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(72) Inventors; and

(71) Applicants: **BRAMBILLA, Nicolo** [IT/US]; 1454 Beacon St., Unit 542, Brookline, MA 02215 (US). **CAO, Wanjun,**

Ben [CN/US]; 96 Bennington Street, Apt #2, Boston, MA 02218 (US). **YAN, Jin** [CN/US]; 411 D Street Unit 312, Boston, MA 02210 (US). **CHEN, Xujie** [CN/US]; 545 Hancock Street, Apt 1, Quincy, MA 02170 (US). **DU, Ting** [CN/US]; 10 Faxon Avenue, Apt. 711, Quincy, MA 02169 (US). **PARK, Kitae** [US/US]; 128 Independence Drive, Chestnut Hill, MA 02467 (US).

(74) Agent: **CANTOR, Michael, A.**; Cantor Colburn LLP, 20 Church Street, Hartford, CT 06103 (US).

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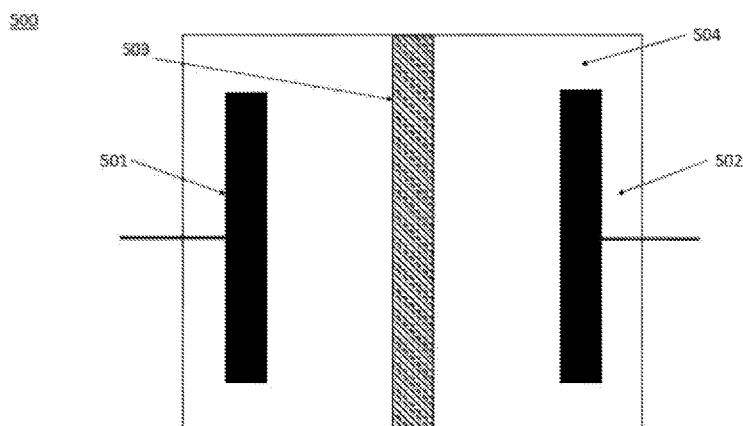


Figure 5

(57) Abstract: An electrode active layer is disclosed that includes a network of high aspect ratio carbon elements (e.g., carbon nanotubes, carbon nanotube bundles, graphene flakes, or the like) that provides a highly electrically conductive scaffold that entangles or enmeshes the active material, thereby supporting the layer. A surface treatment can be applied to the high aspect ratio carbon elements to promote adhesion to the active material and any underlying electrode layers improving the overall cohesion and mechanical stability of the active layer. This surface treatment forms only a thin (in some cases even monomolecular) layer on the network, leaving the large void spaces that are free of any bulk binder material and so may instead be filled with active material. The resulting active layer may be formed with excellent mechanical stability even at large thickness and high active material mass loading.



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ELECTRODES FOR ENERGY STORAGE DEVICES

Related Applications

The present application claims the benefit of and priority to each of U.S. Provisional Pat. App. Serial No. 63/041801 filed June 19, 2020 ; U.S. Provisional Pat. App. Serial No. 62/954771 filed December 30, 2019; U.S. Provisional Pat. App. Serial No. 62/871041 filed July 5, 2019; and U.S. Provisional Pat. App. Serial No. 63/003341 filed April 1, 2020. the entire contents of each of the foregoing references are incorporated herein by reference.

Background

Lithium batteries are used in many products including medical devices, electric cars, airplanes, and consumer products such as laptop computers, cell phones, and cameras. Due to their high energy densities, high operating voltages, and low-self discharges, lithium ion batteries have overtaken the secondary battery market and continue to find new uses in products and developing industries.

Generally, lithium ion batteries ("LIBs" or "LiBs") comprise an anode, a cathode, and an electrolyte material such as an organic solvent containing a lithium salt. More specifically, the anode and cathode (collectively, "electrodes") are formed by mixing either an anode active material or a cathode active material with a binder and a solvent to form a paste or slurry which is then coated and dried on a current collector, such as aluminum or copper, to form a film on the current collector. The anodes and cathodes are then layered or coiled prior to being housed in a pressurized casing containing an electrolyte material, which all together forms a lithium ion battery.

In conventional electrodes binder is used with sufficient adhesive and chemical properties such that the film coated on the current collector will maintain contact with the current collector even when manipulated to fit into the pressurized battery casing. Since the film contains the electrode active material, there will likely be significant interference with the electrochemical properties of the battery if the film does not maintain sufficient contact with the current collector. Further, it has been important to select a binder that is mechanically compatible with the electrode active material(s) such that it is capable of withstanding the degree of expansion and contraction of the electrode active material(s) during charging and discharging of the battery.

Accordingly, binders such as cellulosic binder or cross-linked polymeric binders have been used to provide good mechanical properties. However, such binder materials have disadvantageous effects. For example, the bulk of the binder fills volume in the electrode active layer which otherwise could be used to increase the mass loading of active material and decrease the electrical conductivity of the electrode.

Moreover, binders tend to react electrochemically with the electrolyte used in the cell (especially in high voltage, high current, and/or high temperature applications), resulting in degradation of the performance of the cell.

Summary

5 The applicants have realized that an electrode may be constructed to exhibit excellent mechanical stability without the need for bulk polymer binders. In one aspect, the present disclosure describes embodiments of an electrode active layer that includes a network of high aspect ratio carbon elements (e.g., carbon nanotubes, carbon nanotube bundles, graphene flakes, or the like) that provides a highly electrically conductive scaffold that entangles or enmeshes the active material, thereby supporting the layer. As
10 detailed below, a surface treatment can be applied to the high aspect ratio carbon elements to promote adhesion to the active material and any underlying electrode layers (e.g., a current collector layer) improving the overall cohesion and mechanical stability of the active layer. This surface treatment forms only a thin (in some cases even monomolecular) layer on the network, leaving the large void spaces that are free of any bulk binder material and so may instead be filled with active material. The resulting active
15 layer may be formed with excellent mechanical stability even at large thickness and high active material mass loading.

In another aspect, the present disclosure describes a method including dispersing high aspect ratio carbon elements and a surface treatment material in a solvent to form an initial slurry, wherein said dispersion step results in the formation of a surface treatment on the high aspect ratio carbon; mixing active
20 materials into the first slurry to form a final slurry; coating the final slurry onto a substrate; and drying the final slurry to form an electrode active layer.

Various embodiments may include any of the features or elements described herein, individually or in any suitable combination.

Brief Description of the Drawings

25 Figure 1 is a schematic of an electrode featuring an active material layer.

Figure 2 is a detailed illustration of an embodiment of an active material layer.

Figure 3 is a detailed illustration of another embodiment active material layer.

Figure 4 is an electron micrograph of an active material of the type described herein.

Figure 5 is a schematic of an energy storage cell.

Figure 6 is a flow chart illustrating a method of making the electrode of Figure 1.

Figure 7 shows a schematic of a pouch cell battery

Figure 8 shows a summary of functional parameters for a pouch cell battery for EV applications.

Figure 9 shows a summary of functional parameters for a pouch cell battery.

- 5 Figure 10 shows the results of a comparative performance evaluation of a pouch cell battery featuring a binder free cathode (left plot) and a pouch cell battery featuring a binder based cathode (right plot).

Figure 11 shows the results of a comparative performance evaluation of a pouch cell battery featuring a binder free cathode (upper trace) and a pouch cell battery featuring a binder based cathode (lower trace).

Figure 12 is a schematic of a half cell lithium battery apparatus.

- 10 Figure 13 is a plot showing potential (referenced to the Li/Li⁺ potential) vs specific capacity for binder free cathode half cell (solid traces) and reference binder based cathode half cell (dashed traces) at various current densities

Figure 14 is a plot showing potential (referenced to the Li/Li⁺ potential) vs volumetric capacity for binder free cathode have cell (solid traces) and reference binder based cathode half cell (dashed traces) at various
15 current densities

Figure 15 shows a plot of volumetric capacity vs current density for binder free cathode half cells (upper trace) and reference binder based cathode half cell (lower trace).

- Figure 16 shows a Nyquist plot resulting from electrochemical impedance spectroscopy for several binder free cathode half cells (square, circle and triangle labeled traces) and a reference binder based cathode
20 half cells. The binder free cathode half cells exhibit significantly better performance than the reference cell.

Detailed Description

- Referring to Figure 1, an electrode 10 is shown which includes an active layer 100 disposed on a current collector 101. Some embodiment may include an optional adhesion layer 102 disposed between the active
25 layer 101 and the current collector 102. In other embodiments, the adhesion layer 102 may be omitted.

The current collector 101 may be an electrically conductive layer, such as a metal foil. The optional adhesion layer 102 (which may be omitted in some embodiments) may be a layer of material that promotes adhesion between the current collector 102 and the active layer 100. Examples of suitable

materials for the current collector 101 and the optional adhesion layer 102 are described in International Patent Publication No. WO/2018/102652 published June 7, 2018.

Electrode Active layer

5 In some embodiments, the active layer 100 may include a three-dimensional network 200 of high aspect ratio carbon elements 201 defining void spaces within the network 200. A plurality of active material particles 300 are disposed in the void spaces within the network 200. Accordingly, the active material particles are enmeshed or entangled in the network 200, thereby improving the cohesion of the active layer 100.

10 In some embodiments, a surface treatment 202 (not shown, refer to Figure 2) is applied on the surface of the high aspect ratio carbon elements 201 of the network 200. The surface treatment promotes adhesion between the high aspect ratio carbon elements and the active material particles 300. The surface treatment may also promote adhesion between the high aspect ratio carbon elements and the current collector 100 (also referred to herein as a “conductive layer”) and/or the optional adhesion layer 102.

15 As used herein, the term “high aspect ratio carbon elements” refers to carbonaceous elements having a size in one or more dimensions (the “major dimension(s)”) significantly larger than the size of the element in a transverse dimension (the “minor dimension”).

For example, in some embodiments the high aspect ratio carbon elements 201 may include flake or plate shaped elements having two major dimensions and one minor dimension. For example, in some such embodiments, the ratio of the length of each of the major dimensions may be at least 5 times, 10 times, 20 100 times, 500 times, 1,000 times, 5,000 times, 10,000 times or more of that of the minor dimension. Exemplary elements of this type include graphene sheets or flakes.

For example, in some embodiments the high aspect ratio carbon elements 201 may include elongated rod or fiber shaped elements having one major dimension and two minor dimensions. For example, in some such embodiments, the ratio of the length of the major dimensions may be at least 5 times, 10 times, 25 100 times, 500 times, 1,000 times, 5,000 times, 10,000 times or more of that of each of the minor dimensions. Exemplary elements of this type include carbon nanotubes, bundles of carbon nanotubes, carbon nanorods, and carbon fibers.

In some embodiments, the high aspect ratio carbon elements 201 may include single wall nanotubes (SWNT), double wall nanotubes (DWNT), or multiwall nanotubes (MWNT), carbon nanorods, carbon 30 fibers or mixtures thereof. In some embodiments, the high aspect ratio carbon elements 201 may be formed of interconnected bundles, clusters, or aggregates of CNTs or other high aspect ratio carbon

materials. In some embodiments, the high aspect ratio carbon elements 201 may include graphene in sheet, flake, or curved flake form, and/or formed into high aspect ratio cones, rods, and the like.

In some embodiments, the electrode active layer 100 may contain little or no bulk binder material, leaving more space in the network 200 to be occupied by active material particles 300. For example, in some
5 embodiments, the active layer 200 contains less than 10% by weight, less than 1% by weight, less than 0.1% by weight, less than 0.01% by weight, or less of binder material (e.g., polymeric or cellulosic binder material) disposed in the void spaces.

For example, in some embodiments the electrode active layer is free of or substantially free of polymeric material, or any material other than the active material 300, and the network 200 composed of the high
10 aspect ratio carbon elements 201 and the surface treatment 202 disposed thereon.

In some embodiments, the network 200 is composed largely or even entirely of carbon. For example, in some embodiments the network 200 is at least 90% carbon by weight, at least 95% carbon by weight, at least 96% carbon by weight, at least 97% carbon by weight, at least 98% carbon by weight, at least 99% carbon by weight, at least 99.5% carbon by weight, at least 99.9% carbon by weight, or more.

15 In some embodiments, a size (e.g., the average size, median size, or minimum size) of the high aspect ratio carbon elements 201 forming the network 200 along one or two major dimensions may be at least 0.1 μm , 0.5 μm , 1 μm , 5 μm , 10 μm , 50 μm , 100 μm , 200 μm , 300 μm , 400 μm , 500 μm , 600 μm , 7000 μm , 800 μm , 900 μm , 1,000 μm or more. For example, in some embodiments, the size (e.g., the average size, median size, or minimum size) of the elements 201 forming the network 200 may be in the range of
20 1 μm to 1,000 μm , or any subrange thereof, such as 1 μm to 600 μm .

In some embodiments, the size of the elements can be relatively uniform. For example, in some embodiments, more than 50%, 60%, 70%, 80%, 90%, 95%, 99% or more of the elements 201 may have a size along one or two major dimensions within 10% of the average size for the elements 201 making up the network 200.

25 Applicants have found that an active layer 100 of the type herein can provide exemplary performance (e.g., high conductivity, low resistance, high voltage performance, and high energy and power density) even when the mass fraction of elements 201 making up the network 200 in the layer 100 is quite low, thereby allowing high mass loading of active material particles 300. For example, in some embodiments, the active layer 100 may be at least about 50 wt % (percent by weight), 60 wt %, 70 wt %, 75 wt %, 80 wt
30 %, 85 wt %, 90 wt %, 95 wt %, 96 wt %, 97 wt %, 98 wt %, 99 wt %, 99.5 wt %, or more of active material particles 300.

In some embodiments, the network 200 forms an interconnected network of highly electrically conductive paths for current flow (e.g. electron or ion transport) through the active layer 100. For example, in some embodiments, highly conductive junctions may occur at points where the elements 201 of the network intersect with each other, or where they are in close enough proximity to allow for quantum tunneling of charge carriers (e.g., electrons or ions) from one element to the next. While the elements 201 may make up a relatively low mass fraction of the active layer (e.g., less than 10 wt %, 5 wt %, 4 wt %, 3 wt %, 2 wt %, 1 wt % or less, e.g., in the range of 0.5 wt % to 10 wt % or any subrange thereof such as 1 wt % to 5.0 wt %), the interconnected network of highly electrically conductive paths formed in the network 200 may provide long conductive paths to facilitate current flow within and through the active layer 100 (e.g. conductive paths on the order of the thickness of the active layer 100).

For example, in some embodiments, the network 200 may include one or more structures of interconnected elements 201, where the structure has an overall length along one or more dimensions longer than 2, 3, 4, 5, 10, 20, 50, 100, 500, 1,000, 10,000 or more times the average length of the component elements 201 making up the structure. For example, in some embodiments, network 200 may include one or more structures of interconnected elements 200, where the structure has an overall length in the range of 2 to 10,000 (or any subrange thereof) times the average length of the component elements 201 making up the structure. For example, in some embodiments the network 200 may include highly conductive pathways having a length greater than 100 μm , 500 μm , 1,000 μm , 10,000 μm or more, e.g., in the range of 100 μm – 10,000 μm of any subrange thereof.

As used herein, the term “highly conductive pathway” is to be understood as a pathway formed by interconnected elements 201 having an electrical conductivity higher than the electrical conductivity of the active material particles enmeshed in the network 200.

Not wishing to be bound by theory, in some embodiments the network 200 can be characterized as an electrically interconnected network of elements 201 exhibiting connectivity above a percolation threshold. Percolation threshold is a mathematical concept related to percolation theory, which is the formation of long-range connectivity in random systems. Below the threshold a so called “giant” connected component of the order of system size does not exist; while above it, there exists a giant component of the order of system size.

In some embodiments, the percolation threshold can be determined by increasing the mass fraction of elements 201 in the active layer 100 while measuring the conductivity of the layer, holding all other properties of the layer constant. In some such cases, the threshold can be identified with the mass fraction at which the conductivity of the layer sharply increases and/or the mass fraction above which the

conductivity of the layer increases only slowly with increases with the addition of more elements 201. Such behavior is indicative of crossing the threshold required for the formation of interconnected structures that provide conductive pathways with a length on the order of the size of the active layer 100.

Figure 2 shows a detailed view of high aspect ratio carbon element 201 of the network 200 (as shown in Figure 1), located near several active material particles 300. In the embodiment shown, the surface treatment 202 on the element 201 is a surfactant layer bonded to the outer layer of the surface of the element 201. As shown, the surfactant layer comprises a plurality of surfactant elements 210 each having a hydrophobic end 211 and a hydrophilic end 212, wherein the hydrophobic end is disposed proximal the surface of the carbon element 201 and the hydrophilic end 212 is disposed distal the surface.

