

PCTWORLD INTELLECTUAL
PROPERTY ORGANIZATION
Inter

INTERNATIONAL APPLICATION PUBLISHED UNDER THE PATENT COOPERATION TREATY (PCT)

(51) International Patent Classification ⁶ : C09K 5/04	A1	(11) International Publication Number: WO 96/02603 (43) International Publication Date: 1 February 1996 (01.02.96)
(21) International Application Number: PCT/GB95/01592 (22) International Filing Date: 6 July 1995 (06.07.95) (30) Priority Data: 9414136.3 13 July 1994 (13.07.94) GB (71) Applicant (for all designated States except US): IMPERIAL CHEMICAL INDUSTRIES PLC [GB/GB]; Imperial Chemical House, Millbank, London SW1P 3JF (GB). (72) Inventors; and (75) Inventors/Applicants (for US only): POWELL, Richard, Llewellyn [GB/GB]; 9 Sadler's Wells, Bunbury, Tarporley, Cheshire CW6 9NV (GB). CORR, Stuart [GB/GB]; 31 Foxhills Close, Appleton, Warrington, Cheshire WA4 5DH (GB). MURPHY, Frederick, Thomas [GB/GB]; 53 Fairways, Frodsham, Cheshire WA6 7RY (GB). MORRISON, James, David [GB/GB]; 39 Sandown Crescent, Cuddington, Northwich, Cheshire CW8 2QN (GB). (74) Agents: DEE, Ian, Mark et al.; ICI Chemicals & Polymers Limited, Intellectual Property Dept., P.O. Box 11, The Heath, Runcom, Cheshire WA7 4QE (GB).		(81) Designated States: AU, BR, CA, JP, KR, US, European patent (AT, BE, CH, DE, DK, ES, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE). Published <i>With international search report.</i>
(54) Title: REFRIGERANT COMPOSITIONS (57) Abstract A non-azeotropic (zeotropic) refrigerant composition is described. The refrigerant composition comprises (A) a first component comprising fluoromethane (R-41), (B) a second component comprising at least one hydrofluorocarbon selected from the group consisting of 1,1,1,2-tetrafluoroethane (R-134a) and 1,1,2,2-tetrafluoroethane (R-134), and optionally (C) a third component comprising at least one hydrofluorocarbon selected from the group consisting of difluoromethane (R-32), 1,1,1-trifluoroethane (R-143a) and pentafluoroethane (R-125).		

FOR THE PURPOSES OF INFORMATION ONLY

Codes used to identify States party to the PCT on the front pages of pamphlets publishing international applications under the PCT.

AT	Austria	GB	United Kingdom	MR	Mauritania
AU	Australia	GE	Georgia	MW	Malawi
BB	Barbados	GN	Guinea	NE	Niger
BE	Belgium	GR	Greece	NL	Netherlands
BF	Burkina Faso	HU	Hungary	NO	Norway
BG	Bulgaria	IE	Ireland	NZ	New Zealand
BJ	Benin	IT	Italy	PL	Poland
BR	Brazil	JP	Japan	PT	Portugal
BY	Belarus	KE	Kenya	RO	Romania
CA	Canada	KG	Kyrgystan	RU	Russian Federation
CF	Central African Republic	KP	Democratic People's Republic of Korea	SD	Sudan
CG	Congo	KR	Republic of Korea	SE	Sweden
CH	Switzerland	KZ	Kazakhstan	SI	Slovenia
CI	Côte d'Ivoire	LI	Liechtenstein	SK	Slovakia
CM	Cameroon	LK	Sri Lanka	SN	Senegal
CN	China	LU	Luxembourg	TD	Chad
CS	Czechoslovakia	LV	Latvia	TG	Togo
CZ	Czech Republic	MC	Monaco	TJ	Tajikistan
DE	Germany	MD	Republic of Moldova	TT	Trinidad and Tobago
DK	Denmark	MG	Madagascar	UA	Ukraine
ES	Spain	ML	Mali	US	United States of America
FI	Finland	MN	Mongolia	UZ	Uzbekistan
FR	France			VN	Viet Nam
GA	Gabon				

REFRIGERANT COMPOSITIONS

The present invention relates to non-azeotropic refrigerant compositions and more particularly to non-azeotropic refrigerant compositions which boil over a wide
5 temperature range and thus provide wide temperature glides in the heat exchangers of the heat transfer devices in which they are used.

