(12) INTERNATIONAL APPLICATION PUBLISHED UNDER THE PATENT COOPERATION TREATY (PCT)

(19) World Intellectual Property Organization
International Bureau



(43) International Publication Date 3 June 2010 (03.06.2010)

(10) International Publication Number WO 2010/062717 A1

- (51) International Patent Classification: C08K 3/02 (2006.01) C08K 7/00 (2006.01)
- (21) International Application Number:

PCT/US2009/062975

(22) International Filing Date:

2 November 2009 (02.11.2009)

(25) Filing Language:

English

(26) Publication Language:

English

(30) Priority Data:

61/110,856 3 November 2008 (03.11.2008) US 61/111,229 4 November 2008 (04.11.2008) US

- (71) Applicant (for all designated States except US): UNI-VERSITY OF HOUSTON [US/US]; 316 East Cullen, Houston, TX 77204 (US).
- (72) Inventors; and
- (75) Inventors/Applicants (for US only): CURRAN, Seamus [IE/US]; 2415 Alberton Ln, Pearland, TX 77584 (US). DIAS, Sampath [LK/US]; 7950 Bellfort Street, Apt. 268, Houston, TX 77061 (US). BLAU, Werner [DE/IE]; 2, Sommerville Terrace, Dalkey Avenue, Dalkey, Dublin (IE). WANG, Jun [CN/IE]; Apartment 20, Wicklow Court, 38/40 Georges St., Dublin, 2 (IE). OREMLAND, Ronald, S. [US/US]; 213 Warbler Lane, Brisbane CA 94005 (GB). BAESMAN, Shaun [US/US]; P.o. Box 7012, San Carlos, CA 94070 (US).

- (74) **Agent**: **SHADDOX, Robert, C.**; Winstead Pc, P.o. Box 50784, Dallas, TX 75201 (US).
- (81) Designated States (unless otherwise indicated, for every kind of national protection available): AE, AG, AL, AM, AO, AT, AU, AZ, BA, BB, BG, BH, BR, BW, BY, BZ, CA, CH, CL, CN, CO, CR, CU, CZ, DE, DK, DM, DO, DZ, EC, EE, EG, ES, FI, GB, GD, GE, GH, GM, GT, HN, HR, HU, ID, IL, IN, IS, JP, KE, KG, KM, KN, KP, KR, KZ, LA, LC, LK, LR, LS, LT, LU, LY, MA, MD, ME, MG, MK, MN, MW, MX, MY, MZ, NA, NG, NI, NO, NZ, OM, PE, PG, PH, PL, PT, RO, RS, RU, SC, SD, SE, SG, SK, SL, SM, ST, SV, SY, TJ, TM, TN, TR, TT, TZ, UA, UG, US, UZ, VC, VN, ZA, ZM, ZW.
- (84) Designated States (unless otherwise indicated, for every kind of regional protection available): ARIPO (BW, GH, GM, KE, LS, MW, MZ, NA, SD, SL, SZ, TZ, UG, ZM, ZW), Eurasian (AM, AZ, BY, KG, KZ, MD, RU, TJ, TM), European (AT, BE, BG, CH, CY, CZ, DE, DK, EE, ES, FI, FR, GB, GR, HR, HU, IE, IS, IT, LT, LU, LV, MC, MK, MT, NL, NO, PL, PT, RO, SE, SI, SK, SM, TR), OAPI (BF, BJ, CF, CG, CI, CM, GA, GN, GQ, GW, ML, MR, NE, SN, TD, TG).

Published:

— with international search report (Art. 21(3))



COMPOSITES COMPRISING BIOLOGICALLY-SYNTHESIZED NANOMATERIALS

CROSS-REFERENCE TO RELATED APPLICATIONS

[0001] This application claims priority to United States provisional patent applications 61/110,856, filed November 3, 2008 and 61/111,229, filed November 4, 2008, each of which is incorporated by reference herein in its entirety.

STATEMENT REGARDING FEDERALLY SPONSORED RESEARCH

[0002] This invention was made with government support under Grant Number DE-FG36-08GO88008 awarded by the United States Geological Survey and the National Aeronautics and Space Administration Exobiology Program. The government has certain rights in the invention. Additional support for the invention was provided by Science Foundation Ireland under grant number 08/CE/I1432.

BACKGROUND

[0003] Designed polymer composite materials containing carbon-based nanomaterials such as, for example, carbon nanotubes and fullerenes (including functionalized versions of these species), have emerged over the last decade. Such materials are referred to herein as carbon-based nanocomposites. Carbon-based nanocomposites have beneficial electrical, optical and mechanical properties due to the inclusion of the carbon-based nanomaterials. Carbon-based nanocomposites have been studied for potential uses as photovoltaics, field emission devices, conductive wires and structural members. Although there has been intense interest in carbon-based nanocomposites, development of these systems has been hampered by synthetic obstacles including, for example, yield and chirality control of the carbon-based nanomaterial filler.

[0004] Polymer composites containing inorganic materials such as, for example, glass fibers, have been known for quite some time, but designed polymer composites containing inorganic nanomaterials such as, for example, inorganic quantum dots and nanorods, have been slower to emerge. Such materials are referred to herein as inorganic-based nanocomposites. Chemical syntheses of certain inorganic nanomaterials are hampered by the same synthetic obstacles that encumber organic nanomaterial synthesis. In stark contrast to chemical syntheses, biological

syntheses (particularly of inorganic materials) are known to be highly efficient, environmentally-friendly, and capable of producing structures that cannot be replicated by standard chemical methods. Further, biologically-synthesized inorganic nanomaterials may have properties that meet or exceed those of organic nanomaterials in certain applications.

