

UNITED STATES PATENT OFFICE

2,003,914

PROPELLANT EXPLOSIVE

Edward Whitworth, Saltcoats, Scotland, assignor to Imperial Chemical Industries Limited, a corporation of Great Britain

No Drawing. Application May 15, 1933, Serial No. 671,279. In Great Britain May 18, 1932

8 Claims. (Cl. 52-22)

This invention relates to the production of propellant explosives of the double base solventless type, i. e., those containing nitrocellulose and nitroglycerin or the like and in the manufacture of which no volatile solvent is used.

In the manufacture of this type of propellant it is customary to employ a non-volatile substance which acts as a gelatinizer for the nitrocellulose, and also performs the function of a stabilizer for the explosive as a whole. The substance which has found widest application in this respect is symmetrical di-ethyl diphenyl urea, also known as ethyl centralite, but it has some disadvantages which render the manufacture of the propellant a somewhat difficult operation. In addition, the solubility of ethyl centralite in nitroglycerin at ordinary temperatures is rather low, with the result that the nitroglycerin present in a number of widely used propellants is insufficient to dissolve the desired amount of ethyl centralite.

It is known that symmetrical dimethyl diphenyl urea (methyl centralite) has properties which correspond generally to those of ethyl centralite, but the solubility in nitroglycerin is still lower than that of ethyl centralite and also the setting point of methyl centralite is higher than that of ethyl centralite, so that the difficulties in the manufacture of the propellant are still further increased if methyl centralite is substituted for ethyl centralite. We have now found, however, that by employing a mixture of ethyl centralite and methyl centralite the solubility of the mixture in nitroglycerin is considerably greater than that of the single substances, and also that the setting point of the mixture is definitely lower than that of either substance. Moreover, the gelatinizing power and stabilizing power of the mixture are at least equal to those of the single substances.

Our invention, therefore, consists in an improved method for the manufacture of gelatinized propellant explosives of the type containing nitrocellulose and nitroglycerin or the like nitric ester, according to which nitrocellulose is gelatinized with nitroglycerin or the like nitric ester and a mixture of ethyl centralite and methyl centralite in the absence of a volatile solvent.

We have also found that there is a eutectic mixture of ethyl centralite and methyl centralite, containing approximately 77 parts by weight of ethyl centralite to 23 parts of methyl centralite, this eutectic mixture having the lowest setting point of all the mixtures. The setting points and solubilities in nitroglycerin of ethyl cen-

tralite, methyl centralite and of the eutectic mixture are given in the following table:

Substance	Solubility in 100 gm. nitroglycerin at 20° C.	Setting point
Ethyl centralite.....	17 gm.	70° C.
Methyl centralite.....	4.5 gm.	120° C.
Eutectic mixture.....	50 gm.	61° C.

It will therefore be seen that the employment of the mixed centralites in the manufacture of propellant explosives of the type described offers considerable advantages over the use of either centralite by itself. We prefer to use the eutectic mixture, on account of its superior properties to other mixtures and also because it is easily prepared as a uniform product, but we may also use the mixed centralities in other proportions.

The advantages of the use of the mixed centralites become apparent during the actual process of manufacture of solventless propellants. The preliminary operation of mixing the nitroglycerin, nitrocellulose and centralites may be carried out just as in the case of ethyl centralite, or if desired, the mixed centralites may be dissolved in the nitroglycerin and the solution sprayed on to the water wet nitrocellulose, according to the methods well known in the art. The usual processes of drying and gelatinizing the paste are then carried out, and it is in these that some of the advantages of the mixed centralites become easy to observe. The paste sheets are formed with great ease on hot rolls, and final gelatinization takes place very readily. The operation of re-working faulty sheets or cords is considerably facilitated owing to the flexible nature of the colloid produced when the mixed centralites are employed. If it is desired to press the gelatinized sheets into cord form, the great flexibility and plasticity of the propellant are again in evidence.

