(19) World Intellectual Property Organization

International Bureau



(43) International Publication Date 24 April 2003 (24.04.2003)

PCT

English

(10) International Publication Number WO 03/034532 A1

(51) International Patent Classification7: H01M 10/52, 2/10, 2/02, F21V 33/00, H02K 5/136

(21) International Application Number: PCT/IB02/04023

(22) International Filing Date:

27 September 2002 (27.09.2002)

(25) Filing Language:

(26) Publication Language: English

(30) Priority Data:

01203898.0 12 October 2001 (12.10.2001)

(71) Applicant: KONINKLIJKE PHILIPS ELECTRON-ICS N.V. [NL/NL]; Groenewoudseweg 1, NL-5621 BA Eindhoven (NL).

(72) Inventors: NOTTEN, Petrus, H., L.; Prof. Holstlaan 6, NL-5656 AA Eindhoven (NL). VAN BEEK, Johann, R., G., C., M.; Prof. Holstlaan 6, NL-5656 AA Eindhoven (NL). BRUGGINK, Wilhelmus, H., M.; Prof. Holstlaan 6, NL-5656 AA Eindhoven (NL).

(74) Agent: VAN WERMESKERKEN, Stephanie, C.; Internationaal Octrooibureau B.V., Prof. Holstlaan 6, NL-5656 AA Eindhoven (NL).

(81) Designated States (national): CN, JP.

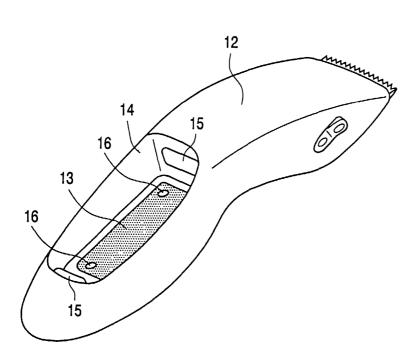
(84) Designated States (regional): European patent (AT, BE, BG, CH, CY, CZ, DE, DK, EE, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE, SK, TR).

Published:

with international search report

For two-letter codes and other abbreviations, refer to the "Guidance Notes on Codes and Abbreviations" appearing at the beginning of each regular issue of the PCT Gazette.

(54) Title: ELECTRICAL APPLIANCE AND BATTERY



(57) Abstract: The invention relates to an electrical appliance provided with a substantially gastight housing, characterized in that the appliance is provided at the inside of its housing with a catalytic surface for converting hydrogen evolved by a hydrogen source present in the electrical appliance. The invention also relates to a battery comprising a housing which is substantially impermeable to mediums and which is provided with a pressure relief valve, and an assembly of an anode, a cathode, and a separator layer positioned between the anode and the cathode, which assembly is positioned in the housing and is at least partly surrounded by an electrolytic solution, characterized in that the battery is provided with a catalytic surface for converting hydrogen gas produced by the assembly at the side of the housing facing away from the assembly.

WO 03/034532 A1

5

10

15

20

25

Electrical appliance and battery

The invention relates to an electrical appliance provided with a substantially gastight housing. The invention also relates to a battery comprising a housing which is substantially impermeable to mediums and which is provided with a pressure relief valve and with an assembly of an anode, a cathode, and a separator layer positioned between the anode and the cathode, which assembly is positioned in the housing and is at least partly surrounded by an electrolytic solution.

Electrical equipment provided with a substantially gastight housing is mostly used in surroundings which may be humid or wet. Examples of this are shaving equipment, toothbrushes, pocket torches, and the like. Since the housing is at least substantially gastight, combustible gas may accumulate at the inside of this equipment, in particular hydrogen gas. This hydrogen gas may be evolved by a battery present in an appliance, but alternative hydrogen sources are also possible such as, for example, a short-circuit of an electrical circuit (printed circuit board) caused by the presence of water (for example condensation). In the prior art, the solution to this problem is sought in a timely removal of the gases present in the housing. This may be achieved, for example, in that a membrane permeable to hydrogen is made to form part of the housing. Among the disadvantages of such membranes, however, are that their presence in a housing is comparatively expensive, that a membrane forms an interruption in the housing, that the membrane is comparatively vulnerable, and that the membrane may at the same time be permeable to liquids and/or gases other than hydrogen.

