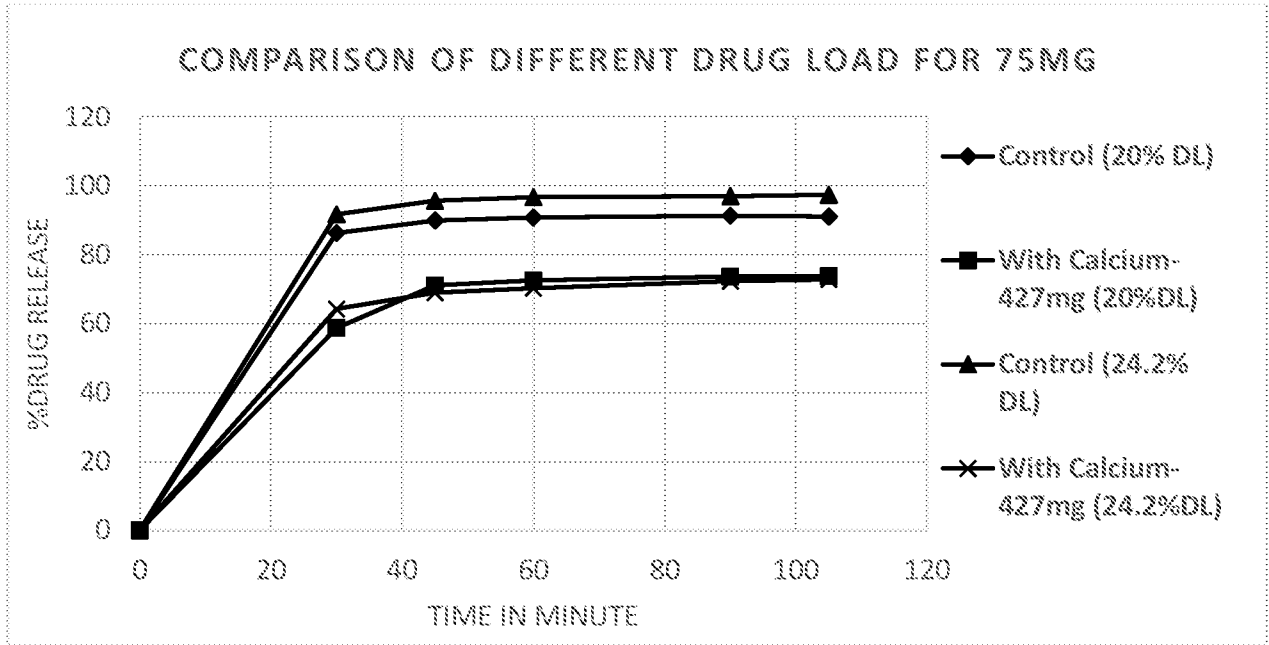


Fig. 2A

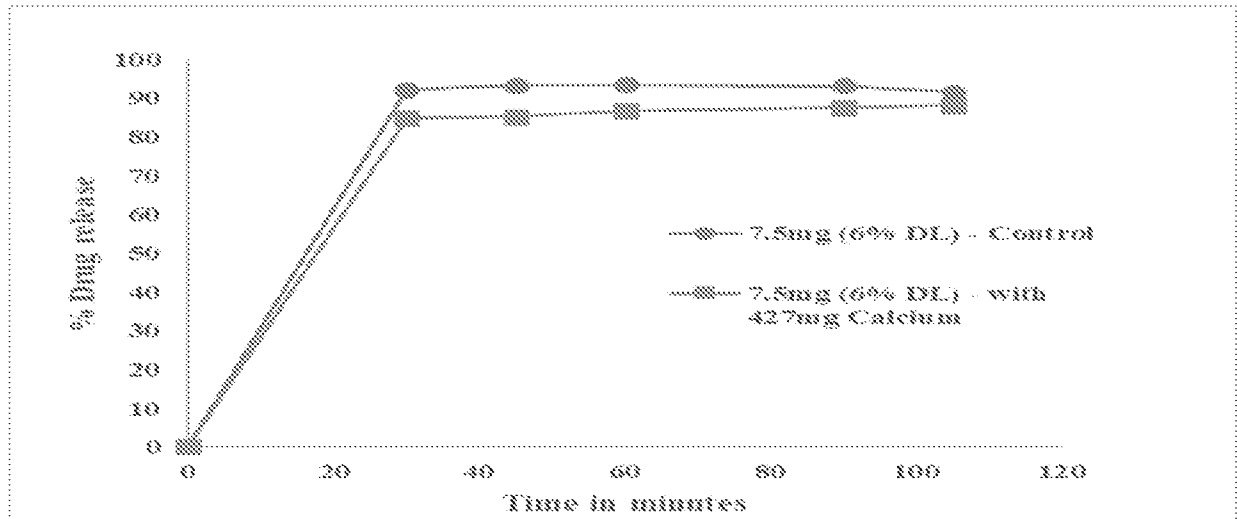


Fig. 2B

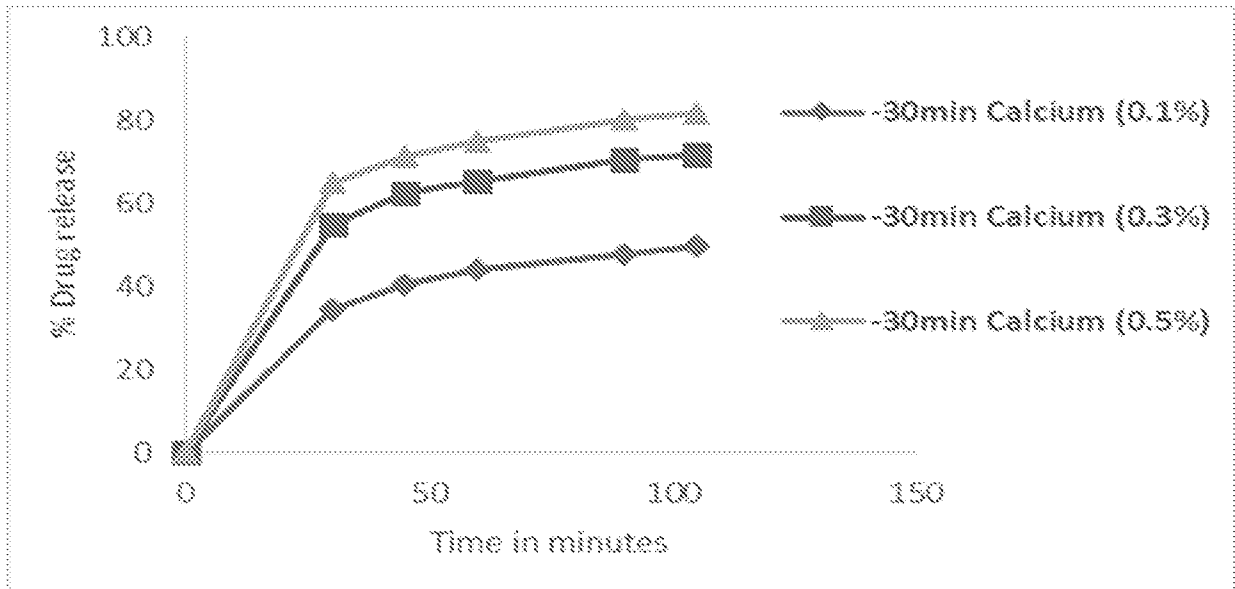


