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3,703,481 AQUEOUS-BASED COSMETIC DETERGENT COMPOSITIONS

COMPOSITIONS
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ABSTRACT OF THE DISCLOSURE

Improved aqueous detergent compositions for cosmetic applications comprising water, a synthetic organic detergent, and from about 1–20%, by weight, of an ammonium, alkanolamine or alkali metal salt of a monester of a C_3 – C_6 saturated aliphatic dicarboxylic acid with a fatty acid alkanolamide. The compositions exhibit improved properties with regard to emolliency, wetting, lathering and the like.

This invention relates to improved detergent and surfactant compositions. More particularly, this invention relates to improved aqueous-based cosmetic detergent compositions having improved characteristics with regard to emolliency, wetting, lathering, hair manageability, counter-irritancy and the like.

Surfactant compositions, including cosmetic surfactant compositions such as shampoos, shaving creams, skin cleansing compositions and the like, containing various derivatives of polycarboxylic acids are known to the art and are disclosed, for example, in U.S. Pats. No. 2,452,-043, issued Oct. 26, 1948 to Lawrence H. Flett; No. 2,490,459, issued Dec. 6, 1949 to Werner Max Lilienfeld; No. 2,562,154, issued July 24, 1951 to Emil A. Vitalis; and 2,878,190, issued Mar. 17, 1959 to Vladimir Dvorkovitz et al. Compounds therein disclosed include, among others, succinic acids, sulfosuccinic acids, sulfosuccinamic acids, esters, sulfosuccinic esters of fatty alkanolamides and salts thereof and various derivatives thereof, and are reported as useful in surfactant compositions, particularly shampoos. However, the aforesaid have numerous deficiencies preventing their use in a wide va- 45 riety of aqueous-based surfactant compositions as opposed to the fatty alkanolamide monoester salt additives more particularly described herein.

In accordance with the present invention, there have been discovered improved aqueous based cosmetic detergent compositions comprising water, a member of the group of synthetic anionic, nonionic and cationic organic detergents and compatible mixtures of same, and from about 1 to 20%, preferably 3 to 10%, by weight, of a member of the group of ammonium, C₂-C₆ alkanolamine 55 and alkali metal salts of the monoester of a C₃-C₆ saturated, unsubstituted, aliphatic dicarboxylic acid with fatty acid mono- and dialkanol-amides wherein the alkanol portion contains up to 8 carbon atoms.

The compositions of the present invention contain, as 60 an essential ingredient, from about 1 to 20%, by weight, of an aforesaid salt of a monoester of a C₃-C₆ saturated, aliphtic dicarboxylic acid such as succinic, malonic, glutaric or adipic acid with a fatty acid alkanolamide.

The fatty acid alkanolamides are well known to the 65 art and generally comprise those prepared from saturated and unsaturated aliphatic monocarboxylic acids containing from about 8 to 22 carbon atoms and preferably those containing from about 10 to 18 carbon atoms. Exemplary fatty acids are caprylic, capric, lauric, myristic, palmitic, stearic, arachidic, behenic, oleic, linoleic and eleostearic acids and the like, as well as mixed

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fatty acids such as tall oil fatty acids, hydrogenated tallow fatty acids, coconut oil fatty acids and vegetable fatty acids such as are derived from cottonseed and soybean oil, as well as rosin fatty acids. The alkanolamines which may be condensed with the aforesaid fatty acids comprise both mono- and dialkanolamines containing up to 8 carbon atoms and are exemplified by monoethanolamine, diethanolamide, isopropanolamine, diisopropanolamine, butanolamine, dibutanolamine, diisobutanolamine and the 10 like. These fatty acid alkanolamides may be prepared by various well known techniques such as by condensing equimolar quantities of fatty acid with alkanolamine, or by condensing a molar excess of alkanolamine with fatty acid or by reacting a fatty acid ester with alkanolamine, or by oxyalkylating, such as oxyethylating, the amide of the fatty acid. Particularly preferred fatty alkanolamides are fatty acid monoalkanolamides, particularly monoethanolamides.