In some embodiments where the carbon element 201 is hydrophobic (as is typically the case with nanoform carbon elements such as CNTs, CNT bundles, and graphene flakes), the hydrophobic end 211 of the surfactant element 210 will be attracted to the carbon element 201. Accordingly, in some embodiments, the surface treatment 202 may be a self-assembling layer. For example, as detailed below, in some embodiments, when the elements 201 are mixed in a solvent with a surfactant elements 210 to form a slurry, the surface treatment 202 layer self assembles on the surface due to electrostatic interactions between the elements 201 and 210 within the slurry.

In some embodiments, the surface treatment 202 may be a self-limiting layer. For example, as detailed below, in some embodiments, when the elements 201 are mixed in a solvent with a surfactant elements 210 to form a slurry, the surface treatment 202 layer self assembles on the surface due to electrostatic interactions between the elements 201 and 210 within the slurry. In some such embodiments, once an area of the surface of the element 201 is covered in surfactant elements 210, additional surfactant elements 210 will not be attracted to that area. In some embodiments, once the surface of the element 201 is covered with surfactant elements 202, further elements are repulsed from the layer, resulting in a self-limiting process. For example, in some embodiments the surface treatment 202 may form in a self-limiting process, thereby ensuring that the layer will be thin, e.g., a single molecule or a few molecules thick.

In some embodiments, the hydrophilic ends 212 of at least a portion of the surfactant elements form bonds with the active material particles 300. Accordingly, the surface treatment 202 can provide good adhesion between the elements 201 of the network 200 and the active material particles. In some embodiments, the bonds may be covalent bonds, or non-covalent bonds such as π - π bonds, hydrogen bonds, electrostatic bonds or combinations thereof.

For example, in some embodiments, the hydrophilic end 212 of the surfactant element 210 has a polar charge of a first polarity; while the surface of the active material particles 300 carry a polar charge of a second polarity opposite that of the first polarity, and so are attracted to each other.

For example, in some embodiments where, during formation of the layer 100, the active material particles 300 are combined in a solvent with carbon elements 201 bearing the surface treatment 202 (as described in greater detail below), the outer surface of the active material particles 300 may be characterized by a Zeta potential (as is known in the art) having the opposite sign of the Zeta potential of the outer surface of the surface treatment 202. Accordingly, in some such embodiments, attractions between the carbon elements 201 bearing the surface treatment 202 and the active material products 300 promote the self-assembly of a structure in which the active material particles 300 are enmeshed with the carbon elements 201 of the network 200.

In some embodiments the hydrophilic ends 212 of at least a portion of the surfactant elements form bonds with a current collector layer or adhesion layer underlying the active material layer 100. Accordingly, the surface treatment 202 can provide good adhesion between the elements 201 of the network 200 and such underlying layer. In some embodiments, the bonds may be covalent bonds, or non-covalent bonds such as π - π bonds, hydrogen bonds, electrostatic bonds or combinations thereof. In some embodiments, this arrangement provides for excellent mechanical stability of the electrode 10, as discussed in greater detail below.

In various embodiments, the surfactant used to form the surface treatment 202 as described above may include any suitable material. For example, in some embodiments the surfactant may include one or more of the following: hexadecyltrimethylammonium hexafluorophosphate (CTAP), hexadecyltrimethylammonium tetrafluoroborate (CTAB), hexadecyltrimethylammonium acetate, hexadecyltrimethylammonium nitrate, cocamidopropyl betaine, N-(cocoalkyl)-N,N,N-trimethylammonium methyl sulfate, and cocamidopropyl betaine. Additional suitable materials are described below.

In some embodiments, the surfactant layer 202 may be formed by dissolving a compound in a solvent, such that the layer of surfactant is formed from ions from the compound (e.g., in a self-limiting process as described above). In some such embodiments, the active layer 100 will then include residual counter ions 214 to the surfactant ions forming the surface treatment 202.

In some embodiments, these surfactant counter ions 214 are selected to be compatible with use in an electrochemical cell. For example, in some embodiments, the counter ions are selected to be unreactive or mildly reactive with materials used in the cell, such as an electrolyte, separator, housing, or the like.

For example, if an aluminum housing is used the counter ion may be selected to be unreactive or mildly reactive with the aluminum housing.

For example, in some embodiments, the residual counter ions are free or substantially free of halide groups. For example, in some embodiments, the residual counter ions are free or substantially free of bromine.

In some embodiments, the residual counter ions may be selected to be compatible with an electrolyte used in an energy storage cell containing the active layer 200. For example, in some embodiments, residual counter ions may be the same species of ions used in the electrolyte itself. For example, if the electrolyte includes a dissolved Li PF_6 salt, the electrolyte anion is PF_6^- . In such a case, the surfactant may be selected as, for example, CTA PF_6 , such that the surface treatment 202 is formed as a layer of anions from the CTA PF_6 , while the residual surfactant counter ions are the PF_6^- anions from the CTA PF_6 (thus matching the anions of the electrolyte).

In some embodiments, the surfactant material used may be soluble in a solvent which exhibits advantageous properties. For example, in some embodiments, the solvent may include water or an alcohol such as methanol, ethanol, or 2-propanol (isopropyl alcohol, sometimes referred to as IPA) or combinations thereof. In some embodiments, the solvent may include one or more additives used to further improve the properties of the solvent, e.g., low boiling point additives such as acetonitrile (ACN), de-ionized water, and tetrahydrofuran.

For example, if a low boiling point solvent is used in the formation of the surface treatment 202, the solvent may be quickly removed using a thermal drying process (e.g., of the type described in greater detail below) performed at a relatively low temperature. As will be understood by those skilled in the art, this can improve the speed and or cost of manufacture of the active layer 202.

For example, in some embodiments, the surface treatment 202 is formed from a material which is soluble in a solvent having a boiling point less than 250°C , 225°C , 202°C , 200°C , 185°C , 180°C , 175°C , 150°C , 125°C , or less, e.g., less than or equal to 100°C .

In some embodiments, the solvent may exhibit other advantageous properties. In some embodiments the solvent may have a low viscosity, such a viscosity at 20°C of less than or equal to 3.0 centipoise, 2.5 centipoise, 2.0 centipoise, 1.5 centipoise, or less. In some embodiments the solvent may have a low surface tension such a surface tension at 20°C of less than or equal to 40 mN/m, 35 mN/m, 30 mN/m, 25 mN/m or less. In some embodiments the solvent may have a low toxicity, e.g., toxicity comparable to alcohols such as isopropyl alcohol.

Notably, this contrasts with the process used to form conventional electrode active layers featuring bulk binder materials such as polyvinylidene fluoride or polyvinylidene difluoride (PVDF). Such bulk binders require aggressive solvents often characterized by high boiling points. One such example is n-methyl-2-pyrrolidone (NMP). Use of NMP (or other pyrrolidone based solvents) as a solvent requires the use of high temperate drying processes to remove the solvent. Moreover, NMP is expensive, requiring a complex solvent recovery system, and highly toxic, posing significant safety issues. In contrast, as further detailed below, in various embodiments the active layer 200 may be formed without the use of NMP or similar compounds such pyrrolidone compounds.

While one class of exemplary surface treatment 202 is described above, it is understood that other treatments may be used. For example, in various embodiments the surface treatment 202 may be formed by functionalizing the high aspect ratio carbon elements 201 using any suitable technique as described herein or known in the art. Functional groups applied to the elements 201 may be selected to promote adhesion between the active material particles 300 and the network 200. For example, in various embodiments the functional groups may include carboxylic groups, hydroxylic groups, amine groups, silane groups, or combinations thereof.

As will be described in greater detail below, in some embodiments, the functionalized carbon elements 201 are formed from dried (e.g., lyophilized) aqueous dispersion comprising nanoform carbon and functionalizing material such as a surfactant. In some such embodiments, the aqueous dispersion is substantially free of materials that would damage the carbon elements 201, such as acids.

Referring to Figure 3, in some embodiments, the surface treatment 202 on the high aspect ratio carbon elements 201 includes a thin polymeric layer disposed on the carbon elements that promotes adhesion of the active material to the network. In some such embodiments the thin polymeric layer comprises a self-assembled and or self-limiting polymer layer. In some embodiments, the thin polymeric layer bonds to the active material, e.g., via hydrogen bonding.

In some embodiments the thin polymeric layer may have a thickness in the direction normal to the outer surface of the carbon elements of less than 3 times, 2 times, 1 times, 0.5 times, 0.1 times that the minor dimension of the element 201 (or less).

In some embodiments, the thin polymeric layer includes functional groups (e.g., side functional groups) that bond to the active material, e.g., via non-covalent bonding such as π - π bonding. In some such embodiments the thin polymeric layer may form a stable covering layer over at least a portion of the elements 201.

In some embodiments, the thin polymeric layer on some of the elements 201 may bond with a current collector 101 or adhesion layer 102 underlying the active layer 200. For example, in some embodiments the thin polymeric layer includes side functional groups that bond to the surface of the current collector 101 or adhesion layer 102, e.g., via non-covalent bonding such as π - π bonding. In some such
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embodiments the thin polymeric layer may form a stable covering layer over at least a portion of the elements 201. In some embodiments, this arrangement provides for excellent mechanical stability of the electrode 10, as discussed in greater detail below.

In some embodiments, the polymeric material is miscible in solvents of the type described in the examples above. For example, in some embodiments the polymeric material is miscible in a solvent that
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includes an alcohol such as methanol, ethanol, or 2-propanol (isopropyl alcohol, sometimes referred to as IPA) or combinations thereof. In some embodiments, the solvent may include one or more additives used to further improve the properties of the solvent, e.g., low boiling point additives such as acetonitrile (ACN), de-ionized water, and tetrahydrofuran.

Suitable examples of materials which may be used to form the polymeric layer include water soluble
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polymers such as polyvinylpyrrolidone. Additional exemplary materials are provided below.

In some embodiments, the polymeric material has a low molecular mass, e.g., less than or equal to 1,000,000 g/mol, 500,000 g/mol, 100,000 g/mol, 50,000 g/mol, 10,000 g/mol, 5,000 g/mol, 2,500 g/mol or less.

Note that the thin polymeric layer described above is qualitatively distinct from bulk polymer binder used
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in conventional electrodes. Rather than filling a significant portion of the volume of the active layer 100, the thin polymeric layer resides on the surface of the high aspect ratio carbon elements, leaving the vast majority of the void space within the network 200 available to hold active material particles 300.

For example, in some embodiments, the thin polymeric layer has a maximum thickness in a direction
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normal to an outer surface of the network of less than or equal to 1 times, 0.5 times, 0.25 times, or less of the size of the carbon elements 201 along their minor dimensions. For example, in some embodiments the thin polymeric layer may be only a few molecules thick (e.g., less than or equal to 100, 50, 10, 5, 4, 3, 2, or even 1 molecule(s) thick). Accordingly, in some embodiments, less than 10%, 5%, 1%, 0.1%, 0.01%, 0.001% or less of the volume of the active layer 100 is filled with the thin polymeric layer.

In yet further exemplary embodiments, the surface treatment 202 may be formed a layer of carbonaceous
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material which results from the pyrolyzation of polymeric material disposed on the high aspect ratio carbon elements 201. This layer of carbonaceous material (e.g., graphitic or amorphous carbon) may

attach (e.g., via covalent bonds) to or otherwise promote adhesion with the active material particles 300. Examples of suitable pyrolyzation techniques are described in U.S. Patent Application Serial No. 63/028982 filed May 22, 2020. One suitable polymeric material for use in this technique is polyacrylonitrile (PAN).

- 5 In various embodiments, the active material particles 300 may include any active material suitable for use in energy storage devices, including metal oxides such as lithium metal oxides. For example, the active material particles 300 may include lithium cobalt oxide (LCO, sometimes called "lithium cobaltate" or "lithium cobaltite," is a chemical compound with one variant of possible formulations being LiCoO_2); lithium nickel manganese cobalt oxide (NMC, with a variant formula of LiNiMnCo); lithium manganese
10 oxide (LMO with variant formulas of LiMn_2O_4 , Li_2MnO_3 and others); lithium nickel cobalt aluminum oxide (LiNiCoAlO_2 and variants thereof as NCA) and lithium titanate oxide (LTO, with one variant formula being $\text{Li}_4\text{Ti}_5\text{O}_{12}$); lithium iron phosphate oxide (LFP, with one variant formula being LiFePO_3), lithium nickel cobalt aluminum oxide (and variants thereof as NCA) as well as other similar other materials. Other variants of the foregoing may be included.
- 15 In some embodiments where NMC is used as an active material, nickel rich NMC may be used. For example, in some embodiments, the variant of NMC may be $\text{LiNi}_x\text{Mn}_y\text{Co}_{1-x-y}$, where x is equal to or greater than about 0.7, 0.75, 0.80, 0.85, or more. In some embodiments, so called NMC811 may be used, where in the foregoing formula x is about 0.8 and y is about 0.1.

In some embodiments, the active material includes other forms of Lithium Nickel Manganese Cobalt
20 Oxide ($\text{LiNi}_x\text{Mn}_y\text{Co}_z\text{O}_2$). For example, common variants such as, without limitation: NMC 111 ($\text{LiNi}_{0.33}\text{Mn}_{0.33}\text{Co}_{0.33}\text{O}_2$); NMC 532 ($\text{LiNi}_{0.5}\text{Mn}_{0.3}\text{Co}_{0.2}\text{O}_2$); NMC 622 ($\text{LiNi}_{0.6}\text{Mn}_{0.2}\text{Co}_{0.2}\text{O}_2$); and others may be used.

In some embodiments, e.g., where the electrode is used as an anode, the active material may include
25 graphite, hard carbon, activated carbon, nanoform carbon, silicon, silicon oxides, carbon encapsulated silicon nanoparticles. In some such embodiments the active layer 100 may be intercalated with lithium, e.g., using pre-lithiation methods known in the art.

In some embodiments, the techniques described herein may allow for the active layer 100 be made of in
large portion of material in the active layer, e.g., greater than 75%, 80%, 85%, 90%, 95%, 99%, 99.5%,
99.8% or more by weight, while still exhibiting excellent mechanical properties (e.g., lack of
30 delamination during operation in an energy storage device of the types described herein). For example, in some embodiments, the active layer may have such aforementioned high amount of active material and a

large thickness (e.g., greater than 50 μm , 100 μm , 150 μm , 200 μm , or more), while still exhibiting excellent mechanical properties (e.g., a lack of delamination during operation in an energy storage device of the types described herein).

The active material particles 201 in the active layer 100 may be characterized by a median particle sized in the range of e.g., 0.1 μm and 50 micrometers μm , or any subrange thereof. The active material particles 201 in the active layer 100 may be characterized by a particle sized distribution which is monomodal, bi-modal or multi-modal particle size distribution. The active material particles 201 may have a specific surface area in the range of 0.1 meters squared per gram (m^2/g) and 100 meters squared per gram (m^2/g), or any subrange thereof.

10 In some embodiments, the active layer 100 may have mass loading of active material particles 300 e.g., of at least 20 mg/cm^2 , 30 mg/cm^2 , 40 mg/cm^2 , 50 mg/cm^2 , 60 mg/cm^2 , 70 mg/cm^2 , 80 mg/cm^2 , 90 mg/cm^2 , 100 mg/cm^2 , or more.

Referring to Figure 4, an electron micrograph of an exemplary active material layer of the type described herein is shown. Tendril like high aspect ratio carbon elements 201 (formed of CNT bundles) are clearly shown enmeshing the active material particles 300. Note the lack of any bulky polymeric material taking up space within the layer.

Energy Storage Cell

Referring to Figure 5, an energy storage cell 500 is shown which includes a first electrode 501 a second electrode 502, a permeable separator 503 disposed between the first electrode 501 and the second electrode 502, and an electrolyte 504 wetting the first and second electrodes. One or both of the electrodes 501, 502 may be of the type described herein.

In some embodiments, the energy storage cell 500 may be a battery, such as a lithium ion battery. In some such embodiments, the electrolyte may be a lithium salt dissolved in a solvent, e.g., of the types described in Qi Li, Juner Chen, Lei Fan, Xueqian Kong, Yingying Lu, Progress in electrolytes for rechargeable Li-based batteries and beyond, Green Energy & Environment, Volume 1, Issue 1, Pages 18-42, the entire contents of which are incorporated herein by reference.

In some such embodiments, the energy storage cell may have an operational voltage in the range of 1.0 V to 5.0 V, or any subrange thereof such as 2.3V – 4.3V.

