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# DESCRIPTION

## Technical Field

**[0001]** The present invention relates to process for immobilizing a lipase on a support having a functional amino group, a process for producing a triglyceride composition using said immobilized lipase and to the use of the lipase in transesterification reactions.

## Background Art

**[0002]** Lipases (E.C. 3.1.1.3), belonging to the group of enzymes, catalyse specifically ester bonds in tri-, di-, and mono-acylglycerols to glycerol and fatty acids. They further catalyse other reactions such as interesterifications, esterifications, acidolysis, alcoholysis and aminolysis. The high costs of lipases make enzymatic processes economically unattractive. Immobilization of the lipases is a way to increase the industrial susceptibility of lipases and allows recovery of the lipase protein. Lipases can be immobilized on different supports applying various ways of pretreatment of the support or the lipase.

**[0003]** Nevena et al. (NEVENA, Z. Immobilization of lipase from *Candida rugosa* on Sepabeads: the effect of lipase oxidation by periodates. *Bioprocess Biosyst Eng.* 2011, no.34, p.803-810.) describes the use of certain Sepabeads® having either amino functional groups or epoxy groups as suitable support for the immobilization of a non-specific lipase from *Candida rugosa*. Sepabeads® having amino functional groups needed activation with glutaraldehyde or sodium-periodate to show improved activity.

**[0004]** Palomo et al. (PALOMO, Jose M, et al. Interfacial adsorption of lipases on very hydrophobic support (octadecyl-Sepabeads): immobilization, hyperactivation and stabilization of the open form of lipases. *Journal of Molecular Catalysis B: Enzymatic.* 2002, vol.19, no.20, p.279-286.) tested the immobilization of various lipases on very hydrophobic support such as octadecyl-Sepabead®.

**[0005]** WO94/28118 describes a process for the immobilization of a lipase on hydrophobic support material (Accurel) using an aqueous solution of lipase enzyme and a non-ionic surfactant such as Tween 20.

**[0006]** Enzyme activity is vulnerable to immobilizations reagents such as glutaraldehyde or immobilization support. To secure enzyme stability and activity after immobilization of the enzyme, often non-lipase proteins are added such as hen egg album or bovin serum albumin. However, these animal proteins are known to cause allergic reactions.

**[0007]** There remains a need for a simplified immobilization method without additional

activation of the support and the right choice of support which will retain lipase activity and stability such as thermo stability to enable the production of commercially relevant triglyceride compositions.

**Summary of invention**

**[0008]** The objective of the present invention is to provide an immobilization process wherein pre-activation of the support could be avoided and if indeed a hydrophobic support such as Sepabeads® having octadecyl groups (EC-OD) are able to perform transesterification reaction to obtain products of commercial importance. Another aim of the present invention was to provide an immobilization process avoiding treatment with non-lipase protein such as animal derived albumin to secure activity and stability of the lipase to produce triglyceride compositions.

**[0009]** According to the present invention, there is provided a process for immobilizing a lipase on a support containing a functional amino group in the presence of a surface-active material in an aqueous solution, wherein the surface-active material is a non-ionic surfactant. The term functional amino group refers to an amino group which is engaged in interacting with or binding to the lipase and optionally, the support.

**[0010]** Further disclosed herein is a process for producing a triglyceride by enzymatic transesterification by using a lipase, which is immobilized on a support having a functional amino group.

**[0011]** Also disclosed herein is the use of the immobilized lipase for producing a triglyceride fat composition comprising at least 15% by weight OPO.

**[0012]** The support having a functional amino group can be any support having an amino group such as amino-epoxy, or alkyl amino having a carbon chain of C1-C24, preferably C2-C10. The support comprises a methacrylic polymer. Preferably the polymer forms a matrix.

**[0013]** A preferred support of the present invention contains a functional alkylamino group such as ethyl amino or hexyl amino.

**[0014]** The mechanism of action between the support and the lipase is either by ionic interaction or chemical binding, wherein the ionic interaction is preferred.