[0005] In view of the foregoing, inorganic-based nanocomposites having biologically-synthesized inorganic nanomaterials dispersed in a polymer matrix may be of considerable benefit in a variety of applications. These inorganic-based nanocomposites may take advantage of property enhancements that are unique to biologically-synthesized inorganic nanomaterials.

SUMMARY

[0006] In various embodiments, the present disclosure describes composite materials including a polymer material and a biologically-synthesized nanoscale material dispersed in the polymer material. Other embodiments of composite materials described in the present disclosure involve composite materials including a polymer material and biologically-synthesized tellurium nanorods dispersed in the polymer material, wherein the biologically-synthesized tellurium nanorods have a non-linear optical limiting response. In other various embodiments, composite materials of the present disclosure are described which include a polymer material and tellurium nanorods dispersed in the polymer material.

[0007] In other various embodiments, the present disclosure also describes optical limiting devices that include a composite material containing a polymer material and tellurium nanorods dispersed in the polymer material. In some embodiments, the tellurium nanorods are biologically synthesized. In some embodiments, the tellurium nanorods are biologically synthesized by *Bacillus selenitireducens*.

[0008] The foregoing has outlined rather broadly the features of the present disclosure in order that the detailed description that follows may be better understood. Additional features and advantages of the disclosure will be described hereinafter, which form the subject of the claims.

BRIEF DESCRIPTION OF THE DRAWINGS

[0009] For a more complete understanding of the present disclosure, and the advantages thereof, reference is now made to the following descriptions to be taken in conjunction with the accompanying drawings describing specific embodiments of the disclosure, wherein:

[0010] FIGURE 1 shows an illustrative schematic demonstrating how scattering centers may form in composite materials of the present disclosure;

[0011] FIGURE 2 shows an illustrative electron micrograph of Te nanorods on the surface of *Bacillus selenitireducens* as individuals and as shards after coalescence;

[0012] FIGURE 3 shows an illustrative electron micrograph of Te nanorods in the form of "rosettes" before being sloughed from the cell surface;

[0013] FIGURES 4A and 4B show illustrative electron micrographs of Te nanorods in the form of rosettes after cleansing to remove bacteria and cellular debris;

[0014] FIGURE 5 shows an illustrative Raman spectrum of solid Te nanorods;

[0015] FIGURE 6 shows an illustrative UV-VIS absorption spectrum of Te nanorods;

[0016] FIGURE 7A shows illustrative UV/VIS absorption spectra of PmPV and a Te nanorod/PmPV nanocomposite dispersed in toluene; FIGURE 7B shows an illustrative UV/VIS absorption spectrum of Te nanorods in a Te nanorod/PmPV composite after subtracting absorption contributions from the PmPV;

[0017] FIGURE 8 shows a schematic of an illustrative z-scan measurement system;

[0018] FIGURE 9A shows an illustrative plot demonstrating the non-linear optical limiting response of a Te nanorod/PmPV nanocomposite at 532 nm and 1064 nm as a function of input energy density; and FIGURE 9B shows an illustrative plot of normalized transmission versus input energy density (J cm⁻²) in which the optical limiting threshold at which transmittance falls to 50% of the normalized linear transmittance is indicated.

DETAILED DESCRIPTION

[0019] In the following description, certain details are set forth such as specific quantities, sizes, etc. so as to provide a thorough understanding of the present embodiments disclosed herein. However, it will be evident to those of ordinary skill in the art that the present disclosure may be practiced without such specific details. In many cases, details concerning such considerations and the like have been omitted inasmuch as such details are not necessary to obtain a complete understanding of the present disclosure and are within the skills of persons of ordinary skill in the relevant art.

[0020] Referring to the drawings in general, it will be understood that the illustrations are for the purpose of describing particular embodiments of the disclosure and are not intended to be limiting thereto. Drawings are not necessarily to scale.

[0021] While most of the terms used herein will be recognizable to those of ordinary skill in the art, it should be understood, however, that when not explicitly defined, terms should be interpreted as adopting a meaning presently accepted by those of ordinary skill in the art. In cases where the construction of a term would render it meaningless or essentially meaningless, the definition should be taken from Webster's Dictionary, 3rd Edition. Definitions and/or interpretations should not be incorporated from other patent applications, patents, or publications, related or not, unless specifically stated in this specification or if the incorporation is necessary for maintaining validity.

[0022] A plethora of biologically-synthesized minerals are known to be produced by a diverse array of bacteria and other animals. Oftentimes, such biologically-synthesized minerals acquire a particle size or crystalline state that is unattainable using standard chemical syntheses. Such biologically-synthesized minerals may embody vastly different properties than either bulk minerals or minerals produced through non-biological routes. Biological syntheses of minerals, particularly nanoscale materials, hold considerable promise for mass production of these species through an inexpensive and environmentally-benign route.