In some such embodiments, the energy storage cell 500 may have an operating temperature range comprising -40 $^{\circ}\text{C}$ to 100 $^{\circ}\text{C}$ or any subrange thereof such as -10 $^{\circ}\text{C}$ to 60 $^{\circ}\text{C}$.

In some such embodiments, the energy storage cell 500 may have a gravimetric energy density of at least 100 Wh/kg, 200 Wh/kg, 300 Wh/kg, 400 Wh/kg, 500 Wh/kg, 1000 Wh/kg or more.

In some such embodiments, the energy storage cell 500 may have a volumetric energy density of at least 200 Wh/L, 400 Wh/L, 600 Wh/L, 800 Wh/L, 1,000 Wh/L, 1,500 Wh/L, 2,000 Wh/L or more.

5 In some such embodiments, the energy storage cell 500 may have a C rate in the range of 0.1 to 50.

In some such embodiments, the energy storage cell 500 may have a cycle life of at least 1,000, 1500, 2,000, 2,500, 3,000, 3,500, 4,000 or more charge discharge cycles.

10 In some embodiments, the energy storage cell 500 may be a lithium ion capacitor of the type described in U.S. Pat. App. Serial No. 63/021492, filed May 8, 2020, the entire contents of which are incorporated herein by reference.

In some such embodiments, the energy storage cell 500 may have an operating temperature range comprising -60°C to 100°C or any subrange thereof such as -40°C to 85°C .

In some such embodiments, the energy storage cell 500 may have a gravimetric energy density of at least 10 Wh/kg, 15 Wh/kg, 20 Wh/kg, 30 Wh/kg, 40 Wh/kg, 50 Wh/kg, or more.

15 In some such embodiments, the energy storage cell 500 may have a volumetric energy density of at least 20 Wh/L, 30 Wh/L, 40 Wh/L, 50 Wh/L, 60 Wh/L, 70 Wh/L, 80 Wh/L or more.

In some such embodiments, the energy storage cell 500 may have a gravimetric power density of at least 5 kW/kg, 7.5 W/kg, 10 kW/kg, 12.5 kW/kg, 14 kW/kg, 15 kW/kg or more.

20 In some such embodiments, the energy storage cell 500 may have a volumetric power density of at least 10 kW/L, 15 kW/L, 20 kW/L, 22.5 kW/L, 25 kW/L, 28 kW/L, 30 kW/L or more.

In some such embodiments, the energy storage cell 500 may have a C rate in the range of 1.0 to 100.

In some such embodiments, the energy storage cell 500 may have a cycle life of at least 100,000, 500,000, 1,000,000 or more charge discharge cycles.

Fabrication Methods

25 The electrode 10 featuring active layer 100 as described herein may be made using any suitable manufacturing process. As will be understood by one skilled in the art, in some embodiments the electrode 10 may be made using wet coating techniques of the types described in International Patent Publication No. WO/2018/102652 published June 7, 2018 in further view of the teachings described herein.

Referring to Figure 6, in some embodiments, the active layer 100 of electrode 10 may be formed using the method 1000. In step 1001 high aspect ratio carbon elements 201 and a surface treatment material (e.g., a surfactant or polymer material as described herein) are combined with a solvent (of the type described herein) to form an initial slurry.

- 5 In step 1002 the initial slurry is processed to ensure good dispersion of the solid materials in the slurry. In some embodiments, this processing includes introducing mechanical energy into the mixture of solvent and solid materials (e.g., using a sonicator, which may be sometimes also be referred to as a “sonifier”) or other suitable mixing device (e.g., a high shear mixer). In some embodiments, the mechanical energy introduced into the mixture is at least 0.4 kilowatt-hours per kilogram (kWh/kg), 0.5 kWh/kg, 0.6 kWh/kg, 0.7 kWh/kg, 0.8 kWh/kg, 0.9 kWh/kg, 1.0 kWh/kg, or more. For example, the mechanical energy introduced into the mixture per kilogram of mixture may be in the range of 0.4 kWh/kg to 1.0 kWh/kg or any subrange thereof such as 0.4 kWh/kg to 0.6 kWh/kg.

In some embodiments an ultrasonic bath mixer may be used. In other embodiments, a probe sonicator may be used. Probe sonication may be significantly more powerful and effective when compared to ultrasonic baths for nanoparticle applications. High shear forces created by ultrasonic cavitation have the ability to break up particle agglomerates and result in smaller and more uniform particles sizes. Among other things, sonication can result in stable and homogenous suspensions of the solids in the slurry. Generally, this results in dispersing and deagglomerating and other breakdown of the solids. Examples of probe sonication devices include the Q Series Probe Sonicators available from QSonica LLC of
15 Newtown, Connecticut. Another example includes the Branson Digital SFX-450 sonicator available commercially from Thomas Scientific of Swedesboro, New Jersey.

In some embodiments, however, the localized nature of each probe within the probe assembly can result in uneven mixing and suspension. Such may be the case, for example, with large samples. This may be countered by use of a setup with a continuous flow cell and proper mixing. That is, with such a setup,
25 mixing of the slurry will achieve reasonably uniform dispersion.

In some embodiments the initial slurry, once processed will have a viscosity in the range of 5,000 cps to 25,000 cps or any subrange thereof, e.g., 6,000 cps to 19,000cps.

In step 1003, the surface treatment 202 may be fully or partially formed on the high aspect ratio carbon elements 201 in the initial slurry. In some embodiments, at this stage the surface treatment 202 may self-assemble as described in detail above with reference to Figures 2 and 3. The resulting surface treatment
30 201 may include functional groups or other features which, as described in further steps below, may promote adhesion between the high aspect ratio carbon elements 201 and active material particles 300.

In step 1004 the active material particles 300 may be combined with the initial slurry to form a final slurry containing the active material particles 300 along with the high aspect ratio carbon elements 201 with the surface treatment 202 formed thereon.

5 In some embodiments, the active material 300 may be added directly to the initial slurry. In other embodiments, the active material 300 may first be dispersed in a solvent (e.g., using the techniques described above with respect to the initial solvent) to form an active material slurry. This active material slurry may then be combined with the initial slurry to form the final slurry.

10 In step 1005 the final slurry is processed to ensure good dispersion of the solid materials in the final slurry. In various embodiments any suitable mixing process known in the art may be used. In some embodiments this processing may use the techniques described above with reference to step 1002. In some embodiments, a planetary mixer such as a multi-axis (e.g., three or more axis) planetary mixer may be used. In some such embodiments the planetary mixer can feature multiple blades, e.g., two or more mixing blades and one or more (e.g., two, three, or more) dispersion blades such as disk dispersion blades.

15 In some embodiments, during this step 1005, the matrix 200 enmeshing the active material 300 may fully or partially self-assemble, as described in detail above with reference to Figures 2 and 3. In some embodiments, interactions between the surface treatment 202 and the active material 300 promote the self-assembly process.

20 In some embodiments the final slurry, once processed will have a viscosity in the range of 1,000 cps to 10,000 cps or any subrange thereof, e.g., 2,500 cps to 6000 cps.

25 In step 1006, the active layer 100 is formed from the final slurry. In some embodiments, final slurry may be cast wet directly onto the current collector conductive layer 101 (or optional adhesion layer 102) and dried. As an example, casting may be by applying at least one of heat and a vacuum until substantially all of the solvent and any other liquids have been removed, thereby forming the active layer 100. In some such embodiments it may be desirable to protect various parts of the underlying layers. For example, it may be desirable to protect an underside of the conductive layer 101 where the electrode 10 is intended for two-sided operation. Protection may include, for example, protection from the solvent by masking certain areas, or providing a drain to direct the solvent away.

30 In other embodiments, the final slurry may be at least partially dried elsewhere and then transferred onto the adhesion layer 102 or the conductive layer 101 to form the active layer 100, using any suitable technique (e.g., roll-to-roll layer application). In some embodiments the wet combined slurry may be

placed onto an intermediate material with an appropriate surface and dried to form the layer (i.e., the active layer 100). While any material with an appropriate surface may be used as the intermediate material, exemplary intermediate material includes PTFE as subsequent removal from the surface is facilitated by the properties thereof. In some embodiments, the designated layer is formed in a press to provide a layer that exhibits a desired thickness, area and density.

In some embodiments, the final slurry may be formed into a sheet, and coated onto the adhesion layer 102 or the conductive layer 101 as appropriate. For example, in some embodiments, the final slurry may be applied to through a slot die to control the thickness of the applied layer. In other embodiments, the slurry may be applied and then leveled to a desired thickness, e.g., using a doctor blade. A variety of other techniques may be used for applying the slurry. For example, coating techniques may include, without limitation: comma coating; comma reverse coating; doctor blade coating; slot die coating; direct gravure coating; air doctor coating (air knife); chamber doctor coating; off set gravure coating; one roll kiss coating; reverse kiss coating with a small diameter gravure roll; bar coating; three reverse roll coating (top feed); three reverse roll coating (fountain die); reverse roll coating and others.

The viscosity of the final slurry may vary depending on the application technique. For example, for comma coating, the viscosity may range between about 1,000 cps to about 200,000 cps. Lip-die coating provides for coating with slurry that exhibits a viscosity of between about 500 cps to about 300,000 cps. Reverse-kiss coating provides for coating with slurry that exhibits a viscosity of between about 5 cps and 1,000 cps. In some applications, a respective layer may be formed by multiple passes.

In some embodiments, the active layer 100 formed from the final slurry may be compressed (e.g., using a calendaring apparatus) before or after being applied to the electrode 10. In some embodiments, the slurry may be partially or completely dried (e.g., by applying heat, vacuum or a combination thereof) prior to or during the compression process. For example, in some embodiments, the active layer may be compressed to a final thickness (e.g., in the direction normal to the current collector layer 101) of less than 90%, 80%, 70%, 50%, 40%, 30%, 20%, 10% or less of its pre-compression thickness.

In various embodiments, when a partially dried layer is formed during a coating or compression process, the layer may be subsequently fully dried, (e.g., by applying heat, vacuum or a combination thereof). In some embodiments, substantially all of the solvent is removed from the active layer 100.

In some embodiments, solvents used in formation of the slurries are recovered and recycled into the slurry-making process.

In some embodiments, the active layer may be compressed, e.g., to break some of the constituent high aspect ratio carbon elements or other carbonaceous material to increase the surface area of the respective layer. In some embodiments, this compression treatment may increase one or more of adhesion between the layers, ion transport rate within the layers, and the surface area of the layers. In various embodiments, compression can be applied before or after the respective layer is applied to or formed on the electrode
5 10.

In some embodiments where calendaring is used to compress the active layer 100, the calendaring apparatus may be set with a gap spacing equal to less than 90%, 80%, 70%, 50%, 40%, 30%, 20%, 10% or less of the layer's pre-compression thickness (e.g., set to about 33% of the layer's pre-compression
10 thickness). The calendar rolls can be configured to provide suitable pressure, e.g., greater than 1 ton per cm of roll length, greater than 1.5 ton per cm of roll length, greater than 2.0 ton per cm of roll length, greater than 2.5 ton per cm of roll length, or more. In some embodiments, the post compression active layer will have a density in the range of 1 g/cc to 10 g/cc, or any subrange thereof such as 2.5 g/cc to 4.0 g /cc. In some embodiments the calendaring process may be carried out at a temperature in the range of 20
15 °C to 140 °C or any subrange thereof. In some embodiments the active layer may be pre-heated prior to calendaring, e.g., at a temperature in the range of 20 °C to 100 °C or any subrange thereof.

Once the electrode 10 has been assembled, the electrode 100 may be used to assemble the energy storage device 10. Assembly of the energy storage device 10 may follow conventional steps used for assembling electrodes with separators and placement within a housing such as a canister or pouch, and further may
20 include additional steps for electrolyte addition and sealing of the housing.

In various embodiments, process 1000 may include any of the following features (individually or in any suitable combination)

In some embodiments, the initial slurry has a solid content in the range of 0.1%-20.0% (or any subrange thereof) by weight. In some embodiments, the final slurry has a solid content in the range of 10.0% -
25 80% (or any subrange thereof) by weight.

In various embodiments, the solvent used may any of those described herein with respect to the formation of the surface treatment 202. In some embodiments, the surfactant material used to form the surface treatment 202 may be soluble in a solvent which exhibits advantageous properties. For example, in some
30 embodiments, the solvent may include water or an alcohol such as methanol, ethanol, or 2-propanol (isopropyl alcohol, sometimes referred to as IPA) or combinations thereof. In some embodiments, the solvent may include one or more additives used to further improve the properties of the solvent, e.g., low boiling point additives such as acetonitrile (ACN), de-ionized water, and tetrahydrofuran.

In some embodiments, if a low boiling point solvent is used the solvent may be quickly removed using a thermal drying process performed at a relatively low temperature. As will be understood by those skilled in the art, this can improve the speed and or cost of manufacture of the electrode 10. For example, in some embodiments, the solvent may have a boiling point less than 250° C, 225° C, 202° C, 200° C, 185° C, 180° C, 175° C, 150° C, 125° C, or less, e.g., less than or equal to 100° C.

In some embodiments, the solvent may exhibit other advantageous properties. In some embodiments the solvent may have a low viscosity, such a viscosity at 20° C of less than or equal to 3.0 centipoise, 2.5 centipoise, 2.0 centipoise, 1.5 centipoise, or less. In some embodiments the solvent may have a low surface tension such a surface tension at 20° C of less than or equal to 40 mN/m, 35 mN/m, 30 mN/m, 25 mN/m or less. In some embodiments the solvent may have a low toxicity, e.g., toxicity comparable to alcohols such as isopropyl alcohol.

In some embodiments, during the formation of the active layer, a material forming the surface treatment may be dissolved in a solvent substantially free of pyrrolidone compounds. In some embodiments, the solvent is substantially free of n-methyl-2-pyrrolidone.

In some embodiments, the surface treatment 201 is formed from a material that includes a surfactant of the type described herein.

In some embodiments, dispersing high aspect ratio carbon elements and a surface treatment material in a solvent to form an initial slurry comprises applying forces to agglomerated carbon elements to cause the elements to slide apart from each other along a direction transverse to a minor axis of the elements. In some embodiments, techniques for forming such dispersions may be adapted from those disclosed in International Patent Publication No. WO/2018/102652 published June 7, 2018 in further view of the teachings described herein.

In some embodiments, the high aspect ratio carbon elements 201 can be functionalized prior to forming a slurry used to form the electrode 10. For example, in one aspect a method is disclosed that includes dispersing high aspect ratio carbon elements 201 and a surface treatment material in an aqueous solvent to form an initial slurry, wherein said dispersion step results in the formation of a surface treatment on the high aspect ratio carbon; drying the initial slurry to remove substantially all moisture resulting in a dried powder of the high aspect ratio carbon with the surface treatment thereon. In some embodiments, the dried powder may be combined, e.g., with a slurry of solvent and active material to form a final solvent of the type described above with reference to method 1000.

In some embodiments, drying the initial slurry comprises lyophilizing (freeze-drying) the initial slurry. In some embodiments, the aqueous solvent and initial slurry are substantially free of substances damaging to the high aspect ratio carbon elements. In some embodiments, the aqueous solvent and initial slurry are substantially free of acids. In some embodiments, the initial slurry consists essentially of the high aspect ratio carbon elements, the surface treatment material, and water.

Some embodiments further include dispersing the dried powder of the high aspect ratio carbon with the surface treatment in a solvent and adding an active material to form a secondary slurry; coating the secondary slurry onto a substrate; and drying the secondary slurry to form an electrode active layer. In some embodiments, the preceding steps can be performed using techniques adapted from those disclosed in International Patent Publication No. WO/2018/102652 published June 7, 2018 in further view of the teachings described herein.

In some embodiments, the final slurry may include polymer additives such as polyacrylic acid (PAA), poly(vinyl alcohol) (PVA), poly(vinyl acetate) (PVAc), polyacrylonitrile (PAN), polyisoprene (PIpr), polyaniline (PANI), polyethylene (PE), polyimide (PI), polystyrene (PS), polyurethane (PU), polyvinyl butyral (PVB), polyvinyl pyrrolidone (PVP). In some embodiments, the active layer may be treated by applying heat to pyrolyze the additive such that the surface treatment 202 may be formed a layer of carbonaceous material which results from the pyrolyzation of the polymeric additive. This layer of carbonaceous material (e.g., graphitic or amorphous carbon) may attach (e.g., via covalent bonds) to or otherwise promote adhesion with the active material particles 300. The heat treatment may be applied by any suitable means, e.g., by application of a laser beam. Examples of suitable pyrolyzation techniques are described in U.S. Patent Application Serial No. 63/028982 filed May 22, 2020.