**[0015]** The surfactant can be formed from sugars, (both mono-di-and polysaccharides), polyols (e.g. sorbitan and sorbitol) or polyethylene glycols having molecular weight from 350 to 35000, such as PEG s 600, 1500, 4000. Very suitable non-ionic surfactants are polyoxyethylene sorbitan C8-C24 fatty acid esters, in particular those derived from lauric acid, such as Tween 20® or derived from oleic acid such as Tween 80®.

**[0016]** The surfactant concentration in the aqueous solution should be sufficient to ensure effective loading of the support by the enzyme. Very good results were obtained by applying an aqueous solution with a surfactant concentration of at least 0.01 wt%, preferably 0.01-10, most preferably 0.1-5wt.%.

**[0017]** An ideal amount of lipase in g to support in g is between 1-20 wt.% by weight, preferably 5-15% by weight.

**[0018]** The contact times applied can vary between wide ranges. Suitably, however, contact times between 1 and 72 hours are applied.

**[0019]** The aqueous lipase solution has preferable a concentration between 1 to 20 g/l.

**[0020]** Although the lipase enzyme can be any prior art lipase, a preference is expressed for a lipase which is selected from 1) 1,3-specific lipases from *Rhizomucor miehei*, *Rhizopus oryzae* and *Thermomyces lanuginosus* 2) lipases from *Penicillium camembertii* specific for the hydrolysis of partial glycerides, such as Amano G, and 3) lipases specific for the hydrolysis of esters or triglycerides, preferably a lipase from *Candida rugosa*. In particular preferred is a 1,3-specific lipase from *Rhizopus oryzae* such as Lipase D from Amano.

**[0021]** Immobilization of the lipase can be performed in many different ways. Suitably, the contact between support, lipase and/or surfactant is performed as a batch process, as a continuous process in a fixed bed, as a continuous process in a fluidized bed or in a continuously stirred tank, while the contacting is performed with a continuous motion of the lipase solution.

**[0022]** The immobilized lipase according to the invention can be applied in any enzymatic conversion process, such as hydrolysis of triglycerides, diglycerides or esters, but also the esterification or transesterification of fatty acids or diglycerides or triglycerides. These processes are also disclosed herein, with the prerequisite that an immobilized lipase according to our invention be used in the process.

**[0023]** Preferred processes for making triglyceride is the production of triglycerides compositions comprising symmetrical triglycerides of the general formula ABA, such as OPO or SOS, wherein O is oleic acid, P is palmitic acid and S is a saturated fatty acid selected from palmitic acid and stearic acid. A particular preferred triglyceride composition of the invention comprises at least 15% by weight OPO.

**[0024]** Triglyceride fats and oils are important commercial products and are used extensively in, for example, the food industry. Some triglycerides are nutritionally important and the triglyceride 1,3-dioleoyl-2-palmitoyl glyceride (OPO) is known to be an important component of human milk fat.

## **Examples**

**[0025]** The following non-limiting examples illustrate the invention and do not limit its scope in any way. In the examples and throughout this specification, all percentages, parts and ratios are by weight unless indicated otherwise.

**Example 1 : Various Sepabeads® with aqueous Lipase D preparation**

**[0026]** Preparation of the lipase solutions: Seven lipase solutions were prepared according to Table 1. Sample N°7 was the control sample. All reagents were mixed at 150 rpm at room temperature between 3 to 24 hours and then centrifuged to receive the immobilized lipase as a pellet.