[0023] One application of biologically-synthesized nanoscale materials that has not been explored heretofore involves the preparation of composite materials. In various embodiments,

the present disclosure describes composite materials including a polymer material and a biologically-synthesized nanoscale material dispersed in the polymer material. In some embodiments, the biologically-synthesized nanoscale material includes, for example, tellurium nanorods, tellurium nanospheres, selenium nanospheres, arsenic (III) sulfide nanotubes, cadmium selenide nanocrystals and zinc selenide nanocrystals. In some embodiments, the nanoscale materials are semiconductors. In other embodiments, the nanoscale materials are metallic. In still other embodiments, the nanoscale materials have optical limiting properties. In some embodiments, the optical limiting properties are non-linear in response to incident light having variable intensity.

[0024] The process of optical limiting involves limiting and attenuating the effects of intense laser pulses and other focused beams of electromagnetic radiation. Optical limiting devices are often used to lower damaging light levels passing through a material, thereby protecting the vision of an operator of a laser or other bright light source such as, for example, an arc welder. One approach to optical limiting makes use of materials whose transmittance decreases at high light levels (high intensities). For protective applications, such as those described above, the response of the optical limiter in decreasing transmittance is ideally rapid with a low saturation threshold. For optimal protection, an optical limiter ideally exhibits broadband optical limiting properties ranging from the visible to the infrared or near infrared region of the electromagnetic spectrum in some embodiments, and extending into the ultraviolet and X-ray region of the electromagnetic spectrum in other embodiments.

[0025] There are two primary mechanisms that produce optical limiting properties in composite materials. A first mechanism is non-linear absorption, and a second is non-linear scattering. Non-linear absorption can be further divided into mechanisms involving multi-photon absorption (e.g., organic molecules or crystals), reverse saturable absorption (e.g., fullerenes, phthalocyanines, porphyrins or chromaphores with heavy metals such as Ag) and free-carrier absorption (e.g., semiconductor nanoparticles or metal nanocomposites). Non-linear scattering may arise from a number of physical phenomena, such as, for example, formation of solvent bubbles as scattering centers upon exposure to intense light, ionization of nanoparticles in the composite material, and/or thermal alterations in the refractive index of the polymer matrix or organic solvents.

[0026] In various embodiments, the composite materials of the present disclosure have optical limiting properties, and in some embodiments, the optical limiting properties are non-linear. In various embodiments, the composite materials of the present disclosure exceed the optical limiting properties of other well studied optical limiting materials such as, for example, carbon nanotubes, phthalocyanines and porphyrins. In various embodiments, the optical limiting absorption properties occur with response times on the order of picoseconds.

[0027] Without being bound by theory or mechanism, it is believed that in the composite materials of the present disclosure, scattering effects play the most significant role in producing the optical limiting properties. According to Mie scattering theory, light beams are not effectively scattered by nanoscale particles alone, and the scattered intensity from Mie scattering centers is not sensitive to the wavelength of the incident light. As will be shown hereinbelow, the non-linear optical limiting properties of the composite materials of the present disclosure are wavelength sensitive.

[0028] Still without being bound by theory or mechanism, it is believed that the scattering mechanism for the composite materials of the present disclosure is based upon polymer bubble formation during microplasma formation. According to present understanding of the scattering mechanism of the optical limiting process, such polymer bubbles may form by two competing routes. FIGURE 1 shows an illustrative schematic demonstrating how polymer bubble formation may occur in composite materials of the present disclosure. As shown in FIGURE 1, photons 1 interact with nanoparticle 2 dispersed in a polymer matrix. According to a first mechanism, photons 1 are absorbed by nanoparticle 2, which results in transfer of thermal energy to the surrounding medium (e.g., the polymer matrix). The transfer of thermal energy results in formation of gas bubbles 3 with nanoparticle 2 as the nucleus. The initially formed gas bubbles 3 rapidly expand due to the large pressure difference at the vapor-matrix interface to produce expanded gas bubbles 4. When the size of expanded gas bubbles 4 increases to the wavelength of the incident photons 1, the expanded gas bubbles 4 effectively scatter the photons, as shown, and consequently reduce the optical transmission. In the non-limiting example shown in FIGURE 1, only one of six incident photons 1 reaches the eye 5 of an observer.

[0029] Still referring to FIGURE 1, in an alternate mechanism, photons 1 may interact with nanoparticle 2 to result in an ionized nanoparticle (not shown). The ionized nanoparticle subsequently forms a microplasm 6 in the polymer matrix. The energy generated by non-linear photonic vibrations and subsequent breakdown of the ionized nanoparticle results in energy transfer to the polymer matrix and formation of the microplasm 6. Rapid expansion of microplasm 6 to form expanded microplasm 7 results in the formation of scattering centers. As in the former mechanism, microplasm 7 effectively scatters photons 1, as shown, such that only a fraction of the incident light reaches the eye 5 of an observer.

[0030] Applicants have discovered that certain biologically-synthesized nanoscale materials have optical limiting behavior that may be superior to that of currently studied optical limiting materials. Certain anaerobic bacteria such as, for example, *Bacillus selenitireducens* and *Sulfurospirillum barnesii* are known to have the ability to respire oxyanions of selenium and tellurium (e.g., TeO₃²⁻ and SeO₃²⁻) and produce elemental selenium and tellurium having nanoscale dimensions as a respiration product. For example, *Bacillus selenitireducens* respires tellurium nanorods having diameters between about 15 and about 25 nm and lengths between about 1 μm and about 2 μm, whereas *Sulfurospirillum barnesii* produces tellurium nanospheres having a diameter on the order of 300 nm. Applicants have discovered that the tellurium nanorods, in particular, have useful nano-photonic properties including, for example, an optical limiting response, which may be non-linear in an embodiment. Accordingly, the tellurium nanorods may be dispersed into polymer materials to form composite materials which may be further used in optical limiting devices.