Surfactants

The techniques described above include the use of surfactants to form a surface treatment 202 on high aspect ratio carbon nanotubes 201 in order to promote adhesion with the active material particles 300. While several advantageously suitable surfactants have been described, it is to be understood that other surfactant material may be used, including the following.

Surfactants are molecules or groups of molecules having surface activity, including wetting agents, dispersants, emulsifiers, detergents, and foaming agents. A variety of surfactants can be used in preparation surface treatments as described herein. Typically, the surfactants used contain a lipophilic nonpolar hydrocarbon group and a polar functional hydrophilic group. The polar functional group can be a carboxylate, ester, amine, amide, imide, hydroxyl, ether, nitrile, phosphate, sulfate, or sulfonate. The surfactants can be used alone or in combination. Accordingly, a combination of surfactants can include

anionic, cationic, nonionic, zwitterionic, amphoteric, and ampholytic surfactants, so long as there is a net positive or negative charge in the head regions of the population of surfactant molecules. In some instances, a single negatively charged or positively charged surfactant is used in the preparation of the present electrode compositions.

- 5 A surfactant used in preparation of the present electrode compositions can be anionic, including, but not limited to, sulfonates such as alkyl sulfonates, alkylbenzene sulfonates, alpha olefin sulfonates, paraffin sulfonates, and alkyl ester sulfonates; sulfates such as alkyl sulfates, alkyl alkoxy sulfates, and alkyl alkoxyated sulfates; phosphates such as monoalkyl phosphates and dialkyl phosphates; phosphonates; carboxylates such as fatty acids, alkyl alkoxy carboxylates, sarcosinates, isethionates, and taurates.
- 10 Specific examples of carboxylates are sodium oleate, sodium cocoyl isethionate, sodium methyl oleoyl taurate, sodium laureth carboxylate, sodium trideceth carboxylate, sodium lauryl sarcosinate, lauroyl sarcosine, and cocoyl sarcosinate. Specific examples of sulfates include sodium dodecyl sulfate (SDS), sodium lauryl sulfate, sodium laureth sulfate, sodium trideceth sulfate, sodium tridecyl sulfate, sodium cocyl sulfate, and lauric monoglyceride sodium sulfate.
- 15 Suitable sulfonate surfactants include, but are not limited to, alkyl sulfonates, aryl sulfonates, monoalkyl and dialkyl sulfosuccinates, and monoalkyl and dialkyl sulfosuccinamates. Each alkyl group independently contains about two to twenty carbons and can also be ethoxylated with up to about 8 units, preferably up to about 6 units, on average, for example, 2, 3, or 4 units, of ethylene oxide, per each alkyl group. Illustrative examples of alky and aryl sulfonates are sodium tridecyl benzene sulfonate (STBS) and
- 20 sodium dodecylbenzene sulfonate (SDBS).
- Illustrative examples of sulfosuccinates include, but are not limited to, dimethicone copolyol sulfosuccinate, diamyl sulfosuccinate, dicapryl sulfosuccinate, dicyclohexyl sulfosuccinate, diheptyl sulfosuccinate, dihexyl sulfosuccinate, diisobutyl sulfosuccinate, dioctyl sulfosuccinate, C12-15 pareth sulfosuccinate, cetearyl sulfosuccinate, cocopolyglucose sulfosuccinate, cocoyl butyl gluceth-10
- 25 sulfosuccinate, deceth-5 sulfosuccinate, deceth-6 sulfosuccinate, dihydroxyethyl sulfosuccinylundecylenate, hydrogenated cottonseed glyceride sulfosuccinate, isodecyl sulfosuccinate, isostearyl sulfosuccinate, laneth-5 sulfosuccinate, laureth sulfosuccinate, laureth-12 sulfosuccinate, laureth-6 sulfosuccinate, laureth-9 sulfosuccinate, lauryl sulfosuccinate, nonoxynol-10 sulfosuccinate, oleth-3 sulfosuccinate, oleyl sulfosuccinate, PEG-10 laurylcitrate sulfosuccinate, sitosereth-14
- 30 sulfosuccinate, stearyl sulfosuccinate, tallow, tridecyl sulfosuccinate, ditridecyl sulfosuccinate, bisglycol ricinosulfosuccinate, di(1,3-di-methylbutyl)sulfosuccinate, and silicone copolyol sulfosuccinates.

Illustrative examples of sulfosuccinamates include, but are not limited to, lauramido-MEA sulfosuccinate, oleamido PEG-2 sulfosuccinate, cocamido MIPA-sulfosuccinate, cocamido PEG-3 sulfosuccinate, isostearamido MEA-sulfosuccinate, isostearamido MIPA-sulfosuccinate, lauramido MEA-sulfosuccinate, lauramido PEG-2 sulfosuccinate, lauramido PEG-5 sulfosuccinate, myristamido MEA-sulfosuccinate, oleamido MEA-sulfosuccinate, oleamido PIPA-sulfosuccinate, oleamido PEG-2 sulfosuccinate, palmitamido PEG-2 sulfosuccinate, palmitoleamido PEG-2 sulfosuccinate, PEG-4 cocamido MIPA-sulfosuccinate, ricinoleamido MEA-sulfosuccinate, stearamido MEA-sulfosuccinate, stearyl sulfosuccinamate, tallamido MEA-sulfosuccinate, tallow sulfosuccinamate, tallowamido MEA-sulfosuccinate, undecylenamido MEA-sulfosuccinate, undecylenamido PEG-2 sulfosuccinate, wheat germamido MEA-sulfosuccinate, and wheat germamido PEG-2 sulfosuccinate.

Some examples of commercial sulfonates are AEROSOL® OT-S, AEROSOL® OT-MSO, AEROSOL® TR70% (Cytec Inc., West Paterson, N.J.), NaSul CA-HT3 (King Industries, Norwalk, Conn.), and C500 (Crompton Co., West Hill, Ontario, Canada). AEROSOL® OT-S is sodium dioctyl sulfosuccinate in petroleum distillate. AEROSOL® OT-MSO also contains sodium dioctyl sulfosuccinate. AEROSOL® TR70% is sodium bistridecyl sulfosuccinate in mixture of ethanol and water. NaSul CA-HT3 is calcium dinonylnaphthalene sulfonate/carboxylate complex. C500 is an oil soluble calcium sulfonate.

Alkyl or alkyl groups refers to saturated hydrocarbons having one or more carbon atoms, including straight-chain alkyl groups (for example, methyl, ethyl, propyl, butyl, pentyl, hexyl, heptyl, octyl, nonyl, decyl, and so on), cyclic alkyl groups (or cycloalkyl or alicyclic or carbocyclic groups) (for example, cyclopropyl, cyclopentyl, cyclohexyl, cycloheptyl, cyclooctyl, and so on), branched-chain alkyl groups (for example, isopropyl, tert-butyl, sec-butyl, isobutyl, and so on), and alkyl-substituted alkyl groups (for example, alkyl-substituted cycloalkyl groups and cycloalkyl-substituted alkyl groups).

Alkyl can include both unsubstituted alkyls and substituted alkyls. Substituted alkyls refers to alkyl groups having substituents replacing one or more hydrogens on one or more carbons of the hydrocarbon backbone. Such substituents can include, alkenyl, alkynyl, halogeno, hydroxyl, alkylcarbonyloxy, arylcarbonyloxy, alkoxy carbonyloxy, aryloxy, aryloxy carbonyloxy, carboxylate, alkylcarbonyl, arylcarbonyl, alkoxy carbonyl, aminocarbonyl, alkylaminocarbonyl, dialkylaminocarbonyl, alkylthiocarbonyl, alkoxy, phosphate, phosphonato, phosphinato, cyano, amino (including alkyl amino, dialkylamino, arylamino, diarylamino and alkylarylamino), acylamino (including alkylcarbonylamino, arylcarbonylamino, carbamoyl and ureido), imino, sulfhydryl, alkylthio, arylthio, thiocarboxylate, sulfates, alkylsulfinyl, sulfonates, sulfamoyl, sulfonamido, nitro, trifluoromethyl, cyano, azido, heterocyclic, alkylaryl or aromatic (including heteroaromatic) groups.

In some embodiments, substituted alkyls can include a heterocyclic group. Heterocyclic groups include closed ring structures analogous to carbocyclic groups in which one or more of the carbon atoms in the ring is an element other than carbon, for example, nitrogen, sulfur or oxygen. Heterocyclic groups can be saturated or unsaturated. Exemplary heterocyclic groups include, aziridine, ethylene oxide (epoxides, oxiranes), thiirane (episulfides), dioxirane, azetidine, oxetane, thietane, dioxetane, dithietane, dithiete, azolidine, pyrrolidine, pyrroline, oxolane, dihydrofuran and furan.

For an anionic surfactant, the counter ion is typically sodium but can alternatively be potassium, lithium, calcium, magnesium, ammonium, amines (primary, secondary, tertiary or quaternary) or other organic bases. Exemplary amines include isopropylamine, ethanolamine, diethanolamine, and triethanolamine.

Mixtures of the above cations can also be used.

A surfactant used in preparation of the present materials can be cationic. Such cationic surfactants include, but are not limited to, pyridinium-containing compounds, and primary, secondary tertiary or quaternary organic amines. For a cationic surfactant, the counter ion can be, for example, chloride, bromide, methosulfate, ethosulfate, lactate, saccharinate, acetate and phosphate. Examples of cationic amines include polyethoxylated oleyl/stearyl amine, ethoxylated tallow amine, cocoalkylamine, oleylamine and tallow alkyl amine, as well as mixtures thereof.

Examples of quaternary amines with a single long alkyl group are cetyltrimethyl ammonium bromide (CTAB), benzyl dodecyl dimethyl ammonium bromide (BddaBr), benzyl dimethyl hexadecyl ammonium chloride (BdhaCl), dodecyl trimethyl ammonium bromide, myristyl trimethyl ammonium bromide, stearyl dimethyl benzyl ammonium chloride, oleyl dimethyl benzyl ammonium chloride, lauryl trimethyl ammonium methosulfate (also known as cocotrimonium methosulfate), cetyl-dimethyl hydroxyethyl ammonium dihydrogen phosphate, bassuamidopropyl konium chloride, cocotrimonium chloride, distearyldimonium chloride, wheat germ-amidopropyl konium chloride, stearyl octyldimonium methosulfate, isostearaminopropyl-konium chloride, dihydroxypropyl PEG-5 linoleammonium chloride, PEG-2 stearamonium chloride, behentrimonium chloride, dicetyl dimonium chloride, tallow trimonium chloride and behenamidopropyl ethyl dimonium ethosulfate.

Examples of quaternary amines with two long alkyl groups are didodecyl dimethyl ammonium bromide (DDAB), distearyldimonium chloride, dicetyl dimonium chloride, stearyl octyldimonium methosulfate, dihydrogenated palmolethyl hydroxyethyl monium methosulfate, dipalmitoylethyl hydroxyethyl monium methosulfate, dioleylethyl hydroxyethyl monium methosulfate, and hydroxypropyl bisstearyldimonium chloride.

Quaternary ammonium compounds of imidazoline derivatives include, for example, isostearyl benzylimidonium chloride, cocoyl benzyl hydroxyethyl imidazolinium chloride, cocoyl hydroxyethylimidazolinium PG-chloride phosphate, and stearyl hydroxyethylimidonium chloride. Other heterocyclic quaternary ammonium compounds, such as dodecylpyridinium chloride, amprolium hydrochloride (AH), and benzethonium hydrochloride (BH) can also be used.

A surfactant used in preparation of the present materials can be nonionic, including, but not limited to, polyalkylene oxide carboxylic acid esters, fatty acid esters, fatty alcohols, ethoxylated fatty alcohols, poloxamers, alkanolamides, alkoxyated alkanolamides, polyethylene glycol monoalkyl ether, and alkyl polysaccharides. Polyalkylene oxide carboxylic acid esters have one or two carboxylic ester moieties each with about 8 to 20 carbons and a polyalkylene oxide moiety containing about 5 to 200 alkylene oxide units. An ethoxylated fatty alcohol contains an ethylene oxide moiety containing about 5 to 150 ethylene oxide units and a fatty alcohol moiety with about 6 to about 30 carbons. The fatty alcohol moiety can be cyclic, straight, or branched, and saturated or unsaturated. Some examples of ethoxylated fatty alcohols include ethylene glycol ethers of oleth alcohol, steareth alcohol, lauryl alcohol and isocetyl alcohol. Poloxamers are ethylene oxide and propylene oxide block copolymers, having from about 15 to about 100 moles of ethylene oxide. Alkyl polysaccharide ("APS") surfactants (for example, alkyl polyglycosides) contain a hydrophobic group with about 6 to about 30 carbons and a polysaccharide (for example, polyglycoside) as the hydrophilic group. An example of commercial nonionic surfactant is FOA-5 (Octel Starreon LLC., Littleton, Colo.).

Specific examples of suitable nonionic surfactants include alkanolamides such as cocamide diethanolamide ("DEA"), cocamide monoethanolamide ("MEA"), cocamide monoisopropanolamide ("MIPA"), PEG-5 cocamide MEA, lauramide DEA, and lauramide MEA; alkyl amine oxides such as lauramine oxide, cocamine oxide, cocamidopropylamine oxide, and lauramidopropylamine oxide; sorbitan laurate, sorbitan distearate, fatty acids or fatty acid esters such as lauric acid, isostearic acid, and PEG-150 distearate; fatty alcohols or ethoxylated fatty alcohols such as lauryl alcohol, alkylpolyglycosides such as decyl glucoside, lauryl glucoside, and coco glucoside.

A surfactant used in preparation of the present materials can be zwitterionic, having both a formal positive and negative charge on the same molecule. The positive charge group can be quaternary ammonium, phosphonium, or sulfonium, whereas the negative charge group can be carboxylate, sulfonate, sulfate, phosphate or phosphonate. Similar to other classes of surfactants, the hydrophobic moiety can contain one or more long, straight, cyclic, or branched, aliphatic chains of about 8 to 18 carbon atoms. Specific examples of zwitterionic surfactants include alkyl betaines such as cocodimethyl carboxymethyl betaine, lauryl dimethyl carboxymethyl betaine, lauryl dimethyl alpha-carboxyethyl

betaine, cetyl dimethyl carboxymethyl betaine, lauryl bis-(2-hydroxyethyl)carboxy methyl betaine, stearyl bis-(2-hydroxypropyl)carboxymethyl betaine, oleyl dimethyl gamma-carboxypropyl betaine, and lauryl bis-(2-hydroxypropyl)alphacarboxy-ethyl betaine, amidopropyl betaines; and alkyl sultaines such as cocodimethyl sulfopropyl betaine, stearylidimethyl sulfopropyl betaine, lauryl dimethyl sulfoethyl betaine, lauryl bis-(2-hydroxyethyl)sulfopropyl betaine, and alkylamidopropylhydroxy sultaines.

A surfactant used in preparation of the present materials can be amphoteric. Examples of suitable amphoteric surfactants include ammonium or substituted ammonium salts of alkyl amphocarboxy glycinate and alkyl amphocarboxypropionates, alkyl amphodipropionates, alkyl amphodiacetates, alkyl amphoglycinates, and alkyl amphopropionates, as well as alkyl iminopropionates, alkyl iminodipropionates, and alkyl amphopropylsulfonates. Specific examples are cocoamphoacetate, cocoamphopropionate, cocoamphodiacetate, lauroamphoacetate, lauroamphodiacetate, lauroamphodipropionate, lauroamphodiacetate, cocoamphopropyl sulfonate, caproamphodiacetate, caproamphoacetate, caproamphodipropionate, and stearoamphoacetate.

A surfactant used in preparation of the present materials can also be a polymer such as N-substituted polyisobutenyl succinimides and succinates, alkyl methacrylate vinyl pyrrolidinone copolymers, alkyl methacrylate-dialkylaminoethyl methacrylate copolymers, alkylmethacrylate polyethylene glycol methacrylate copolymers, polystearamides, and polyethylenimine.

A surfactant used in preparation of the present materials can also be a polysorbate type nonionic surfactant such as polyoxyethylene (20) sorbitan monolaurate (Polysorbate 20), polyoxyethylene (20) sorbitan monopalmitate (Polysorbate 40), polyoxyethylene (20) sorbitan monostearate (Polysorbate 60) or polyoxyethylene (20) sorbitan monooleate (Polysorbate 80).