The invention has for its object to provide an electrical appliance provided with a substantially gastight housing with which hazardous situations resulting from a release of hydrogen gas in the housing of the electrical appliance can be prevented without the disadvantages of the inclusion of a membrane in the housing as described above.

To achieve this object, the invention provides an electrical appliance provided with a substantially gastight housing which is characterized in that the appliance is provided at the inside of its housing with a catalytic surface for converting hydrogen evolved by a hydrogen source present in the electrical appliance. If the appliance is powered by a battery arranged in the appliance, said hydrogen source may be formed by the battery, in particular if this battery is provided with a pressure relief valve for discharging an overpressure from the

5

10

15

20

25

30

PCT/IB02/04023

battery. The catalyst, however, is also capable of neutralizing hydrogen in the housing originating from sources other than a battery. The combustible hydrogen gas can be recombined to water in that the hydrogen gas present in the housing is brought into contact with a suitable catalyst. Hazardous situations such as combustion and/or explosion of the hydrogen gas can be counteracted or prevented through recombination of the combustible hydrogen gas, or at least of a portion thereof. This renders the removal of the hydrogen gas through a membrane unnecessary.

Providing the catalytic surface adjacent a source capable of generating sparks in the electrical appliance prevents at least to a substantial degree that a spark-generating source comes into contact with hydrogen gas. The spark-generating source is thus as it were screened off from the hydrogen gas. If an ignition source is absent, hydrogen gas will not be ignited, which will enhance the safety of the construction still further. Examples of spark-generating sources are: switches, motors, sliding contacts, and the like.

In a preferred embodiment, the catalytic surface is provided at least adjacent an accommodation space for the battery which forms part of the electrical appliance. Such a positioning of the catalyst with respect to the battery to be placed in said accommodation space is favorable because the gases ventilated from the battery come into contact with the catalyst immediately after their discharge and can be subsequently converted.

In another preferred embodiment, the catalytic surface is detachably connected to the housing. The catalyst can be removed from the housing if it does not function optimally any more, and may be replaced, for example, with a new catalyst or one that functions better. If the appliance is not used for a certain time, the catalyst may be uncoupled from the housing and may be placed in another electrical appliance. In an embodiment, the catalyst is shaped such that it can be placed in several types of electrical appliances.

Batteries are in existence with an internal catalyst for converting hydrogen, the presence of said battery being necessary for the operation of the appliance. Thus the international patent application WO 00/36684 describes a catalyst arrangement whereby oxygen and hydrogen gases produced by a so-called valve regulated lead acid (VRLA) battery can be recombined into water. The conversion of oxygen and hydrogen takes place inside the VRLA battery. The catalyst arrangement comprises a substantially airtight housing provided with an opening. An internal chamber is positioned in the housing, in which chamber the catalyst is provided. A microporous disc is positioned in said opening. An excessive generation of oxygen and hydrogen in the battery, such that the capacity of the catalyst arrangement is insufficient for converting the total quantity of evolved gases, will

5

10

15

20

25

30

open a pressure relief valve forming part of the battery owing to a rising pressure in the battery. An excessive gas generation in the VRLA battery may arise from a malfunctioning of the battery owing to, for example, heat and wear, such as a capacity reduction of the catalyst. Hazardous situations can arise because the opening of the pressure relief valve will allow combustible hydrogen gases to enter the atmosphere surrounding the battery. Besides VRLA batteries, rechargeable batteries of other types, such as the aqueous nickel-cadmium (NiCd) and nickel-metal hydride (NiMH) batteries, are capable of generating oxygen and hydrogen gases. If such batteries no longer operate optimally during recharging and over-discharging, for example owing to a reduction in their capacity, the pressure inside the battery may rise substantially. The moment the internal pressure in the battery exceeds a critical battery valve pressure, the (combustible) gases present in the battery will be discharged through the pressure relief valve to the atmosphere surrounding the battery. Such a discharge of gases may lead to hazardous situations if, for example, spark-generating electrical or mechanical components are present in the atmosphere surrounding the battery.