[0031] In any of the various embodiments containing tellurium nanorods that are described herein, the tellurium nanorods may have a length of about 1 μ m to about 2 μ m. The tellurium nanorods may also have a diameter of about 15 nm to about 25 nm in some embodiments, and a diameter of about 20 nm in certain other embodiments.

[0032] One of ordinary skill in the art will recognize that a wide range of polymer materials may be used in any of the composite materials of the present disclosure, depending on the intended end use of the composite material. Such polymer materials may be thermosetting or thermoplastic in various embodiments. In some embodiments, the identity of the polymer

material is not particularly critical, other than that it disperses the tellurium nanorods or other nanoscale material. However, in other embodiments, the identity of the polymer material may be more important. For example, in some embodiments, the polymer material may be semiconducting. In some embodiments, the polymer material is poly[(m-phenylene vinylene)-co-(2,5-dioctyloxy-p-phenylene vinylene)] (PmPV). In certain embodiments the polymer may be, for example, poly(3-hexylthiophene) (P3HT), poly(3-octylthiophene) (P3OT), poly[2-methoxy-5-(2'-ethylhexyloxy-p-phenylene vinylene)] (MEH-PPV), poly[2-methoxy-5-(3,7-dimethyloctyloxy)-p-phenylene vinylene], sodium poly[2-(3-thienyl)-ethoxy-4-butylsulfonate] (PTEBS) and combinations thereof. PmPV may also be combined with any of the aforementioned polymers to make a polymer mixture.

[0033] In various embodiments, the present disclosure describes composite materials including a polymer material and biologically-synthesized tellurium nanorods dispersed in the polymer material. The biologically-synthesized tellurium nanorods have a non-linear optical limiting response. In some embodiments, the tellurium nanorods are synthesized by *Bacillus selenitireducens*.

[0034] In still other various embodiments of the present disclosure, composite materials are described that include a polymer material and tellurium nanorods dispersed in the polymer material. In some embodiments, the tellurium nanorods are biologically synthesized such as, for example, by the microorganism *Bacillus selenitireducens*. In various embodiments, optical limiting devices incorporating such composites are considered by the present disclosure. In some embodiments, a transmission response of the devices is non-linear upon exposure to electromagnetic radiation of variable intensity.

Experimental Examples

[0035] The following examples are provided to more fully illustrate some of the embodiments disclosed hereinabove. It should be appreciated by those of skill in the art that the techniques disclosed in the examples that follow represent techniques that constitute illustrative modes for practice of the disclosure. Those of ordinary skill in the art should, in light of the present disclosure, appreciate that many changes can be made in the specific embodiments that are

disclosed and still obtain a like or similar result without departing from the spirit and scope of the disclosure.

[0036] Example 1: Biological Synthesis of Tellurium Nanorods with *Bacillus selenitireducens*. Biological synthesis of tellurium nanorods and characterization of such nanostructured materials have been described in S.M. Baesman, *et al.*, "Formation of Tellurium Nanocrystals during Anaerobic Growth of Bacteria That Use Te Oxyanions as Repiratory Electron Acceptors", *Appl. Env. Microbiol.*, 73:2007, pp. 2135 – 2143, which is incorporated by reference herein in its entirety. Briefly, nanoscale crystals of elemental tellurium [i.e., Te(0)] were synthesized by growing the haloalkaliphilic anaerobic bacterium *Bacillus selenitireducens* strain MLS 10 in a lactate-tellurite medium. Such growth conditions result in the respiratory biochemical reduction of TeO₃²⁻ to Te(0). The Te(0) first accumulates on the cell surfaces as Tenanorods, which then aggregate and slough off into the surrounding aqueous medium as a black precipitate. The Te(0) nanorods were cleansed of cellular material and debris by ultrasonication, treatments with lysozyme, and repeated washings and centrifugations. The cleansed Te(0) nanorods were re-suspended in deionized water in a stoppered serum bottle and stored under an N₂ atmosphere [to preclude oxidation of Te(0) to Te(IV)] until use.

[0037] FIGURE 2 shows an illustrative electron micrograph of Te nanorods on the surface of *Bacillus selenitireducens* as individuals and as shards after coalescence. As shown in FIGURE 2, the respired Te(0) initially forms as thin (\sim 19 nm x 100 - 300 nm) individual nanorods **11** of Te(0) on the cell surface, which subsequently coalesce to form shards **10** containing intertwined nanorods. The shards **10** then slough off the cell surface, forming aggregates of star-shaped "rosettes", which contain multiple Te(0) shards. FIGURE 3 shows an illustrative electron micrograph of Te nanorods in the form of "rosettes" before being sloughed from the cell surface. After sloughing, the "rosettes" are subsequently cleansed of associated bacteria and cellular debris to provide the final form of the Te nanorods used in the embodiments described herein. FIGURES 4A and 4B show illustrative electron micrographs of Te nanorods in the form of "rosettes" after cleansing to remove bacteria and cellular debris. The Te nanorods in the "rosettes" have dimensions of 20 nm in width and are 1 - 2 μ m in length.

