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#### TITLE

## GRAFT COPOL YMER EMULSIONS AND TWO-PACKAGE WATERBORNE URETHANE COATINGS

#### BACKGROUND OF THE INVENTION

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This invention relates to a graft copolymer emulsion and a waterborne polymer composition containing the graft copolymer emulsion, the emulsion comprising a polymer backbone and a macromonomer compound. This composition is stabilized anionically and contains 0.5 to 30 weight percent of an amine-functional monomer copolymerized in the backbone and/or the macromonomer. The copolymer emulsions of this invention can be cured with a polyisocyanate.

Coating compositions generally comprise a carrier liquid, a film-forming polymer and a curing (crosslinking) agent. For waterborne coatings, the carrier liquid is primarily water. The combination of the film-forming polymer and the crosslinking agent is commonly referred to as a binder. It is desirable that the binder have a low viscosity such that the binder is readily sprayable using automated spraying equipment and a high solids content such that an adequate paint film thickness is yielded with one to two cross-passes of a spray gun.

In a one-package system, the binder is contained within a single package, and curing is typically accomplished by baking the coating after application. In a two-package (2K) system, the binder is contained in two separate packages: a first package contains the film-forming polymer and a second package contains the curing agent. Shortly prior to application, the contents of the two packages are combined, and curing occurs through chemical crosslinking reactions. Once the two packages are combined, the crosslinking reaction is irreversible, and thus two-package systems have a relatively short pot life.

Film-forming polymers most commonly employed are linear addition-type copolymers containing functional groups such as hydroxyl groups, and these copolymers are cured by reaction with crosslinking agents. In two-package urethane technology, curing agents based on isocyanate functional

polymers are used that, upon mixing with a hydroxy-functional copolymer, react to form a crosslinked network.

The most relevant art appears to be WO95/32229. This publication, however, does not disclose the particular graft copolymer emulsion (of this invention) or the coating composition (of this invention) characterized as they are by the combination of excellent dry-film build, Persoz hardness, and solvent-resistance described herein (see Table 11).

#### **SUMMARY OF THE INVENTION**

This invention relates to graft copolymer emulsions which are anionically stabilized by an acid functional macromonomer and in which either macromonomer and/or backbone contain an amine functional monomer, preferably t-butylamino ethyl methacrylate (TBAEMA). We have found that such macromonomers are able to stabilize an overall waterborne polymer composition which has amino functional groups.

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This invention concerns a graft copolymer emulsion which is anionically stabilized and which contains an amine functional monomer. The graft copolymer has a weight average molecular weight of at least 3,000 and further comprises 20 to 95 weight percent of a polymer backbone of 5 to 80 weight percent of a macromonomer. Preferably, the graft copolymer comprises 60 to 95 weight percent of a polymer backbone and 5 to 40 weight percent of a macromonomer. The polymer backbone and/or macromonomer further comprises 0.5 to 30 weight percent of a secondary amino-functional monomer, preferably t-butylaminoethyl methacrylate. The macromonomer further comprises at least about 5 weight percent of an unsaturated acid-functional monomer, preferably methacrylic acid. The macromonomer is prepared using a cobalt-based catalytic chain transfer agent and has a weight average molecular weight of about 1,000 to 30,000.

The graft copolymer emulsion can be used in a waterborne coating composition of this invention which comprises 20 to 95 weight percent solids, based on binder solids, of a graft copolymer and 5 to 80 weight percent solids of a polyisocyanate curing agent and 0 to 75% of other film-forming polymers.

Preferred embodiments of this invention are the following.

A. A graft copolymer emulsion having a weight average molecular weight of at least 3.000 which is anionically stabilized and which comprises

 i) 20 to 95 weight percent of a polymer backbone of polymerized unsaturated compounds;

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ii) 5 to 80 weight percent of a macromonomer comprising at least about 5 weight percent of an unsaturated acid functional monomer, said macromonomer prepared using a cobalt-based catalytic chain transfer agent and said macromonomer having a weight average molecular weight of about 1,000 to 30,000;

so that the backbone comprises 0.5 to 30 weight percent of the total polymerizable unsaturated compounds of an amino-functional polymerizable component.

- B. A waterborne coating composition comprising as binder components:
  - a) 20 to 95 weight percent solids of graft copolymer emulsion A;
  - b) 5 to 80 weight percent solids of a polyisocyanate curing agent; and
  - c) 0 to 75 weight percent of other film-forming polymers.
- C. The waterborne coating composition of B wherein said secondary amino functional monomer is t-butylaminoethyl methacrylate.
  - D. The waterborne coating composition of B wherein said catalytic chain transfer agent is selected from the group consisting of borondifluoro-stabilized chelates of a dialkyl- or diaryl-glyoxime of CoII or CoIII.
- E. The waterborne coating composition of B wherein said
  graft copolymer comprises 60-95 weight percent of said polymer backbone and 5
  to 40 weight percent of said macromonomer.
  - F. The waterborne coating composition of B wherein said polymer backbone comprises 5 to 50 weight percent of a vinyl aromatic. 10 to 70 weight percent of a compound selected from the group consisting of alkyl acrylate, cycloalkyl acrylate, and mixtures thereof, 5 to 40 weight percent of a compound selected from the group consisting of a hydroxy-functional

polymerizable acrylate, hydroxy functional polymerizable methacrylate, and the mixtures thereof, 1 to 8 weight percent of a secondary amino functional monomer.

- G. The waterborne coating composition of B wherein the said macromonomer comprises 10 to 90 weight percent of a compound selected from the group consisting of alkyl methacrylate, cycloalkyl methacrylate, and mixtures thereof, 0 to 40 weight percent of a hydroxy functional methacrylate, 0 to 20 weight percent of a secondary amino functional methacrylate, and 5 to 80 weight percent of an unsaturated acid functional monomer, wherein said macromonomer has a weight average molecular weight of 1,500 to 10,000.
- H. The waterborne coating composition of B wherein said unsaturated acid functional monomer is methacrylic acid.