Compositions which may be prepared according to the present invention consist in general of the following types:—

	Per cent
Nitroglycerin or the like.....	from 50 to 15
Nitrocellulose.....	from 40 to 80
Mixed centralites.....	from 3 to 15

Other ingredients may be introduced according to ballistic requirements, or in order to bring about flashlessness or for anti-fouling purposes. Such ingredients are: dinitrotoluene or trinitro-

toluene for ballistics; nitroguanidine for flashless-
 ness; and tetrabutyl stannane for anti-fouling.
 For many purposes, however, the use of ingredi-
 ents other than nitrocellulose, nitroglycerin and
 the mixed centralites is unnecessary. For ex-
 ample, excellent results are given by the applica-
 tion of the present invention to propellant com-
 positions of the following proportions:

	Parts by weight
10 Nitroglycerin -----	41
Nitrocellulose -----	50
Mixed centralites -----	9
	100

15 The example just given represents proportions
 of nitroglycerin, nitrocellulose and gelatinizer,
 which have long been in use, with ethyl centralite
 being employed as gelatinizer-stabilizer. It is to
 be noted that the use of the mixed centralites in-
 volves only a negligible change in the heat of
 combustion of the powder; also the stability of
 the propellant is not brought into question on
 account of the close similarity of the centralites
 in respect of stabilizing qualities. On the con-
 trary, the increased ease of working produced by
 the use of the mixed centralites makes it possible
 to reduce the temperatures required for the rolling
 and pressing operations, and this reduction of
 temperature is a distinct benefit to the stability
 of the propellant.

25 Having now particularly described and ascer-
 tained the nature of my said invention, I declare
 that what I claim is:

35 1. The process of manufacturing gelatinized
 propellant explosives containing nitrocellulose
 and a liquid nitric ester of a polyhydric aliphatic
 alcohol, which comprises gelatinizing the nitro-
 cellulose with the nitric ester and a mixture of
 diethyl diphenyl urea and dimethyl diphenyl urea.
 40

2. The process of manufacturing gelatinized
 propellant explosives containing nitrocellulose
 and a liquid nitric ester of a polyhydric aliphatic
 alcohol, which comprises gelatinizing the nitro-
 cellulose with the nitric ester, and approximately
 77 parts di-ethyl diphenyl urea together with
 approximately 23 parts di-methyl diphenyl urea
 in each 100 parts of the mixture. 5

3. A propellant explosive containing nitrocel-
 lulose and a liquid nitric ester of a polyhydric
 aliphatic alcohol, which comprises a gelatinized
 nitrocellulose, and a mixture of diethyl diphenyl
 urea and dimethyl diphenyl urea. 10

4. A propellant explosive containing nitrocel-
 lulose and a liquid nitric ester of a polyhydric
 aliphatic alcohol, which comprises a gelatinized
 nitrocellulose and a eutectic mixture of diethyl
 diphenyl urea and dimethyl diphenyl urea. 15

5. The process of manufacturing gelatinized
 propellant explosives containing nitrocellulose
 and nitroglycerin, which comprises gelatinizing
 the nitrocellulose with the nitroglycerin and a
 mixture of diethyl diphenyl urea and dimethyl
 diphenyl urea. 20

6. The process of manufacturing gelatinized
 propellant explosives containing nitrocellulose
 and nitroglycerin, which comprises gelatinizing
 the nitrocellulose with the nitroglycerin, and ap-
 proximately 77 parts of ethyl diphenyl urea to-
 gether with approximately 23 parts of methyl
 diphenyl urea in each 100 parts of the mixture. 25

7. A propellant explosive containing nitrocel-
 lulose and nitroglycerin, which comprises a gelat-
 inized nitrocellulose, and a mixture of diethyl
 diphenyl urea and dimethyl diphenyl urea. 30

8. A propellant explosive containing nitrocel-
 lulose and nitroglycerin, which comprises a gelat-
 inized nitrocellulose, and eutectic mixture of
 diethyl diphenyl urea and dimethyl diphenyl urea. 35

EDWARD WHITWORTH. 40