[0038] Example 2: Optical Characterization of Te Nanorods and Te Nanocomposite Materials. Raman spectra were obtained using a Renishaw InVia Raman spectrometer equipped with a Raman Leica RE02 microscope. The excitation wavelength was 488 nm, produced from an air-cooled Laser-Physics Ar⁺ laser. The Te(0) samples were first suspended in deionized water, drop-cast onto clean silicon substrates and allowed to dry before acquiring the Raman spectra. FIGURE 5 shows an illustrative Raman spectrum of solid Te nanorods. Te(0) possess four zone center phonon modes, two of which are doubly degenerate (E' and E"), one non-degenerate (A1) and a second non-degenerate (A2) which is Raman silent. The second-order Raman spectrum of Te(0) has a peak at 270 cm⁻¹, as shown in FIGURE 5, which corresponds to the second order vibrational mode of the E modes (found typically at 140 cm⁻¹). Te(0) adopts a trigonal geometry that corresponds to 3 tellurium atoms per unit cell, with each unit cell forming a helical chain that orientates specifically to the x-axis. There is a tendency for these chains of Te(0) unit cells to wrap themselves around each other and result in interactions between chains

[0039] FIGURE 6 shows an illustrative UV-VIS absorption spectrum of Te nanorods. UV-Vis absorption data were acquired using a Perkin-Elmer Lambda 20 UV/Vis spectrometer operating over a range of 350 nm – 1000 nm. The Te(0) samples were suspended in deionized water, dropcast on to clean glass microscope slides and allowed to dry before acquiring the absorption spectra.

[0040] To determine the absorption coefficient, a series of Te nanorod/PmPV composite solutions of differing Te(0) concentrations ranging from about 0.0 mg/mL to about 0.5 mg/mL were prepared by diluting a 0.5 mg/mL Te-PmPV solution with a 0.5 mg/mL pure PmPV solution. Linear transmittance (%) and absorbance (cm⁻¹) were measured in 0.1 cm quartz cuvettes using low intensity 532 nm laser pulses of 10 Hz repetition rate. FIGURE 7A shows illustrative UV/VIS absorption spectra of PmPV (curve 60) and a Te nanorod/PmPV nanocomposite (curve 61) dispersed in toluene. As can be seen in FIGURE 7A, strong bands characteristic of PmPV partially obscured the bands from the Te nanorods. FIGURE 7B shows an illustrative UV/VIS absorption spectrum of Te nanorods in a Te nanorod/PmPV composite after subtracting absorption contributions from the PmPV. The subtraction spectrum of FIGURE 7B was qualitatively similar to the UV/VIS spectrum of Te nanorods shown in FIGURE 7A. Further, FIGURE 7B shows that the Te nanorods are broadband absorbers, which makes them

suitable for use in optical limiting materials. The absorption behavior of Te nanorod/PmPV composites followed the Beer-Lambert law. The average absorption coefficient was calculated by fitting the absorbances of all solutions, which yielded a calculated value of 8.41 ± 0.42 mL cm⁻¹.

[0041] Example 3: Non-linear optical measurement of Te nanorod/PmPV composites. For measuring non-linear scattering, z-scan experiments were performed using 6 ns pulses from a Q-switched Nd:YAG laser. FIGURE 8 shows a schematic of an illustrative z-scan measurement system. As shown in FIGURE 8, the total transmittance through the sample as a function of incident laser intensity was measured, while the sample was gradually moved through the focus of a lens 80 (along the z-axis). Scattered and non-scattered radiation was measured with detector 81 and open detector 82, respectively. Effective non-linear extinction coefficients including non-linear absorption and scattering were calculated by fitting the normalized transmittance as a function of position z [T_{norm}(z)], as given by Formula (1).

$$T_{Norm}(z) = Log_e[1+q_0(z)]/1q_0(z)$$
 (1)

In Formula (1), $q_0(z)$ is defined by Formula (2).

$$q_0(z) = q_{00}/[1+(z/z_0)^2]$$
 (2)

In Formula (2), q_{00} is defined by Formula (3).

$$q_{00} = \beta_{eff} I_0 L_{eff} \tag{3}$$

In Formula (3), β_{eff} is the effective intensity-dependent non-linear extinction coefficient, and I_0 is the intensity of the light at focus. L_{eff} is the effective length of the sample defined in terms of the linear absorbance α_0 and the true optical path length L, as given in Formula (4).

$$L_{\rm eff} = [1 - e^{-\alpha_0 L}]/\alpha_0 \tag{4}$$

The effective imaginary third-order optical susceptibility $\text{Im}\{\chi^{(3)}_{\text{eff}}\}$ is directly related to β_{eff} and is expressed as in Formula (5).

$$Im\{\chi^{(3)}_{eff}\} = n_0^2 \epsilon_0 c \lambda \beta_{eff} / (2\pi)$$

In Formula (5), n_0 is the linear refractive index, ε_0 is the permeability of free space, c is the speed of light and λ is the wavelength of the incident light.

[0042] Combining Formulas (1) through (5) allows the non-linear extinction coefficients to be calculated. In conducting the experiments to determine the non-linear extinction coefficients, the laser beam was spatially filtered to remove higher-order modes and tightly focused with a 9 cm focal length lens. The laser was operated at fundamental frequency of 1064 nm and a second harmonic frequency of 532 nm, with a pulse repetition rate of 10 Hz. Simultaneously, a focusing lens setup was arranged at $\sim 30^{\circ}$ to the direct incident beam to monitor scattered light. All samples were tested in 0.1 cm quartz cuvettes.

