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The definitive publisher version is available online at <https://doi.org/10.1016/j.biombioe.2020.105935>

Biogas as an energy vector: a review

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Abstract

Biogas is a sustainable energy vector with diverse input sources (e.g. landfills and anaerobic digestion of waste materials, wastewater treatment sludge, manure from animal production, or energy crops) and diverse applications. The nature of the substrate and the design of the biogas production process determines the composition of raw biogas. All types of biogas must be cleaned and upgraded before delivering to the consumers and in practice, the key challenge of the biogas supply chain is its cleaning and upgrading to consumers quality. The physicochemical technologies used to clean and upgrade the raw biogas are reliable, mature and at high technology readiness levels.

This paper critically reviews the biogas supply chain including feedstock supply, biogas production and upgrading/cleaning processes, potential hazards of biogas contaminants, product specification based on applications, and biogas/biomethane uses. The biogas cleaning and upgrading technologies with emphasis on cost comparison are assessed. In summary, the

24 upgrading technology alternatives and their associated costs are found substantially affected
25 by the project-specific circumstances. For instance, upgrading with chemical scrubbing might
26 be preferred in the availability of cheap on-site thermal energy. If the biomethane is planned to
27 be injected into high-pressure natural gas pipelines, those upgrading methods operating at
28 relatively high pressures (e.g. membranes) would be preferred. If the biomethane injection
29 point to the gas grid is located distant from the production site, the distribution cost will also
30 play a determinative role in the overall biogas supply chain economics. Among all these
31 factors, plant capacity seems to be a pivotal element in the economics of biogas supply chain.
32 Amendments to national and sub-national support schemes are also an important factor
33 affecting investment decisions.

34 **Keywords:** biogas upgrading, biomethane, physicochemical, techno-economics, waste-to-
35 energy.

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45 **Abbreviations**

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47	AD	anaerobic digestion
48	CCS	CO ₂ capture and storage
49	CHP	combined heat and power
50	DCM	di-chloro-methane
51	DEG	di-ethylene glycol
52	DMC	di-methyl carbonate
53	DME	di-methyl ether
54	DMEA	di-methyl ethanolamine
55	EG	ethylene glycol
56	EU	European Union
57	FF	fresh feedstock
58	FT-GTL	Fischer-Tropsch gas-to-liquid
59	GHG	greenhouse gas
60	HRAR	high rate anaerobic reactors
61	MBR	membrane bioreactors
62	IPCC	intergovernmental panel on climate change
63	LF	landfill
64	MCFC	molten-carbonate fuel cell
65	MEA	mono-ethanolamine

66	MSW	municipal solid waste
67	Mtoe	million tonnes of oil equivalent
68	O&M	operating and maintenance costs
69	PAHs	polyaromatic hydrocarbons
70	PSA	pressure swing adsorption
71	ppb	part per billion
72	SOEC	solid oxide electrolysis cell
73	Syngas	synthesis gas
74	TEG	tri-ethylene glycol
75	VOC	volatile organic compounds
76	WtE	waste-to-energy
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120 **1. Background**

121 Meeting the climate change mitigation targets requires a rapid transition from fossil fuels to
122 renewable electricity and renewable fuels. Increased urbanization rate, growing population,
123 and economic evolution are significantly altering the perspective of national solid waste
124 materials in terms of waste composition, generation pace, and waste treatment processes [1].
125 Over a few decades of research and industrial actions, today there is general consensus that
126 conversion of the waste to energy is a promising waste management option.

127 The so-called waste-to-energy (WtE) has multiple advantages. Not only it addresses the waste
128 disposal challenge, but it also offers a good opportunity for energy security, as both the
129 processes for production and consumption of energy can be located in the same geographic
130 location, unlike fossil fuels. WtE can be considered as a semi-renewable source of energy and
131 an alternative (or at least a compliment) to fossil fuels which account for over 80% of the global
132 energy consumption [2]. WtE processes comprise any waste treatment technology that
133 generates any form of energy i.e. heat, electricity, or liquid transport fuels (e.g. diesel, petrol
134 or kerosene) from a waste material feedstock.

135 Bioenergy refers to power, heat, transport fuels, and gas that is produced from biological
136 sources [3]. It is the third principal source of energy in the world, nearly emission-neutral [4]
137 and can have a positive impact on promoting and balancing existing and future energy systems
138 [5]. Dairy waste, agricultural waste, wastewater treatment plants, urban food waste and garden
139 waste, landfill (LF) gas and municipal solid waste are the principal categories for use.

140 When an organic waste is decomposed in the absence of oxygen, a blend of gases (primarily
141 methane and CO₂) is released, known as biogas [6]. Some decades ago, biogas was perceived
142 as “poor man’s fuel” [7], but today it has emerged as one of the major options in the
143 international energy planning context. The biogas production process has several advantages

144 including its feedstock flexibility which can also include MSW. In fact, when MSW is dumped
145 in the nature, it goes under biological dissociation and generates biogas the release of which to
146 the atmosphere creates significant environmental impact. The released methane has over 20–
147 times more global warming effects than CO₂ for a 100-year time horizon. But, once the gas is
148 produced in a process, it becomes a potential alternative energy source, especially for rural
149 communities. Regardless of its energy value, biogas even if flared will release CO₂ which has
150 substantially less environmental impact than methane. Last but not least, access to distributed
151 biogas resources supports the energy (gas and electricity) grid decentralization movement and
152 improves the higher uptake of variable renewable technologies such as photovoltaics (PV) and
153 wind. The actual advantage and also necessity would be in the modulation capability of the
154 renewable electricity production in order to compensate for the variability in the PV or wind
155 energy. Other environmental benefits of biogas are:

- 156 • Protection of the environment by replacing inorganic fertilizer, conservation of forest
157 vegetation, reduction of air & water pollution, and so on [8, 9].
- 158 • Green energy production in form of heat, power, vehicle fuel, and trigeneration [10].
- 159 • Reduction of greenhouse gas emissions by substituting conventional fossil fuels [9].
- 160 • Disposal of organic matters including household wastes, industrial and municipal solid
161 wastes, agricultural residues, etc. [10].
- 162 • Supply of base-load energy to the gas and electricity networks.

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165 Full utilization of the sustainable biogas potential can supply nearly 20% of the worldwide
166 natural gas demand. Currently, the sustainable biogas and biomethane potential is 570 and 730
167 million tonnes of oil equivalent (Mtoe), respectively [11].

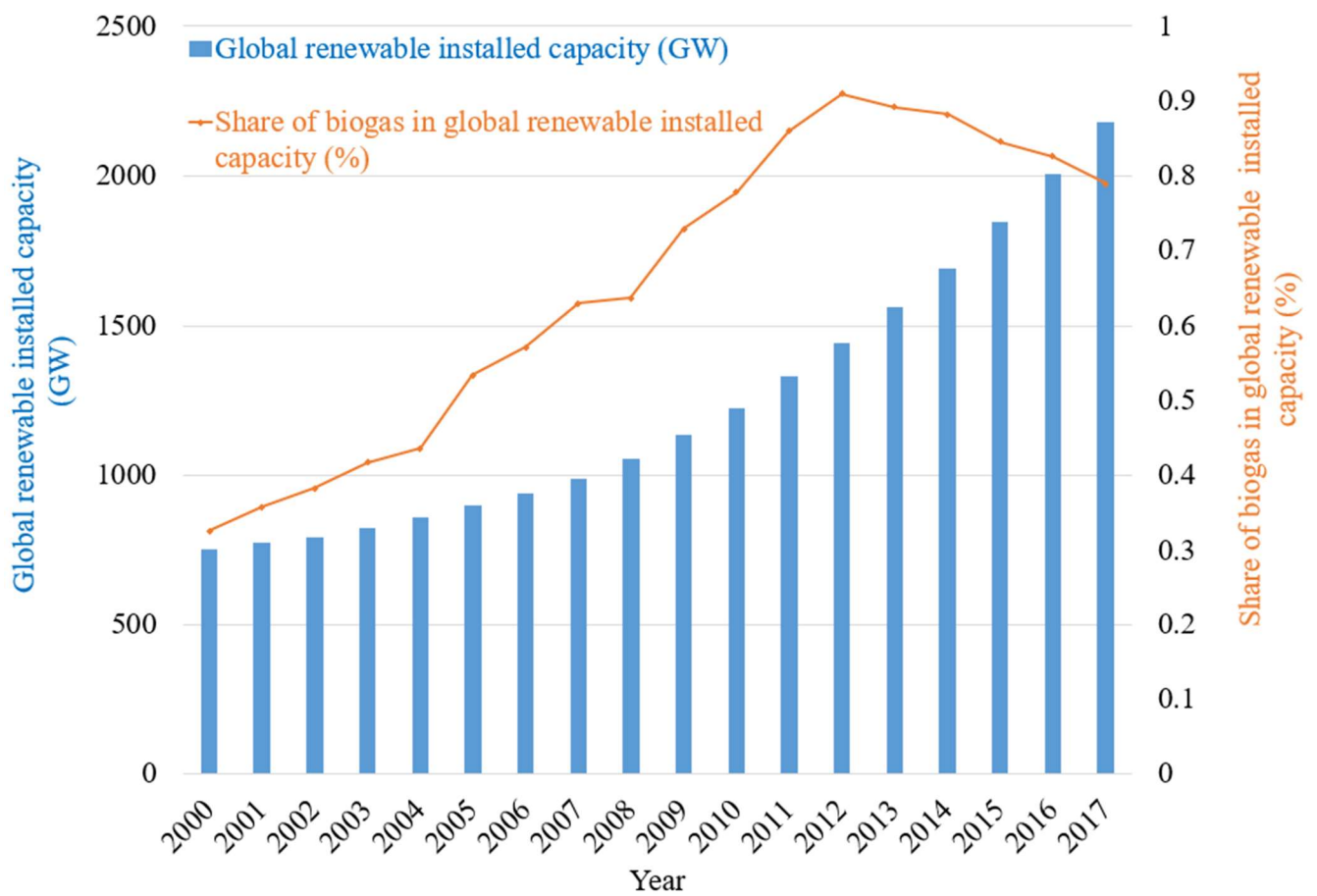
168 Given the increased political prominence of hydrogen and policies and measures to promote it,
169 the contribution of biomethane to energy security and a low emissions future energy mix could
170 be projected. In the search for ‘renewable gas’ the question of the relative economics of
171 biomethane versus renewable hydrogen becomes important. Biomethane has the advantage
172 over hydrogen since natural gas grids and appliances will not require modification, because of
173 the similarity of biomethane to natural gas. The necessary changes of the infrastructure would
174 be rather low up to a H₂ share of around 10% (molar or volumetric concentration) [12]. The
175 mole fractions of hydrogen in non-conventional gas in France is 6%, Austria 4%, Germany
176 5%, and The Netherlands 12% [13].

177 Actual biomethane production in 2018 was about 35 Mtoe. Based on the current policies, the
178 biogas consumption in 2030 and 2040 will be around 95, and 150 Mtoe. The sustainable
179 development scenarios are even more optimistic and project biogas consumption to reach
180 nearly 190 and 325 Mtoe in 2030 and 2040, respectively. Currently, most of the biogas is used
181 for onsite power and heat generation. But, as we move in time, the amount of biogas being
182 upgraded to biomethane increases and biomethane production becomes the main biogas-
183 processing pathway, leaving onsite power and heat application as the second.

184 Figure 1 (a) depicts the evolution of the global renewable installed capacity (bar chart) and the
185 share of biogas (line chart) in the global renewable installed capacity between 2000 and 2017.
186 The installed renewable capacity in 2010 was 754 GW and increased to 2182 GW in 2017. The
187 share of biogas technology in the global renewable installed capacity increased from 0.32% to
188 0.90% during 2000–2012. Thanks to the fast developments in other renewable technologies
189 such as wind and PV, despite growth in biogas installations, its share in the renewable

190 technology mix showed a little decline reaching 0.79 in 2017. Figure 1 (b) represents the global
 191 electricity generation from renewables (bar chart) and the share of electricity generation from
 192 biogas (line chart) between 2000 and 2017 [14]. The global electricity generation from
 193 renewable resources increased from 2850 TWh in 2000 to 6191 TWh in 2017. This changed
 194 the global share of biogas in electricity generation from 0.46% to about 1.42% in this time
 195 period.

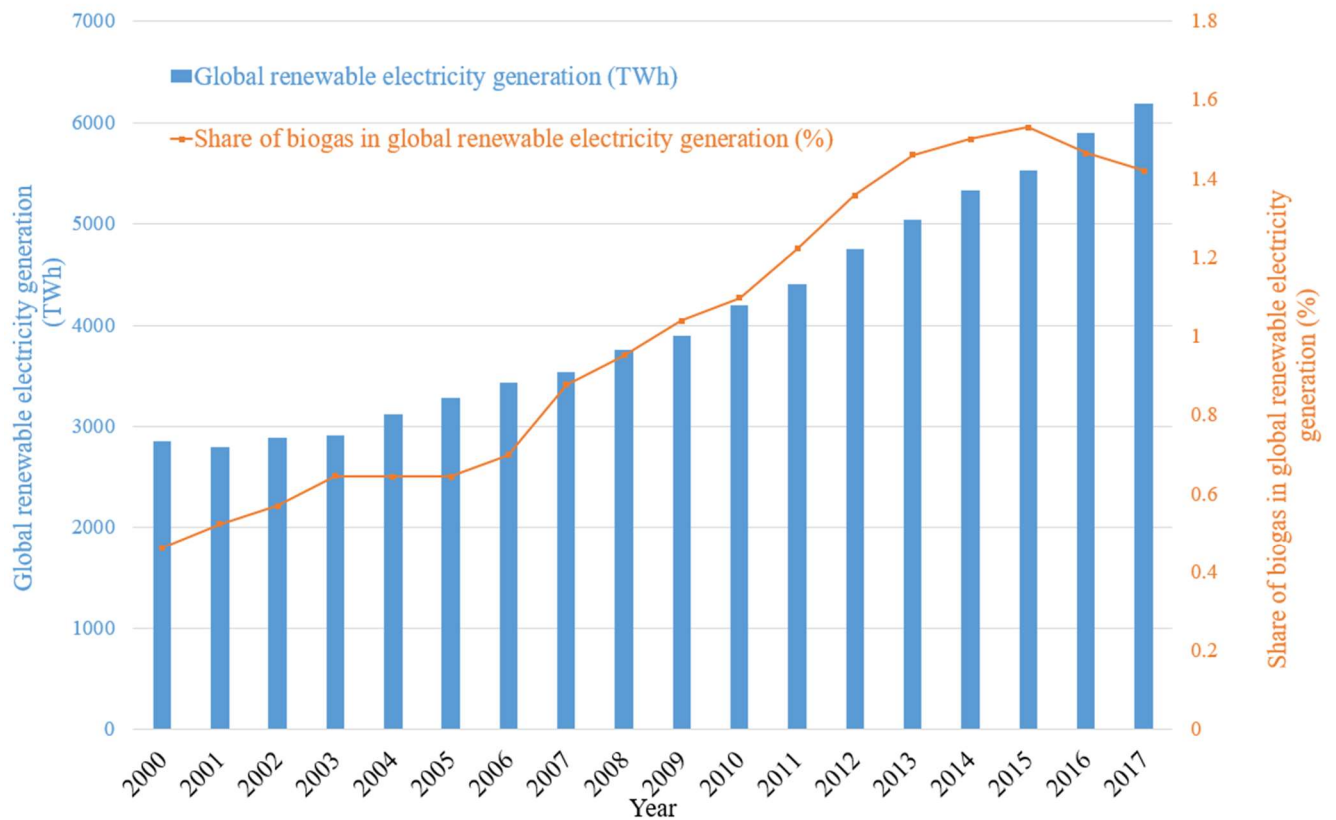
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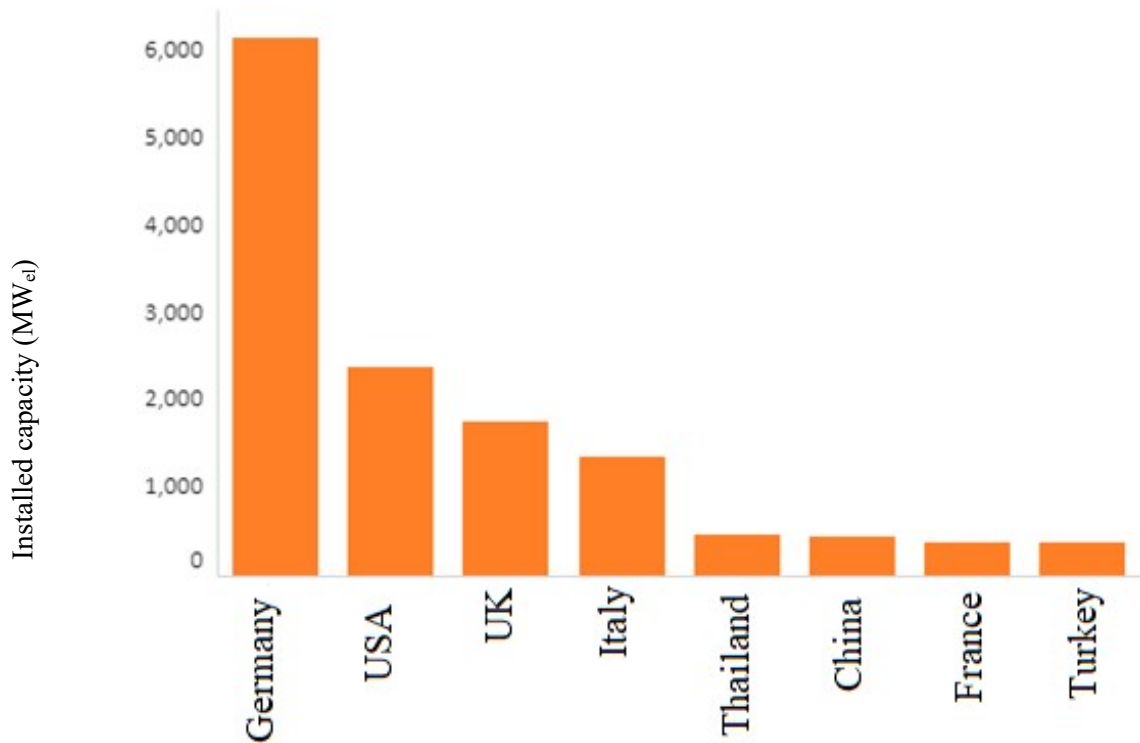
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Figure 1: a) Global renewable installed capacity (left axis) and share of biogas installed capacity (right axis) in the total renewable installed capacity during 2000–2017 (data source: [14]). b) Global electricity production from renewable resources (left axis) and share of biogas in renewable electricity generation (right axis) during 2000–2017 (data source: [14]).

