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(54) Title: METHODS AND COMPOSITIONS FOR CEMENTING SUBTERRANEAN ZONES

(57) Abstract: Methods and compositions for cementing subterranean zones penetrated by well bores are provided. The methods are basically comprised of the steps of providing a cement composition which comprises a hydraulic cement, sufficient water to form a slurry and a dispersing agent comprising a tetrapolymer formed of methacrylic acid, methacrylate, methallyl sulfonic acid and ethoxylated acrylic acid. The cement composition is placed in the subterranean zone to be cemented and allowed to set into an impermeable solid mass therein.



METHODS AND COMPOSITIONS FOR CEMENTING SUBTERRANEAN ZONES

BACKGROUND OF THE INVENTION

1. FIELD OF THE INVENTION

The present invention relates to improved methods and cement compositions for cementing subterranean zones penetrated by well bores.

2. DESCRIPTION OF THE PRIOR ART

Subterranean zones penetrated by well bores are commonly sealed by hydraulic cement compositions. For example, hydraulic cement compositions are used in primary cementing operations whereby strings of pipe such as casings and liners are cemented in well bores. In performing primary cementing, a hydraulic cement composition is pumped into the annular space between the walls of a well bore and the exterior surfaces of a pipe string disposed therein. The cement composition is permitted to set in the annular space thereby forming an annular sheath of hardened substantially impermeable cement therein which supports and positions the pipe string in the well bore and seals the exterior surfaces of the pipe string to the walls of the well bore. Hydraulic cement compositions are also utilized in a variety of cementing operations such as sealing highly permeable zones or fractures in subterranean zones, plugging cracks or holes in pipe strings and the like.

Dispersing agents are commonly included in hydraulic cement compositions used for cementing subterranean zones. The dispersing agents reduce the rheologies of the cement compositions, i.e., the apparent viscosities and other properties of the cement compositions, as well as reducing the water content of the cement compositions. The reduction of the rheology of a cement composition allows the cement composition to be pumped with less friction pressure which utilizes less pump horsepower. In addition, the lower rheology often allows the cement composition to be pumped in turbulent flow. Turbulent flow characteristics are desirable when pumping cement compositions in wells in that drilling fluid is more efficiently removed from surfaces in the well bore as the drilling fluid is displaced by the cement composition being pumped. The inclusion of dispersing agents in cement compositions is also desirable in that the presence of the dispersing agents reduces the water required for preparation of the cement compositions. Cement compositions having reduced water content set into cement masses having improved compressive strength.

A number of dispersing agents have heretofore been utilized in well cement compositions. For example, certain organic acids such as gluconic acid and citric acid have been utilized as cement dispersing agents. However, such organic acids suffer from the disadvantage that they also function as cement composition set retarding agents which is often undesirable. That is, the presence of an organic acid dispersant in a cement composition prevents the cement composition from setting for a longer period of time which is often costly or otherwise detrimental. Other dispersing agents which are commonly used in well cement compositions include polynaphthalene sulfonate which is commercially available under the trade designation "CFR-2TM" from Halliburton Energy Services, Inc. of Duncan, Oklahoma; the condensation product of formaldehyde, acetone and a sulfite which is commercially available under the trade designation "CFR-3TM" from Halliburton Energy Services, Inc.; poly-B-naphthol sulfonate; polymelamine sulfonate; and many others. While the heretofore utilized dispersing agents have been used successfully, they generally require relatively high concentrations in well cement compositions and impart some set retardation to the cement compositions in which they are utilized. Further, some of the heretofore utilized dispersing agents also deteriorate the fluid loss control properties of cement composition fluid loss control agents.

Thus, there are continuing needs for improved methods of cementing and cement compositions wherein the cement compositions include a dispersing agent which can be utilized in smaller amounts, does not impart significant retardation properties to cement compositions and does not deteriorate the fluid loss control properties of fluid loss control agents.

SUMMARY OF THE INVENTION

The present invention provides improved methods and compositions for cementing subterranean zones penetrated by well bores which meet the needs described above and overcome the deficiencies of the prior art. The cement compositions of this invention are comprised of a hydraulic cement, sufficient water to form a slurry and a dispersing agent comprised of a tetrapolymer formed of methacrylic acid, methacrylate, methallyl sulfonic acid and ethoxylated acrylic acid.

The methods of the invention for cementing a subterranean zone penetrated by a well bore comprise the steps of providing a cement composition which comprises a hydraulic cement, sufficient water to form a slurry and a dispersing agent comprising a tetrapolymer

3

formed of methacrylic acid, methacrylate, methallyl sulfonic acid and ethoxylated acrylic acid, placing the cement composition in the subterranean zone to be cemented and allowing the cement composition to set into an impermeable solid mass therein.

The objects, features and advantages of the present invention will be readily apparent to those skilled in the art upon a reading of the description of preferred embodiments which follows.

DESCRIPTION OF PREFERRED EMBODIMENTS

As mentioned above, the well cement compositions of this invention are basically comprised of a hydraulic cement, sufficient water to form a slurry and a dispersing agent comprising a tetrapolymer formed of methacrylic acid, methacrylate, methallyl sulfonic acid and ethoxylated acrylic acid.