Heat transfer devices of the mechanical compression type such as refrigerators, freezers, heat pumps and air conditioning systems are well known. In such devices a refrigerant liquid of a suitable boiling point evaporates at low pressure taking heat from
10 a surrounding heat transfer fluid. The resulting vapour is then compressed and passes to a condenser where it condenses and gives off heat to another heat transfer fluid. The condensate is then returned through an expansion valve to the evaporator so completing the cycle. The mechanical energy required for compressing the vapour and pumping the liquid may be provided by an electric motor or an internal combustion
15 engine.

In addition to having a suitable boiling point and a high latent heat of vaporisation, the properties preferred of a refrigerant include low toxicity, non-flammability, non-corrosivity, high stability and freedom from objectionable odour.

Hitherto, heat transfer devices have tended to use fully and partially halogenated
20 chlorofluorocarbon refrigerants such as trichlorofluoromethane (Refrigerant R-11), dichlorodifluoromethane (Refrigerant R-12), chlorodifluoromethane (Refrigerant R-22) and the azeotropic mixture of chlorodifluoromethane and chloropentafluoroethane (Refrigerant R-115); the azeotrope being Refrigerant R-502. Refrigerant R-22, for example, is widely used in air conditioning systems.

25 However, the fully and partially halogenated chlorofluorocarbons have been implicated in the destruction of the earth's protective ozone layer and as a result the use and production thereof has been limited by international agreement.

Whilst heat transfer devices of the type to which the present invention relates are essentially closed systems, loss of refrigerant to the atmosphere can occur due to
30 leakage during operation of the equipment or during maintenance procedures. It is

important, therefore, to replace fully and partially halogenated chlorofluorocarbon refrigerants by materials having low or zero ozone depletion potentials.

In addition to the possibility of ozone depletion, it has been suggested that significant concentrations of chlorofluorocarbon refrigerants in the atmosphere might contribute to global warming (the so-called greenhouse effect). It is desirable, therefore, to use refrigerants which have relatively short atmospheric lifetimes as a result of their ability to react with other atmospheric constituents such as hydroxyl radicals.

Replacements for some of the chlorofluorocarbon refrigerants presently in use have already been developed. These replacement refrigerants tend to comprise selected hydrofluoroalkanes, i.e. compounds which contain only carbon, hydrogen and fluorine atoms in their structure. Thus, refrigerant R-12 is generally being replaced by 1,1,1,2-tetrafluoroethane (R-134a).

Although suitable replacement refrigerants are available, it could be beneficial in certain heat transfer devices to replace the chlorofluorocarbon refrigerant with a non-azeotropic refrigerant blend which not only exhibits a low or zero ozone depletion potential but which also boils over a reasonably wide temperature range, since in this way the energy efficiency (coefficient of performance) of the device may be increased leading to a reduction in indirect global warming.

The present invention provides a non-azeotropic refrigerant composition which comprises a mixture of compounds having low or zero ozone depletion potentials. The refrigerant composition of the invention can boil over a wide temperature range and this property can be exploited to increase the energy efficiency of the equipment in which the composition is used. Furthermore, the temperature range over which the refrigerant composition of the invention boils, i.e. the so-called temperature glide, can be adjusted by appropriate selection of the components and the amounts thereof forming the composition and in this way it is possible to tailor the composition to its intended application.

According to the present invention there is provided a non-azeotropic (zeotropic) refrigerant composition comprising:

- (A) a first component comprising fluoromethane (R-41) optionally together with carbon dioxide (CO₂);

(B) a second component comprising at least one hydrofluorocarbon selected from the group consisting of 1,1,1,2-tetrafluoroethane (R-134a) and

1,1,2,2-tetrafluoroethane (R-134); and optionally

(C) a third component comprising at least one hydrofluorocarbon selected from

5 the group consisting of difluoromethane (R-32), 1,1,1-trifluoroethane (R-143a) and pentafluoroethane (R-125).