Figure 2 shows the installed biogas capacity (MW_{el}) by country as of 2017. Germany with over 6 GW_{el} is on top, followed by the US, UK, Italy, Thailand, China, France, and Turkey.



207 **Figure 2:** Installed biogas capacity by country as of 2017 (data source: [15])

208 As also evident from Figure 2, Europe is the world leader in terms of the installed biogas plant
 209 capacity and biogas-based power generation [16-18]. In 2017, the total number of biogas plants
 210 in Europe was 17783 [16]. At the same year, the number of biomethane plants in Europe was
 211 540 of which 340 were feeding into grid [16]. The majority of the biomethane generation plants
 212 were in Germany (195 plants) followed by UK (92 plants) and Sweden (70 plants) [10]. The
 213 biogas production in Europe was 18.4 billion Nm³ (N represents normal conditions i.e. $T=20$
 214 °C, $P=1$ atm) that represented a share of 4 % in natural gas use [16]. It has been anticipated that
 215 the biogas production will reach 20 billion Nm³ by 2020 [19]. A review of biogas upgrading
 216 technologies in Europe and share of EU transport sector in 2030 can be found in Ref. [20].
 217 Biogas upgrading technologies are based on pressure swing adsorption, water scrubbing and
 218 chemical scrubbing. Biogas upgrading to biomethane will reach 18 billion cubic meter per year
 219 in 2030 (about 9.5 times higher than that of 2017). This is equivalent to nearly 10% of the EU's
 220 projected natural gas import for 2030. Biomethane will contribute to about 3% of the natural

221 gas consumption of EU countries in 2030 [21]. It is expected that the maritime and road
222 transport customers will use a large share of the produced biomethane in the EU.

223 In summary, biogas is a sustainable energy vector with several benefits including 1) renewable
224 source of energy, 2) lower discharge of methane to the air in comparison to LFs or traditional
225 manure management, and 3) having a high-quality digestate by-product used as fertilizer [22].

226 Although the literature is rich on biogas upgrading technologies, there exists growing attention
227 to the investment analysis and operational cost reduction of the upgrading routes. The current
228 study reviews and evaluates the various aspects of biogas as an energy vector such as biogas
229 production pathways, conventional and prospective upgrading technologies (including
230 physical, chemical, and biological [23]) with their basics of operations,
231 advantages/disadvantages, energy needs, methane recovery efficiency, market penetration,
232 biogas markets/applications in addition to economics of various upgrading technologies. The
233 cost of biogas upgrading depends on raw biogas capacity, the concentration of contaminants in
234 the raw biogas, local circumstances, energy and water cost, envisaged lifetime of the
235 investment, and so on. Furthermore, this paper covers the biogas composition for various
236 feedstocks, standard requirements of upgraded biogas for the grid injection, biogas conversion
237 to chemicals, upgraded biogas product specifications, the potential hazards of biogas
238 contaminants during production or upgrading, and final use of the upgraded biogas and CO₂.

239 The rest of the paper proceeds as follows. Section 2 describes the biogas supply chain including
240 biogas feedstock, biogas production, and the products/by-products (methane, CO₂, chemicals)
241 obtained from biogas. Section 3 addresses the potential hazards of biogas contaminants, final
242 product specification, biogas upgrading processes, in addition to physical/chemical/biological
243 approached for biogas upgrading. Costs of biogas upgrading via various technologies are
244 elaborated in Section 4, followed by conclusions in Section 5.

245 **2. Biogas supply chain**

246 The biogas supply chain is composed of feedstock supply, biogas processing, production and
247 upgrading/purification, in addition to final markets/applications, biogas uses or products (e.g.
248 biomethane injection to gas grid, biomethane conversion to chemicals, etc.). Biogas feedstocks
249 (elaborated in Section 2.1) may be in solid or slurry forms in addition to concentrated and dilute
250 liquid form. Some examples are agricultural residues, livestock manure, organic waste
251 materials, and sewage sludge. The raw biogas must be cleaned and upgraded in various degrees
252 to meet the consumers' gas composition standards (such as mole fractions of H₂, CO₂, O₂, and
253 H₂S in the upgraded biogas). Thereby, the selection of cleaning and upgrading technologies
254 (see Section 3.3 and 3.4) depends on the upgraded biogas demand, the levels of raw biogas
255 contaminants, and project-specific circumstances.

256 A list of existing production and upgrading plants in Europe and around the world based on
257 various features (such as the feedstock, final application of biogas, upgrading technology, the
258 methane content of the upgraded gas, plant capacity and date of plant commencement) is
259 available in Refs. [24, 25]. In addition, a list of providers of upgrading technologies is available
260 in Ref. [25]. This list includes the manufacturers of PSA units, water/chemical scrubbers,
261 organic physical scrubbing units, membrane modules, cryogenic units, and small-scale biogas
262 upgrading installations. Ref. [25] also gives the number of upgrading units by types in several
263 European countries.

264

265 **2.1. Biogas feedstock**

266 The content of fats, proteins, and carbohydrates in the substrates has an obvious direct impact
267 on the biogas production system and the subsequent upgrading sequence. Table 1 lists the
268 methane and biogas yield obtained from various feedstocks. It shows that depending on

269 feedstock type, methane yield can vary between 51% and 65%. The range of produced biogas
 270 volume is even much wider spanning from 25 to 202 m³ per tonne of fresh feedstock.

271 **Table 1:** Methane and biogas yields of various feedstock materials [26-30]

Feedstock	Methane yield [Vol %]	Biogas yield [Nm³/tFF] (FF: fresh feedstock)
Distillers grains with solubles	61	40
Grass silage	54	172
Pig manure	60	60
Sweet sorghum	54	108
Cattle manure	60	45
Corn silage	52	202
Liquid pig manure	65	28
Forage beet	51	111
Organic waste	61	100
Beet	53	88
Liquid cattle manure	60	25
Poultry manure	60	80
Whey	15	330
Cattle slurry	12.8	200
Flotation sludge	21.6	540

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273 Subsequently, the actual greenhouse gas (GHG) emissions reduction depends on the substrate
 274 source used. Table 2 lists the GHG emissions reduction of some feedstock with maize being
 275 the lowest (75%) and manure being the highest (148%).

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282 **Table 2:** The reduction of GHG emissions of biogas source compared to fossil fuels [31]

Feedstock	GHG reduction vs. fossil fuels [%]
Maize	75
Sugar beet (incl. tops)	85
Grass	86
Organic household waste	103
Waste from the food industry	119
Manure	148

283

284 In 2014, the total biomass supply was 59.2 Etta Joule (EJ), i.e. 10.3% of the global energy
 285 supply, with annual growth of 2.3% [32]. The key sector for biomass development is the
 286 forestry sector with 87% of the supply (fuelwood 67% of the biomass feedstock, followed by
 287 charcoal 7%, recovered wood 6%, and wood industry residues 5%). The second sector is the
 288 agriculture sector (animal/agricultural by-products and energy crops) with share of 10% to
 289 biomass supply. The third one contributes to 3% by generating energy from MSW and landfill
 290 gas. In 2015 and in terms of total biogas plants per feedstock in the EU, about 71% of biogas
 291 was produced from agriculture sources, followed by 16% from sewage, 9% from LFs and the
 292 rest from other sources [19]. Europe has more than 50% of the global agricultural area, and
 293 nearly 55% of the global waste to energy conversion [32]. The biomass supply source varies
 294 among the continents and for biogas as a renewable energy source, Europe is an evident leader
 295 [32].

296 **2.2. Biogas production**

297 Biogas is produced from different routes and environments. Its composition is a function of
 298 several factors such as the nature of the substrate used to produce biogas, and the process design
 299 [33]. The various routes of biogas production are:

- 300 • Anaerobic digestion (AD) is known as the biological conversion process in an oxygen-free
 301 environment and is carried out in four steps including hydrolysis, acidogenesis, acetogenesis,

302 and methanogenesis (Figure 3), [34, 35]. Various types of waste materials such as food waste,
303 agricultural and industrial wastes, MSW, wastewater, and crops can be used as feedstock for
304 the AD process [29, 36].

305 • Anaerobic degradation in LFs (natural decomposition of waste). An LF site is a location
306 dedicated for dumping garbage, rubbish or other sorts of solid wastes. With the growing waste
307 production from homes, offices, hospitals, schools, and markets, landfilling has been the most
308 common disposal approach. LFs are either left to pile in heaps or buried. While LFs are the
309 most cost-efficient method of disposing waste materials, they are associated with
310 environmental risks. Soil, water, and predominantly air are dirtied by the deposition of waste
311 materials in the LFs [37]. In addition, decomposition of organic materials in the oxygen-free
312 environment is slow which in long-term has negative effects on the next generations. There are
313 five distinct types of landfilling including LF as a deposit of inert waste, aerobic, semi-aerobic,
314 hybrid, and anaerobic [38]. To produce biogas from LFs, complex biochemical conversion
315 processes including different phases should be designed.

316 • Novel AD technologies including high rate anaerobic reactors (HRAR), membrane
317 bioreactors (MBR), and integrated HRAR- MBR, [39].

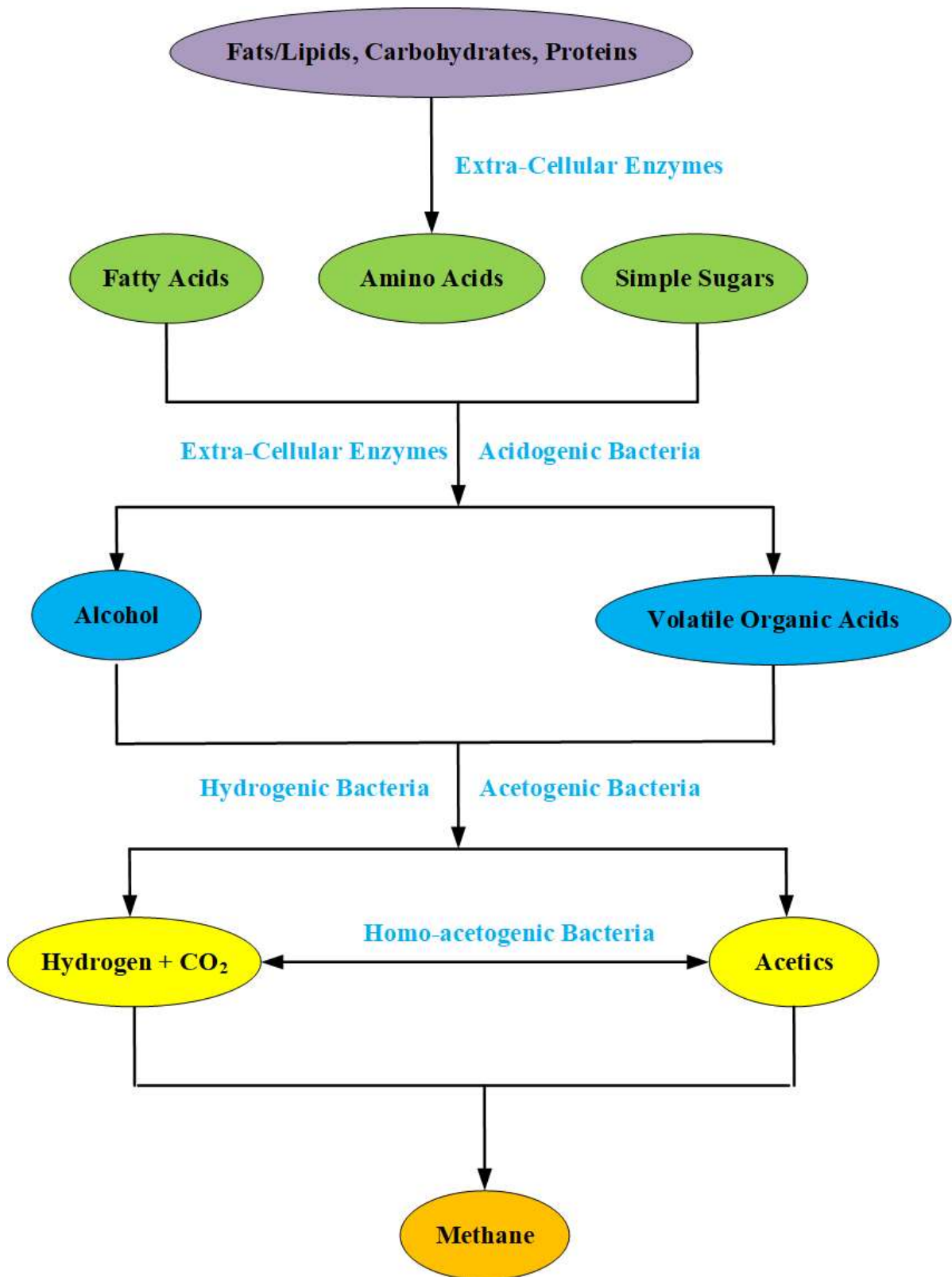


Figure 3: Anaerobic digestion process model [28, 40, 41].

321 Depending on the source, biogas can contain contaminants including sulphur compounds (H₂S,
322 sulphides, disulphides and thiols), halogenated compounds, nitrogen and organic silicon
323 species.

324 The biogas obtained from a conventional LF is a complex mixture of compounds [42]. A
325 typical composition of LF gas may contain: methane (35–65%), CO₂ (15–50%), N₂ (5–40%)
326 that seeps into the LF gas during recovery, O₂ (0–5%), H₂ (0–3%), H₂O (0–5%), CO (0–3%),
327 H₂S (0–100 ppm), NH₃ (0–5 ppm), halocarbons (20–200 ppm), volatile organic compounds,
328 VOC (0–4500 mg/m³), and siloxanes (0–50 mg/m³). A simpler biogas may be obtained from
329 the degradation of livestock manure in an oxygen-free environment, sewage sludge or agro-
330 industrial wastes which encompasses methane (53–70 %), carbon dioxide (30–47 %), N₂ (0–3
331 %), O₂ (0–1 %), H₂O (5–10 %), H₂S (0–10 ppm), NH₃ (0–100 ppm), hydrocarbons (0–200
332 mg/m³), and siloxanes (0–41 mg/m³). CO₂, H₂O and N₂ are the main contaminants of biogas.
333 In Ref. [43], the chloride amount in LF gases was reported to be 118–735 and total fluorine
334 amounted 63–256 mg/m³.