The methods of the present invention for cementing subterranean zones penetrated by a well bore are basically comprised of the following steps. A cement composition which comprises a hydraulic cement, sufficient water to form a slurry and a dispersing agent comprising a tetrapolymer formed of methacrylic acid, methacrylate, methallyl sulfonic acid and ethoxylated acrylic acid is provided. The cement composition is placed in a subterranean zone to be cemented and the cement composition is allowed to set into an impermeable solid mass therein.

The tetrapolymer dispersing agent includes methacrylic acid in an amount in the range of from about 50% to about 60% by weight of the tetrapolymer, methacrylate present in an amount of about 10% by weight of the tetrapolymer, methallyl sulfonic acid present in an amount of about 10% by weight of the tetrapolymer and ethoxylated acrylic acid present in an amount in the range of from about 20% to about 30% by weight of the tetrapolymer.

The ethoxylated acrylic acid in the tetrapolymer dispersing agent is ethoxylated within the range of from about 30 to about 120 moles of ethylene oxide. In a preferred tetrapolymer, the acrylic acid is ethoxylated with about 40 moles of ethylene oxide. In another preferred tetrapolymer, the acrylic acid is ethoxylated with about 100 moles of ethylene oxide.

A preferred tetrapolymer dispersing agent of this invention is comprised of methacrylic acid present in an amount of about 50% by weight of the tetrapolymer, methacrylate present in an amount of about 10% by weight of the tetrapolymer, methallyl

sulfonic acid present in an amount of about 10% by weight of the tetrapolymer and acrylic acid ethoxylated with about 40 moles of ethylene oxide present in an amount of about 30% by weight of the tetrapolymer.

Another preferred tetrapolymer dispersing agent of this invention comprises methacrylic acid present in an amount of about 60% by weight of the tetrapolymer, methacrylate present in an amount of about 10% by weight of the tetrapolymer, methallyl sulfonic acid present in an amount of about 10% by weight of the tetrapolymer and acrylic acid ethoxylated with about 100 moles of ethylene oxide present in an amount of about 20% by weight of the tetrapolymer.

Examples of hydraulic cement which can be utilized in the cement compositions of this invention include, but are not limited to, Portland cements, slag cements, pozzolana cements, gypsum cements, aluminous cements and silica cements. Portland cements and their equivalents are generally preferred for use in accordance with the present invention. Portland cements of the types defined and described in <u>API Specification For Materials And Testing For Well Cements</u>, API Specification 10, 5th Edition, dated July 1, 1990 of the American Petroleum Institute are particularly preferred.

The water in the cement compositions can be fresh water or salt water. The term "salt water" is used herein to mean unsaturated salt solutions and saturated salt solutions including brines and seawater. The water is present in the cement compositions of this invention in amounts sufficient to form a pumpable slurry. Generally, the water is present in an amount in the range of from about 25% to about 100% by weight of hydraulic cement in the compositions.

The tetrapolymer dispersing agent is generally present in a cement compositions of this invention in an amount in the range of from about 0.01% to about 1% by weight of hydraulic cement therein, more preferably in an amount in the range of from about 0.01% to about 0.50%.

As mentioned above, the tetrapolymer dispersing agent of this invention is utilized in small quantities as compared to the quantities required by prior art dispersing agents, causes less set retardation and enhances the fluid loss control properties of fluid loss control agents included in the cement compositions.

An example of a fluid loss control agent, the properties of which are enhanced by the tetrapolymer dispersant is a copolymer of 2-acrylamido-2-methyl propane sulfonic acid and

N,N'-dimethylacrylamide which is commercially available under the trade designation "HALAD – 344TM" by Halliburton Energy Services, Inc. of Duncan, Oklahoma. The fluid loss control agent is generally present in the cement compositions in an amount in the range of from about 0.5% to about 2.0% by weight of the hydraulic cement therein.

A preferred method of this invention for cementing a subterranean zone penetrated by a well bore comprises the steps of: (a) providing a cement composition which comprises a hydraulic cement, sufficient water to form a slurry and a dispersing agent comprising a tetrapolymer formed of methacrylic acid, methacrylate, methallyl sulfonic acid and ethoxylated acrylic acid; (b) placing the cement composition in the subterranean zone to be cemented; and (c) allowing the cement composition to set into an impermeable solid mass therein.

A preferred composition of this invention comprises: a hydraulic cement; sufficient water to form a slurry; and a dispersing agent comprising a tetrapolymer formed of methacrylic acid, methacrylate, methallyl sulfonic acid and ethoxylated acrylic acid.

In order to further illustrate the methods and compositions of this invention, the following examples are given.