The zeotropic refrigerant composition of the invention comprises at least the first and second components defined above.

The first component (component (A)) comprises fluoromethane (R-41) or a mixture
10 of R-41 and carbon dioxide (CO₂). Both of these compounds have a low temperature refrigeration action, with CO₂ subliming at around -78.5°C and R-41 having a boiling point of around -78.4 °C.

The second component (component (B)) comprises at least one hydrofluorocarbon selected from the group consisting of 1,1,1,2-tetrafluoroethane (R-134a) and
15 1,1,2,2-tetrafluoroethane (R-134) which have boiling points of around -26.5°C and -19.7°C respectively. It is apparent that the boiling points of both R-134a and R-134 are appreciably higher than the boiling point of the R-41 and the sublimation temperature of the CO₂ (if included) making up the first component which means that the refrigerant composition of the invention is capable of boiling and condensing over a
20 wide temperature range, i.e. it can exhibit a wide temperature glide in both the evaporator and condenser. Although the second component may contain a mixture of R-134a and R-134, it will preferably contain just one of these compounds and more preferably will contain just R-134a.

The refrigerant composition of the invention may also contain a third component
25 (component (C)) comprising at least one hydrofluorocarbon selected from the group consisting of difluoromethane (R-32), 1,1,1-trifluoroethane (R-143a) and pentafluoroethane (R-125) which have boiling points of around -51.6°C, -47.6°C and -48.5°C respectively. The third component (if included) may contain just one of the three specified compounds or it may comprise a mixture, for example an azeotropic or
30 azeotrope-like mixture, of any two or all three of these compounds. In a preferred

embodiment, the third component (if included) is R-32, R-125 or a mixture comprising R-32 and R-125.

The amounts of the first and second components and the amount of the third component (if included) in the refrigerant composition may be varied within wide limits, but typically the refrigerant composition will comprise from 2 to 45 % by weight of the first component, from 15 to 98 % by weight of the second component and from 0 to 60 % by weight (for example, from 2 to 60 % by weight) of the third component. Preferably, the refrigerant composition will comprise from 2 to 35 % by weight of the first component, from 25 to 98 % by weight of the second component and from 0 to 50 % by weight (for example, from 2 to 50 % by weight) of the third component.

When the optional third component is not included and the second component is R-134a, then the refrigerant composition preferably comprises from 2 to 25 % by weight, more preferably from 2 to 10 % by weight, of the first component and from 75 to 98 % by weight, more preferably from 90 to 98 % by weight, of the second component.

When the optional third component is not included and the second component is R-134, then the refrigerant composition preferably comprises from 2 to 30 % by weight, more preferably from 2 to 15 % by weight, of the first component and from 70 to 98 % by weight, more preferably from 85 to 98 % by weight, of the second component.

When the optional third component is included, then the refrigerant composition preferably comprises from 2 to 25 % by weight, more preferably from 2 to 15 % by weight, of the first component, from 35 to 96 % by weight, more preferably from 75 to 96 % by weight, of the second component and from 2 to 40 % by weight, more preferably from 2 to 20 % by weight, of the third component.

The refrigerant composition of the invention may also be combined with one or more hydrocarbon compounds in an amount which is sufficient to allow the composition to transport a mineral oil or alkyl benzene type lubricant around a refrigeration circuit and return it to the compressor. In this way, inexpensive lubricants based on mineral oils or alkyl benzenes may be used to lubricate the compressor.

Suitable hydrocarbons for use with the refrigerant composition of the invention are those containing from 2 to 6 carbon atoms, with hydrocarbons containing from 3 to 5 carbon atoms being preferred. Propane and pentane are particularly preferred hydrocarbons, with pentane being especially preferred.

5 Where a hydrocarbon is combined with the refrigerant composition of the invention, it will preferably be present in an amount of from 1 to 10 % by weight on the total weight of the refrigerant composition.