335 Table 3 represents the biogas properties obtained from digesters and LFs. For the purpose of
336 comparison, the last three rows in Table 3 provide the market natural gas composition in three
337 markets including Denmark, The Netherlands, and the North Sea. The biogas produced in
338 closed digesters shows higher methane content and considerably lower N₂ and O₂ levels than
339 an LF-derived biogas [44].

340

341

Table 3: Typical composition of biogas from AD, LF gas and natural gas [45]

Biogas	CH ₄ (%)	CO ₂ (%)	O ₂ (%)	N ₂ (%)	H ₂ (%)	H ₂ S (ppm)	Heavy hydrocarbons (%)	Ammonia (ppm)	Other (mg/Nm ³)	Physical properties				Ref.
										Density (kg/m ³)	LHV (MJ/NM ₃)	Wobbe index (MJ/Nm ³)	Methane number	
LF	35–65 (45)	15– 40 (40)	0–5 (1)	5–40 (15)	0–3	0–100 (<100)	0	5	Total chlorine: 20–200	1.3	16	18	>130	[31]
LF	30–60 (45)	15– 40 (40)	0–10 (1)	0–50 (15)	0–2 (1.5)	0–1000 (<100) mg/m ³		0–5 (5) mg/m ³	BTX: 0–500 Total chlorine: 0– 800 Total fluorine: 0– 800 (10) Siloxanes: 0–50	0.8	21	27	144	[13]
LF	47–57	37– 43	< 1	< 1– 17		36–230			Halogenated compounds: 0.3– 1.3 Organic silicon compounds: 0.7–4 Benzene: 0.6–2.3 Toluene: 1.7–5.1					[46]
AD	60–70 (65)	30– 40 (35)	0	– (0.2)	0	0–4000 (<500)	0	100	Total chlorine :0– 5	1.1	23	27	>135	[31]
AD	50–80 (65)	15– 50 (35)	0–1	0–5 (0.2)	0–2	100– 10000 (<600) mg/m ³		0–100 (100) mg/m ³	BTX: 0–20 Total chlorine:0– 100 Total fluorine: 0– 100 (0.5) Siloxanes: 0–50	0.8	22	26	135	[13]
AD	53–70 (63)	30– 47 (47)	0	0.2	0	0–1000 (<1000)	0	<100	0–5	1.2	23	27	>135	[44]

AD	55–58	37–38	< 1	< 1–2		32–169			Organic silicon compounds: < 0.4 Benzene: 0.7–1.3 Toluene: 0.2–0.7					[46]
Danish Natural gas	85–92 (90)	0.2–1.5 (0.7)	-	0.3–1 (0.3)	-	1.1–5.9 (3.1)	9	-	-	0.82	39	55	73	[31]
Dutch Natural gas	81	1		14			3.5			0.8	32	44	-	[24, 44]
North Sea natural gas	86.6 – 88.8 (88.8)	1.9 – 2.3 (2.3)	<0.01	0.9 – 1.1 (1.1)		0 – 5 (1.5) mg/m ³	8.3 – 8.5 (8.3)		BTX:0 – 1750	0.7	35	50	76	[13]

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344 **2.3. Biogas end-use market**

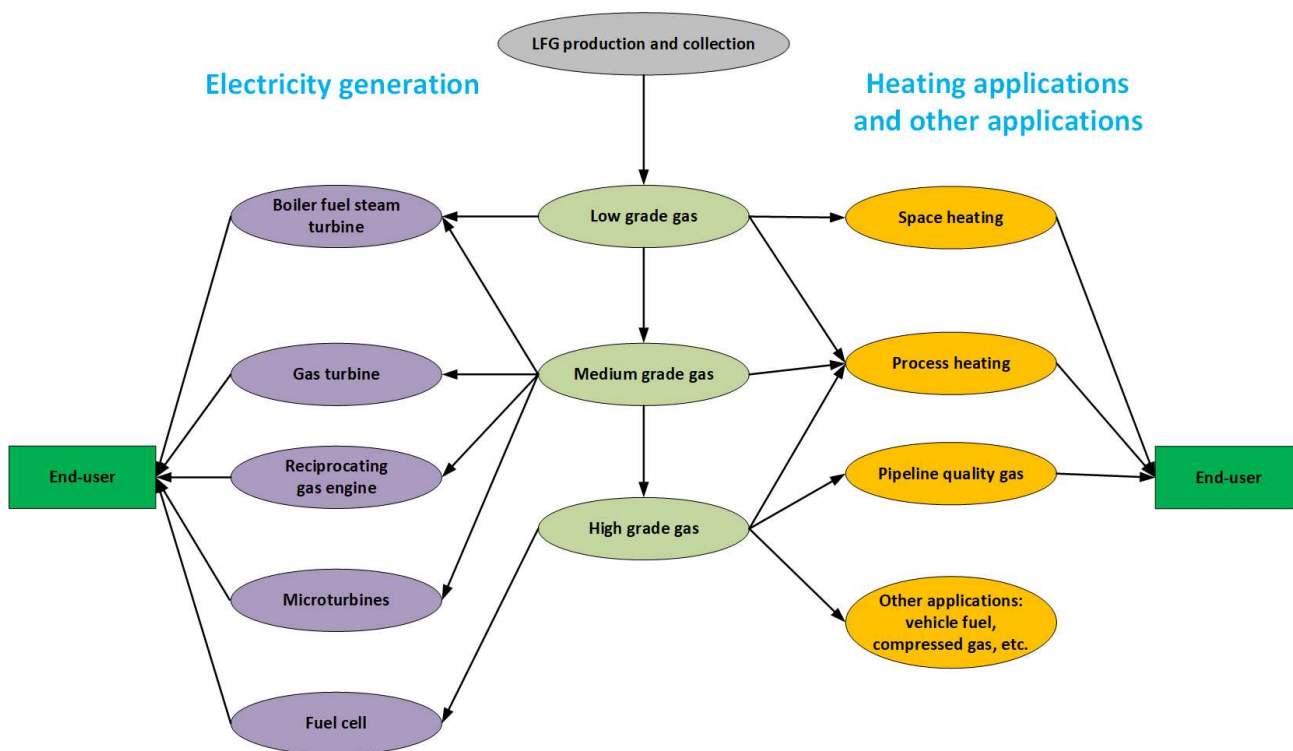
345 Methane and CO₂ are the major constituents of biogas. In this section, we discuss the various
346 uses of methane and CO₂.

347 **2.3.1. Methane**

348 The typical applications of low/medium/high quality gas obtained from cleaning/upgrading of
349 biogas are represented in Figure 4. Upgraded biogas can be utilized to produce:

- 350 • **Heat:** The high quality biogas (biomethane) can be combusted in boilers/stoves. The
351 generated heat can be used for space heating, process heating, and so on. Biogas can
352 also be used in boilers without the need to upgrade it [47]. Chemical absorption
353 upgrading technology is the most suitable way to produce high quality biomethane for
354 domestic stoves [47]. If the purity of the biogas is not of importance, other upgrading
355 technologies can be employed.
- 356 • **Heat/Power:** Otto, diesel engines, and gas turbines can be fueled by biogas. About 30–
357 40% of the biogas energy is converted to power while the remaining energy can be
358 extracted as heat. With the exception of the Otto engines, biogas can be used in dual-fuel
359 engines. In a dual-fuel engine, biogas and diesel are used to maintain the efficiency of
360 the diesel engine as high as possible. Among the continents, Europe is the main
361 producer of bioelectricity and heat from biogas [32].
- 362 • **Fuel for vehicles:** upgraded biogas can fuel light- or heavy-duty vehicles such as cars,
363 buses and trucks. The odorized and pressurized biogas (to nearly 200 bar) is used as
364 fuel in vehicles. A higher H₂ content in biogas is allowed to be used to fuel vehicles.
365 The use of biogas as a vehicle fuel is considered as the best way to reduce fossil fuel
366 consumption [31].

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370

Figure 4: LF gas utilization pathways

371

- 372 • **Injection to gas pipelines:** To inject biogas to the gas transmission networks, it must

373 meet the related standards and requirements listed in Table 4. The injection of a high-

374 quality upgraded biogas (biomethane) to gas grids is the optimal solution to distribute

375 biomethane in the countries with an extensive natural gas network [48]. Injecting

376 biomethane into national gas networks creates new markets and applications.

377 Biomethane and natural gas can be mixed liberally and are wholly interchangeable.

378 Virtual ‘green gas’ distribution networks are simply implementable to produce green

electricity and heat.

Table 4: Standard requirements of upgraded biogas for injecting to grid or fueling vehicles of several countries [13, 24, 31, 34, 49]

Market specifications	France		Germany		Sweden *		Switzerland		Austria	The Netherlands
	L Gas	H Gas	L Gas Grid	H Gas Grid	Biogas type A	Biogas type B	Lim. Injection	Unlim. Injection		
CH₄ (%)					97±1	97±2	>50	>96		>80
CO₂ (%)	<2	<2	<6	<6			<6	<6	<2	
O₂ (%)	<0.01	<0.01	<3	♡	<1	<1	<0.5	<0.5	<0.5	<0.5
H₂ (%)	<6	<6	<5	<5			<5	<5	<4	<12
CO₂+O₂+N₂ (%)					<4	<5				
Sulphur (mg/Nm³)	<100	<100	<30	<30	<23	<23	<30	<30	<5	<45
Water (%)					32 (mg/m ³)	32 (mg/m ³)				
NH₃ (mg/m³)					20	20				
Water dew point (°C)	<-5	<-5	< ground temp	<ground temp	T*-5	T*-5			<-8 (@40 bars)	-10 (@10 bars)
Wobbe index (MJ/Nm³)	42.48– 46.8	48.24– 56.52	37.8–46.8	46.1–56.5	44.7–46.4	43.9–47.3			47.7–56.5	43.46–44.41

* Biogas type A is used in 'lean-burn' engines i.e. heavy vehicles such as trucks and buses while type B is used in stoichiometric combustion engines of private cars. ** T = lowest average daily temperature on a monthly basis

382 **2.3.2. CO₂**

383 CO₂ is one of the key species of the Earth's texture and can be found in its core, crust, as well
384 as in the atmosphere. The increased CO₂ emissions are the main rationale for anthropogenic
385 climate change. CO₂ capture and storage (CCS) has received much attention over the last two
386 decades as one of the main climate change mitigation options. The total CO₂ emitted to the air
387 is nearly 32.2 giga tonnes/annum and the major CO₂ emitter sectors are power/heat production
388 (42.4%), transportation (23%), manufacturing industries and construction (19%), residential
389 (5.8%), services (2.7%), and others (7.1%) [50, 51]. The current global CO₂ utilization amounts
390 up to about 200 million tonnes/annum. Implementing the carbon tax policies make the CO₂
391 capture inevitable and the CO₂ would be available at a low or even negative price. This may
392 interrupt the current trend of CO₂ valorization and consequently increase the CO₂ utilization in
393 the current and/or new industries. CO₂ can be transported via pipelines from CO₂ sources to
394 the CO₂ demanding industries (CO₂ sinks). The concentration of CO₂ to be transported via
395 pipeline must be above 95% [47]. CO₂ utilization pathways can be divided into chemical and
396 physical. Physical CO₂ utilization routes include use of it in carbonated drinks, fire
397 extinguisher, dry ice, refrigerant, solvent, welding medium, process fluid, algae farms for
398 photosynthesis, enhanced oil/gas recovery, etc. CO₂ can be chemically utilized in the following
399 processes: synthesis gas production, methanol production, di-methyl ether (DME) production,
400 urea synthesis, di-methyl carbonate (DMC) production [52], polyurethane production [53],
401 Fischer-Tropsch gas-to-liquid (FT-GTL) products [54, 55], synthetic methane production [56],
402 chemical looping dry reforming [57], mineralization [58], and so on. A literature review of
403 physical and chemical utilization pathways can be found in Ref. [50]. The CO₂ from the biogas
404 upgrading processes can be used in the CO₂ demanding industries such as for chemicals
405 production [54, 55, 59-66]. CO₂ utilization for production of synthetic methane, methanol, and
406 Fischer-Tropsch derived liquids were addressed by Abdin et al. [59]. Assen and co-workers

407 [63] studied the direct and indirect utilization of CO₂ in the polyurethane supply chain. The
408 CO₂ utilization for polyols (direct route) is 0.30 kg CO₂ /kg polyurethane while for indirect
409 route is 1.7 kg CO₂ / kg polyurethane. Conversion of CO₂ to synthesis gas (syngas) via catalytic
410 partial oxidation of methane was considered by Chen [64]. The maximum syngas production
411 was reported at CO₂/O₂ ratio of 0.2 when the oxygen to carbon ratio is one. In addition, 10–
412 41% of CO₂ can be consumed for syngas production in the catalytic partial oxidation of
413 methane. Uner et al. reviewed photocatalytic water splitting and CO₂ reduction to produce
414 methane and methanol [65]. Conversion of CO₂ to FT-derived liquid fuels was addressed in
415 Refs. [54, 55, 60]. CO₂ was fed to the reforming section of the GTL process. CO₂ conversion
416 to methanol was investigated in Ref. [66] where the CO₂ stream came from a power-plant CO₂
417 capture process. Methanol production through CO₂ hydrogenation was studied in Refs. [67,
418 68]. The methanol production rate of the process with an inlet H₂/CO₂ ratio of 3 is nearly 59 %
419 higher than the process with an inlet H₂/CO₂ ratio of 2 [67]. The profit index of a two-stage
420 reactor system is 2.05% higher than the process with one reactor [68].

421 **2.3.3. Biogas conversion to chemicals**

422 In addition to the separation of CO₂ content of biogas to increase its methane content, there
423 exists another approach to valorize the biogas CO₂ content to commodity fuels and chemicals
424 such as methane [69, 70], methanol, hydrogen [70, 71], etc. In the biogas to methane process,
425 the biogas stream is mixed with additional hydrogen coming from a solid oxide electrolysis
426 cell (SOEC) and then is conveyed to a methanation reactor [69]. Boiling water is used in the
427 methanation reactor as the reactor-cooling medium. Operation of a full-scale methanation
428 reactor under favorable conditions for 1000 hours is possible. To obtain pipeline quality gas,
429 the optimal H₂ to CO₂ ratio to the methanation reactor must be 3.9.

430 Tamnitra and co-workers [72] simulated methanol production from biogas. In their simulation,
431 biogas is preheated, mixed with steam, and then is conveyed to a reformer. The reformer

432 outflow is sent to a separator and an absorption unit to remove water. The dried syngas is then
433 fed into a methanol reactor. Methanol production via biological conversion of biogas using
434 methane-oxidizing bacteria (methanotrophs) is investigated in Ref. [73].

435 Hydrogen can be produced via biogas dry reforming. In Ref. [71], the preparation of the
436 catalyst, the optimization of process conditions, types of reactors, and the impact of biogas
437 contaminants were reviewed. A proprietary skid-mounted, small-scale DME production unit
438 that converts biogas (with up to 50% CO₂ content) was developed by Oberon Fuels [74]. The
439 production capacity of the unit is 10,000 gallons/day. Yang et al. [70] reviewed conversion of
440 biogas to hydrogen/syngas, methanol for gasoline production, ethanol and higher alcohols. For
441 biogas conversion to hydrogen/syngas, reforming technologies such as dry reforming, steam
442 reforming, and partial oxidative reforming were discussed. Partial oxidation of methane, photo-
443 catalytic conversion, biological conversion and indirect conversion were addressed for
444 methanol production. For ethanol and higher alcohols production, direct approaches and
445 indirect routes (such as syngas fermentation and catalytic conversion) were elaborated. A novel
446 approach for simultaneous biogas upgrading and co-production of succinic acid and
447 biomethane was addressed by Gunnarsson et al. [75]. Bacterial strain *actinobacillus*
448 *succinogenes* 130Z was used to produce high-purity methane.