EXAMPLE 1

Test cement compositions having densities of 16.4 pounds per gallon were prepared containing fresh water, Joppa API Class H Portland cement and various dispersing agents. The dispersing agents utilized included a prior art dispersing agent comprised of the condensation product of formaldehyde, acetone and sodium bisulfite which is commercially available under the trade designation "CFR-3™" from Halliburton Energy Services, Inc. of Duncan, Oklahoma and two dispersing agents of the present invention. The first dispersing agent of the present invention designated herein as DA-1 was a tetrapolymer consisting of 50% by weight methacrylic acid, 10% by weight methacrylate, 10% by weight methallyl sulfonic acid and 30% by weight acrylic acid ethoxylated with 40 moles of ethylene oxide. The second dispersing agent designated as DA-2 was a tetrapolymer consisting of 60% by weight methacrylic acid, 10% by weight methacrylate, 10% by weight methallyl sulfonic acid and 20% by weight of acrylic acid ethoxylated with 100 moles of ethylene oxide. The test cement compositions included the above described dispersing agents in amounts ranging from 0.05% to 1% by weight of cement in the cement compositions. The rheological properties of samples of the test compositions were determined using a Fann viscometer at

6

temperatures of 80°F, 130°F and 180°F in accordance with the above mentioned API Specification RP10B. The results of the tests are given in Tables I-A, I-B and I-C below.

TABLE I-ARheological Properties Of Cements Including Dispersants At 80°F

Dispersing Agent,	Temp.,	Rheological Properties, rpm.							
% by wt. of cement	°F	600	300	200	100	60	30	6	3
0	80	135	91	75	57	48	39	21	15
"CFR-3TM" - 0.05	80	120	66	54	39	32	26	17	12
DA-1 - 0.05	80	62	25	15	7	4	2	0	0
DA-2 - 0.05	80	70	29	17	8	5	2	0	0
"CFR-3TM" - 0.1	80	102	50	40	29	24	19	14	11
DA-1 - 0.1	80	49	19	12	4	2	1	0	0
DA-2 - 0.1	80	60	23	13	6	4	2	0	0
"CFR-3 ^{тм} " – 0.5	80	84	38	27	13	7	4	1	1
DA-1 - 0.5	80	70	29	18	8	4	2	0	0
DA-2 - 0.5	80	59	24	15	6	3	2	0	0
"CFR-3TM" - 0.75	80	75	36	21	10	5	2	i	1
DA-1 - 0.75	80	64	28	17	8	4	2	1	0
DA-2-0.75	80	68	30	18	9	4	2	0	0
"CFR-3TM" - 1.0	80	71	31	17	9	5	2	1	1
DA-1 - 1.0	80	73	32	19	9	5	2	0	0
DA-2 - 1.0	80	67	30	19	9	5	2	0	0

7

TABLE I-BRheological Properties Of Cements Including Dispersants At 130°F

Dispersing Agent,	Temp.,			Rhec	ological P	roperties,	rpm.	-	
% by wt. of cement	°F	600	300	200	100	60	30	6	3
0	130	170	91	75	57	48	39	21	15
"CFR-3 TM " - 0.05	130	167	123	107	87	76	57	26	17
DA-1 - 0.05	130	65	35	25	16	13	10	7	7
DA-2 - 0.05	130	70	39	29	20	16	12	9	8
"CFR-3TM" - 0.1	130	141	114	101	85	78	60	26	19
DA-1 - 0.1	130	49	22	14	7	5	3	1	1
DA-2 - 0.1	130	45	19	12	6	4	2	1	1
"CFR-3 ^{тм} " – 0.5	130	63	21	12	6	3	2	0	0
DA-1 - 0.5	130	50	18	11	5	3	1	0	0
DA-2 - 0.5	130	38	16	10	4	2	1	0	0
"CFR-3 TM " - 0.75	130	60	20	12	5	3	2	0	0
DA-1 - 0.75	130	53	20	12	5	2	1	0	0
DA-2 - 0.75	130	45	18	11	5	2	1	0	0
"CFR-3TM" - 1.0	130	63	21	12	6	3	1	0	0
DA-1 - 1.0	130	51	23	14	6	4	1	0	0
DA-2-1.0	130	51	19	11	5	3	1	0	0

TABLE I-CRheological Properties Of Cements Including Dispersants At 180°F

Dispersing Agent,	Temp.,			Rhec	ological P	roperties,	rpm.		
% by wt. of cement	°F	600	300	200	100	60	30	6	3
0	180	162	159	139	104	84	54	21	14
"CFR-3 TM " - 0.05	180	135	134	120	95	78	46	15	10
DA-1 - 0.05	180	74	43	35	26	22	20	13	10
DA-2 - 0.05	180	90	45	36	26	22	19	12	9
"CFR-3 ^{тм} " – 0.1	180	190	141	112	101	86	54	12	9
DA-1 - 0.1	180	52	25	18	10	8	5	4	4
DA-2-0.1	180	55	25	17	10	7	5	3	3
"CFR-3TM" - 0.5	180	40	13	9	3	1	0	0	0
DA-1 - 0.5	180	35	10	6	3	2	1	0	0
DA-2 – 0.5	180	31	10	3	2	1	0	0	0
"CFR-3 ^{тм} " – 0.75	180	37	15	8	3	1	0	0	0
DA-1 - 0.75	180	27	11	9	3	1	0	0	0
DA-2 – 0.75	180	30	12	7	3	2	1	0	0
"CFR-3TM" - 1.0	180	38	16	9	3	1	0	0	0
DA-1 - 1.0	180	34	15	8	4	2	0	0	0
DA-2 – 1.0	180	37	12	8	3	2	1	0	0

From Tables I-A, I-B and I-C, it can be seen that the dispersing agents of the present invention, i.e., DA-1 and DA-2, produced better rheological properties at a concentration level of 0.1% by weight of cement than the prior art "CFR-3TM" dispersing agent produced at a concentration of 1% by weight of cement.