A surfactant used in preparation of the present materials can be an oil-based dispersant, which includes alkylsuccinimide, succinate esters, high molecular weight amines, and Mannich base and phosphoric acid derivatives. Some specific examples are polyisobutenyl succinimide-polyethylenepolyamine, polyisobutenyl succinic ester, polyisobutenyl hydroxybenzyl-polyethylenepolyamine, and bis-hydroxypropyl phosphorate.

The surfactant used in preparation of the present materials can be a combination of two or more surfactants of the same or different types selected from the group consisting of anionic, cationic, nonionic, zwitterionic, amphoteric and ampholytic surfactants. Suitable examples of a combination of two or more surfactants of the same type include, but are not limited to, a mixture of two anionic surfactants, a mixture of three anionic surfactants, a mixture of four anionic surfactants, a mixture of two cationic surfactants, a mixture of three cationic surfactants, a mixture of four cationic surfactants, a mixture of two nonionic

surfactants, a mixture of three nonionic surfactants, a mixture of four nonionic surfactants, a mixture of two zwitterionic surfactants, a mixture of three zwitterionic surfactants, a mixture of four zwitterionic surfactants, a mixture of two amphoteric surfactants, a mixture of three amphoteric surfactants, a mixture of four amphoteric surfactants, a mixture of two ampholytic surfactants, a mixture of three ampholytic surfactants, and a mixture of four ampholytic surfactants.

Thin Polymeric Layer Materials

The techniques described above include the use of polymers to form a surface treatment 201 on high aspect ratio carbon nanotubes in order to promote adhesion with the active material particles 300. While several advantageously suitable polymers have been described, it is to be understood that other polymer material may be used, including the following.

The polymer used in preparation of the present materials can be polymer material such a water processable polymer material. In various embodiments any of the follow polymers (and combinations thereof) may be used: polyacrylic acid (PAA), poly(vinyl alcohol) (PVA), poly(vinyl acetate) (PVAc), polyacrylonitrile (PAN), polyisoprene (PIpr), polyaniline (PANI), polyethylene (PE), polyimide (PI), polystyrene (PS), polyurethane (PU), polyvinyl butyral (PVB), polyvinyl pyrrolidone (PVP). In some embodiments. Another exemplary polymer material is fluorine acrylic hybrid Latex (TRD202A), and is supplied by JSR Corporation.

Examples

The following non-limiting examples further describe the application of the teachings of this disclosure. In the following examples, the term “binder free” or “binderless” electrodes reference to electrodes of the type described in detail above featuring a 3D matrix or scaffold of high aspect ratio carbons which a surface treatment thereon which promotes adhesion of active material to the scaffold without the need for bulk polymeric binders such as PVDF.

As used in the following the term C-rate refers to a measure of the rate at which a battery is discharged relative to its maximum capacity. A 1C rate means that the discharge current will discharge the entire battery in 1 hour. For a battery with a capacity of 100 Amp-hrs, this equates to a discharge current of 100 Amps.

Example 1 – Electric Vehicle Battery Cell

The following battery cell suitable for use in Electric Vehicles (“EV”). This cell combines cathode and anode technology of the type described herein for use, e.g., in an EV application. Key high-level benefits include lower cost to manufacture, higher energy density, excellent power density, and wide temperature

range operation. These benefits are derived from the herein described approach to manufacturing battery electrodes, which eliminates the use of PVDF polymer binders and toxic solvents like N-Methyl-2-pyrrolidone (NMP). The result is a substantial performance advantage in range, charging speed, and acceleration for the end-user with a manufacturing process that is lower cost, less capital intensive and safer
5 for the battery producers.

The teachings herein provide a technology platform to manufacture electrodes for energy storage which may exhibit the following advantages: reduction in cost of manufacturing and in the \$/kWh of resulting LIBs, increase in energy density by combining cathodes with thick coatings and high capacity anodes
10 featuring high performance active materials such as Si or SiO_x, fast charging. The teachings herein provide a scalable technology to improve power density in energy storage, by removing conventional polymer binders from the active material coatings.

Conventional electrodes for LIBs are fabricated by mixing an active material, conductive additives and a
15 polymer binder in a slurry. Conventional cathodes are manufactured using NMP-based slurries and PVDF polymer binders. Those binders have very high molecular weight and promote cohesion of active material particles and adhesion to the current collector foil via two main mechanisms: 1) the entanglement promoted by long polymer chains, and 2) hydrogen bonds between the polymer, the active material, and the current collector. However, the polymer binder-based method presents significant drawbacks in performance:
20 power density, energy density, and also cost to manufacture.

The teachings herein provide electrodes that do not have PVDF binders in cathodes, or other conventional binders in anodes. Instead, as detailed above a 3D carbon scaffold or matrix holds active material particles together to form a cohesive layer that is also strongly attached to the metallic current collector. Such active
25 material structure is created during slurry preparation and subsequently in a roll to roll ("R2R") coating and drying process. One of the main advantages of this technology is its scalability and "drop-in" nature since it compatible with conventional electrode manufacturing processes.

The 3D carbon matrix is formed during a slurry preparation using the techniques described herein: high
30 aspect ratio carbon materials are properly dispersed and chemically functionalized using, e.g., a 2-step slurry preparation process (such as the type described above with reference to Fig. 6). The chemical functionalization is designed to form an organized self-assembled structure with the surface of active material particles, e.g. NMC particles for use in a cathode or silicon particles ("Si") or Silicon Oxide ("SiO_x") particles in the case of an anode. The so formed slurry may be based on alcohol solvents for

cathodes and water for anodes, and such solvents are very easily evaporated and handled during the manufacturing process. Electrostatic interactions promote the self-organized structure in the slurry, and after the drying process the bonding between the so formed carbon matrix with active material particles and the surface of the current collector is promoted by the surface treatment (e.g., functional groups on the matrix) as well as the strong entanglement of the active material in the carbon matrix.

5

As will be understood by one skilled in the art, the mechanical properties of the electrodes can be readily modified depending on the application, and the mass loading requirements by tuning the surface functionalization vs. entanglement effect.

10

After coating and drying, the electrodes undergo a calendaring step to control the density and porosity of the active material. In NMC cathode electrodes, densities of 3.5 g/cc or more and 20% porosity or more can be achieved. Depending on mass loading and LIB cell requirements the porosity can be optimized. As for SiOx/Si anodes, the porosity is specifically controlled to accommodate active material expansion during the lithiation process.

15

In some typical applications, the teachings herein may provide a reduction in \$/kWh of up to 20%. By using friendly solvents that are easily evaporated, the electrode throughput is higher, and more importantly, the energy consumption from the long driers is significantly reduced. The conventional NMP recovery systems are also much simplified when alcohol or other solvent mixtures are used.

20

The teachings herein provide a 3D matrix that dramatically boosts electrode conductivity by a factor of 10X to 100X compared to electrodes using conventional binders such as PVDF, which enables fast charging at a battery level. Thick electrode coatings in cathode up to 150um per side (or more) of current collector are possible with this technology. The solvents used in the slurry in combination with a strong 3D carbon matrix are designed to achieve thick wet coatings without cracking during the drying step. Thick cathodes with high capacity anodes are what enable a substantial jump in energy density reaching 400Wh/kg or more.

25

Fast charging is achieved by combining high capacity anodes that are lithiated through an alloying process (Si/SiOx) and by reducing the overall impedance of the cell when combining anodes and cathodes as described herein. The teachings herein provide fast charging by having highly conductive electrodes, and in particular highly conductive cathode electrodes.

30

One exemplary embodiment includes a Li-ion battery energy storage devices in a pouch cell format that combines Ni-rich NMC active material in the cathodes and SiO_x and graphite blend active material in the anodes, where both anodes and cathodes are made using a 3D carbon matrix process as described herein.

5 A schematic of the electrode arrangement pouch cell devices is shown in Figure 7. As shown, a double-sided cathode using polymer binder free cathode layers on opposing sides of an aluminum foil current collector are disposed between two single sided anodes each having a polymer binder free anode layer disposed on a copper foil current collector. The electrodes are be separated by permeable separator material (not shown) wetted with electrolyte (not shown). The arrangement can be housed in a pouch cell of the
10 type well known in the art.

These devices may feature high mass loading of Ni-rich NMC cathode electrodes and their manufacturing method: mass loading = 20-30 mg/cm², specific capacity >210mAh/g. SiO_x/Graphite anode (SiO_x content ≈20 wt.%) based electrodes and their material synthesis and manufacturing method: mass loading 8-14
15 mg/cm², reversible specific capacity ≥ 550 mAh/g. Long life performance specially for SiO_x/Graphite anode based Li-ion based electrolyte for battery: from -30 to 60 °C. High-energy, high-power density, and long cycle life Ni-rich NMC cathode / SiO_x + Graphite/Carbon + based Li-ion battery pouch cells: capacity ≥ 5 Ah, Specific Energy ≥ 300 Wh/kg, Energy Density ≥ 800 Wh/L, with a cycle life of more than 500 cycles under 1C-Rate charge-discharge, and ultra-high-power fast charge-discharge C-Rate (Up to 5C-
20 Rate) capabilities. A summary of performance parameters for a pouch cell of this type are summarized in Figure 8.

Example 2 – Comparative Performance NMC811 Lithium Ion Battery

25 As detailed above, the teachings herein provide electrodes configured with an advanced 3-D high aspect ratio carbon binding structure that eliminates the need for polymer binders, providing greater power, energy density (e.g., via thicker electrodes and higher mass loading of active material), and performance in extreme environments compared to traditional battery electrode designs. The high-performance Li-ion battery energy storage devices are designed and manufactured with an optimized capacity ratio design of binder-
30 free cathode/anode electrodes, anode electrode pre-lithiation, and wide operating temperature electrolyte (e.g., -30 to 60 °C), and optimized test formation processes.

As described herein, the electrodes are manufactured by completely removing high molecular weight polymers such as PVDF and the toxic NMP solvent from the active material layer. This dramatically

improves LiB performance while decreasing the cost of manufacturing and the capital expenditures related to mixing, coating and drying, NMP solvent recovery, and calendaring. In embodiments of the electrodes, a 3D nanoscopic carbon matrix works as a mechanical scaffold for the electrode active material and mimics the polymer chain entanglement. Chemical bonds are also present between the surface of the carbon, the active materials, and the current collector promoting adhesion and cohesion. As opposed to polymers, however, the 3D nanoscopic carbon matrix is very electrically conductive, which enables very high power (high C-rates). This scaffold structure is also more suitable for producing thick electrode active material, which is a powerful way to increase the energy density of LiB cells.

10 In the present example, a binder free cathode was produced according to the teachings of this disclosure featuring a NMC811 as an active material and incorporated in a Li-ion battery (LIB). The cell featured a graphite anode of the conventional type known in the art. The cell was constructed as described above with reference to Figure 7 using the parameters summarized in Figure 9. A conventional electrolyte was used composed of 1M of LiPF₆ in a solvent mixture of ethylene carbonate and dimethyl carbonate with 1% by weight vinyl carbonate additive. As a comparison, an otherwise identical cell was produced using a PVDF binder based cathode. The performance of the cells was compared as described below, showing clear advantages for the binder free cathode cell.

As shown in the results summarized in Figure 10, the binder free cell can reach a specific energy as high as 20 320 Wh/kg based on 20 Ah battery cell design and a graphite anode with more than 2,000 cycle cycle life under 2C-rate charge/discharge. In comparison, the conventional binder-based cathode cell can only achieve 100-250 Wh/kg in specific energy at the cell level.

The binder free cathode cell exhibits ultra-high-power fast charge-discharge C-Rate, up to 5C-Rate with >50% capacity retention. Figure 10 shows a comparison of the charge-discharge curves at various C-rates for the binder free cathode cell (left) and the conventional binder-based cathode cell (right). The binder free cathode cell charge-discharge curve shows over 60% capacity retention of a combined charge-discharge at a 5C rate. Accordingly, separate discharge or charge would exhibit even higher capacity retention. Note in the example provided a conventional graphite anode is used, initial experimental results show that when a Si-dominant anode is combined with NMC811 cathode used in the present example, 10C charge rate is achievable.

Figure 11 shows a comparison of the cycle life of the above described cells. The cells were repetitively cycled between voltages of 2.75V and 4.2V at 25 °C, and the discharge capacity recorded. The binder free

cathode cell exhibits a lifetime of greater than 2,000 cycles with discharge capacity loss of less than 20%. In contrast the binder-based cathode cell experiences greater than 20% discharge capacity loss after only about 1,000 cycles.

5 Example 3 – Pouch Half Cell Comparison

Binder free cathode electrodes of the type described herein can advantageously achieve high mass loadings for example, a mass loading of 45 mg/cm² per side of NMC811 active material is possible. The present example sets forth experimental results showing the performance of such a high mass loading binder free electrode in comparison with a control electrode featuring PVDF binder and an NMC811 active material.

To perform the comparison, half-cells of the type shown in Figure. 12 were constructed using a one sided cathode (either binder free or the binder based control) and a lithium foil on copper substrate as the counter electrode for the cell. The half cells underwent charge rate testing under various current densities and the results summarized below.

Figure 13 is a plot showing potential (referenced to the Li/Li⁺ potential) vs specific capacity for binder free cathode half cell (solid traces) and reference binder based cathode half cell (dashed traces) at various current densities. At all current densities (and thus all C-rates), the binder free cathode half cells show better performance (as indicated by the relative rightward shift of the trace).

Figure 14 is a plot showing potential (referenced to the Li/Li⁺ potential) vs volumetric capacity for binder free cathode have cell (solid traces) and reference binder based cathode half cell (dashed traces) at various current densities. At all current densities (and thus all C-rates), the binder free cathode half cells show better performance (as indicated by the relative rightward shift of the trace).

Figure 15 shows a plot of volumetric capacity vs current density for binder free cathode half cells (upper trace) and reference binder based cathode half cell (lower trace). At all current densities (and thus all C-rates), the binder free cathode half cells show better performance, with the relative performance gap widening at higher C-rates.

Figure 16 shows a Nyquist plot resulting from electrochemical impedance spectroscopy for several binder free cathode half cells (square, circle and triangle labeled traces) and a reference binder based cathode half cells. The binder free cathode half cells exhibit significantly better performance than the reference cell.

It can be seen from the figures that when the current density increases from 0.5 to 10 mA/cm² (1.2 C-Rate), the discharge capacity retention for binder-free NMC811 electrode has a much higher value compared with a binder based PVDF control NMC811 electrode, even though both electrodes have the same mass loading of 45 mg/cm². Note that this C-Rate test under various current densities is presented as a relative comparison between conventional binder based PVDF cathodes and binder free cathodes, and does not reflect the absolute C-rate performance in a full cell configuration, e.g., as presented in the Examples 1 and 2 above.

10 Conclusion

Any terms of orientation provided herein are merely for purposes of introduction and are not limiting of the invention. For example, a “top” layer may also be referred to as a second layer, the “bottom” layer may also be referred to as a first layer. Other nomenclature and arrangements may be used without limitation of the teachings herein.

15 Various other components may be included and called upon for providing for aspects of the teachings herein. For example, additional materials, combinations of materials and/or omission of materials may be used to provide for added embodiments that are within the scope of the teachings herein.

A variety of modifications of the teachings herein may be realized. Generally, modifications may be designed according to the needs of a user, designer, manufacturer or other similarly interested party. The modifications may be intended to meet a particular standard of performance considered important by that party. Similarly, acceptability of performance is to be assessed by the appropriate user, designer, manufacturer or other similarly interested party.

While some chemicals may be listed herein as providing a certain function, a given chemical may be useful for another purpose.

25 When introducing elements of the present invention or the embodiment(s) thereof, the articles “a,” “an,” and “the” are intended to mean that there are one or more of the elements. Similarly, the adjective “another,” when used to introduce an element, is intended to mean one or more elements. The terms “including” and “having” are intended to be inclusive such that there may be additional elements other than the listed elements. As used herein, the term “exemplary” is not intended to imply a superlative example. Rather, “exemplary” refers to an embodiment that is one of many possible embodiments.

The entire contents of each of the publications and patent applications mentioned above are incorporated herein by reference. In the event that the any of the cited documents conflicts with the present disclosure, the present disclosure shall control.

5 Note that it is not intended that any functional language used in claims appended herein be construed as invoking 35 U.S.C. §112(f) interpretations as “means-plus-function” language unless specifically expressed as such by use of the words “means for” or “steps for” within the respective claim.

10 While the invention has been described with reference to exemplary embodiments, it will be understood that various changes may be made and equivalents may be substituted for elements thereof without departing from the scope of the invention. For example, in some embodiments, one of the foregoing layers may include a plurality of layers there within. In addition, many modifications will be appreciated to adapt a particular instrument, situation or material to the teachings of the invention without departing from the essential scope thereof. Therefore, it is intended that the invention not be limited to the particular embodiment disclosed as the best mode contemplated for carrying out this invention, but that the invention will include all embodiments falling within the scope of the appended claims.