449

450 **3. Biogas upgrading**

451 Biogas upgrading is necessary because of two constraints: 1) product safety due to potential
452 hazards caused by contaminants (see Section 3.1 3.2), and 2) product quality enforced by
453 market specifications (see Section 3.13.2).

454 **3.1.Potential hazards of biogas contaminants**

455 Certain actions must be taken to minimize the emissions from AD biogas plants [76]. These
456 actions include the use of flares to avoid methane emissions, enhancing the thermal and
457 electrical efficiency of CHP units, and to avoid leakage to the air [77, 78].

458 CO₂ release to the air takes place during combustion of biogas, transport and storage of
459 biomass, in addition to digestate use. Poeschl et al. [79] studied the CO₂ release to the
460 environment during the production of biogas from supply system of several feedstocks, biogas
461 plant infrastructure, in addition to digestate management. Biogas production from waste
462 materials (such as pomace, cattle manure, food residues, and slaughter waste) is more
463 sustainable than utilization of energy crops. In addition, management of digestate materials
464 results in higher total emission mitigation when an MSW feedstock is used.

465 Methane emissions from biogas processes do not have any health issue and there is no evidence
466 of health issue between biological systems and methane [80, 81]. The main release points of
467 methane in the biogas supply chain are biomass storage, incomplete combustion of biogas, and
468 digestate management.

469 Nitrous oxide emissions from biogas results in a considerable impact on global warming [82].

470 Biogas production and handling is associated with some safety concerns that should be
471 considered during the biogas production process. The following hazards shall be kept to a
472 minimum level [7, 13]:

- 473 ➤ **The health of end-users and employees:** direct toxicity with unburned gas and,
474 indirect toxicity by biogas combustion, chemical products, water and air pollution.
- 475 ➤ **Operation of gas facilities and appliances, and gas grids integrity:** some hazards
476 include corrosion and clogging of the grid equipment, clogging as well as the failure of
477 end-user appliances, etc.

478 In some situations, the treatment processes may suffice to decrease the hazards to acceptable
479 levels. In other occasions, specific treatments of the raw gas may be required. Table 5 indicates
480 the potential hazards of biogas contaminants during the biogas production process, gas
481 treatment equipment, and so on. Water in combination with O₂ or CO₂ can introduce system
482 integrity problems. In addition, H₂S in combination with water and O₂ produces H₂SO₄, that
483 cause corrosion of gas storage tanks, gas pipelines, engines and compressors. The existence of
484 NH₃ and halocarbons corrodes pipelines and engines when biogas is combusted [24].

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Table 5: Potential hazards of biogas contaminants and recommended countermeasures [7, 13, 29, 83]

Hazardous Component/agent	Hazard	Countermeasure
Halocarbons	<ul style="list-style-type: none"> • Corrosive gases: affect the integrity of the system. • Production of furans and dioxins under combustion conditions: health issues. • Toxic and corrosive combustion species: influence on end-user equipment and health issues. 	<ul style="list-style-type: none"> • Sampling and analysis of halocarbons. • Removal of halocarbons. • Exclusion of materials with high halocarbon content.
High content of CO ₂	<ul style="list-style-type: none"> • Change of combustion properties, which affects the performance and safety of final-user equipment. 	<ul style="list-style-type: none"> • Adding heavier hydrocarbons. • Mixing biogas with natural gas. • Maintaining CO₂ concentration within tolerable limits.
Ammonia	<ul style="list-style-type: none"> • Corrosive gas: impact on the integrity of the gas network. • Toxic compound: health issues. • Increased NO_x emissions after combustion. 	<ul style="list-style-type: none"> • Sampling and analyzing for ammonia. • Removing ammonia from the gas.
Biological agents	<ul style="list-style-type: none"> • Bio corrosion: affects the integrity of the system. • Health hazard in case of presence of pathogenic agents 	<ul style="list-style-type: none"> • Sterilization of the substrate material. • Filtration. • Increase digester retention time.
Polyaromatic hydrocarbons (PAHs)	<ul style="list-style-type: none"> • Effect on elastomer and plastic material: system integrity issues. • Carcinogen and toxic: health issues. • Soot formation when PAHs are burnt. • Impact on safety and performance of end-user equipment 	<ul style="list-style-type: none"> • Monitoring and removal.
Siloxanes	Production of silica at combustion conditions which affects the user equipment.	<ul style="list-style-type: none"> • Sampling and analyzing. • Elimination of materials with high silicon content. • Removing siloxanes from the biomass material or the biogas.
Phosphine (PH ₃) and phosgene (COCl ₂)	<ul style="list-style-type: none"> • Toxic compounds: health issues. • Corrosive species: Affect the integrity of the system. 	<ul style="list-style-type: none"> • Sampling and analysis. • Removal of phosphine/ phosgene from the biogas. • Exclusion of sources with high phosphine/ phosgene content.

500

501 Biogas combustion releases pollutants to the air. For example, CO is the key by-product of
502 biogas incomplete combustion. Sulphur dioxide pollutants depend mainly on the efficiency of
503 desulphurization section of the biogas upgrading plant. NO_x emissions, non-methane volatile
504 organic compounds, and formaldehydes are other key pollutants of the biogas combustion
505 process. The emission factors of the mentioned pollutants are given in Refs. [84, 85].

506 The storage, management, and treatment of feedstock and digestate materials are the most
507 important steps from the global warming point of view in biogas production processes [76].
508 Most of nitrous oxide emissions can be avoided when we use closed storage for manure in
509 addition to co-digestion feeding strategy. Outspreading untreated biomass materials on the
510 ground will release large quantities of methane, ammonia, nitrous oxide, volatile hydrocarbons,
511 etc. to the air.

512 **3.2. Requirements for upgraded biogas product specification**

513 Table 6 summarizes the requirements used for biomethane utilization pathways and the
514 associated optional upgrading technologies.

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Table 6: Gas quality and technological recommendations regarding biogas utilization [34, 47]

Utilization pathway	CH ₄ content (%)	CO ₂ content (%)	Contaminants	Cleaning and upgrading technology
Domestic stoves	Heating value comparable to natural gas	-	H ₂ S < 10ppm	H ₂ S: iron hydroxide/oxide. CO ₂ : chemical absorption.
Boiler	-	-	H ₂ S < 250 ppm	H ₂ S: biological desulphurization.
Internal combustion engine	>30	-	H ₂ S: 545–1742 ppm. Halides: 60–491 ppm. Siloxanes: 9–44 ppm. Dew point: T*-6.7 °C.	H ₂ S: biological desulphurization.
Stirling engine	>35	-	H ₂ S < 2800 ppm. Siloxanes: about 0.42 ppm. Halides: 232 ppm. Dew point: T-6.7 °C.	H ₂ S: biological desulphurization.
Gas turbine/micro turbine	>35	-	H ₂ S: 10,000 ppm. Siloxanes: around 0.087 ppm (0.005 for micro-turbine). Dew point: T-6.7 °C.	H ₂ S: biological desulphurization.
Natural gas grid injection	70–98	1.0–8	H ₂ S: 2–15mg/m ³ ; N ₂ : 2–10%. H ₂ : 0.1–4%. O ₂ : 0.01–3%.	H ₂ S: impregnated activated carbon and iron hydroxide/oxide. CO ₂ : PSA + membrane. When O ₂ and N ₂ removal is required. Chemical absorption +PSA if high methane purity is needed.
Vehicle fuel	>96	<3	H ₂ S: 5mg/m ³ .	H ₂ S: impregnated activated carbon along with iron hydroxide/oxide. CO ₂ : chemical absorption /cryogenic separation.
Fuel cell	SOFC: as much as possible.	SOFC: as little as possible. MCFC: <35	H ₂ S: 1–5 ppm (MCFC) and 1ppm (SOFC). Siloxanes: few ppm.	H ₂ S: impregnated activated carbon together with

MCFC: no
specification.

iron
hydroxide/oxide.

528 * T: gas temperature, MCFC: molten-carbonate fuel cell, SOFC: solid oxide fuel cell

529 Gas grids have also certain specifications for the protection of both pipeline assets and end-
530 users. To guarantee the safety, operability and integrity of gas grids, conventional and
531 nonconventional gases should meet the least quality requirements. Each country may have its
532 own standards and requirements for the biomethane grid injection, and vehicle use (see Table
533 4). Table 7 demonstrates the parameters that are constrained in the US legislation or within the
534 EU directives for cross-border gas transmission.

535 **Table 7:** Quality parameters in the proposed harmonized EU H-gas cross-border gas transmission grids [13, 86]

Parameter	Value
Wobbe index	13.6 –15.81 KWh/m ³ (25 °C/0 °C)
Relative density	0.555–0.7
Total Sulphur	< 30 mg /m ³
(H ₂ S + COS)	< 5 mg /m ³
Mercaptans	< 6 mg /m ³
Oxygen	< 10 ppm
Carbon dioxide	< 2.5 % molar
Water dewpoint	< –8°C at 70 bara
Hydrocarbon dewpoint	< –2°C over 1 – 70 bara

536

537 Apart from methane and CO₂, biogas may encompass water, H₂S, O₂, N₂, siloxanes, NH₃, and
538 particles. The energy content of biogas is proportional to its methane content, thereby cleaning
539 and upgrading of the biogas (i.e. removing contaminants) rise the calorific value of the gas. In
540 the upgrading methods that CO₂ is separated from the raw biogas, some of the other
541 contaminants are also removed. However, mechanical wear and corrosion of the upgrading
542 equipment is avoided if the biogas is cleaned before the upgrading process for CO₂ removal.
543 The hydrogen sulphide and halogenated compounds present in biogas can cause corrosion to

544 engines. In the presence of water, sulphur compounds corrode gas storage tanks, compressors,
545 and engines. Engine manufacturers may set minimum limits on methane content (energy
546 density) of the fuel to ensure engine performance. Biogas containing organochloride contribute
547 to corrosion in combustion engines or vehicles while under certain combustion conditions, the
548 formation of furans and dioxins is also possible [43].

549 Depending on the specifications of end-users, the required composition and consequently the
550 type of upgrading technology to be applied is determined. It is also of great importance to
551 minimize methane release from the upgrading process to the air, water leakage from a water
552 scrubber, and any other stream exiting the plant. For instance, in the absorber, some of the
553 methane can be absorbed into the liquid and then be released into the air with the gas stream.
554 The absorption liquid used in the process needs to be treated with other wastewaters.

555 The techniques used to clean, and upgrade biogas are described below. These technologies are
556 available in standardized and pre-fabricated modules [87].

557 **3.3. Biogas upgrading processes**

558 The market of emerging as well as conventional technologies for biogas upgrading is currently
559 changing because of the stringent composition specifications [88]. Biogas upgrading adds
560 investment and operating costs to the overall process. As such, it is essential to choose the most
561 suitable biogas upgrading method, and to optimize the process with regard to minimum energy
562 consumption and maximum methane concentration of the upgraded gas. Physical or chemical
563 upgrading methods need high energy requirements and/or chemicals. As a result, the
564 development of alternative biogas upgrading technologies (e.g. biological methods, microbial
565 electrochemical) with lower operating costs as well as environmental impacts has been
566 triggered [44]. The upgrading costs of the established methods depend on the selected
567 technology, and most prominently on the size of the plant [24]. There are commercial plants

568 for raw gas capacities below ca. 250 Nm³/h, and plants larger than about 2000 Nm³/h are under
569 construction.

570 New technologies such as cryogenic upgrading [24], in-situ methane enrichment in AD [89,
571 90] and ecological lung [24] are also being developed. Note that the stage of development of
572 the upgrading routes might be quite different.

573 The required contaminants concentration of the raw biogas conveyed to upgrading technologies
574 are listed in Table 8 [91]. A detailed process description of the upgrading technologies can be
575 found in Refs. [92-94]. A comparison of different upgrading techniques including PSA, water
576 scrubber, and amine scrubber for different raw gas capacities can be found in Ref. [24].

577 In Europe and in 2015, water scrubbing method accounted for 34% of the upgrading market,
578 followed by chemical absorption, and pressure swing adsorption (PSA) with 25, and 20%,
579 respectively (see Figure 5) [19]. Other mature technologies such as membrane separation, and
580 physical absorption represent 13, and 8 % of the market share, respectively [19].

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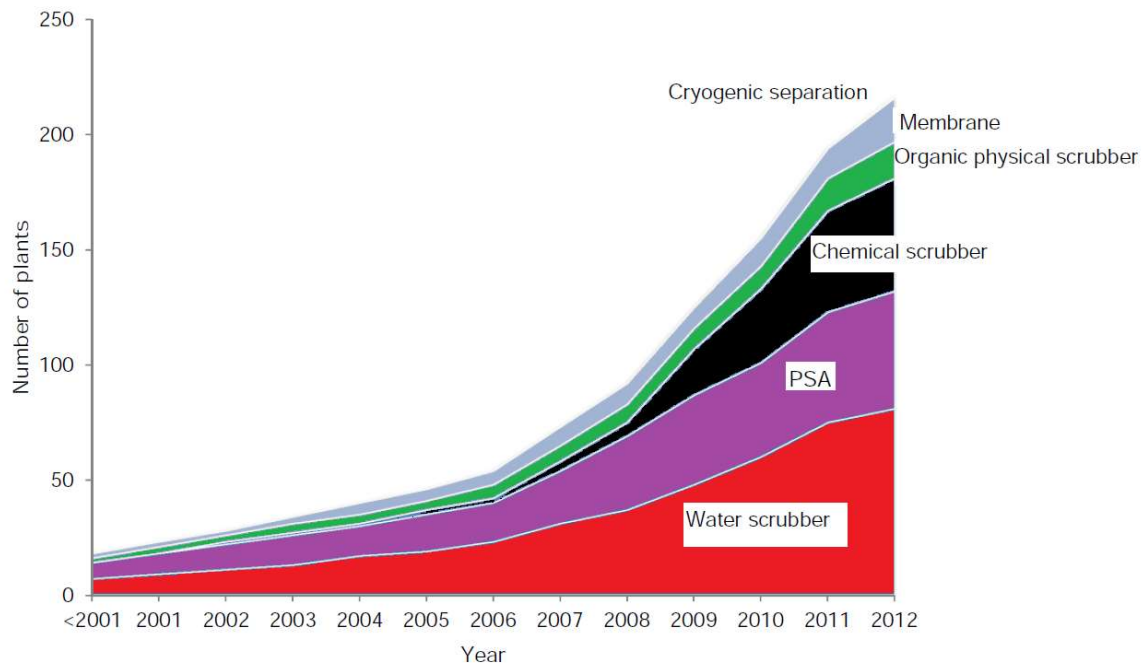
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590 **Table 8:** The required contaminants concentrations in the raw biogas in the various upgrading methods.

Upgrading technology	H₂S	VOC	O₂/N₂/H₂	NH₃
Chemical scrubbing	Moderate concentrations. Main part goes to CO ₂ stream. Polish filter may be needed in upgraded gas.	Moderate concentrations. Main part removed with the CO ₂ and condensate streams.	Go to the upgraded gas.	Moderate concentrations. Main part goes to the CO ₂ stream.
Water scrubbing	Moderate concentrations. Main part goes to the stripper air.	Moderate concentrations. Main part removed with the condensate and stripper air.	Go to the upgraded gas.	Moderate concentrations. Main part removed with process water.
PSA	Low concentrations	Removal from raw gas is required.	O ₂ /N ₂ go to CO ₂ stream, H ₂ goes to product gas	Removal from raw gas is required.
Organic physical scrubbing	Moderate concentrations. main part goes to the stripper air.	Moderate concentrations. main part goes to the stripper air.	Go to the upgraded gas.	Moderate concentrations. main part goes to the stripper air.
Membrane separation	Low concentrations. some amount goes to the product gas.	Removal from raw gas is required.	Go to the CO ₂ and upgraded gas stream.	Usually removed with condensate during drying the raw biogas.
Cryogenic separation	Moderate concentrations. Removed during first stage refrigeration.	Moderate to high concentrations. Removed during first stage refrigeration.	Go to the upgraded gas.	Moderate to high concentrations. Removed during first stage refrigeration.