EXAMPLE 2

Test cement compositions were prepared and tested as described in Example 1 except that sodium chloride salt or calcium chloride salt was added to the water and the cement compositions had densities of 17.0 or 16.7 pounds per gallon. Also, the prior art and inventive dispersing agents were added to the test cement composition samples in amounts of 1% by weight of cement in the test compositions. The rheological properties of the test cement compositions are set forth in Table II below.

TABLE II

Rheological Properties Of Cements Including Dispersants

And Having Various Salt Concentrations At Various Temperatures

Dispersing Agent,	Temp,	Salt (NaCl), %	Density,			Rheol	ogical Pi	operties	, rpm.		
% by wt. of cement	°F	by wt. of water	lb/gal	600	300	200	100	60	30_	6	3
"CFR-3 TM " - 1.0	80	18	17	141	69	46	23	14	7	1	1
DA-1 - 1.0	80	18	17	150	64	41	18	10	4	1	0
DA-2 - 1.0	80	18	17	115	53	33	15	8	3	1	0
"CFR-3TM" - 1.0	130	18	17	92	47	31	16	9	5	1	0
DA-1 – 1.0	130	18	17	77	35	21	9	5	2	Ī	0
DA-2 - 1.0	130	18	17	71	28	18	8	4	2	0	0
"CFR-3TM" - 1.0	180	18	17	78	34	22	11	7	4	1	1
DA-1 - 1.0	180	18	17	80	29	26	17	12	9	5	4
DA-2-1.0	180	18	17	68	54	40	27	20	15	8	6
"CFR-3 TM " - 1.0	80	37	16.7	71	55	36	18	11	6	1	1
DA-1-1.0	80	37	16.7	81	40	26	12	7	3	1	2
DA-2-1.0	80	37	16.7	82	40	26	13	8	4	2	1
"CFR-3TM" - 1.0	130	37	16.7	62	30	20	11	6	4	1	1
DA-1 - 1.0	130	37	16.7	90	67	53	37	29	22	12	8
DA-2-1.0	130	37	16.7	103	55	47	34	27	20	12	7
"CFR-3TM" - 1.0	180	37	16.7	53	23	15	8	5	2	1	1
DA-1 - 1.0	180	37	16.7	171	100	84	60	43	36	9	6
DA-2-1.0	180	37	16.7	180	106	91	72	52	39	11	9
		Salt (CaCl ₂)									
"CFR-3TM" - 1.0	80	2	17.0	176	87	56	28	16	8	2	1
DA-1 - 1.0	80	2	17.0	132	54	33	15	8	4	1	0
DA-2 - 1.0	80	2	17.0	120	56	38	18	12	7	4	4

From the rheological properties set forth in Table II, it can be seen that the dispersing agents of the present invention, DA-1 and DA-2, provide similar rheological properties as the

prior art dispersant "CFR-3TM" in cement compositions containing 18% sodium chloride by weight of the water in the test cement compositions. In the compositions wherein the water was saturated with sodium chloride salt (37% by weight of the water), it can be seen that the prior art dispersing agent "CFR-3TM" out performs the dispersing agents of the present invention. However, in cement compositions containing calcium chloride at a level of 2% by weight of the water, the dispersing agents of the present invention at a concentration of 1% by weight of cement provide better performance than cement compositions containing the prior art "CFR-3TM" dispersing agent in an amount of 1% by weight of cement.

EXAMPLE 3

Rheological data of densified test cement compositions prepared and tested as described in Example 1 except that the cement compositions had densities of 18.5 pounds per gallon, were obtained at a temperature of 80°F. The results of these tests are shown in Table III below.

TABLE III

Rheological Properties Of Densified Cements Including Dispersants

Dispersing Agent,	Temp.,			Rhec	logical P	roperties,	rpm.		
% by wt. of cement	°F	600	300	200	100	60	30	6	3
"CFR-3TM" - 1.0	80	300+	300+	300+	143	77	34	5	2
DA-1 - 0.2	80	300+	267	158	59	28	10	1	1
DA-2 - 0.2	80	300+	202	122	46	21	8	1	1

The rheological properties shown in Table III illustrate that the dispersing agents of this invention produced superior rheological properties at concentrations of 0.2% by weight of densified cement than the prior art "CFR-3TM" dispersing agent produced at a concentration of 1% by weight of densified cement.

EXAMPLE 4

Test high density cement compositions containing iron oxide weighting material and silica flour having densities of 19.5 pounds per gallon were prepared and tested as described in Example 1. The results of these tests are shown in Table IV below.