15

CLAIMS

What is claimed is:

1. An apparatus comprising:

an electrode active layer comprising:

5 a network of high aspect ratio carbon elements defining void spaces within the network;

a plurality of electrode active material particles disposed in the void spaces within the network and enmeshed in the network; and

10 a surface treatment on the surface of the high aspect ratio carbon elements which promotes adhesion between the high aspect ratio carbon elements and the active material particles.

2. The apparatus of any preceding claim, wherein the high aspect ratio carbon elements comprise elements each having two major dimensions and one minor dimension, wherein the ratio of the length of each of the major dimensions is at least 10 times that of the minor dimension.

15 3. The apparatus of any preceding claim, wherein the high aspect ratio carbon elements comprise elements each having two major dimensions and one minor dimension, wherein the ratio of the length of each of the major dimensions is at least 100 times that of the minor dimension.

20 4. The apparatus of any preceding claim, wherein the high aspect ratio carbon elements comprise elements each having two major dimensions and one minor dimension, wherein the ratio of the length of each of the major dimensions is at least 1,000 times that of the minor dimension.

5. The apparatus of any preceding claim, wherein the high aspect ratio carbon elements comprise elements each having two major dimensions and one minor dimension, wherein the ratio of the length of each of the major dimensions is at least 10,000 times that of the minor dimension.

25 6. The apparatus of any preceding claim, wherein the high aspect ratio carbon elements comprise elements each having one major dimension and two minor dimension, wherein the ratio of the length of each the major dimension is at least 10 times that of each of the minor dimensions.

7. The apparatus of any preceding claim, wherein the high aspect ratio carbon elements comprise elements each having one major dimension and two minor dimension, wherein the ratio of the length of each the major dimension is at least 100 times that of each of the minor dimensions.

30 8. The apparatus of any preceding claim, wherein the high aspect ratio carbon elements comprise elements each having one major dimension and two minor dimension, wherein the ratio of the length of each the major dimension is at least 1,000 times that of each of the minor dimensions.

35 9. The apparatus of any preceding claim, wherein the high aspect ratio carbon elements comprise elements each having one major dimension and two minor dimension, wherein the ratio of the length of each the major dimension is at least 10,000 times that of each of the minor dimensions.

10. The apparatus of any preceding claim wherein the high aspect ratio carbon elements comprise carbon nanotubes or carbon nanotube bundles.
11. The apparatus of any preceding claim wherein the high aspect ratio carbon elements comprise graphene flakes.
- 5 12. The apparatus of any preceding claim wherein the electrode active layer contains less than 10% by weight polymeric binders disposed in the void spaces.
13. The apparatus of any preceding claim wherein the electrode active layer contains less than 1% by weight polymeric binders disposed in the void spaces.
14. The apparatus of any preceding claim wherein the electrode active layer contains less than
10 1% by weight polymeric binders disposed in the void spaces.
15. The apparatus of any preceding claim wherein the electrode active layer is substantially free of polymeric material other than the surface treatment.
16. The apparatus of any preceding claim wherein the electrode active layer is substantially free of polymeric material.
- 15 17. The apparatus of any preceding claim, wherein the surface treatment comprises a material which is soluble in a solvent having a boiling point less than 202° C.
18. The apparatus of any preceding claim, wherein the surface treatment comprises a material which is soluble in a solvent having a boiling point less than 185° C.
19. The apparatus of any preceding claim, wherein during the formation of the active layer, a
20 material forming the surface treatment was dissolved in a solvent having a boiling point less than 202° C.
20. The apparatus of any preceding claim, wherein during the formation of the active layer, a material forming the surface treatment was dissolved in a solvent having a boiling point less than 185° C.
- 25 21. The apparatus of any preceding claim, wherein during the formation of the active layer, a material forming the surface treatment was dissolved in a solvent comprising iso-propyl alcohol.
22. The apparatus of any preceding claim, wherein during the formation of the active layer, a material forming the surface treatment was dissolved in a solvent substantially free of n-methyl-2-pyrrolidone.
- 30 23. The apparatus of any preceding claim, wherein during the formation of the active layer, a material forming the surface treatment was dissolved in a solvent substantially free of pyrrolidone compounds.
24. The apparatus of any preceding claim, wherein the network is at least 90% carbon by weight.

25. The apparatus of any preceding claim, wherein the network is at least 95% carbon by weight.
26. The apparatus of any preceding claim, wherein the network is at least 99% carbon by weight.
- 5 27. The apparatus of any preceding claim, wherein the network is at least 99.9% carbon by weight.
28. The apparatus of any preceding claim, wherein the network comprises an electrically interconnected network of carbon elements exhibiting connectivity above a percolation threshold.
- 10 29. The apparatus of any preceding claim, wherein the network defines one or more highly electrically conductive pathways.
30. The apparatus of claim 29, wherein said the pathways have comprising a length greater than 100 μm .
31. The apparatus of claim 29, wherein said the pathways have comprising a length greater than
15 1,000 μm .
32. The apparatus of claim 29, wherein said the pathways have comprising a length greater than 10,000 μm .
33. The apparatus of any preceding claim, wherein the network includes one or more structures formed of the carbon elements, said structure comprising an overall length at least ten times the
20 length of a largest dimension the carbon elements.
34. The apparatus of any preceding claim, wherein the network includes one or more structures formed of the carbon elements, said structure comprising an overall length at least 100 times the length of a largest dimension the carbon elements.
35. The apparatus of any preceding claim, wherein the network includes one or more structures
25 formed of the carbon elements, said structure comprising an overall length at least 1,000 times the length of a largest dimension the carbon elements.
36. The apparatus of any preceding claim, wherein the surface treatment comprises a surfactant layer disposed on the carbon elements.
37. The apparatus of claim 36, wherein the surfactant layer is bonded to the carbon elements.
- 30 38. The apparatus of claim 36 or 37, wherein surfactant layer comprises a plurality of surfactant elements each having a hydrophobic end and a hydrophilic end, wherein the hydrophobic end is disposed proximal a surface one of the carbon elements and the hydrophilic end is disposed distal said surface one of the carbon elements.
- 35 39. The apparatus of claim 38, wherein the hydrophilic ends of at least a portion of the surfactant elements form bonds with the active material particles.

40. The apparatus of claim 39, wherein the bonds comprise ionic bonds.
41. The apparatus of claim 39, wherein the bonds comprise covalent bonds.
42. The apparatus of claim 39, wherein the bonds comprise at least one from the list consisting of: π - π bonds, hydrogen bonds, and electrostatic bonds.
- 5 43. The apparatus of claim 38, wherein:
- the hydrophilic end of the surfactant element has a polar charge of a first polarity; and
- the active material particles carry a polar charge of a second polarity opposite that of the first polarity.
- 10 44. The apparatus of any of claims 36-43, wherein the surfactant layer comprises a water soluble surfactant.
45. The apparatus of any of claims 36-44, wherein the surfactant layer comprises ions from hexadecyltrimethylammonium hexafluorophosphate
46. The apparatus of any of claims 36-45, wherein the surfactant layer comprises ions from at least one from the list consisting of: hexadecyltrimethylammonium tetrafluoroborate, N-
15 (Cocoalkyl)-N,N,N-trimethylammonium methyl sulfate, cocamidopropyl betaine hexadecyltrimethylammonium acetate, and hexadecyltrimethylammonium nitrate
47. The apparatus of any of claims 36-46, wherein the surfactant layer comprises a layer of surfactant ions formed from dissolving an ionic compound in a solvent.
- 20 48. The apparatus of claim 47, wherein the active layer comprises residual counter ions to the surfactant ions formed by dissolving an ionic surfactant compound in a solvent.
49. The apparatus of claim 48, wherein the counter ions are selected to be compatible with use in an electrochemical cell.
50. The apparatus of claim 49, wherein the counter ions are substantially free of halide groups.
- 25 51. The apparatus of any of claims 48-50 wherein the residual counter ions are substantially free of bromine.
52. The apparatus of any of claims 51 wherein the ionic surfactant compound comprises at least one selected from the list consisting of: hexadecyltrimethylammonium tetrafluoroborate, hexadecyltrimethylammonium tetrafluoroborate, N-(Cocoalkyl)-N,N,N-trimethylammonium methyl sulfate, cocamidopropyl betaine hexadecyltrimethylammonium acetate, and
30 hexadecyltrimethylammonium nitrate.
53. The apparatus of any preceding claim, wherein the carbon elements are functionalized.
54. The apparatus of claim 53 wherein the carbon elements are functionalized with a surfactant material.

55. The apparatus of claim 53 or 54, wherein the carbon elements are functionalized are functionalized with functional groups which promote adhesion of the active material particles to the network.
56. The apparatus of claim 55, wherein the functional groups comprise at least one from the list consisting of: carboxylic groups, hydroxylic groups, amine groups, and silane groups.
57. The apparatus of any of claims 48-56, wherein the functionalized carbon elements are formed from dried aqueous dispersion comprising nanoform carbon and surfactant.
58. The apparatus of claim 57, wherein the functionalized carbon elements are formed from a lyophilized aqueous dispersion comprising nanoform carbon and surfactant.
59. The apparatus of any of claims 57-58, wherein the aqueous dispersion is substantially free of acids.
60. The apparatus of any preceding claim, wherein the surface treatment comprises a thin polymeric layer disposed on the carbon elements that promotes adhesion of the active material to the network.
61. The apparatus of claim 60, wherein the thin polymeric layer comprises a self-assembled polymer.
62. The apparatus of any of claims 60-61, wherein the thin polymeric layer bonds to the active material via hydrogen bonding.
63. The apparatus of any of claims 60-62, wherein the thin polymeric layer has a maximum thickness in a direction normal to an outer surface of the network of less than or equal to 1 nm.
64. The apparatus of any of claims 60-63, wherein the thin polymeric layer has a maximum thickness in a direction normal to an outer surface of the network of less than or equal to 10 nm.
65. The apparatus of any of claims 60-64, wherein the thin polymeric layer has a maximum thickness in a direction normal to an outer surface of the network of less than or equal to 50 nm.
66. The apparatus of any of claims 60-65, wherein less than 1% by volume of the void spaces defined by the network are filled with the thin polymeric layer.
67. The apparatus of any of claims 60-66, wherein less than 0.1% by volume of the void spaces defined by the network are filled with the thin polymeric layer.
68. The apparatus of any of claims 60-67, wherein less than 0.1% by volume of the void spaces defined by the network are filled with the thin polymeric layer.
69. The apparatus of any of claims 1-35, wherein the surface treatment comprises a layer of carbonaceous material formed from pyrolyzed polymeric material.
70. The apparatus of claim 69, wherein the layer of carbonaceous material formed from pyrolyzed polymeric material promotes adhesion of the active material particles to the network.

71. The apparatus of any preceding claim wherein the active material particles comprise a metal oxide.
72. The apparatus of any preceding claim wherein the active material particles comprise a lithium metal oxide.
- 5 73. The apparatus of any preceding claim wherein the active material is entangled in the network.
74. The apparatus of any preceding claim, wherein the surface treatment promotes adhesion of the active material layer and a current collector layer.
75. The apparatus of claim 74, wherein the surface treatment includes functional groups bonded
10 with the current collector layer.
76. The apparatus of claim 75, wherein the functional groups are bond with the current collector layer with non-covalent bonds.
77. The apparatus of claim 75, wherein the functional groups are bond with the current collector layer with at least one selected from the list consisting of: π - π bonds, hydrogen bonds, and ionic
15 bonds.
78. The apparatus of any of claims 74-77, wherein the current collector comprises a metal foil.
79. The apparatus of any of claims 74-77, wherein the active material layer has a thickness in the direction normal to the current collector of at least 200 μm .
80. The apparatus of any of claims 74-77, wherein the active material layer has a thickness in
20 the direction normal to the current collector of at least 300 μm .
81. The apparatus of any of claims 74-77, wherein the active material layer has a thickness in the direction normal to the current collector of at least 400 μm .
82. The apparatus of any preceding claim, further comprising an energy storage cell, the energy storage cell comprising:
- 25 a first electrode comprising the active material layer;
- a second electrode;
- a permeable separator disposed between the first electrode and the second electrode;
- and an electrolyte wetting the first and second electrodes.
- 30
83. A method comprising:

dispersing high aspect ratio carbon elements and a surface treatment material in a solvent to form an initial slurry, wherein said dispersion step results in the formation of a surface treatment on the high aspect ratio carbon;

mixing active materials into the first slurry to form a final slurry;

5 coating the final slurry onto a substrate; and

drying the final slurry to form an electrode active layer.

84. The method claim 83, wherein the high aspect ratio carbon elements comprise elements each having two major dimensions and one minor dimension, wherein the ratio of the length of each of the major dimensions is at least 10 times that of the minor dimension.

10 85. The method of claim 83 or 84, wherein the high aspect ratio carbon elements comprise elements each having two major dimensions and one minor dimension, wherein the ratio of the length of each of the major dimensions is at least 100 times that of the minor dimension.

86. The method of any of claims 83-85, wherein the high aspect ratio carbon elements comprise elements each having two major dimensions and one minor dimension, wherein the ratio of the length of each of the major dimensions is at least 1,000 times that of the minor dimension.

87. The method of any of claims 83-86, wherein the high aspect ratio carbon elements comprise elements each having two major dimensions and one minor dimension, wherein the ratio of the length of each of the major dimensions is at least 10,000 times that of the minor dimension.

20 88. The method of any of claims 83-87, wherein the high aspect ratio carbon elements comprise elements each having one major dimension and two minor dimension, wherein the ratio of the length of each the major dimension is at least 10 times that of each of the minor dimensions.

89. The method of any of claims 83-88, wherein the high aspect ratio carbon elements comprise elements each having one major dimension and two minor dimension, wherein the ratio of the length of each the major dimension is at least 100 times that of each of the minor dimensions.

25 90. The method of any of claims 83-89, wherein the high aspect ratio carbon elements comprise elements each having one major dimension and two minor dimension, wherein the ratio of the length of each the major dimension is at least 1,000 times that of each of the minor dimensions.

30 91. The method of any of claims 83-90, wherein the high aspect ratio carbon elements comprise elements each having one major dimension and two minor dimension, wherein the ratio of the length of each the major dimension is at least 10,000 times that of each of the minor dimensions.

92. The method of any of claims 83-91, wherein the high aspect ratio carbon elements comprise carbon nanotubes or carbon nanotube bundles.

93. The method of any of claims 83-92, wherein the high aspect ratio carbon elements comprise graphene flakes.

35 94. The method of any of claims 83-93, wherein the initial slurry has a solid content in the range of 0.1%-20.0% by weight.

95. The method of any of claims 83-94, wherein the final slurry has a solid content in the range of 10.0% - 80% by weight.
96. The method of any of claims 83-95, wherein the solvent has a boiling point less than 202 °C.
97. The method of any of claims 83-96, wherein the solvent has a boiling point less than 185 °C.
- 5 98. The method of any of claims 83-97, wherein the solvent has a boiling point less than 125 °C.
99. The method of any of claims 83-98, wherein the solvent has a boiling point less than or equal to 100 °C.
100. The method of any of claims 83-99, wherein the solvent comprises at least one from the list consisting of: methanol, ethanol, 2-propanol, and water.
- 10 101. The apparatus of any of claims 83-100, wherein during the formation of the active layer, a material forming the surface treatment was dissolved in a solvent substantially free of pyrrolidone compounds.
102. The method of any of claims 83-101, the solvent is substantially free of n-methyl-2-pyrrolidone.
- 15 103. The method of any of claims 83-102, wherein the surface treatment material comprises a surfactant.
104. The method of claim 103, wherein surfactant is substantially free of halide groups.
105. The method of claim 104, wherein surfactant is substantially free of bromine.
106. The method of any of claims 83-105, wherein forming the surface treatment comprises a forming a surfactant layer disposed on the carbon elements.
- 20 107. The method of any of claims 83-106, wherein the surface treatment is a self-assembled layer.
108. The method of claim 106 or 107, wherein forming the surfactant layer comprises disposing a plurality of surfactant elements on a surface of the carbon elements, each of the surfactant elements having a hydrophobic end and a hydrophilic end, wherein the hydrophobic end is disposed proximal a surface one of the carbon elements and the hydrophilic end is disposed distal said surface one of the carbon elements.
- 25 109. The method of claim 108, further comprising causing the hydrophobic ends of at least a portion of the surfactant elements form bonds with the active material particles.
- 30 110. The method of claim 109, wherein the bonds comprise ionic bonds.
111. The method of claim 109, wherein the bonds comprise covalent bonds.
112. The method of claim 109, wherein the bonds comprise at least one from the list consisting of: π - π bonds, hydrogen bonds, and electrostatic bonds.