The refrigerant composition of the invention may also be used in combination with the types of lubricants which have been specially developed for use with
10 hydrofluorocarbon based refrigerants. Such lubricants include those comprising a polyoxyalkylene glycol base oil. Suitable polyoxyalkylene glycols include hydroxyl group initiated polyoxyalkylene glycols, e.g. ethylene and/or propylene oxide oligomers/polymers initiated on mono- or polyhydric alcohols such as methanol, butanol, pentaerythritol and glycerol. Such polyoxyalkylene glycols may also be
15 end-capped with suitable terminal groups such as alkyl, e.g. methyl groups. Another class of lubricants which have been developed for use with hydrofluorocarbon based refrigerants and which may be used in combination with the present refrigerant compositions are those comprising a neopentyl polyol ester base oil derived from the reaction of at least one neopentyl polyol and at least one aliphatic carboxylic acid or an
20 esterifiable derivative thereof. Suitable neopentyl polyols for the formation of the ester base oil include pentaerythritol, poly(pentaerythritols) such as di- and tri(pentaerythritol), trimethylol alkanes such as trimethylol ethane and trimethylol propane, and neopentyl glycol. The esters may be formed with linear and/or branched aliphatic carboxylic acids, such as linear and/or branched alkanolic acids. Preferred acids are selected from the C₅₋₈,
25 particularly the C_{5,7}, linear alkanolic acids and the C₅₋₁₀, particularly the C_{5,9}, branched alkanolic acids. A minor proportion of an aliphatic polycarboxylic acid, e.g. an aliphatic dicarboxylic acid, may also be used in the synthesis of the ester in order to increase the viscosity thereof. Usually, the amount of the carboxylic acid(s) which is used in the synthesis will be sufficient to esterify all of the hydroxyl groups contained in the polyol,
30 although residual hydroxyl functionality may be acceptable.

The single fluid refrigerants and azeotropic refrigerant blends which are used in conventional heat transfer devices boil at a constant temperature in the evaporator under constant pressure conditions, and so produce an essentially constant temperature profile across the evaporator. The temperature of the heat transfer fluid being cooled, which may be air or water for example, drops fairly rapidly on first contacting the cold surfaces provided by the refrigerant evaporating in the evaporator owing to the large difference in temperature between that fluid and the evaporating refrigerant. However, since the temperature of the heat transfer fluid is progressively reduced as it passes along the length of the evaporator, there is a progressive reduction in the temperature differential between the fluid and the evaporating refrigerant and a consequent reduction in the heat transfer or cooling rate.

In contrast, the refrigerant composition of the invention is a non-azeotropic (zeotropic) composition which boils over a temperature range under constant pressure conditions so as to create a temperature glide in the evaporator which can be exploited to reduce the energy required to operate the heat transfer device, e.g. by making use of the Lorentz cycle. One technique for exploiting the temperature glide involves the use of a heat transfer device equipped with a counter current flow evaporator and/or condenser in which the refrigerant and the heat transfer fluid are caused to flow counter currently to each other. With such an arrangement, it is possible to minimise the temperature difference between the evaporating and condensing refrigerant whilst maintaining a sufficiently high temperature difference between the refrigerant and the external fluid(s) to cause the required heat transfer to take place.

The consequence of minimising the temperature difference between the evaporating and condensing refrigerant in the same system is that the pressure difference is also minimised. As a result, the overall energy efficiency of the system is improved as less energy is consumed to bring about the refrigerant pressure rise from evaporator to condenser conditions.

The zeotropic refrigerant composition of the present invention may be used to provide the desired cooling in heat transfer devices such as air conditioning and low temperature refrigeration systems by a method which involves condensing the refrigerant composition and thereafter evaporating it in a heat exchange relationship

with a heat transfer fluid to be cooled. In particular, the refrigerant composition of the invention may be usefully employed as a replacement for refrigerant R-22.

The present invention is now illustrated but not limited with reference to the following example.

5

Example 1

The performance of seven refrigerant compositions of the invention in a refrigeration cycle of the type prevailing in an air conditioning system was evaluated using standard
10 refrigeration cycle analysis techniques in order to assess the suitability thereof as a replacement for R-22. The operating conditions which were used for the analysis were chosen as being typical of those conditions which are found in an air conditioning system, and counter current flow at the heat exchangers was assumed.