10

TABLE IVRheological Properties Of High Density Cements Including Dispersants

Dispersing Agent,	Temp.,			Rhec	logical Pi	operties,	rpm		
% by wt. of cement	°F	600	300	200	100	60	30	6	3
"CFR-3TM" – 1.0	80	300+	300+	300+	235	135	63	9	4
DA-1 - 0.1	80	300+	300+	300+	300+	156	78	18	11
DA-2 - 0.1	80	300+	300+	300+	252	152	78	19	14
DA-1 - 0.2	80	300+	300+	300+	112	53	19	2	1
DA-2 - 0.2	80	300+	300+	256	99	48	28	2	1
DA-1 - 0.3	80	300+	300+	250	95	44	16	2	1
DA-2 - 0.3	80	300+	300+	300+	115	52	18	2	1

The rheological properties of the cement compositions containing the prior art dispersing agent "CFR-3TM" and the dispersing agents of the present invention show that the presence of 0.3% by weight of cement of the dispersing agents of the present invention make the high density cement compositions easy to mix while a high density cement composition containing 1% of the prior art dispersing agent "CFR-3TM" was difficult to mix.

EXAMPLE 5

Test cement compositions were again prepared and tested as indicated in Example 1 at a density of 16.4 pounds per gallon. Prior art fluid loss control additives were added to two groups of the test cement compositions at a concentration of 1% by weight of cement in the compositions. The prior art fluid loss control additives were "HALAD-344TM" described previously herein and "HALAD-23TM" which is hydroxyethylcellulose with 1.5 moles of ethylene oxide substitution and is also commercially available from Halliburton Energy Services, Inc. of Duncan, Oklahoma.

11

TABLE V-A
Rheological Properties Of Cements Including Dispersants And Fluid Loss Control Agents

Dispersing Agent,	"HALAD-23 [™] ",	Temp.,			Rheolo	gical Pro	perties,	rpm.		
% by wt. of cement	% by wt. of cement	. °F	600	300	200	100	60	30	6	3
"CFR-3TM" - 0.5	1	80	300+	300+	300+	300+	300+	205	59	35
DA-1 - 0.2	1	80	300+	300+	300+	300+	244	142	37	21
DA-2 – 0.2	1	80	300+	300+	300+	300+	275	160	42	23
"CFR-3TM" - 0.5	1	130	300+	300+	300+	289	191	107	28	16
DA-1 - 0.2	1	130	300+	300+	300+	256	162	87	19	10
DA-2 – 0.2	1	130	300+	300+	300+	272	172	93	20	10
"CFR-3TM" - 0.5	1	180	300+	300+	300+	214	139	76	20	12
DA-1 - 0.5	1	180	300+	300+	300+	172	105	55	17	6
DA-2 - 0.5	1	180	300+	300+	300+	183	112	58	12	6

TABLE V-B
Rheological Properties Of Cements Including Dispersants And Fluid Loss Control Agents

Dispersing Agent,	"HALAD-344тм",	Temp.,			Rheolog	gical Pro	perties,	rpm.		
% by wt. of cement	% by wt. of cement	°F	600	300	200	100	60	30	6	3
"CFR-3TM" - 0.5	1	80	300+	300+	300+	177	115	64	14	7
DA-1 - 0.2	1	80	300+	300+	300+	182	120	67	15	7
DA-2 - 0.2	1	80	300+	300+	300+	184	122	69	16	8
"CFR-3TM" - 0.5	1	130	300+	300+	284	157	101	54	12	6
DA-1 - 0.2	1	130	300+	300+	285	166	110	62	15	8
DA-2 - 0.2	1	130	300+	300+	300+	192	125	71	17	9
"CFR-3TM" - 0.5	1	180	300+	300+	245	138	89	49	11	6
DA-1 - 0.2	1	180	300+	300+	300+	182	122	70	18	10
DA-2 – 0.2	1	180	300+	300+	300+	210	142	83	22	12

From Tables V-A and V-B, it can be seen that 0.2% by weight of cement of the dispersing agents of the present invention produced better rheological properties in cement compositions containing "HALAD-23TM" fluid loss control additive than the same cement compositions containing the prior art dispersing agent "CFR-3TM" in an amount of 0.5% by weight of cement.

EXAMPLE 6

Test cement compositions were prepared as in Example 1 containing various amounts of the fluid loss control additives "HALAD-23TM", "HALAD-344TM" and "HALAD-9TM". "HALAD-9TM" is hydroxyethylcellulose with 2.5 moles of ethylene oxide substitution and is also commercially available from Halliburton Energy Services, Inc. of Duncan, Oklahoma.

WO 2004/067469

The test cement composition samples were tested for fluid loss at 190°F. The results of these tests are given in Table VI.