Fig. 3A

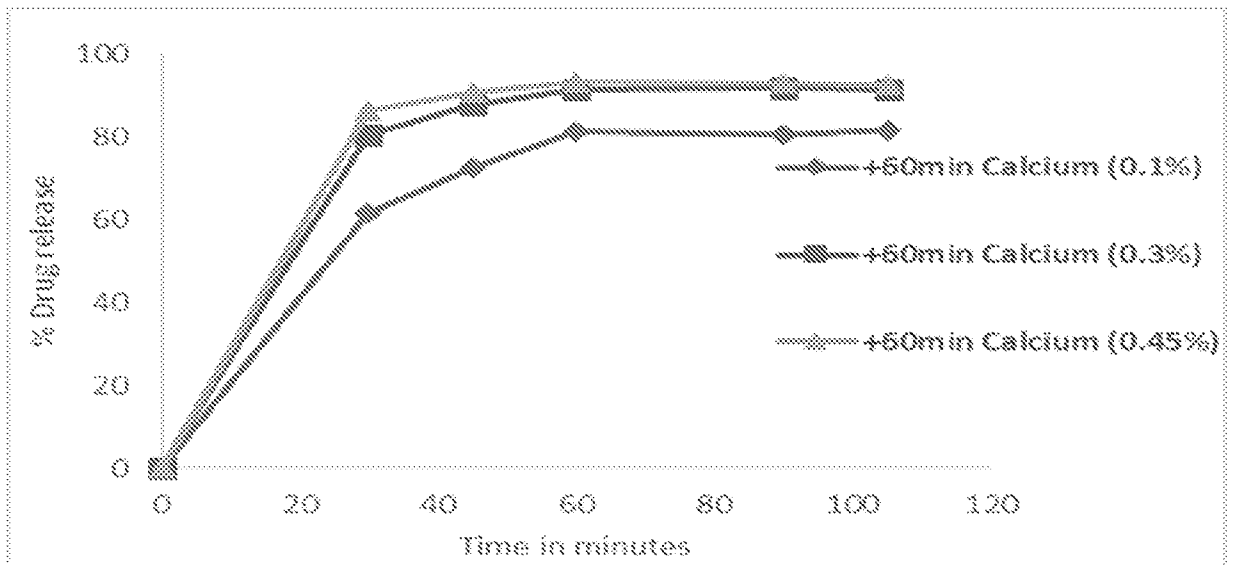


Fig. 3B

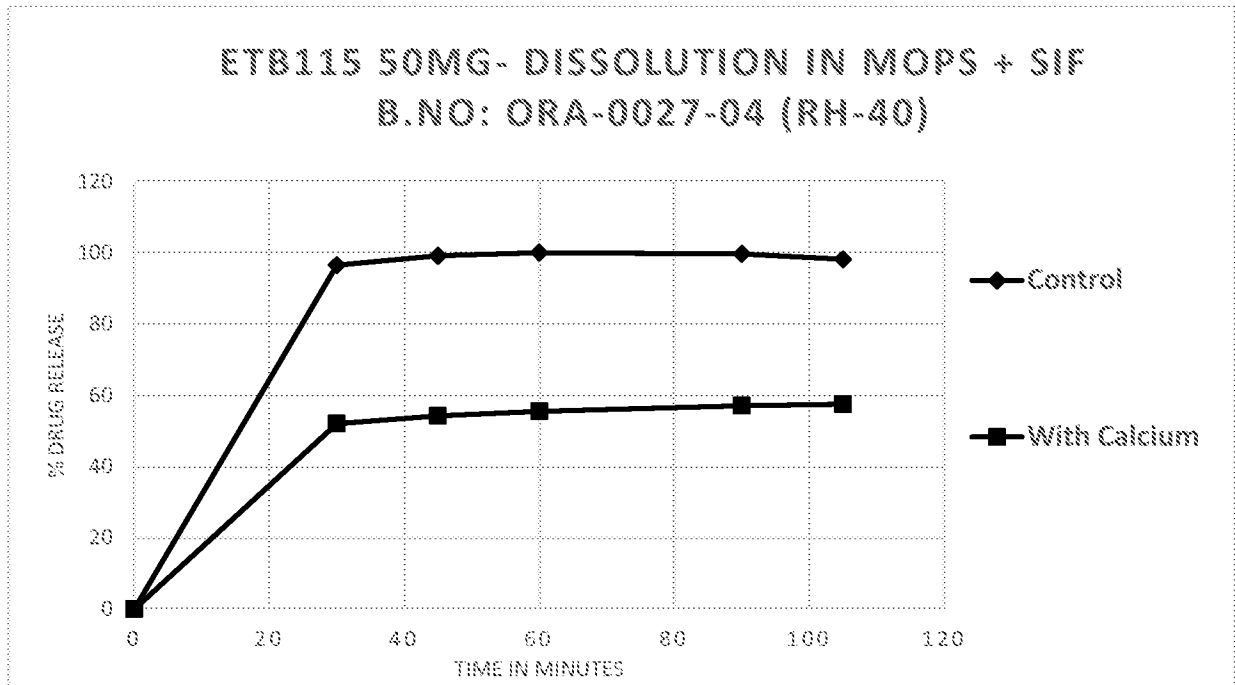


Fig. 4A