The invention also has for its object to provide a battery of the kind mentioned in the opening paragraph by means of which hazardous situations caused by the release of hydrogen gas owing to malfunctioning of the battery can be prevented, while the advantages of the prior art are retained.

The invention for this purpose provides a battery of the kind mentioned in the opening paragraph which is characterized in that the battery is provided with a catalytic surface designed for converting hydrogen gas produced by the assembly at the side of the housing facing away from the assembly. If an excessive (hydrogen) gas production takes place in the battery owing to malfunctioning of the battery or at least of parts of the battery, as described above, the evolved gases will be discharged through the pressure relief valve the moment the internal gas pressure exceeds a critical battery valve pressure. The advantages of a timely neutralization were described above with reference to the electrical appliance according to the present invention. The catalyst may be provided at an outer surface of the battery, preferably adjacent the pressure relief valve.

The catalyst preferably comprises at least one metal. Suitable metals for hydrogen conversion are, for example, Ni, Pd, Pt, Rh, and Ir. In a preferred embodiment, the catalyst is made from an alloy. Examples of suitable alloys are PtPd, PtW, ZrPd, MoNi₃, MoCo₃, etc. Preferably, the catalytic surface is formed by a catalytic material layer. Among the properties of a catalytic material layer is that it forms an extended planar surface on which a catalytic reaction can take place. The material layer may be constructed as a foil.

Alternatively, the material layer may be provided as a thin layer on a side of the housing facing away from the assembly. The thin material layer may be provided by known techniques such as painting, spraying, sputtering, vapor deposition, laser ablation, adhesion, etc. Research has shown that the hydrogen is not absorbed by a catalyst in the form of a hydride, but that it is captured by an electrochemical oxidation reaction in which oxygen acts as the combustion agent. Water is required for a satisfactory conversion of the hydrogen. This, however, will always be the case under normal ambient operational conditions. A large catalytic surface, such as a catalytic material layer, is essential. The required water may form part of the electrolyte and may be carried along with the ventilated gas during ventilation of the battery. The electrochemical conversion may be written as follows as partial reactions and a subsequent total reaction:

$$2 H_2 + 4 OH^{-} \rightarrow 4 H_2O + 4e^{-}$$

$$O_2 + 2H_2O + 4e^{-} \rightarrow 4OH^{-}$$

$$2H_2 + O_2 \rightarrow 2H_2O$$

5

10

20

25

30

The total reaction is exothermic and produces a considerable quantity of heat. To facilitate the above conversion, it is important to provide as large a catalytic surface as possible. Besides the material layer described above, the catalyst may be manufactured from a porous material and/or be present as a material layer provided on a porous material. The catalytic surface is preferably provided with a specific surface area of at least 1 m²/g of catalyst material.

In a preferred embodiment, the catalytic surface is provided on a chamber which merges into the housing and which surrounds the pressure relief valve. In a special preferred embodiment, said chamber closely adjoins the pressure relief valve so that the ventilated gas is forcedly conducted along the catalytic surface. The catalyst may be provided in the chamber in the form of a porous material. Alternatively, it may be positioned as a foil against an inside of the chamber. In an alternative embodiment, the catalyst is formed by particles of a granulate surrounded by the chamber. The chamber is preferably provided with an exhaust opening for gases and/or liquids.

The invention will be explained in more detail with reference to embodiments which are shown by way of example in the appended drawing and to which the invention is by no means limited. In the drawing:

Fig. 1 is a cross-sectional view of a first preferred embodiment of a rechargeable battery according to the invention,

Fig. 2 is a cross-sectional view of a second preferred embodiment of a rechargeable battery according to the invention, and

Fig. 3 is a perspective view of a hair trimmer provided with a catalyst according to the invention.