113. The method of claim 108, wherein:
- the hydrophilic end of the surfactant element has a polar charge of a first polarity;
 - the active material particles carry a polar charge of a second polarity opposite that of the first polarity.
- 5 114. The method of any of claims 103-113, wherein the surfactant material comprises at least one selected from the list consisting of: hexadecyltrimethylammonium tetrafluoroborate, hexadecyltrimethylammonium tetrafluoroborate, N-(Cocoalkyl)-N,N,N-trimethylammonium methyl sulfate, cocamidopropyl betaine hexadecyltrimethylammonium acetate, and hexadecyltrimethylammonium nitrate.
- 10 115. The method of any of claim 83-114, wherein dispersing high aspect ratio carbon elements and a surface treatment material in a solvent to form an initial slurry comprises applying forces to agglomerated carbon elements to cause said elements to slide apart from each other along a direction transverse to a minor axis of the elements.
116. The method of any of claim 83-114, comprising drying the final slurry at a temperature of
15 less than 202° C.
117. The method of any of claim 83-114, comprising drying the final slurry at a temperature of less than 185° C.
118. The method of any of claim 83-114, comprising drying the final slurry at a temperature of less than 125° C.
- 20 119. The method of any of claim 83-114, comprising drying the final slurry at a temperature less than or equal to 100° C.
120. The method of any of claims 83-119, further comprising calendaring the active layer to promote adhesion between the active material and the network.
121. A method comprising:
- 25 dispersing high aspect ratio carbon elements and a surface treatment material in an aqueous solvent to form an initial slurry, wherein said dispersion step results in the formation of a surface treatment on the high aspect ratio carbon; and
- drying the initial slurry to remove substantially all moisture resulting in a dried powder of the high aspect ratio carbon with the surface treatment thereon.
- 30 122. The method of claim 121, wherein drying the initial slurry comprises lyophilizing the initial slurry.
123. The method of claim 121 or claim 122, wherein the aqueous solvent and initial slurry are substantially free of substances damaging to the high aspect ratio carbon elements.
124. The method of claim 123, wherein the aqueous solvent and initial slurry are substantially
35 free of acids.

125. The method of claim 124, wherein the initial slurry consists essentially of the high aspect ratio carbon elements, the surface treatment material, and water.

126. The method of any of claims 121-125, further comprising:

5 dispersing the dried powder of the high aspect ratio carbon with the surface treatment in a solvent and adding an active material to form a secondary slurry;

coating the secondary slurry onto a substrate; and

drying the secondary slurry to form an electrode active layer.

10 127. The method of any of claims 121-126, wherein the high aspect ratio carbon elements comprises elements each having two major dimensions and one minor dimension, wherein the ratio of the length of each of the major dimensions is at least 10 times that of the minor dimension.

15 128. The method of any of claims 121-127, wherein the high aspect ratio carbon elements comprise elements each having two major dimensions and one minor dimension, wherein the ratio of the length of each of the major dimensions is at least 100 times that of the minor dimension.

129. The method of any of claims 121-128, wherein the high aspect ratio carbon elements comprise elements each having two major dimensions and one minor dimension, wherein the ratio of the length of each of the major dimensions is at least 1,000 times that of the minor dimension.

20 130. The method of any of claims 121-129, wherein the high aspect ratio carbon elements comprise elements each having two major dimensions and one minor dimension, wherein the ratio of the length of each of the major dimensions is at least 10,000 times that of the minor dimension.

25 131. The method of any of claims 121-130, wherein the high aspect ratio carbon elements comprise elements each having one major dimension and two minor dimensions, wherein the ratio of the length of each the major dimension is at least 10 times that of each of the minor dimensions.

30 132. The method of any of claims 121-131, wherein the high aspect ratio carbon elements comprise elements each having one major dimension and two minor dimensions, wherein the ratio of the length of each the major dimension is at least 100 times that of each of the minor dimensions.

35 133. The method of any of claims 121-132, wherein the high aspect ratio carbon elements comprise elements each having one major dimension and two minor dimensions, wherein the ratio of the length of each the major dimension is at least 1,000 times that of each of the minor dimensions.

134. The method of any of claims 121-133, wherein the high aspect ratio carbon elements comprise carbon nanotubes or carbon nanotube bundles.

135. The method of any of claims 121-134, wherein the high aspect ratio carbon elements comprise graphene flakes.
136. The method of any of claims 121-135, wherein the solvent has a boiling point less than 202° C.
- 5 137. The method of any of claims 121-135, wherein the solvent has a boiling point less than 185° C.
138. The method of any of claims 121-135, wherein the solvent has a boiling point less than 125° C.
- 10 139. The method of any of claims 121-135, wherein the solvent has a boiling point less than or equal to 100° C.
140. The method of claims 121-139, wherein the secondary solvent comprises at least one from the list consisting of: methanol, ethanol, 2-propanol, and water.
141. The method of any of claims 121-140, wherein the secondary solvent is substantially free of pyrrolidone compounds.
- 15 142. The method of any of claims 121-141, wherein the secondary solvent is substantially free of n-methyl-2-pyrrolidone.
143. The method of any of claims 121-142, wherein the surface treatment material comprises a surfactant.
144. The method of claim 143, wherein surfactant is substantially free of halide groups.
- 20 145. The method of claim 144, wherein surfactant is substantially free of bromine.
146. The method of any of claims 121-145, wherein forming the surface treatment comprises a forming a surfactant layer disposed on the carbon elements.
147. The method of claim 146, wherein the surfactant layer is a self-assembled layer.
148. The method of any of claims 126-147, comprising drying the secondary slurry at a
25 temperature of less than 202° C.
149. The method of any of claims 126-147, comprising drying the secondary slurry at a temperature of less than 185° C.
150. The method of any of claims 126-147, comprising drying the secondary slurry at a temperature of less than 125° C.
- 30 151. The method of any of claims 126-147, comprising drying the secondary slurry at a temperature less than or equal to 100° C.
152. The method of any of claims 121-151, further comprising calendaring the active layer to promote adhesion.

FIGURES

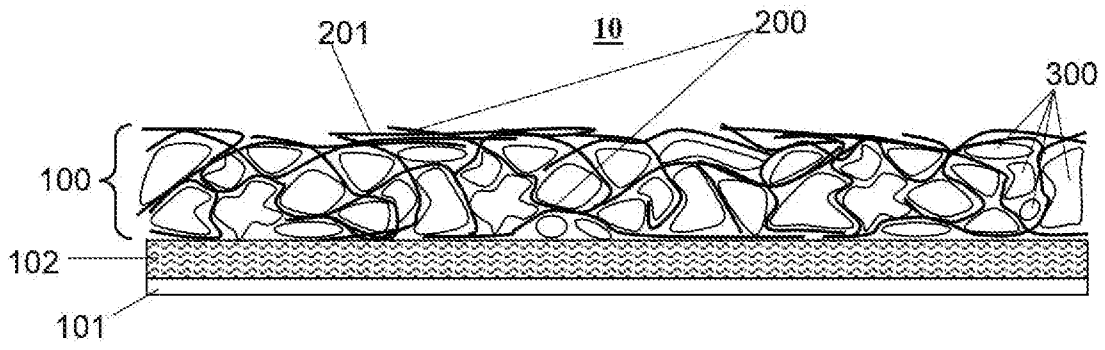


Figure 1

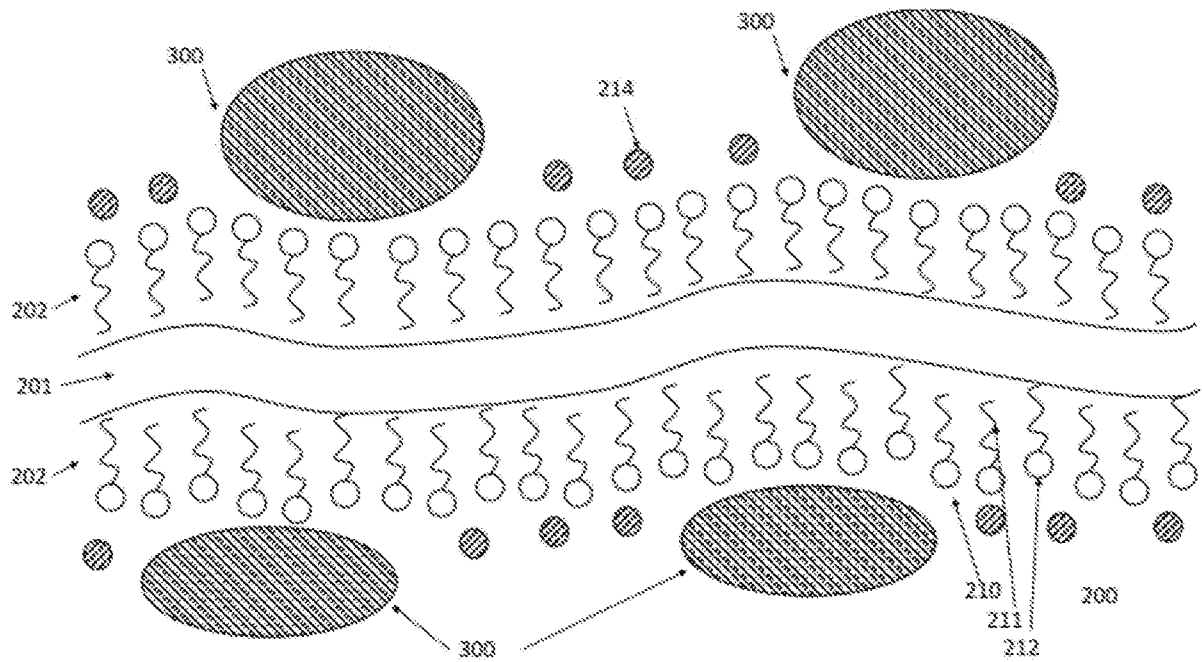


Figure 2

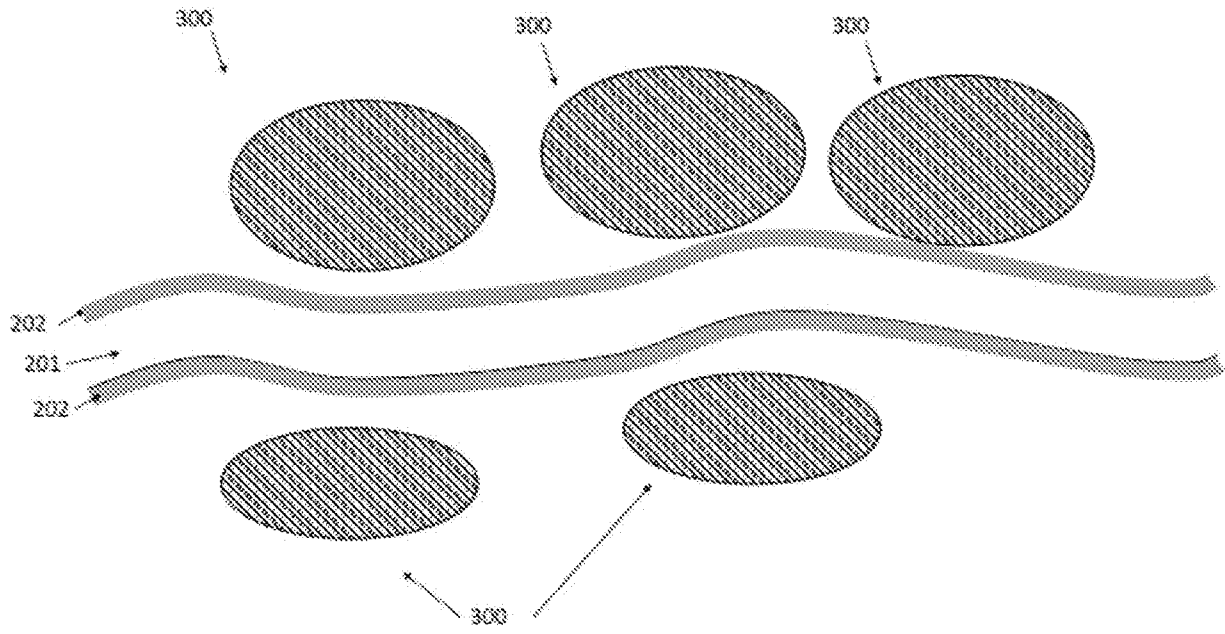


Figure 3

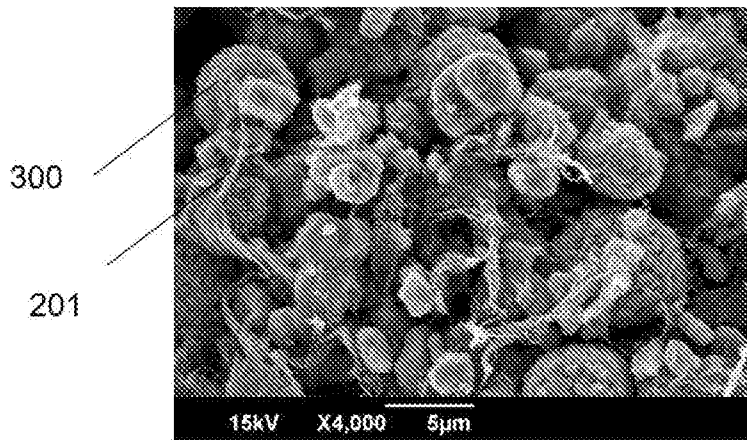


Figure 4

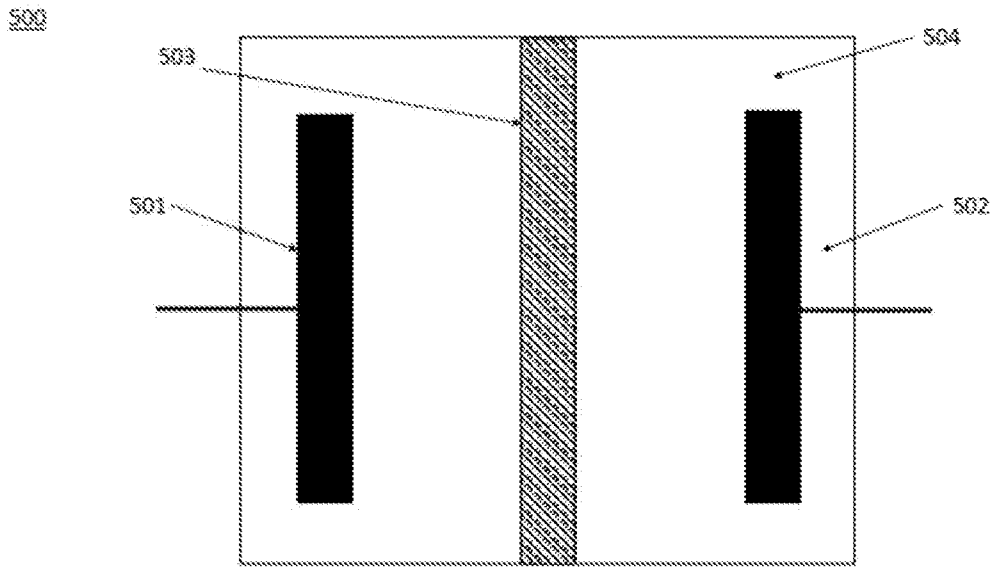


Figure 5

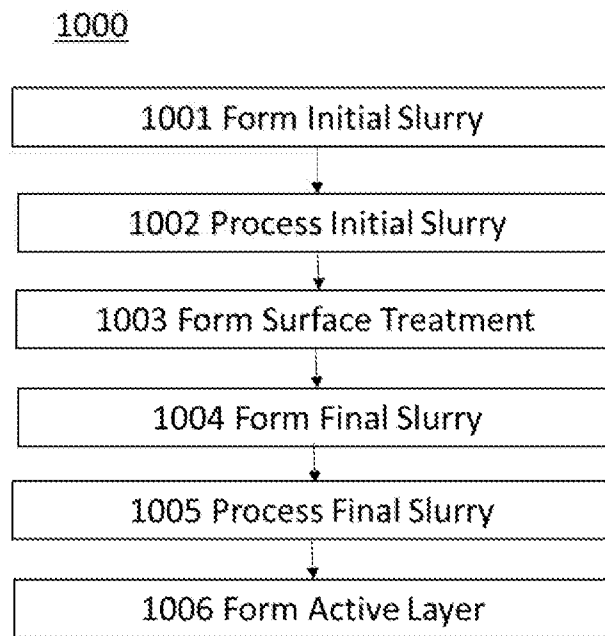


Figure 6

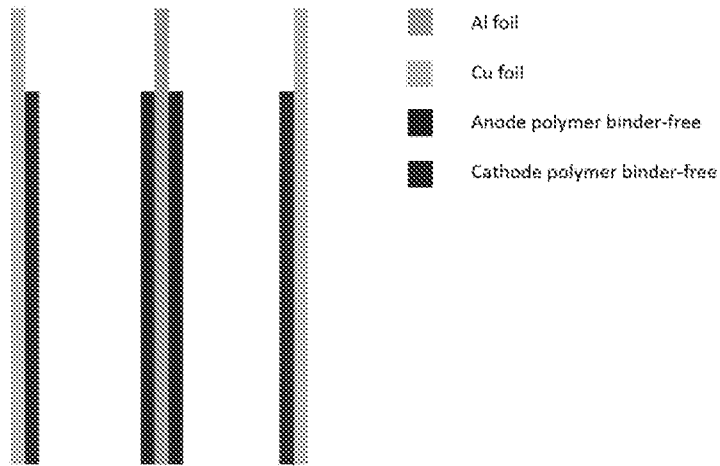


Figure 7

Neocarbonix LIB Pouch Cell Parameters	Descriptions
Cell Capacity	≥ 5 Ah
Specific Energy	≥ 300 Wh/kg
Energy Density	≥ 800 Wh/L
Cycle Life under 1C-Rate Charge-Discharge	≥ 500 cycles
5C-Rate Charge-Discharge Capacity Retention	$\geq 50\%$