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- I. The waterborne coating composition of B wherein said graft copolymer is neutralized with an amine.
- J. The waterborne coating composition of B further

  comprising an oligomeric methacrylate having a degree of polymerization for 2 to

  10, wherein said oligomeric methacrylate is added in an amount to control the

  weight average molecular weight of the graft copolymer to about 3,000 to 80,000.
- K. The waterborne coating composition of B wherein said polyisocyanate curing agent is selected from the group consisting of derivatives of hexamethylene diisocyanate, derivatives of isophorone diisocyanate, derivatives of m-xylylene diisocyanate, and mixtures thereof.
  - L. The waterborne coating composition of B further comprising 0-75 weight percent solids of a compound selected from the group consisting of auxiliary binder and crosslinker, additives, pigments, and extenders.
  - M. The waterborne coating composition of B wherein said polyisocyanate curing agent further comprises 0 to 75 weight percent hydrophobic polyisocyanate and about 25 to 100 weight percent hydrophilic polyisocyanate.
- N. The waterborne coating composition of M wherein said hydrophobic polyisocyanate curing agent is selected from the group consisting of
   biurets or cyclotrimers of hexamethylene diisocyanate and/or isophorone diisocyanate.

O. The waterborne coating composition of M wherein said polyisocyanate curing agent is selected from polyisocyanates such as hexamethylene diisocyanate and isophorone diisocyanate modified with polyethyleneoxide to make the hydrophilic polyisocyanate water-dispersible.

P. The waterborne coating composition of L wherein the auxiliary binders and crosslinkers are selected from carbodiimide, epoxy derivatives, polyurethane and acrylic emulsions.

Other preferred embodiments comprise an (i) component having from 0.5 to 20 weight percent of a secondary amino-functional unsaturated monomer, e.g., tert-butyl amino ethyl methacrylate; the preferred crosslinking agent is a polyfunctional isocyanate having at least two isocyanate functionalities. The compositions of this invention are characterized by especially good drying performance.

#### 15 <u>DETAILS OF THE INVENTION</u>

The graft copolymer is prepared by grafting macromonomers onto a polymer backbone. The macromonomers are prepared via free radical polymerization and comprise at least 5 percent of an unsaturated acid functional monomer. The macromonomers are neutralized with an amine and inverted into an anionic stabilized dispersion. Cobalt chain transfer agents are employed to yield macromonomers having a weight average molecular weight from about 1,000 to 30,000. The backbone monomers and/or macromonomers can comprise up to 30 percent of secondary amino-functional monomers.

#### 25 Macromonomer Synthesis

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The macromonomer contains a single terminal ethylenically unsaturated group which is polymerized into the backbone of the graft copolymer and primarily contains polymerized monomers. Preferably, the macromonomer is based on methacrylate or methacrylamide derivatives.

Examples of methacrylate monomers are alkylmethacrylates as methyl methacrylate, ethyl methacrylate, n-butyl methacrylate, i-butyl

methacrylate, t-butyl methacrylate, 2-ethyl hexyl methacrylate, lauryl methacrylate, stearyl methacrylate. Other examples include cycloalkyl, aryl an aryl alkyl methacrylates as cyclohexyl methacrylate, isobornyl methacrylate, t-butyl cyclohexyl methacrylate, phenyl methacrylate, benzyl methacrylate and the like. Functional methacrylates are hydroxy functional methacrylates as 2-hydroxy ethyl methacrylate. 2- and 3-hydroxypropylmethacrylate, 4-hydroxybutyl methacrylate and the like. An example of an acid functional methacrylate and the preferred one is methacrylic acid. Examples of methacrylamide derivatives are methacrylamide, n-methylol methacrylamide, n-methylol methacrylamide, n-methylol methacrylamide, n-methylol methacrylamide, n-methylol methacrylamide and the like.

Other functional monomers can be copolymerized in the macromonomer compositions such as dimethyl aminoethylmethacrylate, diethylamino ethyl methacrylate, t-butyl amino ethyl methacrylate, methacryloxy ethyl phosphoric acid, sulfo ethyl methacrylate, methacryloxy ethyl, 2,2-spirocyclohexyl oxazolidine, nitrophenol methacrylate, ethyleneureaethyl methacrylate, methacrylonitrile, phtalimido methacrylate, polyethylene glycol methacrylate (varying in molecular weight), polypropyleneglycol methacrylate (varying in molecular weight), aceto acetoxy ethylmethacrylate, M-morpholinoethyl methacrylate, trimethoxy silyl propyl methacrylate.

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Small amounts of polyfunctional methacrylates can be incorporated in the macromonomers such as ethyleneglycol dimethacrylate, I,4-butane diol dimethacrylate, trimethylpropane trimethacrylate, and allylmethacrylate.

Reaction products of methacrylate monomers can be copolymerized. Examples include but are not limited to methacrylic acid reacted with epoxy derivatives such as cyclohexeneoxide, phenyl glycidyl ether, versatic acid epoxy esters such as Cardura® E10 and E5, 1,2-butylene oxide.

Other polymerizable methacrylate derivatives include reaction products of hydroxy alkyl methacrylate with  $\epsilon$ -caprolactones and reaction products of glycidyl methacrylate with acids such as acetic acid, benzoic acid. At least 50 percent and preferably more than 70 percent can be methacrylate type monomers. Up to 40 percent of other polymerizable compounds can be

copolymerized. Examples include acrylate monomers, acrylamide derivatives, vinyl monomers as vinyl aromatics and vinyl esters, maleic acid, fumaric acid, itaconic acid and its derivatives, acrylonitrile and acroleine.