The monoesters are prepared, for instance, by reacting appropriate substantially stoichiometric quantities of C₃-C₆ aliphatic saturated dicarboxylic acids with the aforesaid fatty acid alkanolamides in order to prepare fatty acid alkanolamidedicarboxylic acid monoesters having one free —COOH available for salt formation. The fatty acid alkanolamide is preferably esterified with succinic acid. The monoester salts may be the water-soluble ammonium, C₃-C₆ alkanolamine such as ethanolamine, diethanolamine, triethanolamine, isopropanolamine, disopropanolamine and the like, or alkali metal salts, of the aforesaid monoesters of the fatty acid alkanolamides with dicarboxylic acids. Preferably the monoester salt will be the sodium salt.

The aforesaid salts of fatty acid alkanolamide-dicarboxylic acid monoesters are employable in a wide variety of aqueous-based cosmetic detergent compositions. These cosmetic detergent compositions will generally contain, in addition to the aforesaid salt of a fatty acid alkanolamide-dicarboxylic acid monoester present in amounts varying between 1 and 20%, preferably 3 to 10%, by weight, a quantity of water in the range of about 10 to 90%, by weight, and the balance will be a member of the group of synthetic organic detergents of the anionic, nonionic or cationic type.

The particular synthetic organic detergent, surfactant, soap or mixture thereof chosen for employment in the compositions of the prevent invention will be dependent upon the particular cosmetic preparation sought to be prepared. The compositions of the present invention encompass shampoos, hair preparations such as conditioners and rinses, shaving preparations, especially aerosol shave preparation, skin conditioners and cleansing compositions, dishwashing detergent compositions, hand lotions, bubble-bath compositions and the like.

The synthetic organic detergents or surface-active agents which may be employed in the compositions of the present invention may be selected from a wide variety of water-soluble anionic, nonionic or cationic surfactants such as are generally employed in aqueous cosmetic detergent or surfactant compositions which are meant to come into contact with the skin or hair.

Exemplary anionic materials are the water-soluble, straight and branched chain alkyl aryl sulfonates, particularly the alkyl benzene sulfonates, wherein the alkyl group contains from about 8 to 15 carbon atoms, the lower alkyl aryl or hydrotropic sulfonates such as sodium xylene sulfonate; the olefin sulfonates such as those produced by sulfonating a C_{10} to C_{20} straight chain alphaolefin; hydroxy C_{10} to C_{24} alkyl sulfonates; water-soluble alkene disulfonates containing from about 10 to 24 carbon atoms; the normal and secondary higher alkyl detergents, particularly those having about 8 to 15 carbon

atoms in the alcohol residue such as lauryl or coconut fatty alcohol sulfate; sulfuric acid esters of polyhydric alcohols partially esterified with higher fatty acids such as coconut oil monoglyceride monosulfate, coconut ethanolamide sulfate, lauric acid amide or taurine and the like; the various soaps or salts of fatty acids containing from about 8 to 22, particularly 10 to 18, carbon atoms, such as the sodium, potassium, ammonium and lower alkanolamine, particularly mono-, di- and triethanolamine salts of fatty acids such as stearic acid, oleic acid, coconut fat- 10 ty acid, fatty acids derived from palm oil, soybean oil, tallow and the like. Particularly preferred anionic surfactants include the fatty alcohol and ether alcohol sulfates and the sodium salts of fatty acids containing from about 12 to 18 carbon atoms.