**Table 1**

Sample, N°	Sepabeads® (functional group)	Amount of Sepabeads in g	Lipase in g
1	EC-HA (Hexylamino)	1.5	0.12 in 70 ml
2	EC-OD (Octadecyl)	1.5	0.12 in 70 ml
3	EC-BU (Butyl)	1.5	0.12 in 70 ml
4	EC-HFA (Amino-Epoxy)	1.5	0.12 in 70 ml
5	EC-EA (Ethylamino)	1.5	0.12 in 70 ml
6	EC-EP (Epoxy)	1.5	0.12 in 70 ml
7	No support	0	0.18 in 75 ml (equal to 0.12 in 70 ml)

**Example 2 (comparative)**

**[0027]** The acidolysis reaction was performed at 60 °C with all seven lipase preparations using the following acidolysis assay:

1 g Immobilized enzyme (use the pellet after centrifugation)

35 g Palm oil stearin fraction (Feedstock)

49 g Oleic acid

0.126 g H<sub>2</sub>O

**[0028]** Composition Feedstock to be found in Table 2.

Table 2

	Feedstock
Carbon number	
C48	62.1
C50	24.3
C52	8.6
C54	1.9
C56	0.0

[0029] The carbon number was determined by GC according to AOCS Ce 5.86.

[0030] Table 3 provides the results of the various acidolysis reactions of feedstock after 24 h.

Table 3

	HA	OD	BU	HFA	EA	EP	Control
Carbon number							
C48	59.9	60.1	60.0	57.4	60.3	60.1	60.3
C50	25.4	25.3	25.3	26.8	25.1	25.3	25.2
C52	9.2	9.2	9.2	10.2	9.1	9.2	9.1
C54	2.1	2.1	2.2	2.5	2.1	2.1	2.1
C56	0.2	0.1	0.2	0.2	0.2	0.2	0.2

[0031] After 24 h nearly no product OPO or OOP (C52) was produced for all lipase preparations

### Example 3

[0032] 70 ml of the lipase preparation of Example 1 was mixed with 2.4 g hen egg albumin, 0.65 g Tween 20® and 1.5 g of the respective supports. The acidolysis reaction was performed according to example 2.

[0033] Table 4 shows the results after acidolysis (24 hours) by using various sepabeads with aqueous lipase D solution in the presence of hen egg albumin and TWEEN 20.

Table 4

	HA	OD	BU	HFA	EA	EP
Carbon number						
C48	7.2	9.5	61.2	6.9	7.2	61.4

	HA	OD	BU	HFA	EA	EP
Carbon number						
C50	29.3	31.3	24.8	29.0	25.0	24.7
C52	41.5	41.5	8.8	42.8	42.3	8.8
C54	20.4	16.9	2.0	20.4	20.7	1.9
C56	0.5	0.4	0.0	0.5	0.4	0.0

#### Example 4

**[0034]** 70 ml of the lipase preparation of Example 1 was mixed with 250 mg PEG 1500, 0.65 g Tween 20® and 1.5 g of the respective supports. The acidolysis reaction was performed according to example 2. As comparison immobilization on polypropylene (Accurel) under same reaction conditions was performed.

**[0035]** Table 5 shows the results after acidolysis (24 hours) by using various sepabeads with aqueous lipase D solution in the presence of PEG 1500 and Tween 20®.

**Table 5**

	HA	OD	HFA	EA	Accurel
Carbon number					
C48	6.7	61.9	6.8	6.8	58.2
C50	27.3	24.4	27.6	27.3	25.2
C52	42.4	8.5	42.5	42.4	10.0
C54	22.5	1.9	22.0	22.4	2.3
C56	0.2	0.0	0.5	0.4	4.1

#### Example 5

**[0036]** Multiple usage of Lipase D immobilized on support EC-HA.

**[0037]** 70 ml of the lipase preparation of Example 1 was mixed with 30 mg PEG 600, 0.65 g Tween 20® and 1.5 g of support EC-HA. The acidolysis reaction was performed according to example 2. After 3.5 hours the acidolysis reaction was stopped and the immobilized lipase separated from the reaction mixture by filtration. The immobilized lipase is collected and used for the second run of the acidolysis assay. These runs were repeated eight times. At each run a sample (~ 2 ml) at time 3.5 hours were taken for carbon number analysis.