In a first step, the macromonomer is prepared by polymerization of a monomer blend in an organic solvent or solvent blend (including a blend with water) using a free radical initiator and a cobalt chain transfer agent at an elevated temperature between 40°C and 190°C. Typical solvents that can be used to form the macromonomer are ketones such as methylethyl ketone, isobutyl ketone, ethylamyl ketone, acetone, alcohols such as methanol, ethanol, isopropanol, esters such as ethyl acetate, glycols such as, ethyleneglycol monobutyl, ethers such as tetrahydrofuran, aromatic solvents as toluene, xylene and the like.

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To insure the resulting macromonomer only has one terminal ethylenically unsaturated group which will polymerize with the backbone monomers to form the graft copolymer, the macromonomer is polymerized by using a catalytic chain transfer agent that contains a Co+2 or a Co3+ group. Preferred cobalt chain transfer agents are borondifluoro stabilized chelates of a dialyl, borondifluoro stabilized chelates of a diarylglyoxime, and mixtures thereof. More specifically, preferred cobalt chain transfer agents are pentacyanocobaltate (II), diaquabis (borondifluoro dimethyl-glyoximato) cobaltate (II), and diaquabis borondifluorophenylglyoximato) cobaltate (II). Typically, cobalt chain transfer agents are used at concentrations of about 5 to 5,000 ppm depending upon the particular monomers used.

In subsequent steps, additional monomers and cobalt catalyst and conventional azo type polymerization initiators such as 2,2-azobis(2-methylbutane nitrile) and 2,2'-azobis(2,4'-dimethylpentane nitrile), 2,2'-azobis(2,4-dimethyl-4-methoxyvalero nitrile) are added and polymerization is continued until a macromonomer is formed of the desired molecular weight.

Any of the above mentioned azo-type catalysts can be used as can other suitable initiators such as peroxides, peroxyesters, peroxycarbonates and hydroperoxides. Typical of such catalysts are di-tetriarybutyl peroxide, di-cumyl

peroxide, di-tertiaryamyl peroxide, cumene hydroperoxide, di-n-propyl peroxydicarbonate, peresters such as amyl peroxyacetate and the like.

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Preferred alkyl methacrylates that can be used to produce the macromonomer have 1-18 carbon atoms in the alkyl group such as methyl methacrylate, ethyl methacrylate, propyl methacrylate, isopropyl methacrylate, butyl methacrylate, pentyl methacrylate, hexyl methacrylate, 2-ethylhexyl methacrylate, lauryl methacrylate and the like.

Other polymerizable nonmethacrylate based monomers that can be used in amounts less than 50 percent by weight are styrene, alpha-methyl styrene, methacrylamide and methacrylonitrile. Preferred hydroxy functional monomers that can be used are 2-hydroxyethyl methacrylate, 2- and 3-hydroxypropyl methacrylate, and the like. Preferred amino functional monomers are diethylaminoethyl methacrylate, t-butylaminoethyl methacrylate, methacryloxyethyl (2-2'spirocyclohexyl) oxazolidine.

The macromonomer can comprise 3 to 100 percent by weight, preferably about 20 to 50 percent by weight, based on the weight of the macromonomer, of an unsaturated acid functional monomer, i.e., a polymerized ethylenically unsaturated acid. Methacrylic acid is preferred particularly if it is the sole constituent. Other acids that can be used in less than 50 percent by weight are ethyleneically unsaturated carboxylic acids such as acrylic acid, itaconic acid, maleic acid and the like. Ethylenically unsaturated sulfonic, sulfinic, phosphoric or phosphonic acid and esters thereof also can be used such as styrene sulfonic acid, acrylamido methylpropane sulfonic acid, vinyl phosphonic acid and the like.

A preferred macromonomer comprises 10 to 90 weight percent of alkyl methacrylate, cycloalkyl methacrylate, or mixtures thereof, 0 to 40 weight percent of a hydroxy functional methacrylate, 0 to 20 weight percent of a secondary amino functional methacrylate, and 5 to 80 weight percent of an unsaturated acid functional monomer, wherein said macromonomer has a weight average molecular weight of about 1,500 to 10,000. Another more preferred macromonomer having a weight average molecular weight of about 2,000 to

5,000 comprises about 20 to 80 percent by weight of polymerized methyl methacrylate, 10 to 50 percent by weight of polymerized methacrylic acid, and 10 to 40 percent by weight of 2-hydroxyethyl methacrylate and 0 to 20 percent of tbutylamino ethyl methacrylate.

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#### Graft Copolymer Synthesis

The graft copolymer is prepared by emulsion polymerization wherein the macromonomer is first neutralized with a base and subsequently dispersed in water, and the backbone is formed in the presence of the macromonomer dispersion. This yields a graft copolymer emulsion without the necessity for emulsifiers. Typical bases are amines and include amino methyl propanol, amino ethyl propanol, dimethyl ethanol amine, triethyl amine and the like. A preferred base is dimethyl ethanol amine, and ammonium hydroxide. Other bases that can be used are metal hydroxides of sodium, potassium, lithium and the like. The backbone monomers can be a blend of monomers as described for the macromonomer.

Alkyl methacrylates can be used in the backbone of the graft copolymers, and typically have 1 to 18 carbon atoms in the alkyl group such as methyl methacrylate, ethyl methacrylate, propyl methacrylate, isopropyl methacrylate, butyl methacrylate, pentyl methacrylate, hexyl methacrylate, 2-ethyl hexyl methacrylate, lauryl methacrylate and the like. Cycloaliphahtic methacrylates also can be used such as trimethylcyclohexyl methacrylate, isobutylcyclohexyl methacrylate, isobornyl methacrylate and the like.