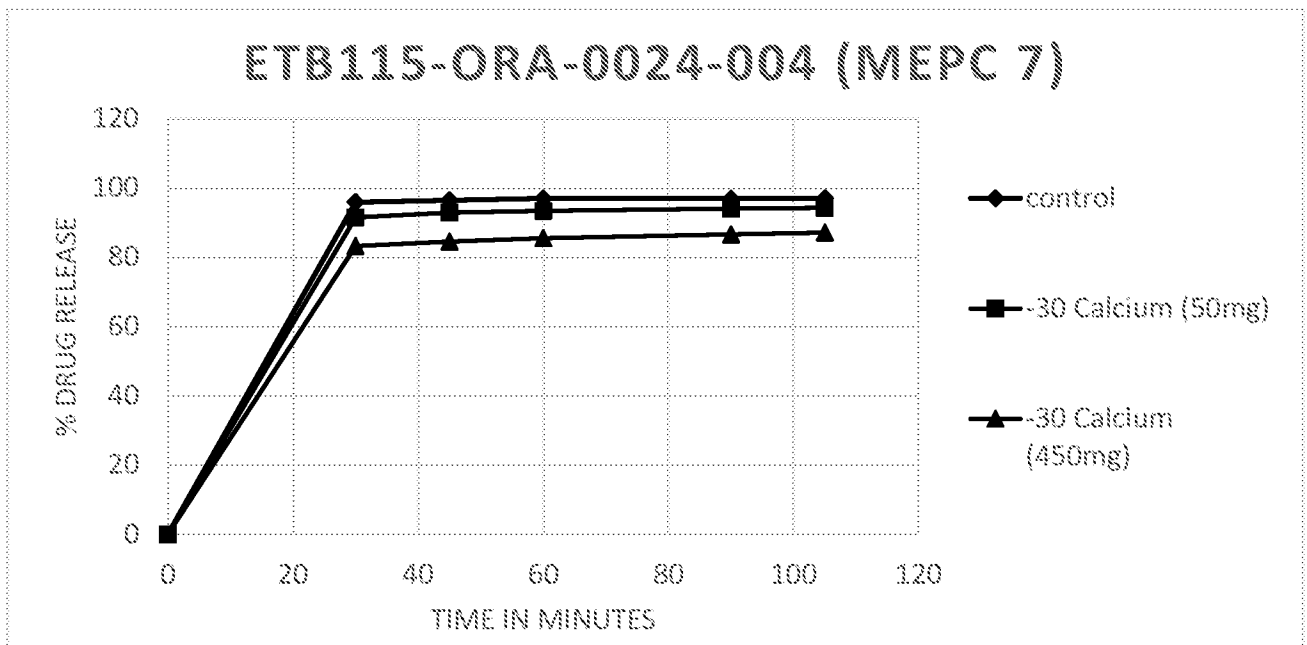


Fig. 4B

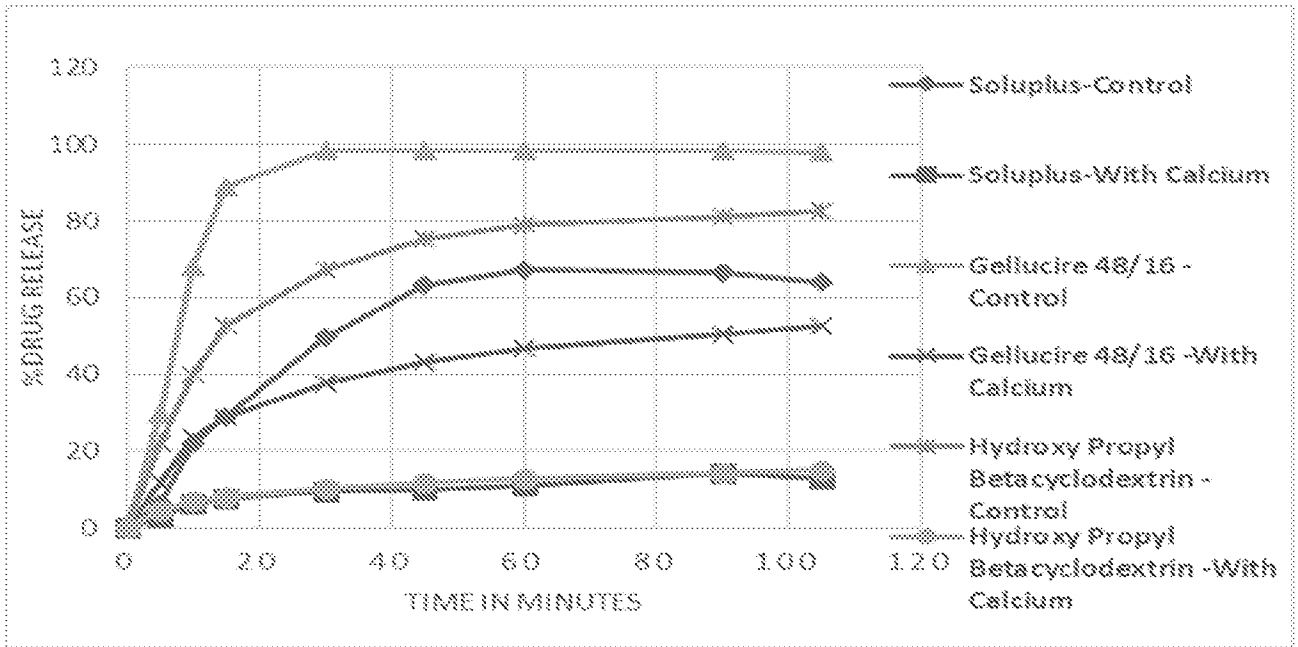


Fig. 4C

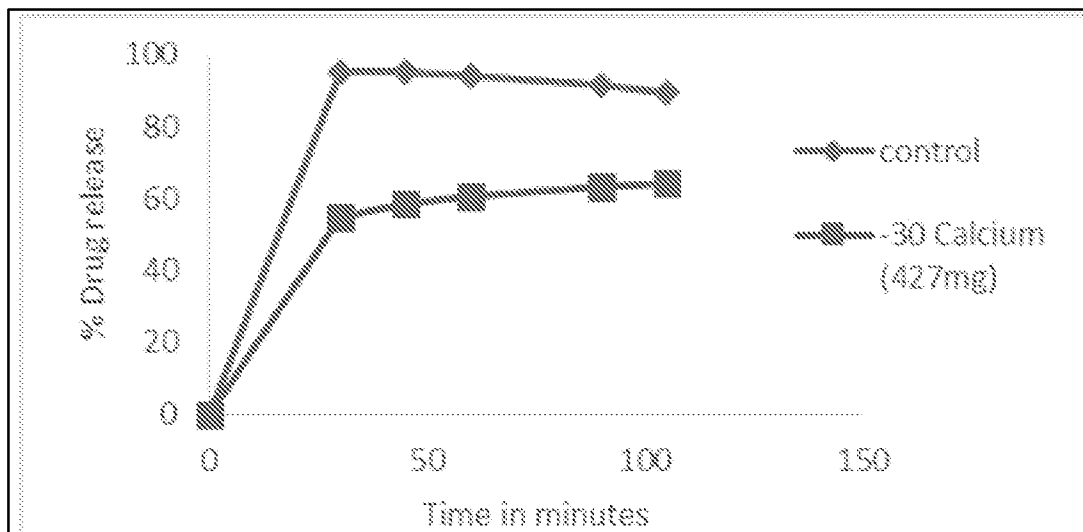