In order to illustrate the benefit of the zeotropic refrigerant compositions of the
15 invention in terms of their improved energy efficiency, it was first necessary to define the inlet and outlet temperatures of the heat transfer fluids at each heat exchanger (evaporator and condenser). The temperatures in the evaporator and condenser, assuming that a single fluid refrigerant was used in the cycle, were then chosen and these temperatures together with the inlet and outlet temperatures of the heat transfer
20 fluids referred to above were used to determine a target log mean temperature difference for each heat exchanger. In the cycle analysis itself, the refrigerant inlet and outlet temperatures at both the evaporator and condenser were adjusted until the target log mean temperature difference was achieved for each heat exchanger. When the target log mean temperature difference for each heat exchanger was achieved, the
25 various properties of the refrigerant composition in the cycle were recorded.

The following refrigerant compositions were subjected to the cycle analysis:

- (1) A composition comprising 5 % by weight R-41 and 95 % by weight R-134a.
- 30 (2) A composition comprising 10 % by weight R-41 and 90 % by weight R-134a.

- (3) A composition comprising 5 % by weight R-41, 5 % by weight R-125 and 90% by weight R-134a.
- (4) A composition comprising 5 % by weight R-41, 10 % by weight R-125 and 85 % by weight R-134a.
- 5 (5) A composition comprising 5 % by weight R-41, 5 % by weight R-32 and 90 % by weight R-134a.
- (6) A composition comprising 5 % by weight R-41, 5 % by weight R-32, 5 % by weight R-125 and 85 % by weight R-134a.
- (7) A composition comprising 10 % by weight R-41, 5 % by weight R-125 and 85
10 % by weight R-134a.

The following operating conditions were used in the cycle analysis.

EVAPORATOR:

15

Evaporator Temperature:	10°C
Inlet Temperature of Heat Transfer Fluid	25°C
Outlet Temperature of Heat Transfer Fluid	15°C
Log Mean Temperature Difference for Evaporator	9.1°C

20

CONDENSER:

Condenser Temperature:	42°C
Inlet Temperature of Heat Transfer Fluid	30°C
25 Outlet Temperature of Heat Transfer Fluid	40°C
Log Mean Temperature Difference for Condenser	5.58°C

Amount of Superheat:	15°C
Amount of Subcooling:	5°C
30 Isentropic Compressor Efficiency:	75 %
Cooling Duty:	1 kW

The results of analysing the performance of the seven refrigerant compositions in an air conditioning cycle using these operating conditions are given in Table 1.

The performance parameters of the refrigerant compositions which are presented in Table 1, i.e. condenser pressure, evaporator pressure, discharge temperature, refrigeration capacity (by which is meant the cooling duty achieved per unit swept volume of the compressor), coefficient of performance (COP) (by which is meant the ratio of cooling duty achieved to mechanical energy supplied to the compressor), and the glides in the evaporator and condenser (the temperature range over which the refrigerant composition boils in the evaporator and condenses in the condenser), are all art recognised parameters.

The performance of refrigerant R-22 and R-134a under the same operating conditions is also shown in Table 1 by way of comparison.

It is apparent from Table 1 that the refrigerant compositions of the invention boiled over a wide temperature range in the evaporator and condensed over a wide temperature range in the condenser, i.e. they exhibited wide glide behaviour in both heat exchangers, and that this property enhanced the energy efficiency of the air conditioning cycle as is evident from the higher values recorded for the coefficient of performance in the case of the refrigerant compositions of the invention. It is also apparent from Table 1 that the refrigerant composition of the invention can exhibit a performance in an air conditioning system which, except for the wide glide behaviour and its attendant advantages, is comparable to that of refrigerant R-22.