Aryl methacrylates such as benzyl methacrylate also can be used in 25 the backbone of the graft copolymers. Other polymerizable monomers that can be used are styrene, alpha methyl styrene, methacrylamide and methacrylonitrile. Hydroxy-functional monomers can be used such as hydroxyethyl methacrylate, hydroxypropyl methacrylate, and the like. Amino-functional monomers can be used such as diethylaminoethyl methacrylate, t-butylamino methacrylate, methacryloxyethyl (2-2'-spirocyclohexyl) oxazolidine, 2-(4-pyridyl)ethyl methacrylate and the like.

Alkyl acrylates can be used, preferably having 1 to 18 carbons in the alkyl group such as methyl acrylate, ethyl acrylate, propyl acrylate, isopropyl acrylate, butyl acrylate pentyl acrylate, hexyl acrylate, 2-ethyl hexyl acrylate, nonyl acrylate, lauryl acrylate and the like. Functional monomers such as hydroxyalkyl acrylates, aminoalkyl acrylates, acrylamide and acrylonitrile can be used, including 2-hydroxyethyl acrylate, 2- and 3-hydroxypropyl acrylate, 2- and 4-hydroxybutyl acrylate, n-butoxymethyl acrylamide, n-methoxymethyl methacrylamide, dimethylaminoethyl acrylate, n-methylol acrylamide etc. Other functional monomers also include the reaction product of acid functional monomers with monoepoxy ethers or monoepoxy esters and reaction products of glycidyl functional monomers with mono amines or mono acids. Cycloaliphatic acrylates can be used such as trimethylcyclohexyl acrylate, t-butylcyclohexyl acrylate and the like. Aryl acrylates such as benzyl acrylate also can be used. Vinylaromatics can be used as styrene, vinyl toluene, t-butyl styrene, α-methyl 15 styrene and the like. Vinyl derivatives can be used as vinyl acetate, vinylpropionate, versatic esters of vinylalcohol (VEOVA-Shell) vinylchloride, vinylsilane and the like. Other polymerizable monomers include those mentioned under macromonomer compositions.

The backbone monomers preferably include a secondary amino

functional monomer as t-butyl amino ethyl methacrylate. Secondary amino

functionality can also be achieved by hydrolysis of methacryloxyethyl-(2,2'
spirocyclohexyl)oxazolidine in water to yield 2-hydroxyethyl amino ethyl

methacrylate. Another way of getting secondary amino functionality can be added

to a polymer chain according to the reaction shown in Formula 1, wherein

copolymerized glycidylmethacrylate is reacted with a monofunctional amine.

$$R-CH-CH_2 + R'NH_2 \longrightarrow R-CH-CH_2-NH-R'$$

Formula 1

Acid functionality can also be converted to primary amine functionality by imination and further reaction with a monoepoxide would give a secondary amine, as shown in Formula 2.

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#### Formula 2

Typical cosolvents in blends with water coming from the macromonomer dispersion that can be used in the synthesis of the graft copolymer are ketones such as methyl ethyl ketone, isobutyl ketone, ethyl amyl ketone, acetone, alcohols such as methanol, ethanol, isopropanol, esters such as ethyl acetate, glycols such as ethylene glycol mono butyl ether and the like.

Polymerization initiators are used in the preparation of the graft copolymer. Some suitable thermal initiators are water miscible and include potassiumpersulfate, sodiumpersulfate, ammoniumpersulfate, and 4,4'-azobis (4-cyanovaleric acid). Other thermal initiators include azo compounds such as 2,2'-azobis(2-methylbutane nitrile) and 2,2'-azobis (2,4'-dimethylpentane nitrile) 2,2'-azobis(2,4-dimethyl-4-methoxyvalero nitrile). A preferred azo catalyst is 4,4'-azobis, 4-cyanovaleric acid. Other suitable thermal catalysts include peroxides and hydroperoxides such as di-tertiarybutyl peroxide, di-cumyl peroxide, ditertiary amyl peroxide, cumene hydroperoxide, di(n-propyl peroxydicarbonate, peresters such as tertiary amyl peroxyacetate and the like.

Redox initiators can be used and include peroxides combined with suitable reducing agents, said reducing agents including ascorbic acid; sodium formaldehyde sulphoxylate; thiosulfates; bisulphates; hydrosulphates; water soluble amines such as diethylene triamine, etc.; and salts as the sulphates of

metals which are capable of existing in more than one valence state, said metal including cobalt, cerium, iron, nickel, copper, and the like.

Chain transfer agents, such a mercaptans and halogenated derivatives, can be used to control the molecular weight of the copolymer. Other ways of controlling the molecular weight is the use of low molecular weight macromonomers. A preferred chain transfer agent is oligomeric methacrylate having a degree of polymerization from 2 to 10, wherein the oligomeric methacrylate is added in an amount to control the weight average molecular weight of the graft copolymer to about 10,000 to 80,000. The graft copolymerization process of this invention is typically carried out below 140°C and preferably at 20° to 100°C depending on the initiator. The graft copolymerization process can be either batch, semi-batch or continuous feed.

A preferred polymer backbone comprises 5 to 50 weight percent of a vinyl aromatic, 10 to 70 percent of alkyl acrylate, cycloalkyl acrylate, or mixtures thereof, 5 to 40 weight percent of hydroxy functional polymerizable acrylate, hydroxyfunctional polymerizable methacrylate, or mixtures thereof, and 1 to 8 weight percent of a secondary amino functional monomer.

Another more preferred backbone polymer for use in a two component coating formulation comprises 1 to 10 weight percent t
butylaminoethyl methacrylate, 5 to 40 weight percent of a vinyl aromatic, 5 to 40 weight percent of an alkyl or cycloalkyl acrylate, and 5 to 40 weight percent of a hydroxy functional monomer, preferably hydroxy isopropyl methacrylate.

#### Package A - Binder

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In a two-package waterborne system, the aqueous emulsion graft copolymers comprise a single package, designated herein as package A. The emulsion graft copolymer can be used directly as binder or mixed with other auxiliary polymeric binders. The auxiliary binders can but need not have hydroxy and/or secondary amine functionality.