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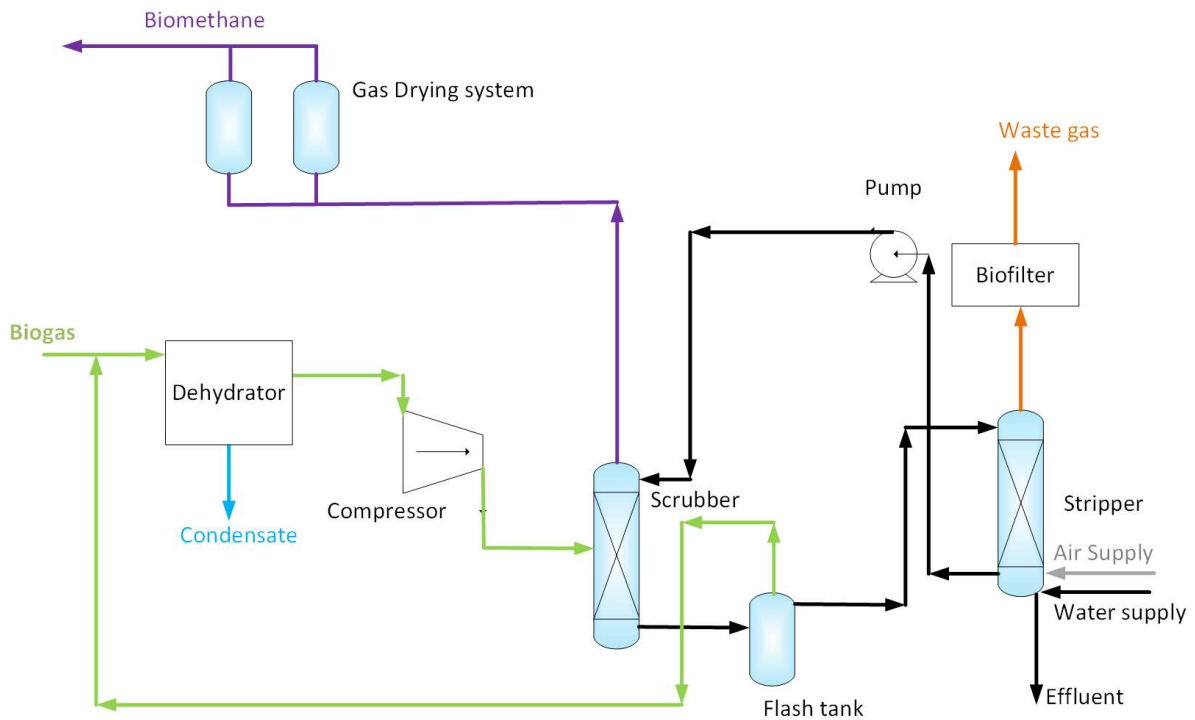
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593 **Figure 5:** Evolution of the operational biogas upgrading plants in Europe during 2001–2012.

594 3.3.1. Water scrubbing

595 CO₂ removal via water scrubbers is a traditional approach in chemical engineering. This
 596 process is based on the reality that the water solubility of CO₂ is approximately 26 times (at 25
 597 °C) higher than that of CH₄, [42]. The availability of cheap water supply is a determinative
 598 element in the selection of this technology. The CO₂ removal from the biogas produced in
 599 wastewater treatment plants can be performed in single-pass water scrubbers operating at 6–10
 600 bar. However, LF biogas can be treated in a sequential water scrubber with tap water coupled
 601 to a two-stage desorption column in order to regenerate water. The absorption column is filled
 602 with Pall or Raschig rings random packing and counter-current flow of water and gas
 603 minimizes the energy consumption and the methane loss [95]. Water flowrate depends on
 604 operational pressure. In the desorption column, CO₂ is removed from water by addition of
 605 atmospheric air. The regenerated absorbent is then returned to the absorption unit (Figure 6).

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607

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Figure 6: Schematic of biogas upgrading using water scrubber

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A typical flow of about 0.1–0.2 m³ water per Nm³ of raw biogas is reported for single-pass water scrubbers. The water consumption of process designs with water recycling lies within the range of 0.18–0.23 m³/Nm³ of biogas. Higher operating pressures result in lower water flowrates but increases compression, and pumping costs. Off-gas treatment processes such as biofilters, incinerators, or activated carbon filters used to minimize H₂S, and CH₄ emissions from the desorption column entail additional costs to the process.

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3.3.2. Organic solvent scrubbing

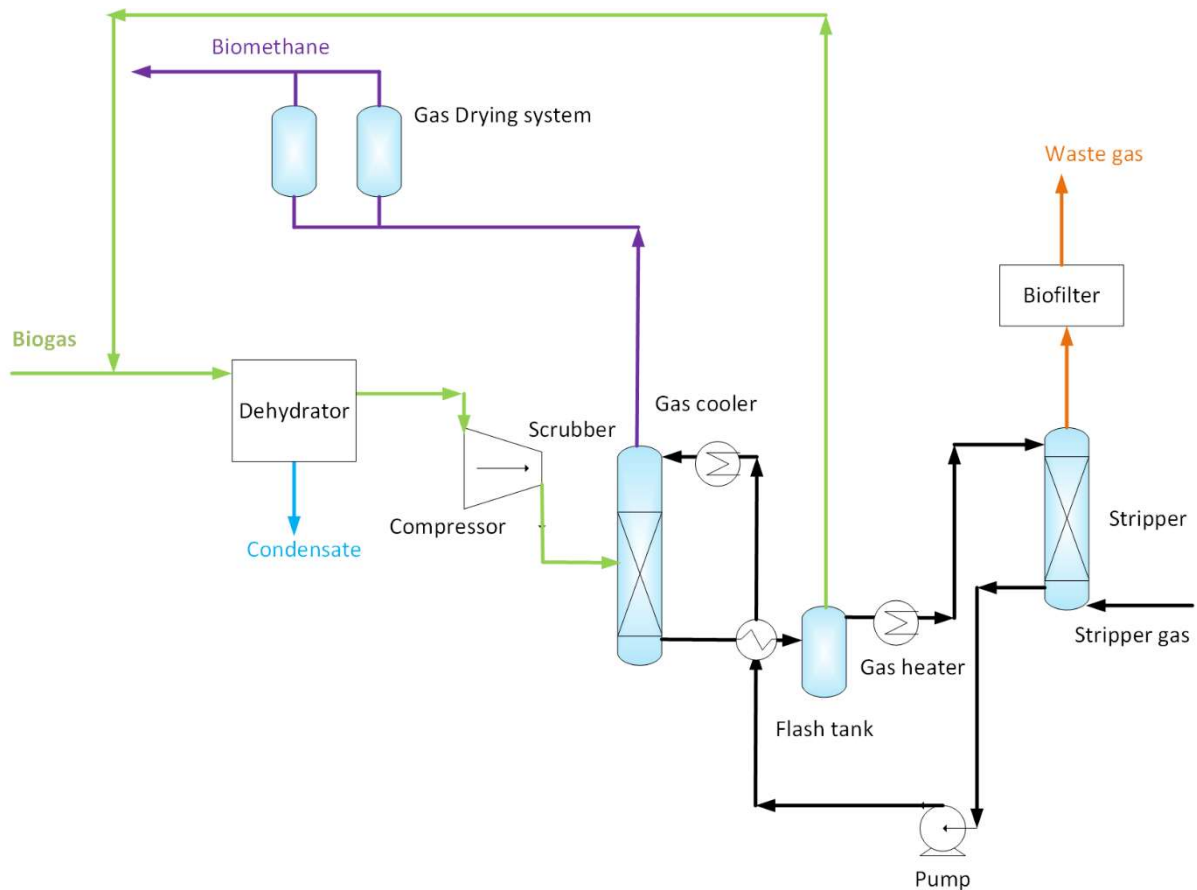
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This upgrading technology is very similar to water scrubbers, except that the use of solvent increases the CO₂ solubility and thus capture. The organic solvent-based scrubbing needs a gas pre-treatment unit (to separate water) and several inter-cooling/-heating stages to guarantee an efficient operation i.e. the solvent is cooled before absorption and heated before desorption (see Figure 7). In addition, CO₂ solubility in Selexol and Genosorb solvents are 5 and 17 times

620

621 higher than in water. Thereby, the diameters of the scrubbers are smaller than those of water
622 scrubbers, since a lower flowrate of the organic solvent is needed.

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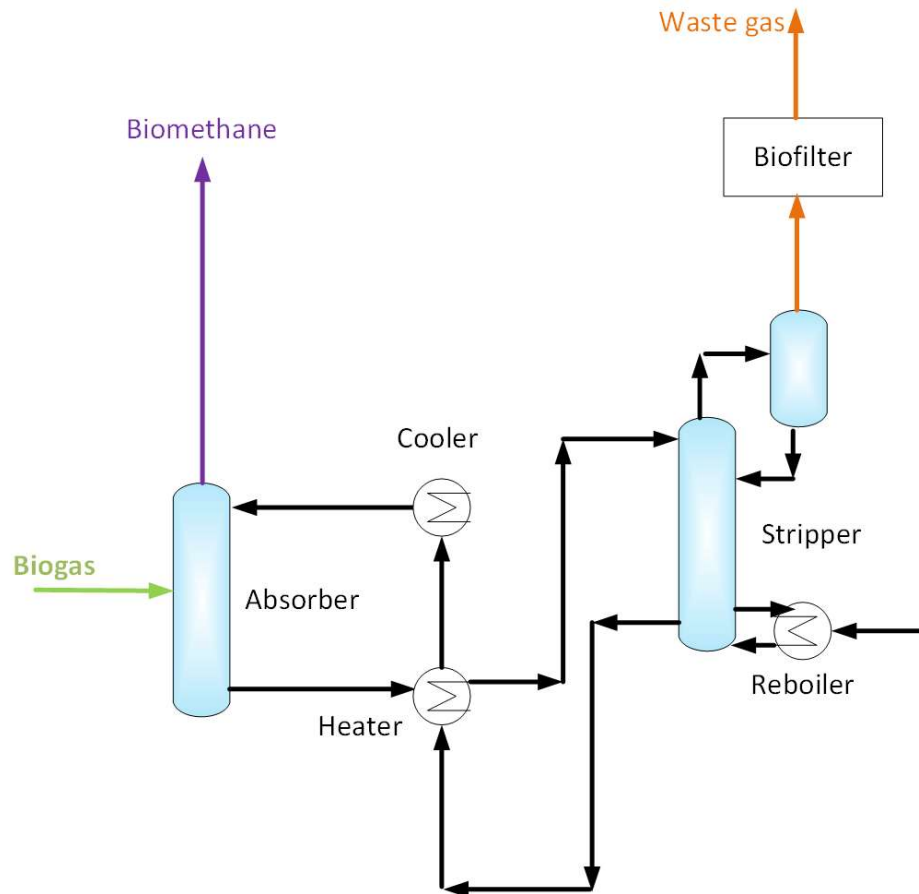
625 **Figure 7:** Schematic of biogas upgrading using organic solvent scrubber

626 3.3.3. Chemical scrubbing

627 This upgrading process is fundamentally similar to water or organic solvent scrubbing (Figure
628 8). The process configuration consists of a packed bed (filled with structured or random
629 packings) absorption unit (in which the CO₂ is separated from the biogas) plus a stripper
630 (regenerator) equipped with a reboiler. This process is simpler than water/organic scrubbing
631 due to the enhanced process performance of using CO₂-reactive absorbents such as alkali
632 aqueous solutions or alkanol amines. The CO₂ (and H₂S) of the raw biogas react with the amine.

633 The absorber operates at 1–2 bar while the operating pressure of the stripper is usually 1.5–3
634 bar. Steam can be used in the reboiler of the stripper.

635



636

637 **Figure 8:** Schematic of biogas upgrading using chemical scrubber

638 3.3.4. Pressure swing adsorption (PSA)

639 This technique is a dry method which is based on the selective adsorption of CO₂ over CH₄
640 onto zeolite and polymeric sorbents, silica-gel, activated alumina, or activated carbon [96].

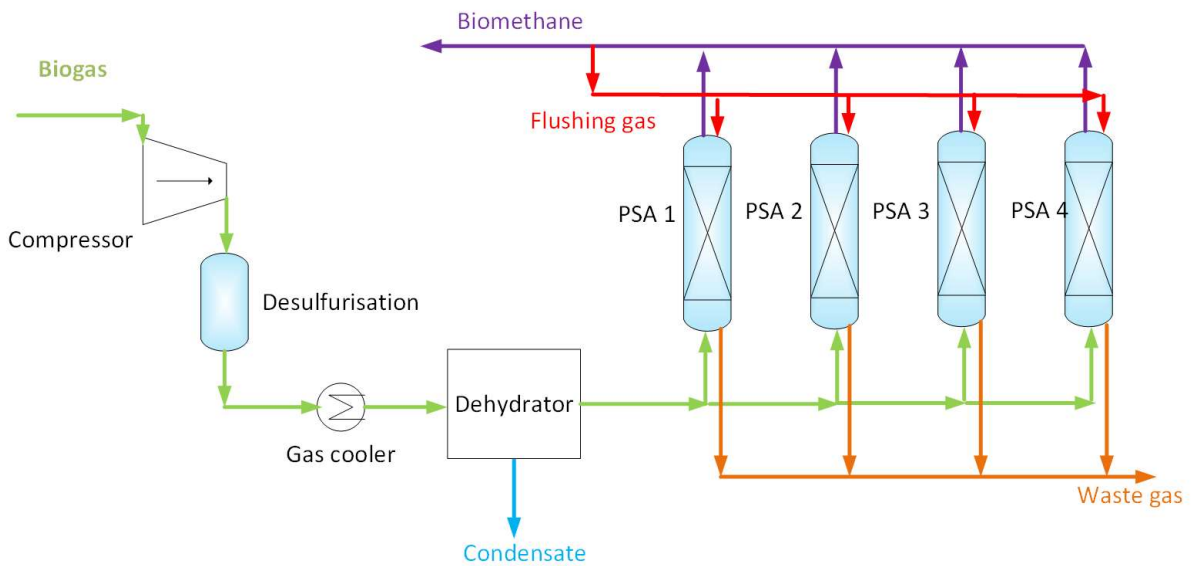
641 Packed bed columns operating in parallel under pressurization, feed, blowdown and purge

642 regime. The compressed raw biogas enters the adsorption column where CO₂ is retained and

643 methane flows through the bed. When the adsorbent is saturated with CO₂, the feeding is

644 stopped, the blowdown phase is initiated by decreasing the pressure to release CO₂ which can

645 be directed into an off-gas stream. For continuous operation of the process, four packed bed
646 columns are being closed and opened consecutively (Figure 9).



647

648

Figure 9: Biogas upgrading using PSA

649 **3.3.5. Membrane separation**

650 This separation technique is based on the selective permeation of gas components flowing
651 through a semi-permeable membrane. Polymeric materials are preferred membranes over non-
652 polymeric materials for upgrading the biogas due to the lower cost, stability at high pressures,
653 easy manufacture, and easy scalability. Most of the methane is retained while most of the CO₂
654 permeates through the membrane. The biogas is cleaned (to remove particles, H₂O, H₂S, VOCs,
655 NH₃, and siloxanes [97, 98]) to avoid deterioration and clogging of the membrane stages,
656 compressed to 6–20 bar, and then is fed to membrane modules. The process may have several
657 membrane stages.