10

15

20

25

30

5

Fig. 1 is a cross-sectional view of a first preferred embodiment of a rechargeable battery 1 according to the invention. The battery 1 comprises a housing 2 in which an assembly (not shown) of an anode, a cathode, and a separator is accommodated, and a pressure relief valve 3. The pressure relief valve 3 may at the same time act as the positive pole of the battery 1. The battery is provided with a catalytic material layer 4 at the outside of the housing 2. If a gas produced by the battery 1, including hydrogen, exceeds a critical battery valve pressure, the pressure relief valve 3 will open and the gases present in the battery will be discharged through an opening in the pressure relief valve 3 to the atmosphere surrounding the battery 1. There at least a portion of the discharged hydrogen gas can come into contact with the catalytic material layer 4. The contact between the hydrogen gas and the catalytic material layer 4 stimulates the conversion of combustible hydrogen gas to (non-combustible) water. The catalytic material layer 4 shown in Fig. 1 is provided around the outside of the housing 2 in the form of a foil layer. The catalyst has a high surface/volume ratio so as to achieve a satisfactory operation; the hydrogen conversion taking place at the surface of a catalyst can thus be effective.

Fig. 2 is a cross-sectional view of a second preferred embodiment of a rechargeable battery 5 according to the invention. The battery 5 again comprises a housing 6 in which an assembly (not shown) of an anode, a cathode, and a separator is accommodated, and a pressure relief valve 7. The battery 5 is also provided with a recombination chamber 8, which recombination chamber 8 narrowly adjoins the housing 6 and also surrounds the pressure relief valve 7. The recombination chamber 8 is provided with at least one opening 9 for discharging any excess gases and/or liquids present in the recombination chamber 8. The recombination chamber 8 is at least partly filled with a catalyst 10 in the form of granulate

particles which may or may not be porous. An alternative option is that a porous material (not shown in the Figure) is provided with an active component which acts as a catalyst 10. The catalyst 10 is designed so as to facilitate the conversion of hydrogen to water. A microporous disc 11 is provided between the opening 9 and the catalyst 10. The disc 11 prevents a (too) easy escape of hydrogen gas from the recombination chamber 8 to the atmosphere surrounding the battery 5. A construction of the battery 5 as shown in Fig. 2 has the advantage that the gases discharged through the pressure relief valve 7 come into contact with the catalyst 10 for a major portion, so that the hydrogen conversion rate will be high. The disc 11 in a preferred embodiment may at the same time have the property that it is substantially non-permeable to hydrogen gas, such that the hydrogen gas discharged from the housing 6 remains trapped in the recombination chamber 8 until it has been converted into water.

Fig. 3 is a perspective view of a hair trimmer 12 provided with a catalyst 13 according to the invention. The catalyst 13 is positioned in an accommodation space 14 for a battery (not shown). The accommodation space 14 may be closed by a cover element (not shown). The battery may be placed on the catalyst 13 and is preferably designed such that it applies itself to contact elements 15. The catalyst 13 is detachably connected to the accommodation space 14 for the battery by means of coupling elements 16. In alternative embodiments, however, it is possible to provide the catalyst 13 in other locations of the trimmer 12.

PCT/IB02/04023

7 CLAIMS:

1. An electrical appliance provided with a substantially gastight housing, characterized in that the appliance is provided with a catalytic surface at the inside of its housing for converting hydrogen evolved by a hydrogen source present in the electrical appliance.

5

- 2. An electrical appliance as claimed in claim 1, characterized in that the appliance is powered by a battery arranged inside the appliance.
- 3. An electrical appliance as claimed in claim 1 or 2, characterized in that the catalytic surface is provided on a housing which forms part of the electrical appliance.
 - 4. An electrical appliance as claimed in any one the preceding claims, characterized in that the catalytic surface is provided at least adjacent to a source which is capable of generating sparks and which is present inside the electrical appliance.

15

- 5. An electrical appliance as claimed in any one of the claims 2 to 4, characterized in that the catalytic surface is provided at least adjacent an accommodation space for a battery which forms part of the electrical appliance.
- 20 6. An electrical appliance as claimed in any one of the preceding claims, characterized in that the catalytic surface is detachably connected to the housing.
 - 7. A battery comprising:
- a housing which is substantially impermeable to mediums and which is provided with a pressure relief valve, and
 - an assembly of an anode, a cathode, and a separator layer positioned between the anode and the cathode, which assembly is positioned in the housing and is at least partly surrounded by an electrolytic solution,

characterized in that the battery is provided with a catalytic surface designed for converting hydrogen gas produced by the assembly at the side of the housing facing away from the assembly.