The aqueous or emulsion graft copolymer can also be used in a formulation which incorporates additional compounds such as auxiliary binders,

pigments, extenders and additives. Preferably, the waterborne coating composition comprise 0 to 75 weight percent solids of auxiliary binders, additives, pigments and extenders. Examples of pigments are organic and inorganic pigments such as phthalocyanines, quinacridones, iron oxides, titanium dioxide, aluminum silicates, talcs and anticorrosion type pigments such as zinc phosphates and so forth. Examples of additives include but are not limited to defoamers, thickeners, leveling agents, dispersants, UV absorbers, antioxidants and the like.

The formulations containing the graft copolymer emulsion and auxiliary polymers and agents can be diluted further with water or cosolvent to bring it to a suitable viscosity. Typical solvents include alcohols, ethers, ketones, esters, aromatics, aliphatics. Acids or bases can be added to adjust the pH of the formulation.

#### Package B - Curing Agent

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In a two-package waterborne system, the polyisocyanate curing agent comprises a single package, designated herein as package B. The curing agent can be any organic polyisocyanate or mixtures thereof which have aliphatically, cycloaliphatically, aryl-aliphatically, or aromatically bound isocyanate groups. The polyisocyanate curing agent will have at least two isocyanate functional (NCO) groups. If necessary, the polyisocyanates can be used together with small quantities of inert solvents to lower the viscosity.

Preferred polyisocyanates are those containing biuret, urethane, uretdione, and/or isocyanurate groups prepared from hexamethylene diisocyanate (HDI), isophorone diisocyanate, 1-isocyanto-3,3,5-trimethyl-5-isocyanato methylcyclohexane (IPDI), bis-(isocyanato cyclohexyl)-methane (HDPDI) and mtetramethyl xylene diisocyanate (mTMXDI), for example N,N',N'-tri-(6-isocyanatohexyl)-biuret and/or N,N',N'-tri-(6-isocyanatohexyl)-isocyanurate; also, polyisocyanates prepared from 2.6-diisocyanatotoluene and 4,4'-diisocyanatodiphenylmethane or mixtures thereof. Polyisocyanates rendered hydrophilic with polyethylene oxide can be used as curing agents. Furthermore, hydrophobic polyisocyanates can be blended with the hydrophilic polyisocyanate

up to a weight ratio of 75 parts hydrophobic polyisocyanate to 25 parts hydrophilic polyisocyanates. Suitable hydrophilic polyisocyanates include Desmodur® LS2032 from Bayer.

## 5 Package A combined with Package B - Coating Compositions

Before application on a substrate, the coating composition of this invention is prepared by combining the emulsion graft copolymers in package A with a polyisocyanate curing agent contained in package B and stirring at room temperature. Emulsifiers can be added to assist dispersion and stabilization of the coating composition. The copolymers in package A contain both hydroxy (OH) functional and secondary amino (NH) functional groups, and the curing agent of package B contains isocyanate (NCO) functional groups. The quantity of polyisocyanate curing agent used is calculated to result in an isocyanate to hydroxy and secondary amino (NCO/(OH + NH) equivalent ration of 0.5/1 to 5/1, preferably 0.8/1 to 2/1.

The coating compositions according to the invention are suitable for all fields of applications. Examples include coatings for concrete, wood, paper and metals. The coating compositions can also be used as primers or fillers or as binders for top coats in automotive paints. After reducing the coating composition to a suitable viscosity with water and/or cosolvents, the coating composition is applied on the substrate and dried at room temperature or under stoving conditions up to 220°C. The following Examples illustrate the present invention in more detail. All parts and percentages are expressed by weight. Molecular weights are determined using gel permeation chromatography.

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## **EXAMPLES AND PROCEDURES**

#### Procedures 1 to 10 Macromonomers

These procedures illustrate the use of CoII chelate in the synthesis of the following macromonomers. The chelate is BF<sub>2</sub> bridged CoII (1,2-diphenyl-1,2-dioxoiminoethane)2 ( $H_2O$ )<sub>2</sub> chelate. Mixture 1 of Table 1 below was heated at reflux ( $\pm 80$ °C) in a reaction vessel that was kept under nitrogen. Mixture 2 was

added over 4 hours. Simultaneously with mixture 2, mixture 3 was added over 90 min followed immediately by mixture 4. Mixture 5 was added, for rinsing, followed by a 5 min hold. Afterwards, mixture 6 was added over 30 min followed by another rinsing step and held for 60 min. During the total process, the temperature was kept at reflux.

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	Mixture 1	Methyl ethyl ketone	<u>Mixture 2</u> Methyl Methacrylate	2-Ethyl Hexyl Methacrylate	t-Butyl Amino Ethyl Methacrylate	z-nydroxy Etnyl Methacrylate Methacrylic Acid	Methyl Ethyl Ketone	Mixture 3	Methyl Ethyl Ketone	COII Chelate	VAZO® 67 initiator	Mixture	Methyl Ethyl Ketone	COII Chelate	VAZO® 52 initiator	Mississe	Methyl Ethyl Ketone	Missing	Tiononov@ 25075	VA 200 52 initiation	VACOUS DE IMITIATUR	Methyl Ethyl Ketone	Methyl Ethyl Ketone	
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Procedure 6 5.114	100		Procedure 6	49.7%	24-1/3	65	0069	15400			
Procedure 5	100		Procedure 5	not measured	more than Z6	250	2000	4100			
Procedure 4 5.102	100		Procedure 4	48.7%	less than A	133	1700	3400			
Procedure 3	001	TABLE 2	Procedure 3	20.0%	Σ	89	3200	6400	hylpentanenitrile)	utanenitrile)	
Procedure 2	001		Procedure 2	50.2%	1-1/4	143	1500	3000	2,2-Azo bis(2,4-dimethylpentanenitrile)	2,2-Azo bis(2-methylbutanenitrile)	t-butylperoxypivalate 75% in mineral spirits
<u>Procedure 1</u> 5.111	001		Procedure 1	81.8%	23	252	2300	4800	2,2-4		t-but 75%
									DuPont	DuPont	Akzo
Final thinning Methyl Ethyl Ketone TOTTAL			Test Results	Solids	Viscosity (Gardner Holdt)	Acid value	Molecular Weight Number	Molecular Weight Weight	VAZO® 52	VAZO® 67	Trigonox® 25C75
5		10				15				20	