Figure 8

CATHODE		RESULTS	
Cathode active layer thickness (um)	150	Cathode active layer thickness (um)	150
Cathode battery material specific capacity (mAh/g)	220	Anode active layer thickness (um)	135.23
Cathode nano-carbon material specific capacity (mAh/g)	33	Separator thickness (um)	20
Battery material to total cathode weight (%)	97	Total cathode weight (g/cm ²) (+electrolyte)	0.0549
Binder to total cathode weight (%)	0	Total anode weight (g/cm ²) (+electrolyte)	0.0413
Cathode electrode porosity (%)	30	Total electrolyte weight (g/cm ²)	0.0143
Al foil thickness (um) (X2)	12		
ANODE			
Anode active material specific capacity (mAh/g)	372		
Binder to total anode weight (%)	0		
Lithium to anode material weight (%)	0		
Anode electrode porosity (%)	30		
Anode capacity/cathode capacity	1.1	Specific energy (w/o package) (Wh/kg)	382.45
Cu foil thickness (um) (X2)	8	Specific energy (including package) (Wh/kg)	325.08
SEPARATOR			
Separator porosity (%)	50	Energy density (w/o package) (Wh/L)	1302.68
Separator thickness (um)	20	Energy density (including package) (Wh/L)	852.23
CELL VOLTAGE			
Upper voltage (V)	4.20		
Lower voltage (V)	3.00		
Battery cell voltage (V)	3.7		
Package efficiency (%)	85		

Figure 9

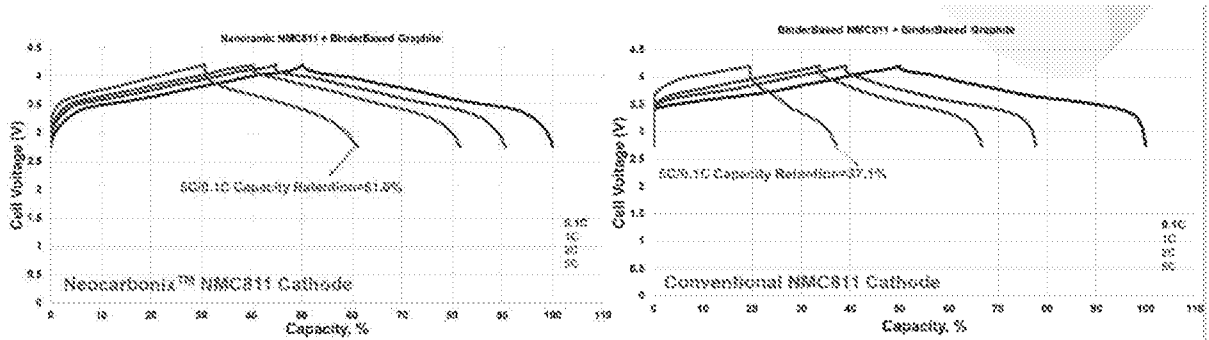


Figure 10

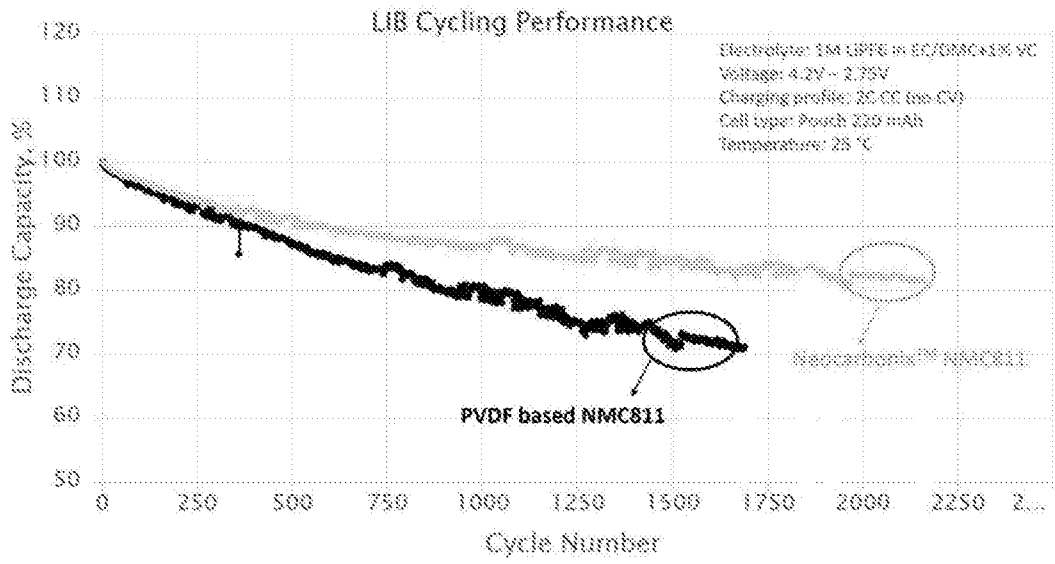


Figure 11

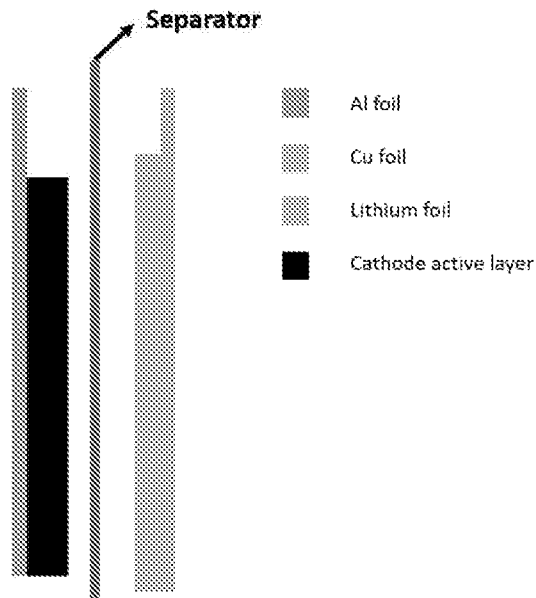


Figure 12

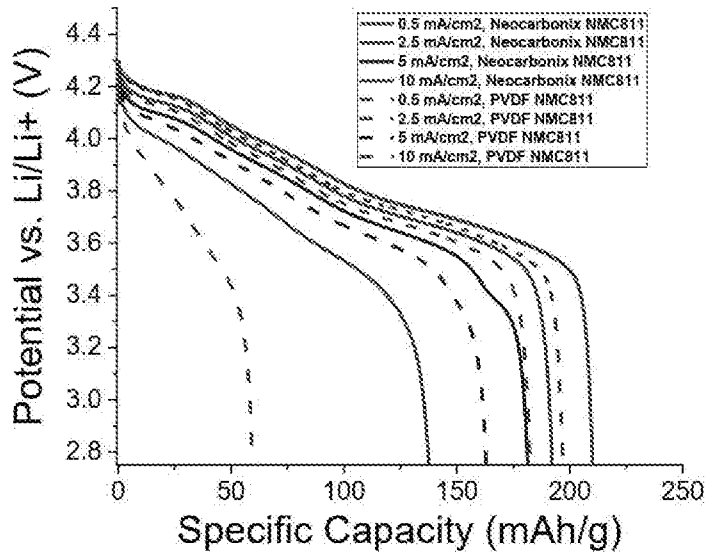


Figure 13

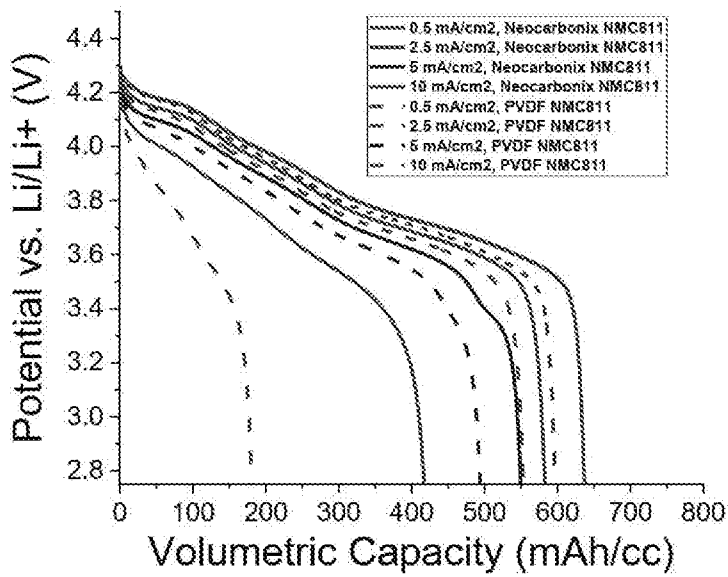


Figure 14

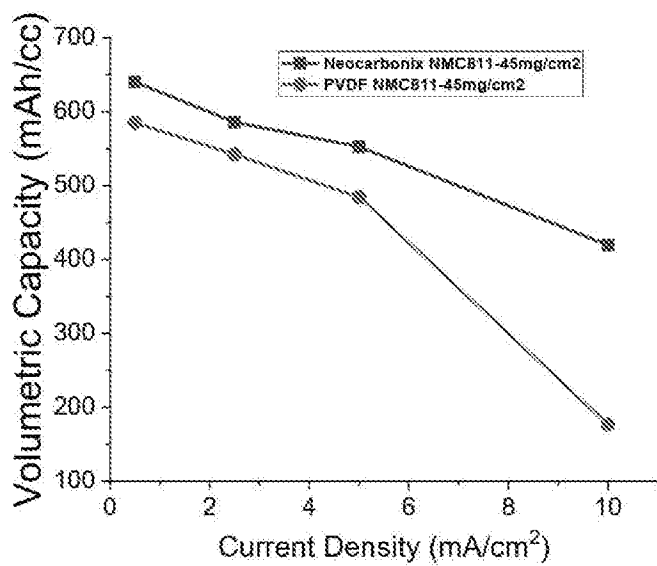


Figure 15

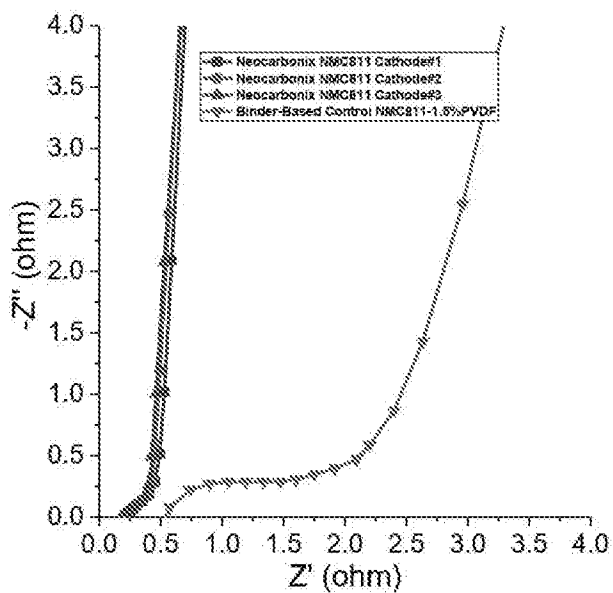


Figure 16

INTERNATIONAL SEARCH REPORT

International application No.

PCT/US 2020/040943

A. CLASSIFICATION OF SUBJECT MATTER		
<i>H01 M 4/133 (2010.01)</i> <i>H01 M 4/139 (2010.01)</i> <i>H01G 11/36 (2013.01)</i>		
According to International Patent Classification (IPC) or to both national classification and IPC		
B. FIELDS SEARCHED		
Minimum documentation searched (classification system followed by classification symbols)		
H01M 4/00-4/139, 10/00-10/0525, H01G11/00-11/36		
Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched		
Electronic data base consulted during the international search (name of data base and, where practicable, search terms used)		
PatSearch (RUPTO internal), USPTO, PAJ, Esp@cenet, DWPI, EAPATIS, PATENTSCOPE		
C. DOCUMENTS CONSIDERED TO BE RELEVANT		
Category*	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
X	WO 2011/117530 A1 (ARKEMA FRANCE) 29.09.2011, claims 1, 2, 12, 14	1-3, 83-85, 121, 123-125
Y		122
Y	CN108950736 A (GUIZHOU SENHUAN ACTIVATED CARBON CO LTD) 07.12.2018, abstract	122
A	RU 2282919 C1 (FEDOROV MIKHAIL ANATOLEVICH et al) 27.08.2006	1-3, 83-85, 121-125
A	EP 2081244 A1 (SAMSUNG ELECTRONICS CO LTD) 22.07.2009	1-3, 83-85, 121-125
<input type="checkbox"/> Further documents are listed in the continuation of Box C. <input type="checkbox"/> See patent family annex.		
*	Special categories of cited documents:	“T”
“A”	document defining the general state of the art which is not considered to be of particular relevance	later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention
“D”	document cited by the applicant in the international application	“X”
“E”	earlier document but published on or after the international filing date	document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step when the document is taken alone
“L”	document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified)	“Y”
“O”	document referring to an oral disclosure, use, exhibition or other means	document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art
“P”	document published prior to the international filing date but later than the priority date claimed	“&”
		document member of the same patent family
Date of the actual completion of the international search		Date of mailing of the international search report
24 September 2020 (24.09.2020)		01 October 2020 (01.10.2020)
Name and mailing address of the ISA/RU: Federal Institute of Industrial Property, Berezhkovskaya nab., 30-1, Moscow, G-59, GSP-3, Russia, 125993 Facsimile No: (8-495) 531-63-18, (8-499) 243-33-37		Authorized officer E. Kovaleva Telephone No. 8 499 240 25 91

Box No. II Observations where certain claims were found unsearchable (Continuation of item 2 of first sheet)

This international search report has not been established in respect of certain claims under Article 17(2)(a) for the following reasons:

1. Claims Nos.:
because they relate to subject matter not required to be searched by this Authority, namely:

2. Claims Nos.:
because they relate to parts of the international application that do not comply with the prescribed requirements to such an extent that no meaningful international search can be carried out, specifically:

3. Claims Nos.: 4-82, 85-120, 126-152
because they are dependent claims and are not drafted in accordance with the second and third sentences of Rule 6.4(a).

Box No. III Observations where unity of invention is lacking (Continuation of item 3 of first sheet)

This International Searching Authority found multiple inventions in this international application, as follows:

1. As all required additional search fees were timely paid by the applicant, this international search report covers all searchable claims.
2. As all searchable claims could be searched without effort justifying additional fees, this Authority did not invite payment of additional fees.
3. As only some of the required additional search fees were timely paid by the applicant, this international search report covers only those claims for which fees were paid, specifically claims Nos.:
4. No required additional search fees were timely paid by the applicant. Consequently, this international search report is restricted to the invention first mentioned in the claims; it is covered by claims Nos.:

Remark on Protest

- The additional search fees were accompanied by the applicant's protest and, where applicable, the payment of a protest fee.
- The additional search fees were accompanied by the applicant's protest but the applicable protest fee was not paid within the time limit specified in the invitation.
- No protest accompanied the payment of additional search fees.