- 5 8. A battery as claimed in claim 7, characterized in that the catalyst comprises at least one metal.
 - 9. A battery as claimed in claim 7 or 8, characterized in that the catalytic surface is formed by a layer of catalytic material.
 - 10. A battery as claimed in any one of the claims 7 to 9, characterized in that the catalytic surface has a specific surface area of at least $1 \text{ m}^2/\text{g}$ of catalyst material.

10

11. A battery as claimed in any one of the claims 7 to 10, characterized in that the catalytic surface is provided on a chamber which merges into the housing and which surrounds the pressure relief valve.

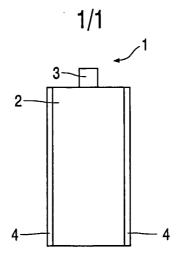


FIG. 1

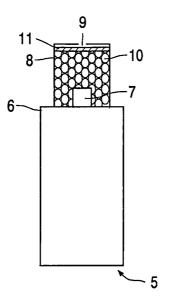


FIG. 2

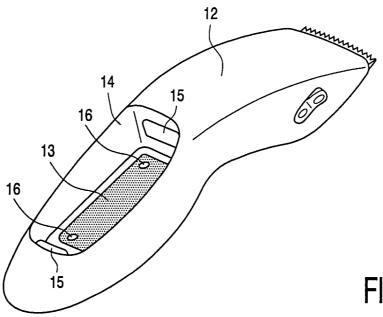


FIG. 3

INTERNATIONAL SEARCH REPORT



A. CLASSIFICATION OF SUBJECT MATTER IPC 7 H01M10/52 H01M H01M2/10 H01M2/02 F21V33/00 H02K5/136 According to International Patent Classification (IPC) or to both national classification and IPC B. FIELDS SEARCHED Minimum documentation searched (classification system followed by classification symbols) H01M F21L F21V H05K H02K Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched Electronic data base consulted during the international search (name of data base and, where practical, search terms used) EPO-Internal, PAJ C. DOCUMENTS CONSIDERED TO BE RELEVANT Citation of document, with indication, where appropriate, of the relevant passages Relevant to claim No. EP 1 122 590 A (FUJI PHOTO FILM CO LTD) Χ 1-68 August 2001 (2001-08-08) column 6, line 14 -column 8, line 40; claims 1-12 WO OO 62366 A (BALLARD GENERATION SYSTEMS 1-5 Χ INC ; KNAGGS LESLIE BRENON (CA)) 19 October 2000 (2000-10-19) page 9, line 25 -page 13, line 24; claims 1 - 36χ EP 0 949 706 A (EVEREADY BATTERY INC) 1,2,7-913 October 1999 (1999-10-13) column 8, line 54 -column 9, line 31; claims 1-13 Patent family members are listed in annex. Further documents are listed in the continuation of box C. Special categories of cited documents: "T" later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the *A* document defining the general state of the art which is not considered to be of particular relevance invention *E* earlier document but published on or after the international "X" document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step when the document is taken alone *L* document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified) "Y" document of particular relevance; the claimed invention cannot be considered to involve an invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art. *O* document referring to an oral disclosure, use, exhibition or document published prior to the international filing date but later than the priority date claimed "&" document member of the same patent family Date of the actual completion of the international search Date of mailing of the international search report 17/01/2003 10 January 2003 Authorized officer Name and mailing address of the ISA European Patent Office, P.B. 5818 Patentiaan 2 NL - 2280 HV Rijswijk Tel. (+31-70) 340-2040, Tx. 31 651 epo nl, Fax: (+31-70) 340-3016 De Vos, L