	Procedure 10	20	. 1		27.5	•	2.5	10	9	. 4	1	•	4	0.010	0.35			9	0.005	0.3	!	1.23	
TABLE 3	Procedure 9	•	20		•	6.5		30	70			3 6	C. 4	0.0075		0.1875		12.5	0.0125	0.3125		-	-
	Procedure 8	20			<u>8</u>	r	i	22	10	9		7	, 000	0.000	0.33			10	0.003	0.3		1.23	1
	Procedure 7	20	•		22.5	•	, ;	17.5	01	9		4	0.006	0.000		•		10	0.003	0.3		1.23	
	Mixture 1	Methyl ethyl ketone	Isopropanoi	Mixture 2	Memyl Melhacrylate	EButyl Amino Febral Machamalan	2-Hydroxy Edyal Master alac	Methodalic Acid	Memacrylic Acid	Methyl Ethyl Ketone	Mixture 3	Methyl Ethyl Ketone	COII Chelate	VAZO® 67 Initiator	VAZO® 52 Initiator		Mixture 4	Methyl Ethyl Ketone	COII Chelate	VAZO® 52 Initiator	Mixture 5	Methyl Ethyl Ketone	isopropanoi
	Ś				IJ	2					12					ć	20				;	52	

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	Procedure 10	0.1	1.9	. :	0.1 6.11	100			Procedure 10	46.9%	9Z <<	138	1500	6300
,   	Procedure 9	• 0.1	6:1	. ;	7.48	Ξ		TABLE 4	Procedure 9	45%	%+X	231	1800	3500
	Procedure 8	- 0.1	6.1	. ;	3.11	100			Procedure 8	48.4%	¥	125	1900	3300
	Procedure 7	0.1	6.1		II .	100			Procedure 7	49.9%	¥	141	1800	3300
	Mixture 6	Trigonox® 25C75 VAZO® 52 Initiator	Methyl Ethyl Ketone Methyl Ethyl Ketone	Final thinning Methyl Bithyl Ketone	Isopropanol	TOTAL			Test Results	Solids	Viscosity (Gardner Holdt)	Acid value	Molecular Weight Number	Molecular Weight Weight
	S			10			15	19		20				

# PROCEDURES 11 AND 12. OLIGOMERIC METHACRYLATES TABLE 5

Mixture 1 was brought to reflux under nitrogen inserting. Mixture 2 was added as a shot followed by addition of mixture 3 over 4 hours.

5		Procedure 11	Procedure 12
	Mixture 1 2-Ethyl Hexyl Methacrylate Methyl Ethyl Ketone Methyl Methacrylate	- 6 14	14 6
10	Mixture 2 COII Chelate Methyl Ethyl Ketone Methyl Ethyl Ketone (Rinse)	0.01 2.5	0.02 2.5 1
15	Mixture 3 2-Ethyl Hexyl Methacrylate Methyl Methacrylate Methyl Ethyl Ketone VAZO® 52	56 18.93 0.56	56 - 18.92 0.56
20	Final thinning Methyl Ethyl Ketone TOTAL Test Result	1 100	1 100
	Viscosity (Gardner Holdt) MN	less than A 230	less than A
25	MW	330 (three peaks)	800 (three peaks)

## EXAMPLES 1 TO 12. GRAFT COPOLYMER EMULSIONS

These Examples illustrate the preparation of a graft copolymer

30 emulsion according this invention.

Mixture 1 was heated to reflux (±95°C) after the pH has been adjusted to 7.5-8. The solvent methyl ethyl ketone and/or isopropanol was stripped-off azeotropically. Mixture 2 and 3 were further added simultaneously over 4 hours. Mixture 4 was added as a rinse and the reactor contents were further held at reflux for 1 hour. Mixture 5 was added and the solvent (methyl ethyl

ketone and/or isopropanol) was further distilled azeotropically. Finally, the pH was adjusted to about  $8\,$ 

	Example 6	26	· •		22.50		י רי			9 (	5.5			<u>.</u>	5,0	C7.7	0.50	6.43					8.0		8.0	18.4
	Example 5	26	•	,	22.50		3.3	!		11.5	0.0	•		15	375	77.7	0.30	6.43	•		0.64	50.0	•	•	ı	19.36
	Example 4	26.74	11.25		ı	2.1				13.5	300	13.723	7.65		275	05.0	9.5	3.21			,	8 (	9.0	0.0	,	18.6
TABLE 6	Example 3	27.99	•	22.50		2.1				13.5	01.0	9.10	7.65		2.25	0.50	50.5	17.6			,	80	9:0	2	•	18.6
	Example 2	26.79		22.50	•		3.3			13.5	8 10	2 1	7.65		2.25	0.50	3.21	13:0				8.0		80	5	18.4
	Example 1	26.79	•	22.50		5.1	•			13.5	01.8			•	•	0.50	121	2.25	4			8.0	9.0	•		9'81
	Mixture 1	Deionized water	Macromonomer Procedure 1	Macromonomer Procedure 2	Macronionomer Procedure 4	Dimethylethanolamine	Dimethylaminomethylpropanol	(80% in water)	Mixture 2	Styrene	2-Ethylhexyl Acrylate	2. Hydroxympophy Mathematica	Addiso A = 15: A = 1	Adduct Acrylic Acid and Cardura® E10*	t-Butylaminoethyl Methacrylate	Butyl Cellosolve	Oligomer of Procedure 11	Methacryloxyethyl (2.2'-Spirocyclo		Mixture 3	Ammoniumpersulfate	Azo-Carboxy**	Dimethylethanolamine	Dimethylaminoethylpropanol	(80% in water)	Deionized water
		S				,	0.7					~	•		2	2		20						25		