[0043] FIGURE 9A shows an illustrative plot demonstrating the non-linear optical limiting response of a Te nanorod/PmPV nanocomposite at 532 nm (curve 72) and 1064 nm (curve 73) as a function of input energy density. In a material having a linear optical response, a plot of output energy density versus input energy density will be a straight line. However, as clearly shown in FIGURE 9A, non-linear behavior was observed. For the Te nanorod/PmPV nanocomposite tested in FIGURE 9A, the concentration of PmPV was 0.5 mg/mL, and the Te concentration was estimated to be ~0.28 mg/mL using the linear absorption coefficient calculated above.

[0044] FIGURE 9A also shows that the Te nanorod/PmPV nanocomposite had high linear transmittances of 78.8% at 532 nm (curve **70**) and 67.0% at 1064 nm (curve **71**). It can be clearly seen from FIGURE 9A that the Te nanorod/PmPV nanocomposite solutions exhibited exceptional optical limiting performances at both 532 nm and 1064 nm, which is characteristic of a broadband optical limiting response from the visible to the near infrared (NIR).

[0045] The non-linear extinction coefficients N_{eff} , deduced from z-scan curves, were 141 ± 14 cm GW^{-1} and 21 ± 3 cm GW^{-1} at 532 nm and 1064 nm, respectively. FIGURE 9B shows an illustrative plot of normalized transmission versus input energy density (J cm⁻²) in which the optical limiting threshold at which when transmittance falls to 50% of the normalized linear transmittance is indicated. As shown in FIGURE 9B, the optical limiting threshold at 532 nm was 2.2 J cm⁻² (curve **74**) and at 1064 nm was 15.4 J cm⁻² (curve **75**), as indicated by the solid lines. The scattered signals [curve **76** (532 nm) and curve **77** (1064 nm)] increased significantly

along with the decrease of transmission, indicating that non-linear scattering was responsible for the optical limiting.

[0046] Increased scattering is typically synchronous with the decrease in transmission for lower concentration Te nanorod/PmPV solutions (90%). However, the delay in non-linear scattering seen in FIGURE 9B was observed for higher concentration composite samples. Such a delay is believed to be due to a residual non-linear absorption mechanism (e.g., multi-photon absorption or free-carrier absorption) which becomes evident in more concentrated Te nanorod composite solutions. Nonetheless, non-linear scattering appeared to be the primary mechanism operating in the Te nanorod/PmPV composites. As a further note, the non-linear scattering effect of pristine PmPV solutions at 532 nm showed only very slight optical limiting ($\beta_{eff} \sim 0.2$ cm GW⁻¹) due to two-photon absorption which is so low it can be ignored.

Limiting Materials. To evaluate the optical limiting performance of Te nanorod/PmPV composites, several representative materials having known optical limiting properties were selected for experimental comparison. Well-known reverse saturable absorbers C₆₀ and indium phthalocyanine (tBu₄PcInCl) that respond favorably at an excitation wavelength of 532 nm were examined. As shown in Table 1 below, Te nanorod/PmPv composites had a larger non-linear extinction coefficient than either of these known optical limiting materials, even though the concentration of Te nanorods in the polymer composites was quite low.

Table 1: Non-Linear Extinction Coefficients of Te Nanorod/PmPV and Other Optical Limiting Materials

Material	Host	Conc.	T	œ	β_{eff}	Im {χ ⁽³⁾ }	Laser	
			[%]	[cm ⁻¹]	[cm GW ⁻¹]	[×10 ⁻¹¹ , esu]	parameter	
Те	PmPV in	0.28 g/L,	*0.0	2.38	141±14	5.32±0.52	532nm, 6ns	
	toluene	0.028 wt.%	78.8					
C ₆₀	toluene	1.8 g/L		2.81	66±9	2.1±0.4	532 n m, 6ns	
∂Bu₄PcInCl	toluene	0.5 g/L		0.53	44±9	1.6±0.3	532nm, 6ns	
CNTs	PmPV in					المية به	**** * * * *	
	toluene	5.9 wt.%		3.754	1.16		532nm, 6ns	
Те	PmPV in	0.28 g/L,		4.01	21.1±3.3		***************************************	
	toluene	0.028 wt.%	67.0			1.59±0.25	1064nm, 6ns	
ASPT dye	Epoxy rod	0.004 M/L			6		1064nm, 8ns	

For optical limiting at 1064 nm, Te nanorod/PmPV was compared with a two-photon adsorption dye trans-4-[p-(N-ethyl-N-hydroxyethylamino)styryl]-N-methylpyridinium tetraphenylborate (ASPT). Under similar laser conditions, the non-linear absorption coefficient of ASPT was 6 cm GW⁻¹, while that of the Te nanorod/PmPV composite was as high as 21 cm GW⁻¹.

[0048] From the foregoing description, one of ordinary skill in the art can easily ascertain the essential characteristics of this disclosure, and without departing from the spirit and scope thereof, can make various changes and modifications to adapt the disclosure to various usages and conditions. The embodiments described hereinabove are meant to be illustrative only and should not be taken as limiting of the scope of the disclosure, which is defined in the following claims.