Fig. 4D

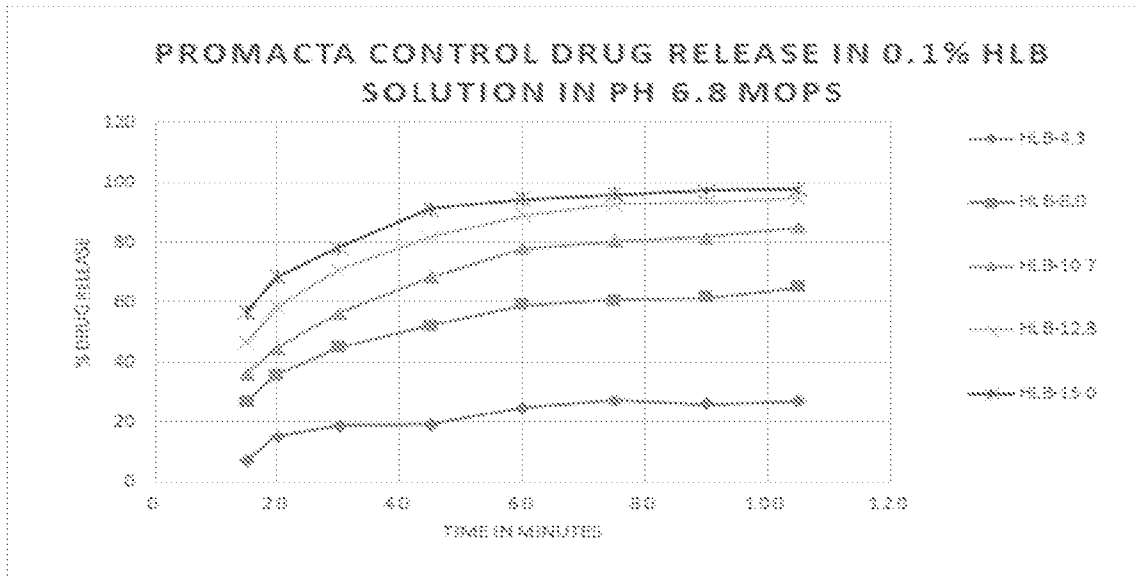


Fig. 5A

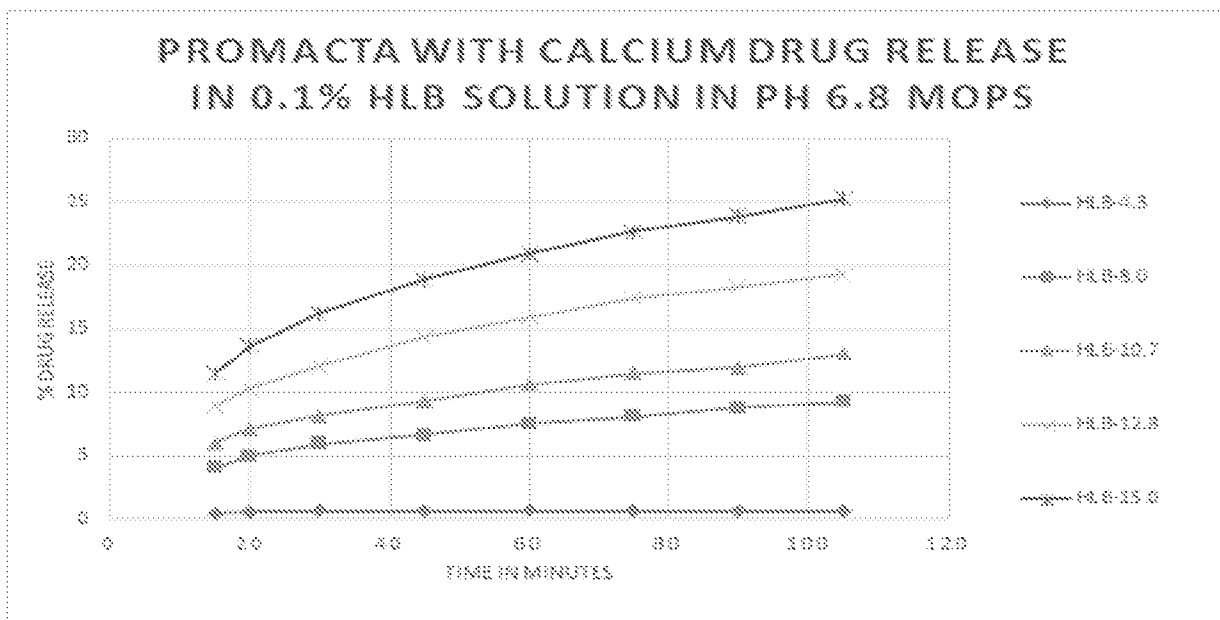


Fig. 5B