TABLE VI
Fluid Loss From Cements Including Dispersants And Fluid Loss Control Additives

Dispersing Agent, % by wt. of cement	Fluid Loss Control Agent, % by wt. of cement	Temp., °F	Fluid Loss, cc/30 min API
"CFR-3 TM " – 0.66	"HALAD-23 TM " – 0.33	190	190
DA-1 - 0.66	"HALAD-23TM" - 0.33	190	235
DA-2 - 0.66	"HALAD-23TM" - 0.33	190	396
"CFR-3TM" - 0.5	"HALAD-23 TM " – 1	190	86
DA-1 - 0.2	"HALAD-23 TM " – 1	190	242
DA-2 - 0.2	"HALAD-23 ^{тм} " – 1	190	255
"CFR-3TM" - 0.5	"HALAD-344 TM " – 1	190	28
DA-1 - 0.2	"HALAD-344 TM " – 1	190	20
DA-2 - 0.2	"HALAD-344 ^{тм} " – 1	190	20
"CFR-3TM" - 1	"HALAD-344TM" – 1	190	173
DA-1-1	"HALAD-344 ^{тм} " – 1	190	40
DA-2 – 1	"HALAD-344 TM " - 1	190	20
"CFR-3TM" - 0.66	"HALAD-9 ^{тм} " – 0.33	190	244
DA-1 - 0.66	"HALAD-9 TM " – 0.33	190	367
DA-2 – 0.66	"HALAD-9TM" - 0.33	190	419

From Table VI it can be seen that none of the prior art or inventive dispersing agents had a significant effect on the fluid loss properties of "HALAD-23TM" or "HALAD-9TM". However, when the fluid loss control properties of "HALAD-344TM" were measured in the presence of the three dispersing agents, measurable differences in the fluid loss were observed. Increasing the concentration of the prior art CFR-3 dispersing agent resulted in higher fluid loss with "HALAD-344TM" fluid loss control additive. On the other hand, increasing the concentration of either of the dispersing agents of the present invention provided excellent fluid loss control as compared to the prior art dispersing agent "CFR-3TM".

EXAMPLE 7

Test cement compositions having a density of 17 lb/gal were prepared as described in Example 1. The thickening times of the cement compositions containing the dispersing agents of the present invention and the prior art dispersing agent "CFR-3TM" were tested for thickening time at 125°F. The results of these tests are shown in Table VII.

13

TABLE VIIThickening Time Tests

Dispersing Agent, % by wt. of cement	Salt (NaCl), % by wt. of cement	Temp., °F	Thickening Time, hr:min
"CFR-3TM" - 1.0	0	125	7:05
DA-1 - 0.1	0	125	2:39
DA-2 - 0.2	0	125	5:10
"CFR-3 TM " - 0.75	18	125	4:12
DA-1 - 0.75	18	125	28:49
DA-1 - 0.75	18	125	24:16

From Table VII it can be seen that long thickening times are caused by the dispersing agents of the present invention at concentrations of 0.75% by weight of cement in the cement composition. While the test cement composition sample containing the prior art "CFR-3TM" dispersing agent caused a relatively short set retardation, when small concentrations of the dispersing agents of the present invention were used to disperse the cement, the set retardation properties observed were comparable to those caused by the prior art "CFR-3TM" dispersing agent.

EXAMPLE 8

Test cement compositions having a density of 18 lb/gal were prepared as described in Example 1. The compressive strengths of the test cement compositions including the dispersing agents of the present invention and the prior art "CFR-3TM" dispersing agent were determined at 140°F. The results of these tests are shown in Tables VIII-A and VIII-B.

TABLE VIII-ACompressive Strength Tests

Dispersing Agent,	Temp.,	Crushed Compres	sive Strength (psi)
% by wt. of cement	°F	24 Hr, psi	48 Hr, psi
"CFR-3TM" - 0.5	140	7037.7	8446.5
DA-1 - 0.1	140	5474.3	6774.5
DA-2 - 0.1	140	4825.8	6554.9

14

TABLE VIII-B
Compressive Strength Tests

Dispersing Agent,	Temp.,	Compr	ressive Strengths on UCA		
% by wt. of cement	°F	12 Hr	24 Hr	72 Hr	
"CFR-3 TM " - 0.5	140	4891	6413	7800	
DA-2-0.1	140	5698	7413	9385	

From Tables VIII-A and VIII-B, it can be seen that 0.1% by weight of cement of a dispersing agent of the present invention produced higher 12 hr., 24 hr. and 72 hr. compressive strengths than the prior art "CFR-3TM" dispersing agent at a concentration of 0.5% by weight of cement.

From the data produced in Examples 1-8, it is clearly shown that the dispersing agents of this invention perform better than the prior art dispersing agent "CFR-3TM" in all areas. Generally, the dispersing agents of the present invention require approximately 10 times less concentration to provide the same results as the prior art dispersing agent "CFR-3TM".

Thus, the improved methods and composition of this invention are well adapted to carry out the objects and attain the ends and advantages mentioned as well as those which are inherent therein. While numerous changes may be made by those skilled in the art, such changes are encompassed within the spirit of this invention as defined by the appended claims.