25

30

TABLE I

Refrigerant % by weight	R22 100	R134a 100	41/134a 5/95	41/134a 10/90	41/125/134a 5/5/90	41/125/134a 5/10/85	41/32/134a 5/5/90	41/32/125/134a 5/5/85	41/125/134a 10/5/85
Evaporator Pressure (bar)	6.79	4.14	5.44	6.67	5.63	5.83	5.85	6.04	6.87
Condenser Pressure (bar)	16.14	10.66	13.36	16.32	13.75	14.16	14.2	14.6	16.73
Discharge Temperature (°C)	79.7	79.7	72.4	78.5	72.3	72.2	74.3	74.1	78.1
Coefficient of Performance (COP)	5.34	5.36	5.77	5.64	5.75	5.74	5.77	5.76	5.63
COP Relative to R22	1	1.01	1.08	1.06	1.08	1.07	1.08	1.08	1.05
Refrigeration Capacity (KJ/m ³)	4,348	2,857	3,846	4,545	4,000	4,000	4,000	4,167	4,762
Refrigeration Capacity Relative to R-22	1	0.66	0.88	1.05	0.92	0.92	0.92	0.96	1.1
Evaporator Glide °C	0	0	7.2	8	7.5	7.7	7.9	8.1	12.1
Condenser Glide °C	0	0	9.8	14.8	9.9	9.9	10	10	14.5

5

10

15

Claims:

1. A non-azeotropic refrigerant composition comprising:
 - (A) a first component comprising fluoromethane; and
 - 5 (B) a second component comprising at least one hydrofluorocarbon selected from the group consisting of 1,1,1,2-tetrafluoroethane and 1,1,2,2-tetrafluoroethane.
2. A non-azeotropic refrigerant composition as claimed in claim 1 wherein the first component (A) additionally comprises carbon dioxide.
- 10 3. A non-azeotropic refrigerant composition as claimed in claim 1 or claim 2 wherein the second component is 1,1,1,2-tetrafluoroethane.
4. A non-azeotropic refrigerant composition as claimed in claim 3 which comprises from 2 to 25 % by weight of the first component (A) and from 75 to 98 % by weight of the second component (B).
- 15 5. A non-azeotropic refrigerant composition as claimed in claim 4 which comprises from 2 to 10 % by weight of the first component (A) and from 90 to 98 % by weight of the second component (B).
6. A non-azeotropic refrigerant composition as claimed in claim 1 or claim 2 wherein the second component is 1,1,2,2-tetrafluoroethane.
- 20 7. A non-azeotropic refrigerant composition as claimed in claim 6 which comprises from 2 to 30 % by weight of the first component (A) and from 70 to 98 % by weight of the second component (B).
8. A non-azeotropic refrigerant composition as claimed in claim 7 which comprises from 2 to 15 % by weight of the first component (A) and from 85 to 98 % by weight of
- 25 the second component (B).
9. A non-azeotropic (zeotropic) refrigerant composition comprising:
 - (A) a first component comprising fluoromethane;
 - (B) a second component comprising at least one hydrofluorocarbon selected from the group consisting of 1,1,1,2-tetrafluoroethane and 1,1,2,2-tetrafluoroethane; and
 - 30 optionally

- (C) a third component comprising at least one hydrofluorocarbon selected from the group consisting of difluoromethane, 1,1,1-trifluoroethane and pentafluoroethane.
10. A non-azeotropic refrigerant composition as claimed in claim 9 wherein the
5 first component (A) additionally comprises carbon dioxide.
11. A non-azeotropic refrigerant composition as claimed in claim 9 or claim 10 which comprises from 2 to 45 % by weight of the first component (A), from 15 to 98 % by weight of the second component (B) and from 0 to 60 % by weight of the third component (C).
- 10 12. A non-azeotropic refrigerant composition as claimed in claim 11 which comprises from 2 to 35 % by weight of the first component (A), from 25 to 98 % by weight of the second component (B) and from 0 to 50 % by weight of the third component (C).
- 15 13. A non-azeotropic refrigerant composition as claimed in claim 9 or claim 10 which comprises the optional third component (C).
14. A non-azeotropic refrigerant composition as claimed in claim 11 which comprises the optional third component (C) in an amount of from 2 to 60 % by weight on the total weight of the refrigerant composition.
- 20 15. A non-azeotropic refrigerant composition as claimed in claim 12 which comprises the optional third component (C) in an amount of from 2 to 50 % by weight on the total weight of the refrigerant composition.
- 25 16. A non-azeotropic refrigerant composition as claimed in claim 13 which comprises from 2 to 25 % by weight of the first component (A), from 35 to 96 % by weight of the second component (B) and from 2 to 40 % by weight of the third component (C).
17. A non-azeotropic refrigerant composition as claimed in claim 16 which comprises from 2 to 15 % by weight of the first component (A), from 75 to 96 % by weight of the second component (B) and from 2 to 20 % by weight of the third component (C).
- 30 18. A refrigeration or air conditioning system containing a non-azeotropic refrigerant composition as claimed in any one of claims 1 to 17.