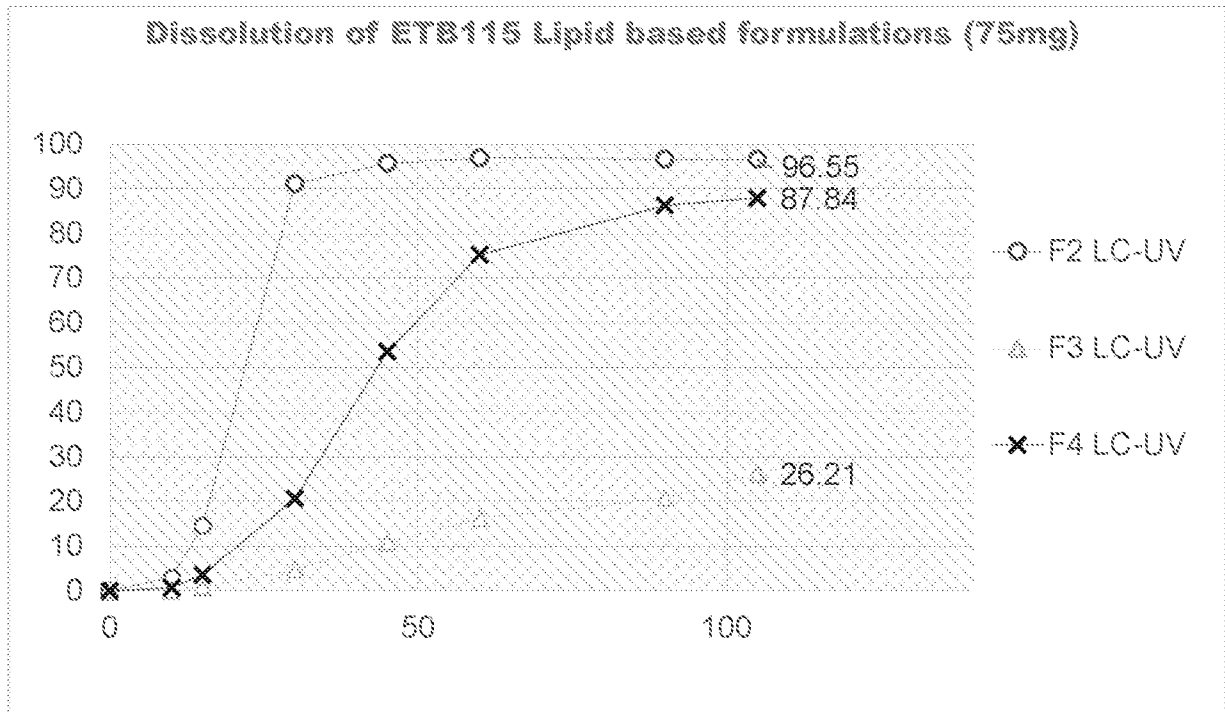


Fig. 6

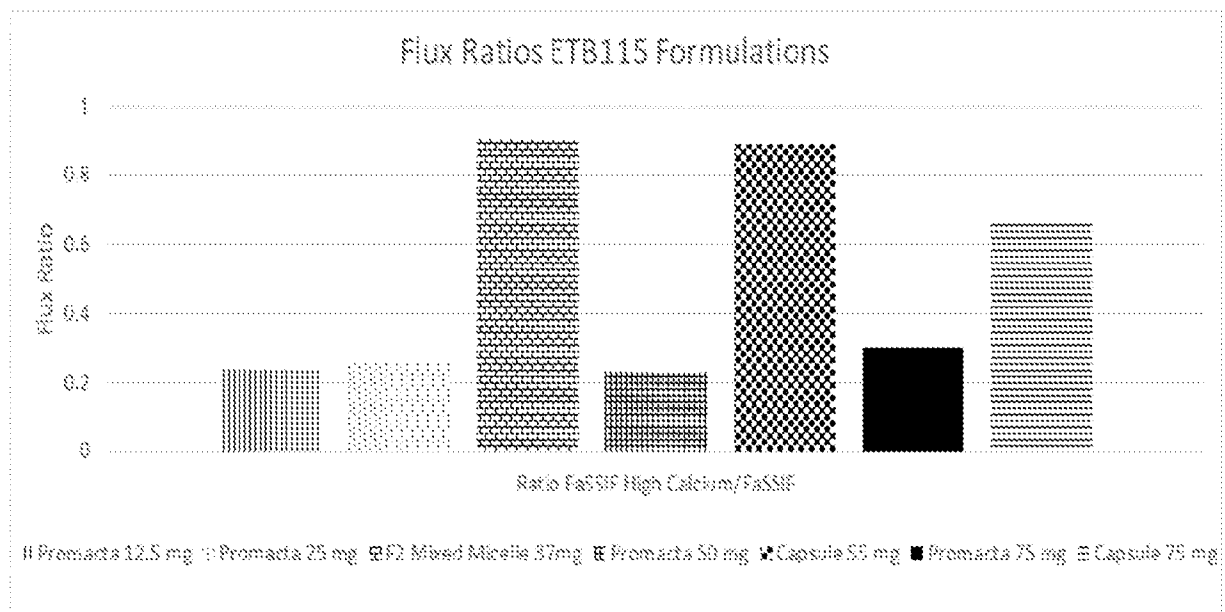


Fig. 7

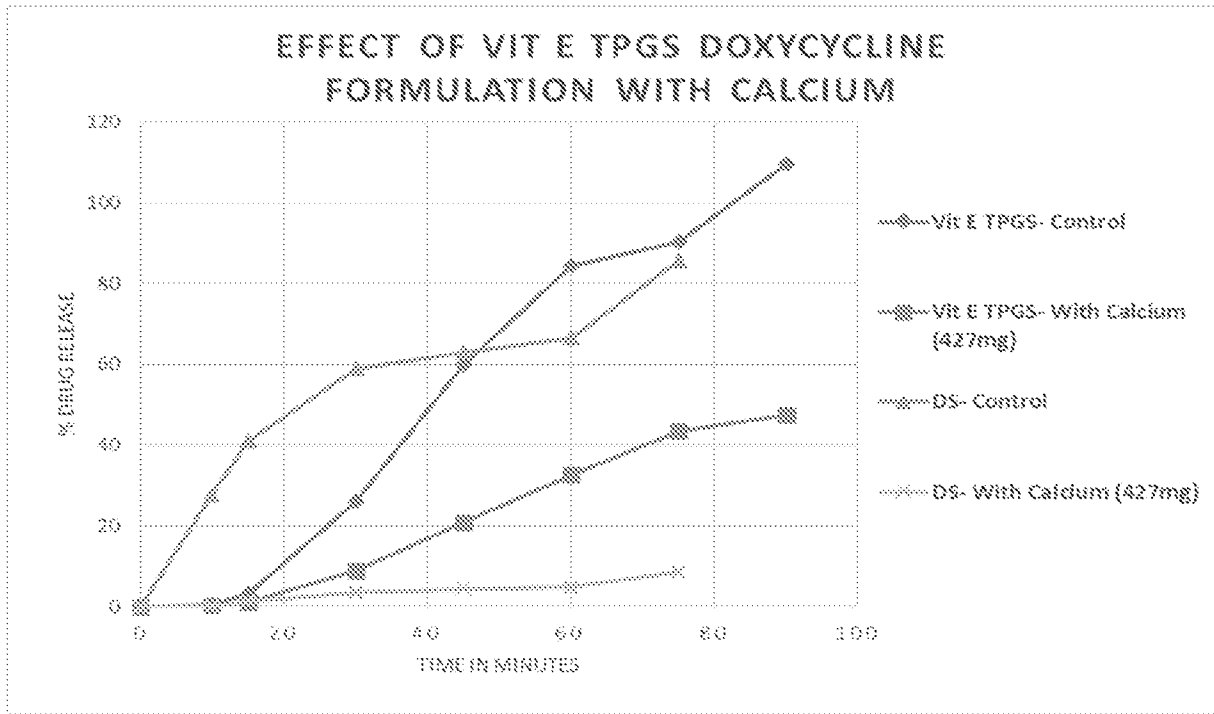


Figure 8

INTERNATIONAL SEARCH REPORT

International application No
PCT/IB2022/052473

A. CLASSIFICATION OF SUBJECT MATTER
INV. A61K9/48 A61K9/107 A61K47/22 A61K9/00
ADD.

According to International Patent Classification (IPC) or to both national classification and IPC