Exemplary nonionic detergents which may be employed in the compositions of the present invention are the polyalkylene oxide condensation products with a hydrophobic organic compound. The hydrophobic organic group usually contains from about 8 to 30 carbon atoms and is con- 20 densed with from about 5 to 50 alkylene oxide groups such as ethylene oxide, propylene oxide, butylene oxide, and the like. Specific condensation products include the polyoxyethylene ethers of alkyl phenols wherein the alkyl group contains from about 6 to 12 carbon atoms and 25 containing from about 8 to 18 mols of ethylene oxide per mol of said alkyl phenols; polyalkylene oxide ethers of fatty alcohols such as lauryl, myristyl, cetyl, stearyl and oleyl alcohols, and C_{10} to C_{20} oxo alcohols, which are condensed with from about 10 to 30 mols of ethylene 30 $\,$ oxide per mol of said alcohols; polyoxyalkylene esters of organic acids such as the higher fatty acids, rosin acids, tall oil acids and the like, said esters containing from about 12 to 30 mols of ethylene oxide and about 8 to 22 carbon atoms in the acyl group; polyalkylene oxides condensates 35 with higher fatty acid amides, such as the higher fatty acid primary amides, mono- and diethanolamides as well as condensates of higher fatty acid sulfonamides.

Further suitable nonionic surfactants are the higher fatty acid alkanolamides, such as the monoethanolamides, diethanolamides and isopropanolamides wherein the acyl radical has an average of from about 10 to 18 carbon atoms. Examples are coconut fatty acid, capric, lauric, oleic and myristic diethanolamides and the corresponding monoethanolamides and isopropanolamides; the fatty tertiary amine oxides which may be represented by the general formula $R_1R_2R_3N\rightarrow O$ wherein R_1 represents C_{10} to C₁₈ alkyl such as decyl, dodecyl, tetradecyl, octadecyl and the like and R2 and R3 represent methyl or ethyl.

Particularly preferred members of the above group of 50 nonionic surfactants include the fatty acid alkanolamides and the polyoxyalkylene, particularly the ethylene oxide condensates thereof.

Exemplary cationic surfactants which may be employed in the compositions of the present invention are quaternary ammonium salts having long chain hydrophobic radicals such as C_{10} to C_{20} alkyl radicals as exemplified by 1-methyl-1-higher alkyl amido-ethyl-2-higher alkyl imidazolinum salts wherein the alkyl groups have 10 to 20 carbon atoms, ethoxylated long chain fatty quaternary 60 ammonium compounds, higher alkyl benzyldimethylammonium salts, higher alkyl isoquinolinium salts, di (higher alkyl) dimethylammonium salts, and the like. The anion of the water-soluble quaternary ammonium compound is most usually a chloride or methosulfate, but other ions 65 may be used such as bromide, phosphate, dialkyl phosphate or acetate ions.

The aforesaid anionic, cationic and nonionic surfactants are generally present in amounts ranging from about 5 to 85%, by weight, depending upon the particular cos- 70 metic formulation prepared. Admixtures of the aforesaid may be employed so long as such admixtures are compatible and do not cause precipitation or otherwise adversely affect the stability of the system.

poos containing water, 5-10% fatty amine oxide, 3-5% sodium fatty alcohol sulfate and about 3-5% of said C3-C₆ dicarboxylic acid-fatty acid alkanolamide monoester salt; dishwashing liquids containing 15-20% alkyl benzene sulfonate, 5-10% sulfated ethoxylated aliphatic alcohol, 3-10% hydrotropic sulfonate and from 5-10% of said C₃-C₆ aliphatic dicarboxylic acid-fatty acid alkanolamide monoester salt; shampoo bars containing 60-70% mixed fatty acid soaps and 10-20% of said C₃-C₆ aliphatic dicarboxylic acid-fatty acid alkanolamide monoester and the like.

Various supplemental materials may be added to the compositions of the present invention such as are generally employed in cosmetic preparations. Exemplary are additives such as lanolin, ethoxylated lanolin, glycerol and polyglycerols, lower alcohols such as ethanols, dyes, perfumes, antibacterial agents such as hexachlorophene, preservative agents such as formalin, sequestering agents such as water-soluble phosphate compounds and the like.