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Deionized Water Butylcellosolve	example 1 0.5	Example 2 0.5	<u>Example 3</u> 0.5	Example 4 0.5	Example 5 0.5	Example 6 0.5	
	3	C n	C.O	0.5	0.5	0.5	
	2.45	2.45	2.45	3.90	2.45	2.45	
	112.213	112.213	112.213	105.625	112.93	112.93	
	12.213	12.213	12.213	6.588	12.93	12.93	
	100	100	001	99.037	100	001	
			TABLE 7				
	Example 1	Example 2	Example 3	Example 4	Example 5	Example 6	
	45.3%	47.3%	45.3%	44.27%	40.1%	42.7%	
	400 cps	14,800 cps	360 cps	720 cps	45 cps	520 cps	
	7.9	8.2	8.3	8.3	7.3	8.2	
	5900	2600	5700	8800	3000	2900	
	31,200	36,400	39,300	55,700	14,500	18,300	

<u>≤</u> 23

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	Example 12		78	• .	1.68	•	•	9				13 6	 	<b>э</b>	7.65	3.25	7.7	0.5	3.21	20.5	7.63		8:0	0.6	200	0.0		5.0	2 6	C.O.	,	2011
	Example 11	č	97	٠,	7.7	•			22.5			13.5	27.0	0.00	7.65	1.69		C.U	3.21	•			<b>8</b> .0	9.0	16.30			0.5	20	ĵ		7
	Example 10	38	07	· c		· ;	22.5					13.5	o	,	5.625	3.375	0.5	3 6	3.2]			•	<b>8</b> :0	9.0	18.6			0.5	0.5	<b>!</b>	3 403	7.407
TABLE 8	Example 9	2.8	; '	2.1	i			•	22.5			13.5	10.35	99.6	20.7	,	0.5		3.21	•		Ġ	0.0	9.0	16.39		•	6.5	0.5		29 87	
	Example 8	28	,	2.1	200	i		•			:	13.5	6	37.9	0.75	2.25	0.5	171	7:41			80	9 0	0.0	9.81		3.0	0.0	0.5		3.403	
	Example 7	28	11.25	2.1	•		1	•				13.5	9.225	11 02 \$	20.1.	5.5/5	0.5	3.21		•		80	90	9	9.6		9.0	7 .	0.5		3.403	
	Mixture 1	_	Macromononier Procedure 1	Dimethylethanolamine	Macromonomer Procedure 7	Macromonoer Procedure 8		Macromonomer Drocedure 10	of althoughter trocedure to	Mixture 2	Styrene	Colorate  7 Establishment A and the		2-Hydroxypropyl Methacrylate	t-Burylaminoethyl Methachylate	Burd Callacatus	Daily 1 Callosolive	Oligomer of Procedure 11	Methyl Methacrylate		Mixture 3	Azo-Carboxy (See Table 7)	Dimethylethanol amine	Deionized Water		Mixture 4	Deionized Water	Butyloellosolva	Surjection of the	Mixture 5	Deionized Water	
		5					10						7.	C			2	4			20						25					

	Example 12	106.463 6.463 100		Example 12	43 5%	120 cm	sch 071	5.0	20,600
	Example 11	136.17 12.21 123.96		Example 11 Ex	34.9%		8 4	•	
ned)	Example 10	112.213 12.213 100		Example 10	44.9%	1400 cps	4.8	2200	11,700
TABLE 8 (Continued)	Example 9	136.17 12.21 123.96	TABLE 9	Example 9	36.2%	88,000 cps	7.9	5800	47,300
	Example 8	112.213 12.213 100		Example 8	44.8%	560 cps	8.3	2800	14,300
	Example 7	106.588 6.588 100		Example 7	49%	75,000 cps	8.2	5900	26,600
		TOTAL LOSS YIELD		Test results	Solids	Viscosity (Brookfield)	рН	MN	MW
		'n	01			15	25		

EXAMPLES 13 TO 17

Compositions of the graft copolymer on a weight basis 100 is shown below:

	Example 17	25	30 25.5 24.5 2.5	100		45.5	4600 cps	. &	7700	37,800
	Example 16	12.5	30 25 10 5	001		48.5	400 cps	9.2	5100	25,500
TABLE 10	Example 15	25	33 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5	100		45.3	350 cps	8.1	4600	21,400
	Example 14	25 5	. 40 8	. 001		48.5	800 cps	∞	7500	65,000
	Example 13	15 5 5	30 22 5 5	001		45.7	3350 cps	8.3	8100	49,200
		Macromonomer of Procedure 2 Macromonomer of Procedure 4 Macromonomer of Procedure 1 Oligomer of Procedure 9	Styrene 2-Ettyl Hexyl Acrylate 2-Hydroxypropylmethacrylate t-Butylaminoethylmethacrylate	Total	Test result	Solids	Viscosity	Hd	ΔN	MW
2		10	15	26	20				;	25

### EXAMPLES 18 AND 19

The acrylic emulsions were blended with a water-dispersible polyisocyanate (Desmodur® 2032) in such a ratio that the ratio of NCO over OH-functional groups was 1.5 over 1. Drawdowns were prepared over glass and the panels were air-dried (AD) and infrared dried (IR). Dry-film builds are indicated in microns (µ). The Persoz hardness after AD and solvent-resistance both AD and IR were measured after several hours (H) to days (D).