658 **3.3.6. Cryogenic separation**

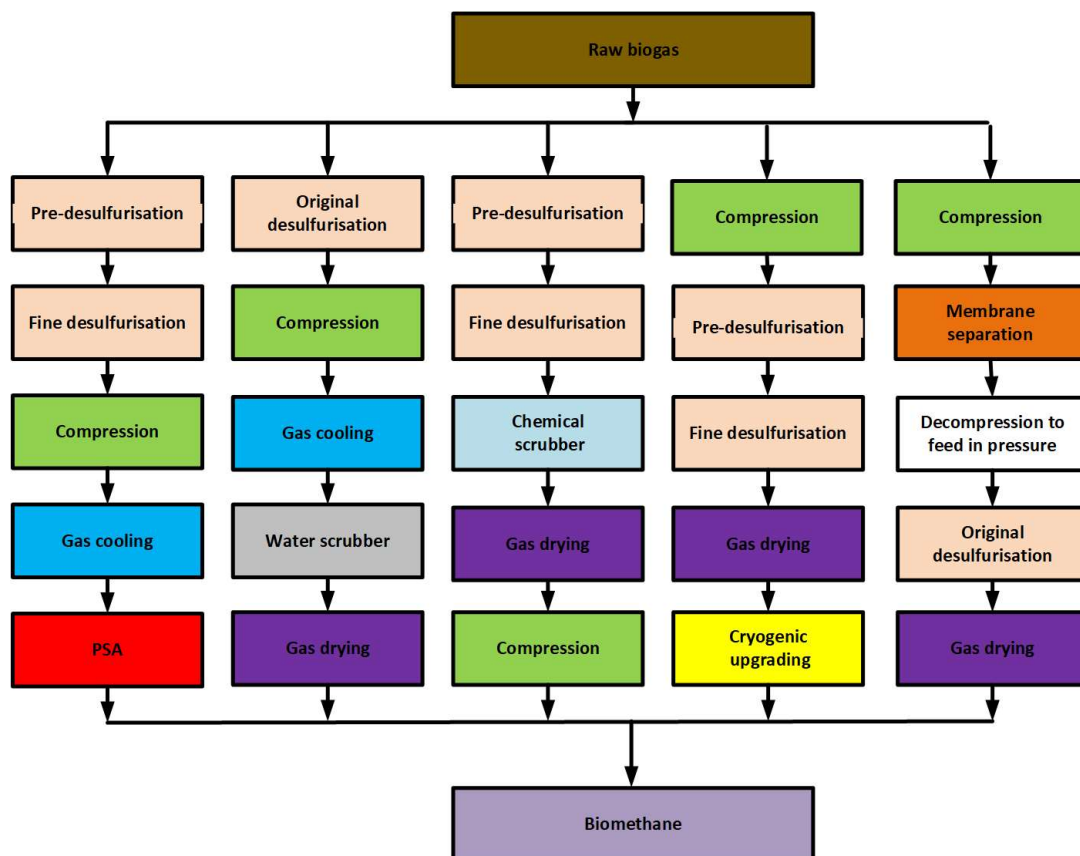
659 In this method, contaminants like H₂S and CO₂ are liquefied and separated in three successive
660 stages to remove them and also to optimize the energy recovery [99]. The temperature of the

661 compressed biogas to the first separator is -45 °C, to the second one is -70 °C and to the third
 662 separator is -120 °C.

663 **3.3.7. Process configuration**

664 No single technology can remove all contaminants in biogas. Generally, each upgrading
 665 technology is capable of removing one or two contaminants. Ultimately, based on the
 666 biomethane requirement aspects, a combination of processes is used to build the biogas
 667 upgrading plant. The combination of the techniques used for cleaning and upgrading biogas is
 668 illustrated in Figure 10.

669



670

671 **Figure 10:** The combinations of biogas cleaning and upgrading processes

672 The physical/chemical processes used to remove biogas contaminants are described in Section
 673 3.4.

674 **3.4. Physical/Chemical processes for contaminants removal**

675 **3.4.1. Water removal**

676 The water content of biogas may condensate or form hydrates in transmission lines and make
677 corrosion and erosion issues [24]. Water can be separated by cooling, compression, adsorption
678 or absorption. By decreasing the temperature or increasing the pressure, water condenses and
679 is then removed, [45]. Water removal by adsorption can be achieved by SiO₂, molecular sieves
680 or activated carbon. The regeneration process can be done by heating or a decrease in pressure.
681 Absorption can be performed in glycol solutions (regenerated by heating) such as ethylene
682 glycol (EG), di-ethylene glycol (DEG), tri-ethylene glycol (TEG) or by the use of hygroscopic
683 salts.

684 **3.4.2. H₂S removal technologies**

685 The H₂S present in the raw biogas can be removed by adsorption onto several materials
686 including activated carbon iron oxide or hydroxide[100], membrane separation, calcium
687 hydroxide Ca(OH)₂ [101], ethylenediaminetetraacetic acid Fe-EDTA [102], in-situ
688 precipitation in the digester via iron salt addition, and absorption [103]. The most
689 straightforward method of controlling H₂S concentration in the biogas is the in-situ
690 desulphurization occurring in the biogas digester. It can be done by dosing air/pure oxygen to
691 the digester gas stream or iron hydroxides/chlorides to the digester liquid phase [104, 105]. Use
692 of sodium and calcium hydroxide in the process causes the formation of salts with elemental
693 sulphur which cannot be regenerated.

694 The H₂S content of biogas can be eliminated by adsorption/oxidation to elemental sulphur or
695 conversion to SO₂ [33]. Mezmur and Bogale used KOH/ NaOH and activated carbon to reduce
696 CO₂- and H₂S- content of biogas by 82 and 99%, respectively [106]. In addition, charcoal and

697 silica gel were employed to remove the moisture. The cleaned biogas was used for power
 698 generation.

699 Membrane-based separation can also be used for the selective H₂S removal from biogas [107].
 700 Polyimide, poly-sulphone and cellulose acetate are commonly used as membrane materials for
 701 the biogas upgrading purposes [108]. Significant improvement in the desulphurization
 702 efficiency can be achieved if special rubbery polymeric membrane material is used [109].

703 Table 9 represents the pros and cons of the physical/chemical H₂S removal routes.

704 **Table 9:** Pros and cons of physical/chemical H₂S removal technologies

H ₂ S removal technology	Advantages	Disadvantages
Adsorption using iron oxide or hydroxide	<ul style="list-style-type: none"> • Low operating cost. • High H₂S removal efficiency, i.e. >99%. 	<ul style="list-style-type: none"> • Temperature should be controlled. • Regeneration of adsorbent is expensive. • H₂S content of raw gas must be <100 ppm.
Adsorption on activated Carbon	<ul style="list-style-type: none"> • High H₂S removal Efficiency. 	<ul style="list-style-type: none"> • Activated carbon has a short lifetime. • Carbon regeneration is performed at a high temperature.
Absorption (sodium hydroxide washing)	<ul style="list-style-type: none"> • Operation under ambient pressure and temperature. • Water can be used as solvent. • High H₂S removal efficiency of 90–100%. 	<ul style="list-style-type: none"> • Low liquid to biogas ratio is needed.
Membrane separation	<ul style="list-style-type: none"> • High H₂S removal efficiency. • Simultaneous CO₂ removal from the raw gas. 	<ul style="list-style-type: none"> • Concentration of H₂S in the raw gas must be <2%.
In-situ precipitation	<ul style="list-style-type: none"> • Efficient at high H₂S content of raw gas. • Low investment cost. 	<ul style="list-style-type: none"> • High operating costs. • Not efficient at low H₂S concentrations. • Concentration of H₂S in the raw gas must be >100–150 ppm.

705

706 H₂S and CO₂ can also be removed from the raw biogas using cryogenic separation [110]. For
 707 large-scale biogas desulfurization, there exist three patented H₂S removal processes i.e.
 708 Biopuric[®], Thiopaq[®], and H₂SPLUS SYSTEM[®], [111-113]. In these processes, the
 709 combination of bioreactors and chemical scrubbers are used.

710 **3.4.3. Halocarbons, volatile organic compounds (VOC), siloxanes removal**
711 **technologies**

712 Halogenated compounds contain one or several halogen atoms including chlorine, fluorine,
713 iodine or bromine [114]. Examples of halogenated compounds are carbon chlorobenzene,
714 tetrachloride, tri-fluoromethane and chloroform. Halocarbons are often seen in the LF biogas
715 but rarely found in the AD biogas or from sewage sludge, and organic wastes. Halocarbons are
716 oxidized to corrosive products when biogas is combusted. The corrosive combustion products
717 together with water can cause corrosion issues in downstream pipelines and appliances. In
718 addition, furans (polychlorinated dibenzofurans and polychlorinated dibenzo-p-dioxins), and
719 dioxins can form if the time and temperature of combustion are sufficient. Siloxanes are
720 chemical species that contain a silicon-oxygen bond (Si-O) and are used in shampoos and
721 deodorants. So, siloxanes can be found in raw biogas originated from both sewage sludge
722 treatment plants and LFs. Siloxanes, VOCs and halocarbons can be separated from the biogas
723 by adsorption on activated carbon [42, 105]. Siloxanes can also be removed by cooling the
724 biogas, silica gel/activated aluminum, or absorption in a mixture of liquid hydrocarbons [24].
725 Siloxanes can also be separated during the H₂S removal process. The adsorption on activated
726 carbon must be run under high pressure with low moisture contents of biogas. There are some
727 technical problems with the regeneration of activated carbon materials. The reported siloxanes
728 removal efficiency via adsorption on activated carbon are 95% [105] and 74–83% [115]. In
729 addition, using cryogenic condensation of siloxanes, a removal efficiency of 25.9 and 99.3%
730 can be obtained when the biomethane temperature is dropped to -25 and -70 °C, respectively
731 [42]. The cryogenic condensation route needs high investment and operating costs [113].

732 Use of ionic liquids in removing VOC compounds from raw biogas was studied by Privalova
733 et al. [116]. Experimental results suggest that 1-butyl-3-methylimidazolium acetate [BMIM]

734 [AC] ionic liquid has a better performance compared to aqueous amines solutions as it captures
 735 65 wt% of the identified VOC compounds, whereas amine solutions retain nearby 35 wt% only.

736

737 **3.4.4. N₂ and O₂ removal technologies**

738 N₂ and O₂ are present at high molar concentrations from LFs when vacuum generation is used
 739 to collect the raw biogas as a consequence of air infiltration. The technologies used for both N₂
 740 and O₂ removal are adsorption with activated carbon, molecular sieves, pressure swing
 741 adsorption, membrane and cryogenic separation. A fraction of N₂ and O₂ can be separated
 742 during the desulphurization processes or via some of the upgrading routes. Removal of both
 743 N₂ and O₂ is cumbersome and expensive. The concentration of these species in the upgraded
 744 biogas should be too low unless the biogas is used for boilers or CHP units. The corresponding
 745 pros and cons of the upgrading process routes used to remove N₂ and O₂ are listed in Table 10.

746 **Table 10:** Pros and cons of N₂ and O₂ removal technologies

N ₂ and O ₂ removal technology	Advantages	Disadvantages
Membrane separation	<ul style="list-style-type: none"> • Low energy requirements. • Compact design and light in weight. • Easy operation. 	<ul style="list-style-type: none"> • Complex maintenance. • High purchasing cost of membrane. • Low CH₄ separation efficiency and high CH₄ loss.
PSA	<ul style="list-style-type: none"> • Low energy demand. • CO₂, N₂ and O₂ can be separated simultaneously. 	<ul style="list-style-type: none"> • Water and H₂S have to be removed before the pressure swing adsorption process. • Periodical regeneration of the adsorbent is required.
Cryogenic separation	<ul style="list-style-type: none"> • Simultaneous removal of multiple contaminants. • CO₂ is produced as a by-product. 	<ul style="list-style-type: none"> • High capital cost and energy demand.

747

748 **3.4.5. CO₂ removal technologies**

749 The CO₂ content of biogas is about 25–50% on a volume basis and its removal from biogas is
 750 essential to enhance the biogas energy density and heating value, and to mitigate the costs of

751 pipeline/road/rail transportation. Physical/chemical methods used to remove CO₂ from the raw
752 biogas are based on the physical or chemical transfer of the CO₂ to another gas, liquid- or solid-
753 state material. The technologies used for CO₂ removal are PSA, water scrubbing, organic
754 physical/chemical scrubbing, adsorption of gas molecules on adsorbent materials, cryogenic
755 separation, thermo-catalytic methanation, and membranes separation. These technologies are
756 mature and have higher efficiency compared to the biological routes used for CO₂ removal
757 [110]. CH₄ recovery is an important factor from the economic and ecological points of view
758 [117].

759 The water-scrubbing route is the most popular biogas upgrading process. Commercial water
760 scrubbers from several suppliers can be found in the market for a wide range of gas capacities
761 [24]. In the pressurized water scrubbing process, the raw biogas is compressed and contacted
762 with water counter-currently in a vertical column. In this method, CO₂, some quantities of
763 methane and H₂S of the raw biogas are transferred to the water stream. The regeneration step
764 involves a flash depressurization, to recover methane [70], and a desorption column operating
765 under atmospheric pressure. The water scrubbing process is simple and robust but the existence
766 of nitrogen and oxygen in the desorption column is a drawback. In addition, bacterial growth
767 takes place and cause clogging. Foaming in the scrubbers is another potential operational
768 problem.

769 Organic physical scrubbing process is extremely similar to water scrubbing technique unless
770 an organic solvent such as polyethylene glycol (with higher solubility for CO₂/H₂S in addition
771 to lower vapor pressure) is used as an alternative to water [118]. Commercial processes using
772 polyethylene glycol (as solvent) are Purisol™, Selexol™, and Genosorb™ [100]. Use of solvents
773 reduce the size of absorption vessels, lessen pumping, and mitigate the solvent loss.

774 In chemical scrubbing technique, on the other hand, aqueous solutions of several alkanol
775 amines such as mono-ethanolamine (MEA), diethanolamine (DEA), di-glycolamine (DGA),
776 and di-methyl ethanolamine (DMEA) interact with CO₂ that increase the selectivity between
777 CO₂ and methane resulting in a very low absorption of CH₄ s[105]. Most of amine scrubbers
778 operate near atmospheric pressure. The absorption step and chemical reactions are exothermic
779 which heat up the solution by around 25 °C. Note that absorption equilibrium favored by low
780 operating temperatures while chemical reactions favored by high operating temperatures. So,
781 optimal operating temperatures must be found [100]. The solvent leaving the absorber is heated
782 up for regeneration purposes and then is conveyed to the stripper column operating at high
783 temperature. A pre-desulphurization unit is needed to avoid poisoning the alkanol amine
784 solvent. Thermal or oxidative degradation of the solution increases the equipment corrosion
785 potential, chemical consumption, and release of hazardous degradation species [119-121].

786 Several adsorbent materials such as activated carbon materials, zeolites, titano-silicates, silica
787 gels, etc. can separate CO₂ from methane [122]. Water removal is necessary before the process
788 to prevent the poisoning of the adsorbents. The biogas feed must be pressurized to about 10 bar
789 to have enough driving force in the process. Regeneration process can be carried out in a
790 cycling operation [70]. Two to nine cycling adsorbents are used in parallel arrangement to assure
791 a continuous operation of the process [24]. O₂ and N₂ can be separated simultaneously with
792 CO₂ [47].

793 Cryogenic separation can be used to remove contaminants from raw biogas. In this approach,
794 ammonia, H₂S are separated and then CO₂ is removed in liquid form. Further cooling causes
795 CO₂ sublimation and its removal in solid form. This technology is very energy intensive [100].

796 In the thermo-catalytic methanation process, the CO₂ and H₂ content of biogas can be converted
797 to methane on nickel catalyst under 10–20 bar [123, 124]. The mole fraction of methane in the

798 biomethane stream is 96%. Kirchbacher et al. simulated four process configurations to
799 investigate the impact of fermentation setup, recycling of off-gas and multi-stage membrane
800 separation, pressure and gas hourly space velocity [125]. The process configurations are
801 process with no recycle, process with basic recycle, process with recycle + preceding methane
802 removal, in addition to process with recycle + two-stage membrane separation. The results
803 show that a two-stage biogas fermentation is extremely beneficial, as it increases the hydrogen
804 storage capacity by about 70 %. Upgrading of raw biogas obtained from the organic matter of
805 MSW by solar/wind-derived hydrogen was analyzed in Ref. [126]. Two case studies were
806 conducted for UK (high wind availability) and Spain (high solar availability). The results
807 suggest that the cost of biogas upgrading in the UK is lower than in Spain with the current
808 prices.

809 Membrane separation route is one of the conventional methods used to upgrade the LF gas.
810 The first plants were constructed in the late 1970s. This technology is based on dissimilar
811 solubility and diffusivity of gaseous species in each membrane. The pre-treated raw biogas is
812 pressurized to 5–30 bar and is conveyed to the membrane module [100]. Membrane-based
813 separation is simple to operate, safe and robust, the scale-up flexibility is high, and there is no
814 need of hazardous chemicals [70]. High CO₂ removal efficiency (>95%), as well as low CH₄
815 loss, are achieved when physicochemical technologies are used for biogas upgrading. Other
816 chemical species can also be captured via physicochemical methods from the raw biogas with
817 CO₂. However, it is recommended to install a biogas pre-treatment process for the partial
818 removal of those compounds prior to CO₂ separation to avoid operational issues.