The aqueous-based cosmetic detergent compositions of the present invention possess a number of highly significant advantages. They exhibit excellent emolliency when applied to the skin as hand lotions, skin cleansing compositions, shaving creams, or in the form of dishwashing detergents. Emolliency refers to the desirable after-feel imparted to the skin and the conditioning and protective effects on the skin by the retention of natural greases or oils and the prevention and relief of dryness. Shampoos formulated in accordance with the present invention impart an observable improved manageability and luster to the hair and emolliency to the scalp, and in addition, they have a diminished degreasing effect and are non-irritating to the scalp and skin. Moreover, compositions such as shaving preparations also show excellent wetting and lathering properties and produce a soothing and non-irritating effect upon the skin. Generally the compositions of the present invention have the capacity to reduce and suppress the irritating tendency of conventional detergents commonly employed in detergent compositions. Even in hard water, wetting and lathering properties are not substantially diminished.

The invention is further illustrated by the following examples, which are not to be considered as limitative of its scope. Percentages are by weight.

EXAMPLE I

A shampoo is prepared containing the following ingredients:

	Percen	
(a) Sodium lauryl sulfate	(7
(b) Lauric diethanolamide		
(c) Triethanolamine salt of the monoester of suc	cinic	
acid and coconut monoethanolamide		3
(d) Water	8′	7

The above formulation is found to be high lathering even in hard water, and non-drying to the hair and scalp. It is observed that after use, the hair has good manageability and luster, and does not require the use of a hair conditioning rinse.

EXAMPLE II

A liquid dishwashing detergent is prepared containing the following components:

Perc	
(a) Ammonium dodecyl benzene sulfonate	18
(b) Ammonium sulfate of 4 mol ethylene oxide adduct	
of nonyl phenol	8
(c) Sodium xylene sulfonate	5
(d) Ethanol	
(e) Ammonium salt of the monoester of lauric mono-	
ethanolamide and succinic acid	5
(f) Water	51

The above formulation is found not only to be nonirritating to the skin but produces an emollient effect on Exemplary cosmetic detergent compositions are sham- 75 the skin after use. The identical composition, except for the

ammonium lauric monoethanolamide monosuccinate is EXAMPLE IX prepared for purposes of comparison and is found to pro-A hairdressing composition is prepared by combining the duce a pronounced drying effect on the hands with little following ingredients: or no emollient effect on the skin perceptible. Percent **EXAMPLE III** (a) Polyoxyethylene ether of lauryl alcohol having 23 oxyethylene units _____ A soap bar is prepared by combining the following in-(b) Lauric diethanolamide _____ gredients: (d) Sodium salt of the monoester of succinic acid (a) Tallow soap _____ 54 10 and coconut monoethanolamide _____ 5.0 (b) Coconut soap _____ 16 (e) Glycerine _____ 10.0 (c) Ethoxylated lanolin _____ (f) Water _____ (d) Sodium salt of the monoester of succinic acid and coconut monoethanolamide _____ 18 EXAMPLE X (e) Water _____ 10 15 A shaving cream suitable for use in aerosol-type dis-This soap bar forms a lather with water which is quite pensers is prepared by combining the following ingrecopious, similar to that of a shampoo lather and has a dients: desirable after-feel on the skin. (a) Glycerol ______ 10.0 (b) Triethanolamine _____ 4.0 EXAMPLE IV A shampoo is prepared by combining the following in-(c) Stearic acid _____ 4.5 (d) Coconut fatty acids ______(e) Glyceryl monostearate _____ (a) Coconut diethanolamide _____ 5.0 (f) Triethanolamine salt of coconut monoethanolam-(b) Sodium lauryl sulfate _____ 9.0 3.0 ide succinic acid monoester _____ (c) Triethanolamine dodecyl benzene sulfonate ____ 8.0 (g) Water _____ 66.0 (d) Perfume _____ 0.4 (h) Propellent (dichlorodifluoromethane-dichloro-(e) Monosodium phosphate difluoroethane mixture) ______ (f) Sodium salt of the monoester of malonic acid with oleic isopropanolamide _____ 2.0 EXAMPLE XI (g) Water _____ 75 An emollient skin cream is prepared by combining the The product when applied to the hair significantly imfollowing ingredients: proves manageability with no drying effect on the scalp.