19. The use in a refrigeration or air conditioning system of a non-azeotropic refrigerant composition as claimed in any one of claims 1 to 17.

20. A method for providing cooling which comprises condensing a non-azeotropic refrigerant composition as claimed in any one of claims 1 to 17 and thereafter
5 evaporating it in a heat exchange relationship with a fluid to be cooled.

10

15

20

25

30

INTERNATIONAL SEARCH REPORT

International Application No
PCT/GB 95/01592

A. CLASSIFICATION OF SUBJECT MATTER
IPC 6 C09K5/04

According to International Patent Classification (IPC) or to both national classification and IPC

B. FIELDS SEARCHED

Minimum documentation searched (classification system followed by classification symbols)
IPC 6 C09K

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Electronic data base consulted during the international search (name of data base and, where practical, search terms used)

C. DOCUMENTS CONSIDERED TO BE RELEVANT

Category *	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
Y	EP,A,0 299 614 (DU PONT DE NEMOURS) 18 January 1989 see abstract; claims 1-4; table 2 ---	1-20
Y,P	WO,A,94 17153 (IMPERIAL CHEMICAL INDUSTRIES) 4 August 1994 see page 8, line 17 - page 9, line 1; claims 1-11 see abstract ---	1-20
A,P	US,A,5 340 490 (DECAIRE) 23 August 1994 see abstract; claims 1-3; example 1 ---	1-20
A	DE,A,41 16 274 (FORSCHUNGSZENTRUM FUER KAELTE TECHNIK) 19 November 1992 see abstract; claims 1-4; example 1 ---	1-20
	-/--	

Further documents are listed in the continuation of box C.

Patent family members are listed in annex.

* Special categories of cited documents :

- "A" document defining the general state of the art which is not considered to be of particular relevance
 "E" earlier document but published on or after the international filing date
 "L" document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified)
 "O" document referring to an oral disclosure, use, exhibition or other means
 "P" document published prior to the international filing date but later than the priority date claimed

- "T" later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention
 "X" document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step when the document is taken alone
 "Y" document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art.
 "&" document member of the same patent family

Date of the actual completion of the international search

12 September 1995

Date of mailing of the international search report

22.09.95.

Name and mailing address of the ISA

European Patent Office, P.B. 5818 Patentlaan 2
NL - 2280 HV Rijswijk
Tel. (+ 31-70) 340-2040, Tx. 31 651 epo nl,
Fax: (+ 31-70) 340-3016

Authorized officer

Nicolas, H

INTERNATIONAL SEARCH REPORT

International Application No
PCT/GB 95/01592

C.(Continuation) DOCUMENTS CONSIDERED TO BE RELEVANT		
Category *	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
A	<p>WO,A,93 14174 (IMPERIAL CHEMICAL INDUSTRIES) 22 July 1993 see page 3, line 1 - page 4, line 7; claims 1-8 see abstract</p> <p style="text-align: center;">-----</p>	1-20

1

INTERNATIONAL SEARCH REPORT

Information on patent family members

International Application No

PCT/GB 95/01592

Patent document cited in search report	Publication date	Patent family member(s)	Publication date
EP-A-299614	18-01-89	US-A- 4810403	07-03-89
		AU-B- 597221	24-05-90
		AU-B- 1752288	15-12-88
		CN-B- 1025214	29-06-94
		DE-A- 3870525	04-06-92
		JP-A- 64001787	06-01-89

WO-A-9417153	04-08-94	AU-B- 5838594	15-08-94

US-A-5340490	23-08-94	AU-B- 6919394	13-02-95
		WO-A- 9502652	26-01-95

DE-A-4116274	19-11-92	NONE	

WO-A-9314174	22-07-93	AU-B- 3168893	03-08-93
		CA-A- 2126187	22-07-93
		EP-A- 0620838	26-10-94
		JP-T- 7503741	20-04-95