INTERNATIONAL SEARCH REPORT

In ational Application No
PCT/IB 02/04023

	1/1B 02/04023
	Relevant to claim No.
US 5 535 107 A (PROK ROBERT A) 9 July 1996 (1996-07-09) column 2, line 36 -column 3, line 62; claims 1-10	1,2,5
US 5 998 325 A (SHEPODD TIMOTHY JON) 7 December 1999 (1999-12-07) column 2, line 16 -column 3, line 11; claims 1-8	1,7
US 4 078 893 A (BRAMHALL PAUL J ET AL) 14 March 1978 (1978-03-14) the whole document	1,7
PATENT ABSTRACTS OF JAPAN vol. 008, no. 209 (E-268), 22 September 1984 (1984-09-22) & JP 59 096648 A (MATSUSHITA DENKO KK), 4 June 1984 (1984-06-04) abstract	1,2
US 4 127 134 A (USHAKOFF ALEXIS E) 28 November 1978 (1978-11-28) column 2, line 31 -column 4, line 19; claims 1-7	1,2
PATENT ABSTRACTS OF JAPAN vol. 012, no. 403 (E-674), 26 October 1988 (1988-10-26) & JP 63 146346 A (FUJITSU LTD), 18 June 1988 (1988-06-18) abstract	1,2
PATENT ABSTRACTS OF JAPAN vol. 2000, no. 09, 13 October 2000 (2000-10-13) & JP 2000 164179 A (HONDA ACCESS CORP), 16 June 2000 (2000-06-16) abstract	7
GB 1 503 736 A (BOSCH GMBH ROBERT) 15 March 1978 (1978-03-15) page 1, line 90 -page 2, line 86; claims 1-11	7,8,11
PATENT ABSTRACTS OF JAPAN vol. 1996, no. 02, 29 February 1996 (1996-02-29) & JP 07 272770 A (MATSUSHITA ELECTRIC IND CO LTD), 20 October 1995 (1995-10-20) abstract	7
US 4 378 413 A (WINSEL AUGUST ET AL) 29 March 1983 (1983-03-29) the whole document	7,11
	Citation of document, with indication, where appropriate, of the relevant passages

INTERNATIONAL SEARCH REPORT

Information on patent family members

Ir ational Application No PCT/IB 02/04023

							- -,
	ent document in search report		Publication date		Patent family member(s)		Publication date
ΕP	1122590	A	08-08-2001	JP CN	2000292889	Α	20-10-2000 08-08-2001
				EP US	1122590 2001005452		08-08-2001 28-06-2001
WO	0062366	Α	19-10-2000	US AU	6372983 3799800		16-04-2002 14-11-2000
				WO GB	0062366 2363674	A1	19-10-2000 02-01-2002
EP	0949706	Α	13-10-1999	US AU	2001038939 3469799		08-11-2001 25-10-1999
				CA	2327860		14-10-1999
				EP	0949706		13-10-1999
				TW	428337		01-04-2001
				WO	9952169	A1	14-10-1999
US	5535107	Α	09-07-1996	AU	4601396		10-07-1996
				EP	0870155		14-10-1998
				WO 	9619696 	A1 	27-06-1996
US 	5998325 	A 	07-12-1999	NONE			
US 	4078893	Α	14-03-1978	NONE			
JP	59096648	Α	04-06-1984	JP	1516803		07-09-1989
				JP	63067749	B 	27-12-1988
US	4127134	Α	28-11-1978	CA	1094168		20-01-1981
				DE	2815558		19-10-1978
				FR	2387029		10-11-1978
				GB JP	1583610 54001990		28-01-1981 09-01-1979
				NL	7803755		13-10-1978
				SE	7804087		12-10-1978
JP	63146346	Α	18-06-1988	NONE			
JP	2000164179	Α	16-06-2000	NONE			
GB	1503736	Α	15-03-1978				
				FR	2262416 4074024		19-09-1975 14-02-1978
				US US	4074024		04-07-1978
	07272770 9 			NONE			
US	4378413	Α	29-03-1983	DE	3011421	A1	01-10-1983
				SE SE	439215 8101849	ט	03-06-1985 26-09-1983
				3E	0101049	M	ZU-U9-190.