CLAIMS

What is claimed is the following:

1. A composite material comprising:

a polymer material; and tellurium nanorods dispersed in the polymer material.

2. A composite material comprising:

a polymer material; and

biologically-synthesized tellurium nanorods dispersed in the polymer material; wherein the biologically-synthesized tellurium nanorods have a non-linear optical limiting response.

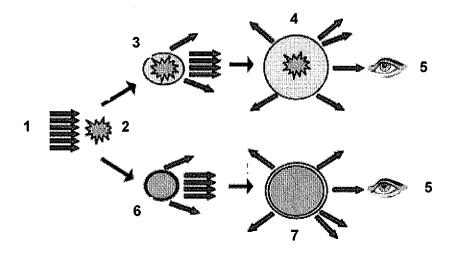
- 3. The composite material of claim 1, wherein the tellurium nanorods are biologically synthesized.
- 4. The composite material of claim 2 or claim 3, wherein the tellurium nanorods are biologically synthesized by *Bacillus selenitireducens*.
- 5. The composite material of claim 1 or claim 2, wherein the tellurium nanorods have a diameter of about 15 nm to about 25 nm.
- 6. The composite material of claim 1 or claim 2, wherein the tellurium nanorods have a diameter of about 20 nm.
- 7. The composite material of claim 1 or claim 2, wherein the tellurium nanorods have a length of about 1 μ m to about 2 μ m.
- 8. The composite material of claim 1 or claim 2, wherein the polymer material comprises a semiconducting polymer.
- 9. The composite material of claim 8, wherein the semiconducting polymer is selected from the group consisting of poly(3-hexylthiophene) (P₃HT), poly(3-octylthiophene) (P₃OT), poly[2-methoxy-5-(2'-ethylhexyloxy-p-phenylene vinylene)] (MEH-PPV), poly[2-

methoxy-5-(3,7-dimethyloctyloxy)-p-phenylene vinylene], sodium poly[2-(3-thienyl)-ethoxy-4-butylsulfonate] (PTEBS) and combinations thereof.

- 10. The composite material of claim 1 or claim 2, wherein the polymer material comprises poly[(m-phenylene vinylene)-co-(2,5-dioctyloxy-p-phenylene vinylene)] (PmPV).
- 11. A composite material comprising:
 - a polymer material; and
 - a biologically-synthesized nanoscale material dispersed in the polymer material.
- 12. The composite material of claim 11, wherein the biologically-synthesized nanoscale material is selected from the group consisting of tellurium nanorods, tellurium nanospheres, selenium nanospheres, arsenic (III) sulfide nanotubes, cadmium selenide nanocrystals and zinc selenide nanocrystals.
- 13. The composite material of claim 11, wherein the biologically-synthesized nanoscale material has a non-linear optical limiting response.
- 14. An optical limiting device comprising the composite material of claim 1.
- 15. An optical limiting device comprising the composite material of claim 2.
- 16. The optical limiting device of claim 14, wherein a transmission response of the device is non-linear upon exposure to electromagnetic radiation of variable intensity.
- 17. The optical limiting device of claim 14 or claim 15, wherein the tellurium nanorods are biologically synthesized by *Bacillus selenitireducens*.

1/7

FIGURE 1



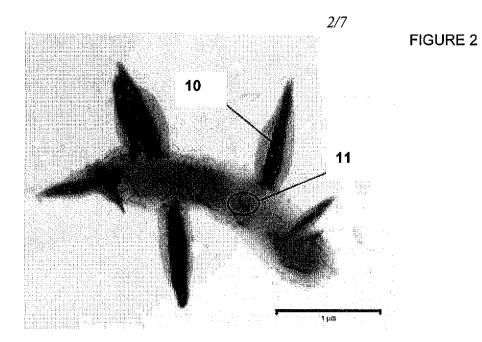


FIGURE 3

3/7

FIGURE 4A

FIGURE 4B

4/7

FIGURE 5

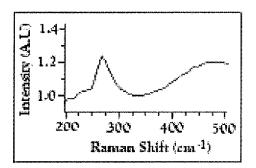
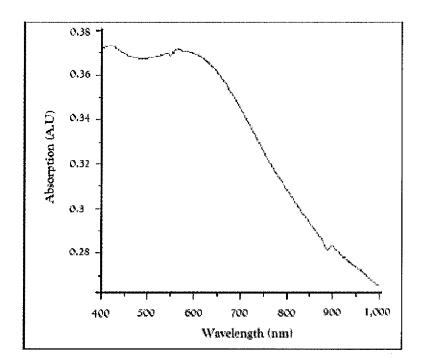