B. FIELDS SEARCHED
 Minimum documentation searched (classification system followed by classification symbols)
A61K

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Electronic data base consulted during the international search (name of data base and, where practicable, search terms used)
EPO-Internal, BIOSIS, EMBASE, WPI Data

C. DOCUMENTS CONSIDERED TO BE RELEVANT

Category*	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
X	US 5 891 845 A (MYERS GARRY [US]) 6 April 1999 (1999-04-06) examples 1-10 -----	1-8, 21-25
X	US 2012/190653 A1 (GILBARD JEFFREY P [US] ET AL) 26 July 2012 (2012-07-26) paragraph [0048]; table 1 -----	1-8, 21-25
A	CONGLIAN YANG ET AL: "Recent Advances in the Application of Vitamin E TPGS for Drug Delivery", THERANOSTICS, vol. 8, no. 2, 1 January 2018 (2018-01-01) , pages 464-485, XP055649557, AU ISSN: 1838-7640, DOI: 10.7150/thno.22711 abstract -----	1-8, 21-25

Further documents are listed in the continuation of Box C. See patent family annex.

* Special categories of cited documents :

"A" document defining the general state of the art which is not considered to be of particular relevance "E" earlier application or patent but published on or after the international filing date "L" document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified) "O" document referring to an oral disclosure, use, exhibition or other means "P" document published prior to the international filing date but later than the priority date claimed	"T" later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention "X" document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step when the document is taken alone "Y" document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art "&" document member of the same patent family
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Date of the actual completion of the international search 2 June 2022	Date of mailing of the international search report 03/08/2022
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Name and mailing address of the ISA/ European Patent Office, P.B. 5818 Patentlaan 2 NL - 2280 HV Rijswijk Tel. (+31-70) 340-2040, Fax: (+31-70) 340-3016	Authorized officer Konter, Jörg
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INTERNATIONAL SEARCH REPORT

International application No.
PCT/IB2022/052473

Box No. II Observations where certain claims were found unsearchable (Continuation of item 2 of first sheet)

This international search report has not been established in respect of certain claims under Article 17(2)(a) for the following reasons:

1. Claims Nos.:
because they relate to subject matter not required to be searched by this Authority, namely:

2. Claims Nos.:
because they relate to parts of the international application that do not comply with the prescribed requirements to such an extent that no meaningful international search can be carried out, specifically:

3. Claims Nos.:
because they are dependent claims and are not drafted in accordance with the second and third sentences of Rule 6.4(a).

Box No. III Observations where unity of invention is lacking (Continuation of item 3 of first sheet)

This International Searching Authority found multiple inventions in this international application, as follows:

see additional sheet

1. As all required additional search fees were timely paid by the applicant, this international search report covers all searchable claims.

2. As all searchable claims could be searched without effort justifying an additional fees, this Authority did not invite payment of additional fees.

3. As only some of the required additional search fees were timely paid by the applicant, this international search report covers only those claims for which fees were paid, specifically claims Nos.:

4. No required additional search fees were timely paid by the applicant. Consequently, this international search report is restricted to the invention first mentioned in the claims; it is covered by claims Nos.:
1-8, 24 (completely); 21-23, 25 (partially)

Remark on Protest

- The additional search fees were accompanied by the applicant's protest and, where applicable, the payment of a protest fee.
- The additional search fees were accompanied by the applicant's protest but the applicable protest fee was not paid within the time limit specified in the invitation.
- No protest accompanied the payment of additional search fees.

FURTHER INFORMATION CONTINUED FROM PCT/ISA/ 210

This International Searching Authority found multiple (groups of) inventions in this international application, as follows:

1. claims: 1-8, 24 (completely); 21-23, 25 (partially)

Pharmaceutical compositions comprising vitamin E TPGS.

2. claims: 9-20 (completely); 21-23, 25 (partially)

Pharmaceutical compositions comprising micelle forming agent.

INTERNATIONAL SEARCH REPORT

Information on patent family members

International application No

PCT/IB2022/052473

Patent document cited in search report	Publication date	Patent family member(s)	Publication date
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			AU 1301199 A 15-06-1999
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			EP 1032373 A1 06-09-2000
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			WO 2012100109 A2 26-07-2012