[0038] Table 6 shows the results after acidolysis by reusing the immobilized lipase D on EC-HA support in subsequent 8 runs

**Table 6**

	Run 1	Run 2	Run 3	Run 4	Run 5	Run 6	Run 7	Run 8
C46	0.8	1.2	1.3	1.5	1.4	1.5	1.5	1.6
C48	11.2	19.6	22.7	25.3	24.7	26	27.7	28.5
C50	34.1	35.9	36.4	35.9	35.8	35.6	35.3	35.2
C52	42.4	34.2	31.3	29.4	30.1	29.1	28	27.4
C54	11.2	9.1	8.2	7.8	8	7.8	7.6	7.3

### Example 6

[0039] Lipase D solution (0.9 g/77 ml) was mixed with various Tween in amounts provided in Table 7 and stirred for 15 min. To each of the preparations 1,5 g of Sepabead EC-HA was added and the mixture was stirred for 24 hours. Then immobilized enzyme was filtered off and tested in the acidolysis reaction as described in example 2. Table 7 shows the results of 5 different Tween's after acidolysis after 3.5 h.

**Table 7**

Carbonnumber	Tween 20	Tween 40	Tween 60	Tween 80	Tween 85
Amount in g	0.650	0.676	0.693	0.694	0.974
C46	1.5	2.14	2.79	1.34	2.4
C48	16.6	41.19	54.37	18.02	46.5
C50	34.2	31.4	27.49	33.26	30.5
C52	36.9	18.73	11.85	36.11	15.8
C54	10.6	5.78	3.48	10.77	4.5
C56	0.3	0.53	0	0.38	0.3

### Example 7

[0040] Example 6 was repeated with Tween 80® with the difference that the premixing of the lipase solution with Tween 80® was skipped. Lipase solution, Tween 80® and support material were put together and the mixture was stirred for 24 hours. Then immobilized lipase was filtered off and tested in the acidolysis reaction as described in example 2. Table 8 shows the results after acidolysis after 3.5 h.

**Table 8**

Carbonnumber	With premixing	No premixing
C46	1.3	1.51
C48	13.8	16.38
C50	33	32.32
C52	40.2	38.34
C54	11.4	11.14
C56	0.3	0.3

[0041] The results demonstrate that premixing of Tween 80® with lipase solution is not required.

## REFERENCES CITED IN THE DESCRIPTION

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### Patent documents cited in the description

- [WO9428118A](#) [0005]

### Non-patent literature cited in the description

- **NEVENA, Z.** Immobilization of lipase from *Candida rugosa* on Sepabeads: the effect of lipase oxidation by periodates *Bioprocess Biosyst Eng.*, 2011, vol. 34, 803-810 [0003]
- **PALOMO, JOSE M et al.** Interfacial adsorption of lipases on very hydrophobic support (octadecyl-Sepabeads): immobilization, hyperactivation and stabilization of the open form of lipases *Journal of Molecular Catalysis B: Enzymatic*, 2002, vol. 19, 20279-286 [0004]

**Patentkrav**

- 5           **1.** Fremgangsmåde til immobilisering af en lipase på en bærer, der indeholder en funktionel aminogruppe, som omfatter at bringe lipasen i kontakt med bæreren i nærvær af et overfladeaktivt materiale i en vandig opløsning, hvor det overfladeaktive materiale er et ikke-ionisk overfladeaktivt stof.
- 2.** Fremgangsmåde ifølge krav 1, hvor lipasen er en 1,3 specifik lipase.
- 10           **3.** Fremgangsmåde ifølge krav 1 eller 2, hvor lipasen er afledt af *Rhizopus oryzae*.
- 4.** Fremgangsmåde ifølge et af de foregående krav, hvor den funktionelle aminogruppe er en funktionel alkylaminogruppe, som har C1- C22 carbonatomer.
- 15           **5.** Fremgangsmåde ifølge et af de foregående krav, hvor det overfladeaktive materiale er udvalgt fra gruppen bestående af polyethyleneglycoler, methoxy-polyethyleneglycoler, polysorbater og blandinger deraf.
- 20           **6.** Fremgangsmåde ifølge et af de foregående krav, hvor den vandige opløsning har en koncentration af overfladeaktivt materiale på mindst 0,01 vægt-%.
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