FIGURE 6



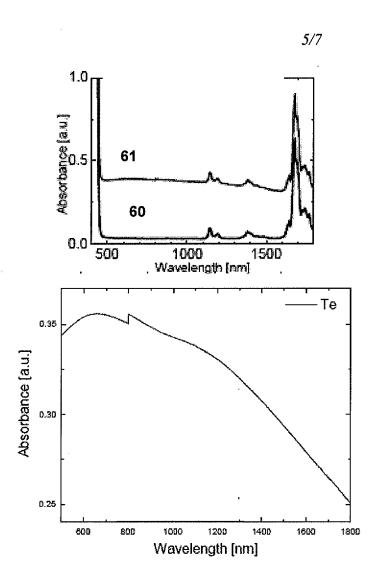


FIGURE 7A

FIGURE 7B

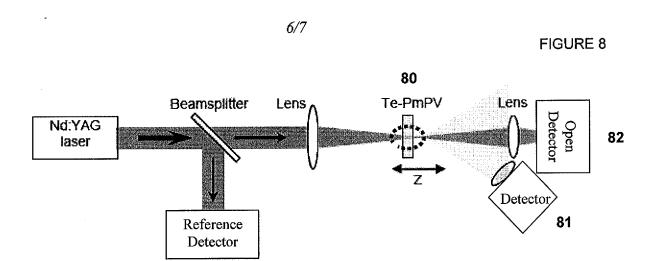


FIGURE 9A

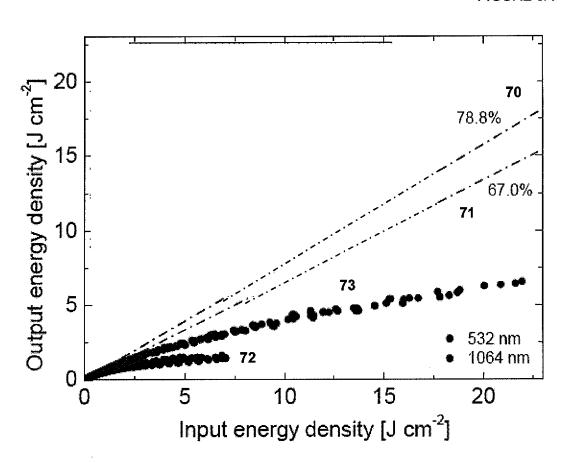
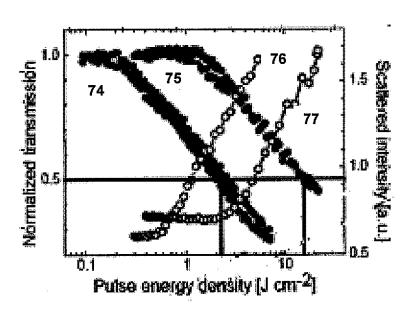


FIGURE 9B



INTERNATIONAL SEARCH REPORT

International application No
PCT/US2009/062975

			101/032003/0023/3				
A. CLASSIFICATION OF SUBJECT MATTER INV. C08K3/02 C08K7/00							
According to International Patent Classification (IPC) or to both national classification and IPC							
B. FIELDS SEARCHED							
Minimum do COSK	ocumentation searched (classification system followed by classification	on symbols)					
Documentat	tion searched other than minimum documentation to the extent that s	such documents are incl	uded in the fields searched				
	ata base consulted during the international search (name of data basternal, WPI Data	se and, where practical	, search terms used)				
C. DOCUM	ENTS CONSIDERED TO BE RELEVANT						
Category*	Citation of document, with indication, where appropriate, of the rele	Relevant to claim No.					
X	DATABASE WPI Week 200631 Thomson Scientific, London, GB; A 2006-295398 XP002565758 & CN 1 687 251 A (UNIV TIANJIN) 26 October 2005 (2005-10-26) abstract	1–17					
X	YING-JIE ZHU: "Poly(vinylpyrrolinew reductant for preparation of nanorods, nanowires, and tubes from NANOTECHNOLOGY, vol. 17, 2006, pages 645-650, XPC page 645, paragraph 1	1-17					
X Furti	her documents are listed in the continuation of Box C.	X See patent far	nily annex.				
* Special c	ategories of cited documents :	PTI lotor description 1	Enhad oftentho inter-street Street				
"A" docume consid "E" earlier of filling d "L" docume which citation "O" docume other other in the country of t	ent defining the general state of the art which is not letered to be of particular relevance document but published on or after the international late ent which may throw doubts on priority claim(s) or is cited to establish the publication date of another nor other special reason (as specified) ent referring to an oral disclosure, use, exhibition or means	"T" later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention "X" document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step when the document is taken alone "Y" document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combinated with one or more other such documents, such combination being obvious to a person skilled in the art. "&" document member of the same patent family					
Date of the actual completion of the international search Date of mailing of the international search report							
2	8 January 2010	19/02/2010					
Name and r	mailing address of the ISA/ European Patent Office, P.B. 5818 Patentlaan 2 NL – 2280 HV Rijswijk	Authorized officer					
	Tel. (+31-70) 340-2040, Fax: (+31-70) 340-3016	Siemens, Thomas					

INTERNATIONAL SEARCH REPORT

International application No
PCT/US2009/062975

Continu	tion). DOCUMENTS CONSIDERED TO BE RELEVANT	PCT/US2009/062975			
Category*		Delevent to elejo No			
A Lategory	SHAUN M. BAESMAN: "Formation of Tellurium Nanocrystals during Anaerobic Growth of Bacteria That use te Oxyanions as respiratory electron acceptors" APPLIED AND ENVIRONMENTAL MICROBIOLOGY, vol. 73, no. 7, April 2007 (2007–04), pages 2135–2143, XP002565757 page 2140; figure 6	Relevant to claim No.			

INTERNATIONAL SEARCH REPORT

Information on patent family members

International application No
PCT/US2009/062975

Pa cited	tent document in search report		Publication date		Patent family member(s)	Publication date	
CN	1687251	Α	26-10-2005	NONE			
		÷					
						÷ .	;
							:
						-	