 (a) Glyceryl monostearate
 10.0

 (b) Glycerol
 15.0

 (c) Cetyl trimethyl ammonium chloride
 0.2

 EXAMPLE V Example IV is repeated, except that for ingredient (f) (d) Sodium salt of the monoester of succinic acid there is employed the ammonium salt of the monoester of and coconut monoethanolamide _____ glutaric acid with lauric diethanolamide and equivalent re- 40 sults are obtained. EXAMPLE VI **EXAMPLE XII** A shampoo is prepared by combining the following in-A various emollient hand lotion is prepared by comgredients and is observed by the user to impart a greater bining the following ingredients: Percent 45 degree of manageability and luster to the hair. ___ 2.0 (a) Lanolin --(b) Coconut diethanolamide _____ (a) Polyoxyethylene sorbitan monolaurate _____ (c) Sodium salt of the monoester of succinic acid (b) Cetyl benzyl dimethyl ammonium bromide ____ with coconut butanolamide _____ 5.0 0.4 (c) Perfume ______ (d) Hexachlorophene _____ 1.0 50 (d) Triethanolamine salt of the monoester of suc-(e) Water _____ cinic acid and lauric monoisopropanolamide ____ (e) Water _____ 76.6 **EXAMPLE VII** What is claimed is: A composition suitable as a skin-cleansing liquid or bub-1. An aqueous-based cosmetic detergent composition ble-bath composition is prepared employing the following which consists essentially of (a) from about 10% to 90%, components: by weight, of water; (b) from about 1% to 20%, by weight, of a member of the group consisting of ammo-(a) Oleic diethanolamide _____ 3.0 nium, C2-C6 alkanolamine and alkali metal salts of mono-(b) Ammonium coconut monoglyceride _____ 20.0 esters of C₃-C₆ saturated, unsubstituted, aliphatic di-(c) Sodium lauryl sulfate _____ carboxylic acids with fatty acid mono- and di-alkanol-(d) Diethanolamine salt of the monaester of stearic amides wherein the alkanol portion contains up to 8 carmonoethanolamide and adipic acid _____ 10.0 bon atoms; and (c) from 5% to 85%, by weight, of a (e) Water _____ member of the group consisting of synthetic anionic, nonionic and cationic organic detergents and compatible mix-**EXAMPLE VIII** tures of same with the proviso that said (c) ingredient is A shampoo is prepared by combining the following indistinct from said (b) ingredient. gredients: 2. A composition according to claim 1 wherein said aliphatic dicarboxylic acid is succinic acid. Percent (a) Triethanolammonium coconut sulfate _____ 8.0 3. A compostion according to claim 1 wherein said (b) Sodium alkyl (C_{10} – C_{20}) sulfate _____ 7.0 fatty acid alkanolamide is a monoalkanolamide. (c) Sodium t-octylphenl sulfonate _____ 6.0 4. A composition according to claim 1 wherein said (d) Potassium salt of the monoester of succinic acid fatty acid contains from 10 to 18 carbon atoms. and lauric monoethanolamide _____ 1.0 5. A composition according to claim 1 wherein said

(e) Water ____ 78 75 salt is a sodium salt.

6. A composition according to claim 1 wherein said salts of said monoesters are present in an amount of from about 3% to about 10% by weight.

7. A composition according to claim 1 wherein said synthetic organic detergent is a fatty alcohol sulfate.

- 8. A composition according to claim 1 wherein said synthetic organic detergent is a member of the group consisting of fatty acid salts, fatty acid soaps and mixtures thereof.
- 9. A composition according to claim 1 wherein said 10 RICHARD D. LOVERING, Primary Examiner synthetic organic detergent is a fatty acid alkanolamide.
- 10. A composition according to claim 6 wherein said 252—90, 117, 134, 356, 527, 548, DIG. 5, DIG. 13;
 11. A composition according to claim 6 wherein said 15 monoester salt is the sodium salt of the monoester of succinic acid and coconut monoethanolamide.

monoester salt is the triethanolamine salt of the monoester of succinic acid and lauric monoethanolamide.

References Cited

UNITED STATES PATENTS

2,173,448		Katzman et al 260-404
2,733,212	1/1956	Epstein et al 252—117
2,343,431		Wells et al 260—404 X

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