819 The highest methane recoveries can be achieved with chemical absorption compared to all
820 other upgrading approaches (Table 11). Employing complex designs with membrane
821 separation can yield methane recovery values of up to 99–99.5% at a higher investment cost

822 [44]. It is important to note that lower methane loss can be obtained at the expense of higher
823 energy demand.

824 The fixed capital investment costs of physicochemical CO₂ removal processes depend on the
825 concentration of contaminants and plant size (i.e. the economy of scale). Thereby, the higher
826 the biogas upgrading capacity, the lower the investment cost. The capital costs of membrane
827 separation process rapidly increase as the process is scaled down. The main operating cost of
828 the separation technologies are the cost of electricity needed for mechanical equipment e.g.
829 biogas compression and pre-treatment, liquid pumping, cost of energy for solvent regeneration
830 at high temperatures in case of chemical absorption, etc. The maintenance cost of a membrane-
831 based upgrading route is about 3–4% of the fixed investment costs, while this value is nearly 2
832 to 3% for absorption and adsorption processes [42].

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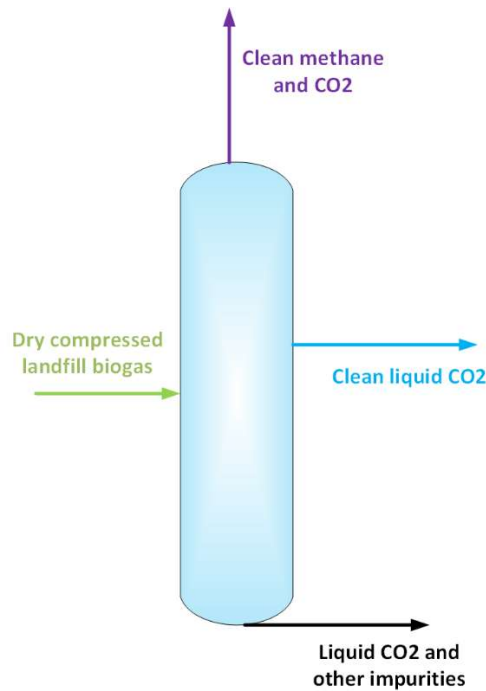
843 **Table 11:** Investment costs, plant capacity, energy use, methane loss and final methane content of the upgraded
 844 biogas physical/chemical technologies [42, 44, 47, 127]

CO ₂ removal technology	Working pressure (bar) [24]	Operating temperature (°C) [24]	Methane loss (%)	Methane concentration of the upgraded biogas (%)	Power demand consumption (kWh/Nm ³)*	Heat demand (kWh _{th} /m ³ raw biogas) [87]	Plant capacity (Nm ³ /h), [investment cost (€ h /Nm ³)]** [88]
Water scrubbing	4–7	No	<2	>96	0.2–0.3	0	100–500–1000 [5500–2500–2000]
Chemical absorption	No pressure	160	0.1–1.2	>99	0.12–0.15	0.6	600–1800 [3200–1500]
PSA***	4–7	No	-	96–98	0.25–0.6	0	600–2000 [2700–1500]
Membrane separation	8	10–30 [128]	-	96–98	0.2–0.38	0	100–400–1000 [6000–2500–2000]
Physical separation	4–7	55–80	<2	96–98.5	0.2–0.51	0	250–1000–1500 [4500–2000–1500]

* Electricity consumption can be stated as the amount of energy per volume of cleaned gas, raw gas, ton of CO₂, etc. [47].
 ** Capital costs, operating and maintenance costs (O&M) of upgrading technologies are available in Ref. [47].
 *** H₂S and water needs to be separated before the PSA-column [24].

845

846 Acrion Technologies Inc. has built a process named CO₂ Wash[®] for the upgrading of the LF
 847 gas (Figure 11) [129]. This process removes sulphur and halogenated compounds, siloxanes,
 848 and non-methane hydrocarbons from the LF biogas. The removed contaminants dissolved in
 849 the liquid CO₂ of column bottom stream can be sent to incinerators together with the LF gas.
 850 The purified liquid CO₂ with purity of 99.99% and a gas stream comprising mainly methane
 851 and CO₂ are the other streams from the CO₂ Wash[®] process. The siloxanes, chlorinated
 852 compounds as well as sulphur contents of the clean methane and CO₂ stream (top stream of the
 853 tower) are all below the detection levels of 5, 10 and 100 part per billion (ppb), respectively.



854

855

Figure 11: Schematic view of the CO₂ Wash[®] (Acrlon)

856

857 **3.4.6. Other compounds**

858 Solid particulates, as well as oil-like compounds present in the raw biogas, are separated via

859 dust collectors [114]. Sludge and foam are separated via cyclones. A filter with 2 to 5-micron

860 mesh size is appropriate for most downstream applications of the biogas. Ammonia (NH₃) is

861 removed when the biogas is dried or upgraded and a separate cleaning unit is not required. The

862 methane loss of the upgrading process (i.e. the methane present in the off-gas stream) can be

863 avoided by mixing the off-gas with air to be used for combustion [83]. Methane can be seen in

864 the off-gas of PSA columns, in air/water from water scrubbers with/without water recirculation.

865 Separating methane from the off-gas is done to make the upgrading plant economically viable

866 and to avoid/minimize the methane slip to the environment due to its strong GHG emission.

867 **3.5. Biological biogas upgrading technologies**

868 There are two biological approaches for biogas upgrading: 1) chemoautotrophic, and 2)
869 photosynthetic. Most of these methods are at pilot stage or early stage of full-scale
870 implementation [23].

871 **3.5.1. Chemoautotrophic approaches for CO₂ conversion**

872 The CO₂ content of biogas can be converted to methane by methanation process. Bioconversion
873 of CO₂ to methane is a cutting-edge solution for upgrading a raw biogas [130]. This approach
874 not only reduces the CO₂ concentration but also increases the energy content of the effluent
875 stream by increasing the methane concentration. It is possible to upgrade a raw biogas with
876 CO₂ molar concentration of 60% to biomethane with methane mole fraction of 90% under
877 thermophilic operation at 65 °C [131] and 98% under mesophilic condition (at 37 °C) [132].
878 In the hydrogenotrophic CO₂ removal route, the CO₂ part of raw biogas is microbiologically
879 upgraded by reduction of CH₄ with H₂ [133]:



881 Microbiological upgrading of CO₂ in biogas to biomethane is performed by each of the
882 following three technologies:

- 883 • In-situ (biogas upgrading) by adding H₂ from an external source into anaerobic digester
884 enabling methanogenic Archaea to transform CO₂ (40–60% molar concentration) to
885 methane [134]. In this method, CO₂ can be directly captured from the reactor, it has
886 simple operation, and is an alternative for storing excess renewable electricity [130].
887 Depending on the optimal temperature, there exist four methanogens, 1)
888 psychrotolerant (about 18 °C), 2) mesophilic (about 37 °C), 3) thermophilic (about 55
889 °C), and 4) hyperthermophilic (about 65 °C) [130]. The main challenges of the in-situ

890 biological upgrading method are the methanogenesis inhibition at pH values above 8.5,
891 and oxidation of alcohols in addition to volatile fatty acids [23].

892 • Ex-situ methanation refers to the provision of CO₂ and H₂ in an anaerobic reactor
893 containing hydrogenotrophic to produce methane [23, 135]. The advantages of this
894 approach over in-situ technique are; 1) stability of the process since upgrading
895 operation is taking place in a separate section (unit), 2) simpler biochemical process, 3)
896 it is not dependent on biomass, 4) syngas can be used instead of pure CO₂, and 5)
897 feasibility of supplying power to remote areas. The efficiency of the biogas upgrading
898 process is highly dependent on the reactor type [23]. In addition, the operating
899 temperature is an important factor for the efficiency of the bio-methanation process.
900 The main challenge of this technology is the low mass transfer rate of the present gas
901 and liquid phases.

902 • Hybrid of in-situ and ex-situ processes: in this method, a fraction of partially upgraded
903 biogas from the in-situ upgrading process is conveyed to an upgrading unit (ex-situ).
904 This technique is less developed compared to the in-situ and ex-situ approaches.

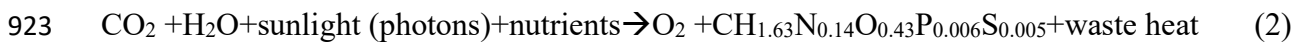
905 CO₂ removal through microbial electrochemical is an innovative method to upgrade biogas
906 [136, 137]. In microbial electrolysis cells, electrons are released by bacteria and can be
907 combined with protons to produce hydrogen in the cathode [138]. The produced hydrogen can
908 upgrade biogas [139]. The in-situ upgrading approach is more effective than the ex-situ system
909 [136]. The performance of the in-situ method can be further enhanced under continuous mode
910 of operation.

911

912 **3.5.2. Photoautotrophic approaches for H₂S and CO₂ removal**

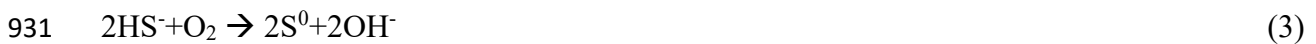
913 The photosynthetic CO₂ removal route is an alternative technique that sequester CO₂ to
914 increase the methane concentration. H₂S removal can be achieved by these methods, while over
915 54% of the raw biogas CO₂ content is consumed [23]. Phototrophic organisms such as algae in
916 enclosed or open photobioreactors can catalyze the process. The advantage of closed systems
917 over open systems are high photosynthetic performance and low land/water needs. The main
918 bottlenecks of closed systems are high energy requirement and cost. An open system needs low
919 resources for its construction and operation.

920 Photoautotrophic microorganisms such as prokaryotic cyanobacteria and eukaryotic
921 microalgae can convert CO₂ utilizing water, sunlight, and nutrients to produce oxygen, heat
922 and biomass [42]:

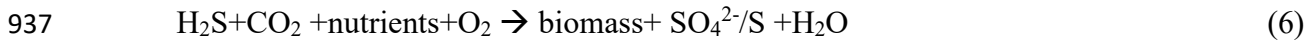


924 The produced active biomass for the production of value-added compounds [140] or as a
925 feedstock to produce biogas [141]. Several cyanobacteria or microalga such as Chlorella,
926 Arthrospira and Spirulina have high photosynthetic efficiency and are mostly studied for biogas
927 upgrading.

928 In the photosynthetic H₂S removal method, biogas is upgraded in algal-bacterial photo-
929 bioreactors [24, 44]. It is a favorable replacement for the concurrent removal of CO₂ and H₂S
930 in a single-step process.



934 In this technology, H₂S is oxidized to sulphate by sulphur oxidizing bacteria (SOB) using the
935 oxygen which is photo-synthetically produced during bio-fixation of CO₂ via microalgae. This
936 process is described as Eq. (6) [142]:



938 The H₂S concentration decreases from the range of nearly 3000–5000 ppm to 50–100 ppm
939 [87]. Safety measures and precautions are required to prevent the production of explosive gas
940 mixtures [105]. The dosing of iron hydroxides/chlorides forms and precipitates FeS, which in
941 turn significantly reduce the H₂S concentration in the gas phase.

942 Biogas can be biologically converted to liquid products such as acetate, ethanol, and butanol
943 [143, 144]. A variety of microorganisms such as acetobacterium woodii, clostridium
944 scatologenes, and the like can convert CO₂ and hydrogen to liquid compounds [145].

945 **3.5.3. Halocarbons, VOC, siloxanes removal technologies**

946 LF biogas halocarbons including 1,1,1-trichloroethane, methylene chloride (di-chloro-
947 methane, DCM), carbon tetrachloride (CCl₄), dichlorodifluoromethane, and
948 tetrachloroethylene are biodegraded at low concentrations under anaerobic, aerobic conditions
949 [146, 147]. The biodegradability of VOC materials present in biogas such as benzene, toluene,
950 volatile fatty acids is reported in Ref. [148].

951 The removal of hexa-methyl-cyclo-tri-siloxane with a removal efficiency of up to 20% is
952 reported in Ref. [149]. The removal of octamethylcyclotetrasiloxane under anaerobic and
953 aerobic conditions was addressed in Ref. [150]. The removal efficiency of 50–60% for aerobic
954 conditions and 15% under anaerobic conditions was reported. In another study, the removal
955 efficiency of 74% for octamethylcyclotetrasiloxane was achieved by Li et al. [151].

956 **4. Costs of biogas upgrading**

957 In nations and regions where policies and legislation have been introduced to incentivize the
958 production of renewable gas, or renewable heat, the cost of biogas upgrading can be offset or
959 partially offset by the magnitude of the incentives, depending on their mode of application. For
960 example, in Germany, a Feed-in Tariff is offered for electricity feed-in associated with biogas
961 installations including those involving CHP [152]. In the UK, the Renewable Heat Incentive is
962 offered [153]. If these incentives are withdrawn it will obviously affect the relative economic
963 attractiveness of biogas upgrading. Amendments to national and sub-national support schemes
964 are also important factors affecting investment decisions. There is evidence that investment in
965 biomethane plants has slowed in Germany since amendments to the EEG law in 2018 [10].

966 Currently, the price gap between natural gas and biomethane varies significantly by region
967 [11]. For example, the price gap in Asian developing countries is the narrowest one while the
968 price gap in the North American countries is the highest one.

969 The differences in upgrading costs of technologies are very much affected by the project-
970 specific circumstances. For instance, upgrading with chemical scrubbing might be of much
971 interest if cheap heat is available on-site. If the biomethane is injected to the high-pressure
972 natural gas network, those upgrading processes working at relatively high pressure such as
973 membranes would be preferred because the injection costs can be reduced. Biogas production
974 and biomethane consumption may take place at different locations. If the biomethane injection
975 point to the gas grid is located away from the production plant, the distribution cost should be
976 considered.

977 The investment cost of upgrading technologies to remove CO₂ from the raw gas against the
978 capacity of biogas is presented in Table 11 [88]. By constructing a power law equation for each
979 technology, we find that the chemical adsorption technology has the best economy of scale

980 (when the plant capacity is doubled, the investment cost increases by about 24%). On the other
981 hand, water scrubbing route has the poorer economy of scale (when the plant capacity is
982 doubled, the investment cost increases by about 44%). To remove CO₂ from a raw biogas with
983 flowrate of 50 Nm³/h, the water scrubbing technology is the cheapest one, while chemical
984 absorption is the most expensive one. At a high biogas flowrate of 2000 Nm³/h, the cheapest
985 and most expensive technologies for removing CO₂ are respectively physical separation, and
986 chemical absorption.

987 In Ref. [92], the investment costs of five upgrading routes including PSA, water scrubbing,
988 amine scrubbing, membrane, and Genosorb[®] scrubbing versus the capacity of raw biogas are
989 presented. All investment costs excluding those for chemical scrubber include the cost of off-
990 gas treatment technology for methane degradation. In addition, the costs of different upgrading
991 technologies were discussed with the following assumptions [92]:

- 992 • Investment and maintenance costs were based on the price of technology providers in
993 2013.
- 994 • Plant annual availability of 96%.
- 995 • Operating lifetime of plants: 15 years.
- 996 • Interest rate: 5%.
- 997 • Costs for planning, permission and additional construction was set to 10% of
998 investment costs.
- 999 • Insurance costs was assumed to be 0.5% of investment costs.
- 1000 • Personnel cost for each technology and any plant size was €35 /h.
- 1001 • Methane recovery rates and energy consumptions (w.r.t 55 and 65% methane content
1002 of the raw biogas) were based on warranty values.

1003 • Desulphurization process (if required): H₂S reduction by 100 ppm, €5/m³ of raw biogas
1004 capacity/year.

1005 • Costs for extra cleaning steps (e.g. NH₃ or siloxanes removal) were not considered.