What is claimed is:

WO 2004/067469

- 1. A method of cementing a subterranean zone comprising the steps of:
- (a) providing a cement composition which comprises a hydraulic cement, sufficient water to form a slurry and a dispersing agent comprising a tetrapolymer formed of methacrylic acid, methacrylate, methallyl sulfonic acid and ethoxylated acrylic acid;
- (b) placing said cement composition in said subterranean zone to be cemented; and
 - (c) allowing said cement composition to set into therein.
- 2. The method of claim 1 wherein said methacrylic acid is present in said tetrapolymer in an amount in the range of from about 50% to about 60% by weight of said tetrapolymer.
- 3. The method of claim 1 wherein said methacrylate is present in said tetrapolymer in an amount of about 10% by weight of said tetrapolymer.
- 4. The method of claim 1 wherein said methallyl sulfonic acid is present in said tetrapolymer in an amount of about 10% by weight of said tetrapolymer.
- 5. The method of claim 1 wherein said acrylic acid in said tetrapolymer is ethoxylated with in the range of from about 30 to about 120 moles of ethylene oxide.
- 6. The method of claim 1 wherein said acrylic acid in said tetrapolymer is ethoxylated with about 40 moles of ethylene oxide.
- 7. The method of claim 1 wherein said acrylic acid in said tetrapolymer is ethoxylated with about 100 moles of ethylene oxide.
- 8. The method of claim 1 wherein said ethoxylated acrylic acid is present in said tetrapolymer in an amount in the range of from about 20% to about 30% by weight of said tetrapolymer.

9. The method of claim 1 wherein said hydraulic cement in said cement composition is selected from the group consisting of Portland cements, slag cements, pozzolana cements, gypsum cements, alumina cements and silica cements.

16

- 10. The method of claim 1 wherein said hydraulic cement in said cement composition is Portland cement.
- 11. The method of claim 1 wherein said water in said cement composition is selected from the group consisting of fresh water and salt water.
- 12. The method of claim 1 wherein said water is present in said cement composition in an amount in the range of from about 25% to about 100% by weight of cement therein.
- 13. The method of claim 1 wherein said tetrapolymer dispersing agent is present in said cement composition in an amount in the range of from about 0.01% to about 1% by weight of cement therein.
- 14. The method of claim 1 wherein said cement composition further comprises a fluid loss control agent selected from the group consisting of a copolymer of 2-acrylamid o-2-methyl propane sulfonic acid and N,N'-dimethylacrylamide and hydroxyethylcellulose with 1.5 moles of ethylene oxide substitution.
- 15. The method of claim 14 wherein said fluid loss control agent is a copolymer of 2-acrylamido-2-methyl propane sulfonic acid and N,N'-dimethylacrylamide.
- 16. The method of claim 14 wherein said fluid loss control agent is present in said cement composition in an amount in the range of from about 0.5% to about 2% by weight of cement therein.
 - 17. A cement composition comprising:
 - a hydraulic cement;
 - sufficient water to form a slurry; and
- a dispersing agent comprising a tetrapolymer formed of methacrylic a cid, methacrylate, methallyl sulfonic acid and ethoxylated acrylic acid.

- 18. The composition of claim 17 wherein said methacrylic acid is present in said tetrapolymer in an amount in the range of from about 50% to about 60% by weight of said tetrapolymer.
- 19. The composition of claim 17 wherein said methacrylate is present in said tetrapolymer in an amount of about 10% by weight of said tetrapolymer.
- 20. The composition of claim 17 wherein said methallyl sulfonic acid is present in said tetrapolymer in an amount of about 10% by weight of said tetrapolymer.
- 21. The composition of claim 17 wherein said acrylic acid in said tetrapolymer is ethoxylated within the range of from about 30 to about 120 moles of ethylene oxide.
- 22. The composition of claim 17 wherein said acrylic acid in said tetrapolymer is ethoxylated with about 40 moles of ethylene oxide.
- 23. The composition of claim 17 wherein said acrylic acid in said tetrapolymer is ethoxylated with about 100 moles of ethylene oxide.
- 24. The composition of claim 17 wherein said ethoxylated acrylic acid is present in said tetrapolymer in an amount in the range of from about 20% to about 30% by weight of said tetrapolymer.
- 25. The composition of claim 17 wherein said hydraulic cement is selected from the group consisting of Portland cements, slag cements, pozzolana cements, gypsum cements, alumina cements and silica cements.
- 26. The composition of claim 17 wherein said hydraulic cement is Portland cement.
- 27. The composition of claim 17 wherein said water is selected from the group consisting of fresh water and salt water.
- 28. The composition of claim 17 wherein said water is present in an amount in the range of from about 25% to about 100% by weight of cement in said composition.

18

- 29. The composition of claim 17 wherein said tetrapolymer dispersing agent is present in an amount in the range of from about 0.1% to about 1% by weight of cement in said composition.
- 30. The composition of claim 17 which further comprises a fluid loss control agent selected from the group consisting of a copolymer of 2-acrylamido-2-methyl propane sulfonic acid and N,N'-dimethylacrylamide and hydroxyethylcellulose with 1.5 moles of ethylene oxide substitution.
- 31. The composition of claim 30 wherein said fluid loss control agent is 2-acrylamido-2-methyl propane sulfonic acid and N,N'-2 methylacrylamide.
- 32. The composition of claim 30 wherein said fluid loss control agent is present in an amount in the range of from about 0.5% to about 2% by weight of cement in said composition.