						1700	-1								
Based On					Persoz Har	Persoz Hardness (AD)						Solvent Resistance	tance		
	Without	lno				With Poly	With Polyisocyanate								
	Polyisocyanate	yanate													
	2H	4H		4H	=		\$		ē	,					
							F		7	מי	2	<b></b>	Q	<u>~</u>	
	70°	705	30п	155	30μ	55	30μ	SSµ	30-	55µ					
Ex. 18 Emulsion of Example 3	103	113	170	162	257	234	304	17.6	36.7	287	2	200			
									ì	107	:	V	,	ζ.	
Ex. 19 Emulsion of Example 2	114	821	203	92.	273	22	317	204	303	881	P.F	vG-EX	F.G	0	

Where the terms "comprise", "comprises", "comprised" or "comprising" are used in this specification, they are to be interpreted as specifying the presence of the stated features, integers, steps or components referred to, but not to preclude the presence or addition of one or more other feature, integer, step, component or group thereof.





The claims defining the invention are as follows:-

- 1. A polyisocyanate curable graft copolymer emulsion having a weight average molecular weight of 3,000 to 65,000 which is anionically stabilized and which comprises:
  - i) 20 to 95 weight percent of a polymer backbone of polymerized unsaturated compounds;
  - 5 to 80 weight percent of a macromonomer comprising at least about 5 weight percent of an unsaturated acid-functional monomer, said macromonomer prepared using a cobalt-based catalytic chain transfer agent and said macromonomer having a weight average molecular weight of about 1,000 to 30,000;

in which the backbone comprises 0.5 to 30 weight percent, of total polymerizable unsaturated compounds, of an amino-functional polymerizable component.

- 2. A two-package waterborne coating composition comprising as binder components:
  - a) 20 to 95 weight percent solids of a polyisocyanate curable graft
     copolymer emulsion according to claim 1;
  - b) 5 to 80 weight percent solids of a polyisocyanate curing agent; and
  - c) 0 to 75 weight percent of other film-forming polymers.
- 3. The waterborne coating composition of claim 2, wherein said amino functioned component is a secondary amino functional monomer.
- 4. The waterborne coating composition of claim 3, wherein said secondary amino functional monomer is t-butylaminoethyl methacrylate.
- The waterborne coating composition of claim 2, wherein said catalytic
   chain transfer agent is selected from the group consisting of borodifluoro-stabilized chelates of a dialkyl- or diaryl-glyoxime of Coll or Coll.



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6. The waterborne coating composition of claim 2, wherein said graft copolymer comprises 60 to 95 weight percent of polymer backbone and 5 to 40 weight percent of macromonomer wherein:

the polymer backbone comprises 5 to 50 weight percent of a vinyl aromatic, 10 to 70 weight percent of a compound selected from the group consisting of alkyl acrylate, cycloalkyl acrylate, and mixtures thereof, 5 to 40 weight percent of a compound selected from the group consisting of a hydroxy-functional polymerizable acrylate, hydroxy-functional polymerizable methacrylate, and mixtures thereof and 1 to 8 weight percent of a secondary amino-functional monomer; and

the macromonomer comprises 10 to 90 weight percent of a compound selected from the group consisting of alkyl methacrylate, cycloalkyl methacrylate, and mixtures thereof, 0 to 40 weight percent of a hydroxy-functional methacrylate, 0 to 20 weight percent of a secondary amino functional methacrylate, and 5 to 80 weight percent of an unsaturated acid-functional monomer, wherein said macromonomer has a weight average molecular weight of 1,500 to 10,000.

- 7. The waterborne coating composition of claim 2, wherein said polyisocyanate curing agent is selected from the group consisting of derivatives of hexamethylene diisocyanate, derivatives of isophorone diisocyanate, derivatives of m-xylylene diisocyante and mixtures thereof; the composition further comprising an oligomeric methacrylate having a degree of polymerization of 2 to 10, wherein said oligomeric methacrylate is added in an amount to control the weight average molecular weight of the graft copolymer to about 3,000 to 80,000.
- 8. The waterborne composition of claim 2, wherein said unsaturated acidfunctional monomer is methacrylic acid and the graft copolymer is neutralized with an amine.
- 9. The waterborne coating composition of claim 2 further comprising 0 to 75 weight percent solids of a compound selected from the group consisting of auxiliary binder and crosslinker, additives, pigments, and extenders; and 0 to 75 weight percent of a hydrophobic polyisocyanate; and 25 to 100 weight percent of a hydrophilic polyisocyanate.

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- 10. A polyisocyanate curable graft copolymer emulsion having a weight average molecular weight of 3,000 to 65,000 which is anionically stabilized and which comprises:
  - i) 20 to 95 weight percent of a polymer backbone of polymerized unsaturated compounds;
  - ii) 5 to 80 weight percent of a macromonomer comprising at least about 5 weight percent of an unsaturated acid-functional monomer, said macromonomer prepared using a cobalt-based catalytic chain transfer agent and said macromonomer having a weight average molecular weight of about 1,000 to 30,000;

in which the backbone comprises 0.5 to 30 weight percent, of total polymerizable unsaturated compounds, of an amino-functional polymerizable component substantially as herein described with reference to at least one of the accompanying Examples and/or Procedures, but excluding comparative Examples and/or Procedures.

- 11. A two-package waterborne coating composition comprising as binder components:
  - a) 20 to 95 weight percent solids of a polyisocyanate curable graft copolymer emulsion according to claim 1;
  - b) 5 to 80 weight percent solids of a polyisocyanate curing agent; and
- c) 0 to 75 weight percent of other film-forming polymers substantially as herein described with reference to at least one of the accompanying Examples and/or Procedures, but excluding comparative Examples and/or Procedures.

DATED this 31st day of May, 2001.

#### E.I. DU PONT DE NEMOURS AND COMPANY

30 By their Patent Attorneys: CALLINAN LAWRIE

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