INTERNATIONAL SEARCH REPORT

Internatic cation No PCT/GB 03/01874

A. CLASSIFICATION OF SUBJECT MATTER IPC 7 C04B24/16 E21B33/13

According to International Patent Classification (IPC) or to both national classification and IPC

B. FIELDS SEARCHED

 $\begin{array}{ccc} \text{Minimum documentation searched (classification system followed by classification symbols)} \\ IPC & 7 & C04B & E21B \end{array}$

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Electronic data base consulted during the international search (name of data base and, where practical, search terms used)

EPO-Internal

С. DOCUM	ENTS CONSIDERED TO BE RELEVANT	
Category °	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
Х	EP 0 940 374 A (TAKEMOTO OIL & FAT CO LTD) 8 September 1999 (1999-09-08) claim 1	1-13, 17-29
Υ		14-16, 30-32
Y	US 4 784 693 A (MCLACHLAN WILLIAM C ET AL) 15 November 1988 (1988-11-15) column 2, line 3-17	14-16, 30-32
А	EP 0 217 608 A (HALLIBURTON CO) 8 April 1987 (1987-04-08) claim 1	1,14
A	EP 0 331 308 A (TAKEMOTO OIL & FAT CO LTD) 6 September 1989 (1989-09-06) claims 1,5	1,4, 9-13,17, 20,25-29
	-/	

لتا	اتها		
Special categories of cited documents: A' document defining the general state of the art which is not considered to be of particular relevance	*T* later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the		
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Date of the actual completion of the international search 16 September 2003	Date of mailing of the international search report $01/10/2003$		
Name and mailing address of the ISA European Patent Office, P.B. 5818 Patentlaan 2 NL – 2280 HV Rijswijk Tel. (+31-70) 340-2040, Tx. 31 651 epo nl, Fax: (+31-70) 340-3016	Authorized officer Daeleman, P		

Patent family members are listed in annex.

Further documents are listed in the continuation of box C.

INTERNATIONAL SEARCH REPORT

Internati :ation No
PCT/GB 03/01874

		PC1/GB 03/018/4		
	etion) DOCUMENTS CONSIDERED TO BE RELEVANT	la		
Category °	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.		
A	EP 0 924 174 A (TAKEMOTO OIL & FAT CO LTD) 23 June 1999 (1999-06-23)	1-4, 9-13, 17-20, 25-29		
	claim 1 	23 23		
Α	US 3 483 007 A (HOOK FRED E) 9 December 1969 (1969-12-09) claim 1			
Α	WO 95 16643 A (FUJITA SHUICHI ;KAO CORP (JP); SATOH HARUYUKI (JP); YAMATO FUJIO () 22 June 1995 (1995-06-22)			
A	US 5 466 289 A (YONEZAWA TOSHIO ET AL) 14 November 1995 (1995-11-14)			

INTERNATIONAL SEARCH REPORT

Internatio lication No
PCT/GB 03/01874

				rci/db	03/010/4
Patent documer cited in search rep		Publication date		Patent family member(s)	Publication date
EP 0940374	Α	08-09-1999	JP JP	3375279 B2 11246250 A	10-02-2003 14-09-1999
			JP	3322838 B2	09-09-2002
			JP	2000086314 A	28-03-2000
			ĔΡ	0940374 A1	08-09-1999
			TW	506953 B	21-10-2002
			ÜS	6176921 B1	23-01-2001
US 4784693	A	15-11-1988	AU	2449088 A	04-05-1989
			CA	1317754 C	18-05-1993
			CN	1034190 A	26-07-1989
			EP	0314118 A2	03-05-1989
			NO	884820 A	02-05-1989
			US 	4892589 A . 	09-01-1990
EP 0217608	Α	08-04-1987	US	4640942 A	03-02-1987
			AT DE	64981 T 3680061 D1	15-07-1991 08-08-1991
			EP	0217608 A2	08-08-1991
EP 0331308	Α	06-09-1989	JP	1226757 A	11-09-1989
			JP	2038550 C	28-03-1996
			JP	5011057 B	12-02-1993
			DE	68908211 D1	16-09-1993
			DE	68908211 T2	25-11-1993
			EP KR	0331308 A2 9106893 B1	06-09-1989 10-09-1991
			US	4962173 A	09-10-1990
			US	5087648 A	11-02-1992
EP 0924174	А	23-06-1999	JP	3278391 B2	30-04-2002
		20 00 2000	ĴΡ	11171619 A	29-06-1999
			DE	69801737 D1	25-10-2001
			DE	69801737 T2	04-07-2002
			EP	0924174 A1	23-06-1999
			US	6140440 A	31-10-2000
			US	6228962 B1	08-05 - 2001
US 3483007	Α	09-12-1969	NONE		
WO 9516643	Α	22-06-1995	JP	2774445 B2	09-07-1998
			JP	7223852 A	22-08-1995
			CN	1137263 A ,B	04-12-1996
			DE	69417576 D1	06-05-1999
			DE	69417576 T2	26-08-1999
			EP ES	0734359 A1 2129190 T3	02-10-1996
			MO F2	2129190 13 9516643 A1	01-06-1999 22-06-1995
			US	5707445 A	13-01-1998
 IIS 5466280	Δ	14-11-1995	JP.	2646449 R2	27-08-1007
 US 5466289	Α	14-11-1995	JP JP	2646449 B2 6191918 A	27-08-1997 12-07-1994