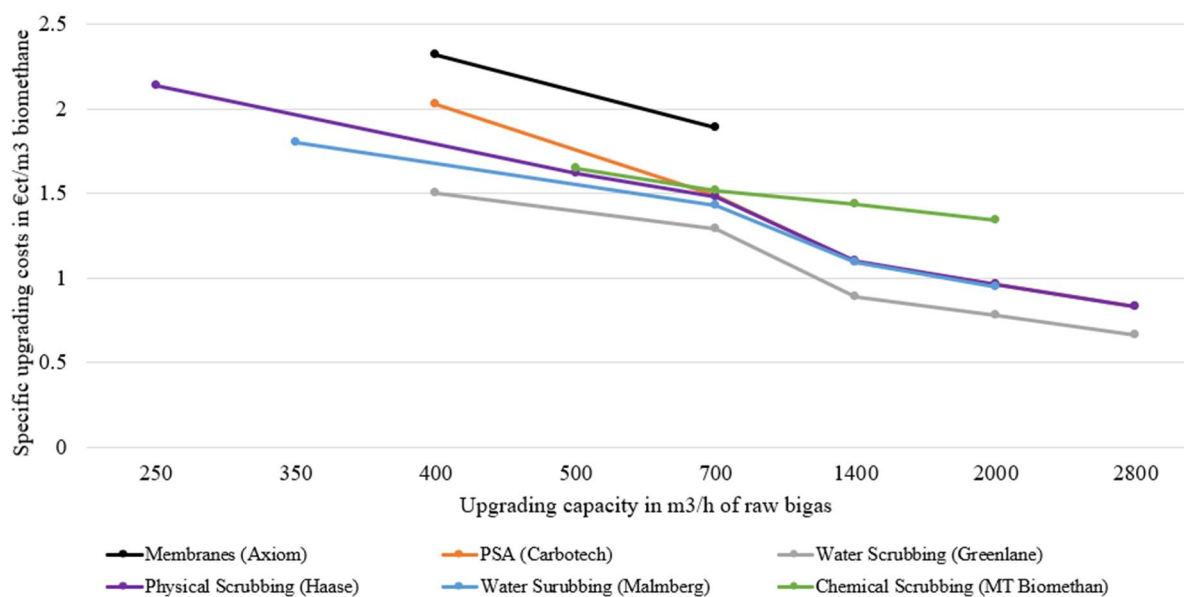
1006 Methane content of the raw biogas were changing from of 55 to 65%. The heat cost of 3–5
1007 €cent/kWh_{th} and electricity cost of 12–18 €cent/kWh_{el} were assumed.

1008 For the PSA system, the upgrading costs for raw biogas capacities are in the range 350–2800
1009 Nm³/h. For water scrubbing technology, the raw gas capacities vary between 300–1400 Nm³/h.
1010 The raw gas capacities of amine scrubber system and membrane technology range within 250–
1011 2000 Nm³/h, and 250–750 Nm³/h, respectively. The comparison of the investment costs of the
1012 five upgrading technology suggests that the PSA route has the highest investment cost for
1013 upgrading capacities of 250–2800 Nm³/h. At high upgrading capacities, the Genosorb®
1014 scrubbing technology is the cheapest one. On the other hand, at low biogas upgrading
1015 capacities, amine scrubbing is the cheapest in most of cases.

1016 Figure 12 illustrates the costs for biogas upgrading (per Nm³/h of biogas) to biomethane based
1017 on a company survey in 2012 [154, 155]. As evident from this figure, the higher the plant
1018 capacity the lower the specific investment cost. When the plant capacity is over 250 Nm³/h,
1019 biogas upgrading using membranes results in higher investment costs. On the other hand, for
1020 any upgrading capacity, water scrubbers offer the minimum upgrading cost among other
1021 technologies [155]. For the cases that the upgrading capacity is 20 Nm³/h, the specific
1022 investment cost based on a rough calculation is about \$1.07/Nm³ CH₄ while for the upgrading
1023 capacity of 200 Nm³/h, the specific investment cost drops to \$0.25/ Nm³ CH₄[87].

1024 Miltner and co-workers performed the techno-economic analysis of biogas upgrading (per
1025 standard m³/h of biogas) to biomethane via pressurized water scrubbing, amine scrubbing,
1026 PSA, and membrane-based gas permeation [100]. Total upgrading cost comprises fixed

1027 investment costs (plant lifetime of 15 years), electricity price of 0.15 €/kWh, annual availability
 1028 of 98 %, and maintenance costs. At low biogas capacities, membrane-based gas permeation
 1029 has the lower upgrading cost, while at high plant capacities the pressurized water scrubbing
 1030 technology is the cheapest technology. Upgrading cost for a plant with a capacity of 250 Sm³
 1031 biogas/h is about 0.25 €/Sm³ biomethane, and it drops to 0.15 €/Sm³ biomethane for upgrading
 1032 capacities above 2000 Sm³ biogas/h.



Haase: Based on Genosorb[®] solvent; MT Biomethane: Current name is Hitachi Zosen Inova

1033 **Figure 12:** Specific costs for biogas upgrading in 2012

1034 For the cases that biomethane is injected to natural gas grid, the most important factors are
 1035 network pressure, volumetric flow and distance to the gas grid. The highest investment costs
 1036 are contributed to compressor stations (ca. 60% of the investment [156]), metering/gas quality
 1037 measurement equipment and pipes construction costs. The major operating costs are the costs
 1038 of the gas conditioning (heating value adjustment by adding LPG) and power consumption of
 1039 compressors. The fixed capital investment for connection to a 16 bar (medium pressure)
 1040 network drops from 12,240 to 2,457 €/Sm³/h as the biomethane flowrate increases from 125 to
 1041 700 Sm³/h. The operating costs are dominated by the fixed capital costs (up to 350 Sm³/h

1042 biomethane injection). Injection into a distribution network (pressure <1 bar) imposes low
1043 investment and operating costs. When the biogas pressure increases from 16 bar to 55 bar (high-
1044 pressure pipelines), the compression costs will increase by up to 50%. Table 16.8 and figure
1045 16.4 of Ref. [156] illustrate examples of costs for gas grid connections.

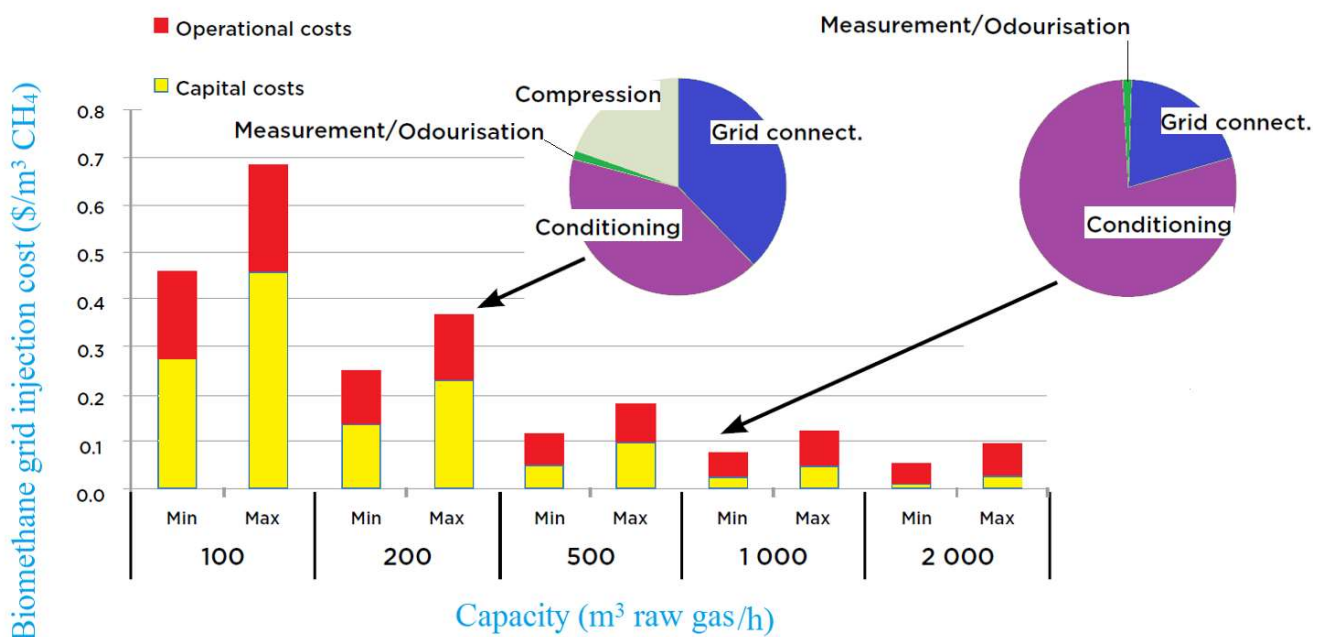
1046 Stürmer et al. [157] compared total cost of biogas upgrading with four technologies including
1047 pressurized water scrubbing, PSA, gas permeation, and amine scrubbing for three biomethane
1048 production capacities of 80, 150, and 500 Nm³/h. The scaling factor of the upgrading
1049 technologies are pressurized water scrubbing 0.38, PSA 0.49, gas permeation membrane
1050 separation 0.61, and amine scrubbing 0.56. The scaling factor of the upgrading technologies +
1051 biomethane injection facilities (compression units, transfer station, and gas pipe) are 0.38, 0.46,
1052 0.55, and 0.51, respectively. This means that when the biomethane production capacity is
1053 doubled, the increment cost increases of biogas upgrading and injection using water scrubbing
1054 is the lowest one. In addition, biogas upgrading via the membrane is the cheapest technology
1055 at biomethane production capacities of 80, and 150 Nm³/h. On the other hand, pressurized
1056 water scrubbing becomes the cheapest technology at biomethane production capacity of 500
1057 Nm³/h. In another study, Paturska and co-workers performed an economic study for
1058 biomethane supply based on the Latvian natural gas infrastructure [158]. The authors
1059 considered three scenarios and five biogas upgrading technologies including water scrubbing,
1060 amine scrubbing, membrane separation, physical scrubbing with organic solvents and PSA. In
1061 the first scenario, it was assumed that each biogas plant delivers the produced biomethane via
1062 its in-situ upgrading infrastructure to the grid. In the second scenario, it was presumed that the
1063 raw biogas of all biogas plants is conveyed to a central biogas upgrading plant for the grid
1064 injection. The last scenario states that biomass conversion to biogas, biogas upgrading, and
1065 grid injection take place in a central facility. The results indicate that the price of the injected

1066 biomethane of Scenario 3 (the most favorable scenario) is nearly 19 % higher than the natural
 1067 gas.

1068 Figure 13 depicts the biomethane grid injection cost vs. the capacity of the raw biogas. For
 1069 each capacity, the minimum and maximum costs represent the situations where the injection is
 1070 to low- and high-pressure gas networks. The economy of scale is mainly attributed to the capital
 1071 costs. Operational costs are less sensitive to the injection capacity. As illustrated in Figure 13,
 1072 the difference between the minimum and maximum costs stems from different compression
 1073 costs of the biomethane. The biomethane injection costs into the gas grid increase considerably
 1074 when the gas injection capacity is reduced. Further costs for biomethane distribution via the
 1075 gas network arise as a fee when the gas is transported via the public gas grid. In Germany, for
 1076 example, the fee varies between from \$0.05–0.26 / Sm³ methane.

1077

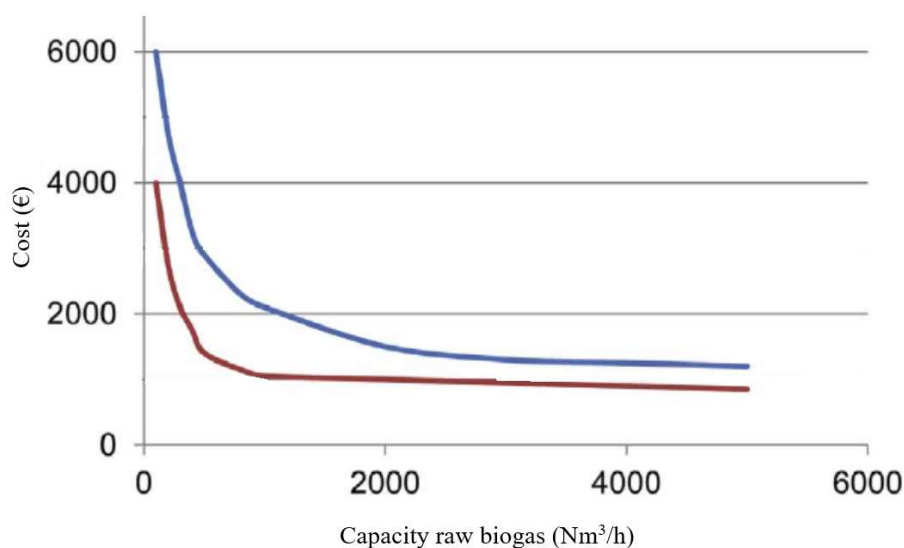
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1079

1080 **Figure 13:** Biomethane grid injection costs based on cost analysis in Germany, 2010

1081 Techno-economic comparison of biogas production from various feedstocks in different
1082 countries is presented in Ref. [159]. The total investment of two plants in Sweden with annual
1083 capacity of 100 and 20 ktonne of biomethane from pinewood and forest residues in 2014 were
1084 estimated to be 65.1, and 60.5 M€, respectively. In addition, the investment cost of a plant in
1085 Sweden with capacity of 110,000 m³ MSW/year + 3500 m³/day raw biogas in 2012 was 49.2
1086 M\$. In Ref. [87], the total production costs of biomethane for vehicle fueling by feedstock and
1087 capacity scale ranging from 100–2000 m³/h is addressed. The feedstock are energy crops,
1088 manure and industrial waste. The total production costs include costs of biogas production,
1089 cleaning and upgrading, as well as distribution via the gas grid. In another study, the economic
1090 feasibility of biogas upgrading of physicochemical upgrading processes (section 3.3) on the
1091 market is studied (Figure 14), [91]. In this figure, the lower and upper lines show the low and
1092 high fixed capital investment of all existing physicochemical upgrading technologies versus
1093 the plant capacity. The lines approach to each other as the capacity of raw biogas increases.
1094 Techno-economic analysis of LF biogas upgrading with membranes and utilization for vehicle
1095 fuel is addressed in Ref. [160]. The impact of feed composition, feed flow rate and pressure on
1096 gas processing cost was investigated. The raw biogas from the LF contains 55% CH₄, 38.9%
1097 CO₂, 5% N₂, 0.4% O₂, 0.002% H₂S and 0.66% H₂O.



1099

1100 **Figure 14:** Cost range of physicochemical upgrading technologies versus raw biogas capacity (Note: the lower
 1101 and upper lines show the low and high fixed capital investment of all existing physicochemical upgrading
 1102 technologies).

1103 5. Conclusions

1104 Biogas production from waste materials and renewables is a favorable solution to the energy
 1105 and environmental issues facing communities. Biogas upgrading faces substantial challenges
 1106 concerning investment costs, energy consumption along with operating costs. The alternatives
 1107 to biogas upgrading to biomethane are flaring or burning of raw biogas for power production
 1108 or concurrent heat and power generation. Methane recovery is an important factor from the
 1109 economic and ecological points of view. The biomethane obtained from biogas upgrading is a
 1110 replacement of fossil fuels. Biomethane can be used for heating, power and steam generation,
 1111 as vehicle fuel, in chemical plants, for injections to gas networks, etc. Only if the economic
 1112 returns from direct injection to grid or sale for use in transport applications are significantly
 1113 higher (for example if electricity prices are low, and incentives for renewable electricity

1114 generation from biogas are low or non-existent), then investment in upgrading for biomethane
1115 uses can be made.

1116 This paper addresses the biogas supply chain including feedstock supply, biogas production
1117 and upgrading processes (physical, chemical, and biological), requirements for product
1118 specification based on applications, upgrading efficiency, methane recovery/loss, capital costs,
1119 and biogas/biomethane uses. In addition, the CO₂ utilization in the CO₂ demanding industries
1120 is covered.

1121 The cost of biogas upgrading depends on several factors including raw biogas capacity, the
1122 concentration of contaminants in the raw biogas, local circumstances, energy and water cost,
1123 interest rate, general investment climate, envisaged lifetime of the investment, reliability of the
1124 biogas source, other economic risks including the projected cost curves of competitor
1125 technologies, and final utilization aspects of the upgraded biogas. Upgrading with chemical
1126 scrubbing might be of much interest if cheap heat is available on-site. In case of the grid
1127 injection, it will depend on the quality standards (e.g. Wobbe Index, heating value, and other
1128 criteria) that is required by the applicable gas safety laws and the gas distribution company in
1129 each jurisdiction.

1130 The main findings of this paper are that the biogas upgrading cost depends very much on the
1131 raw biogas capacity. For example, a company survey showed that water scrubbing is the
1132 cheapest biogas upgrading technology at any biogas capacity. At biogas capacities lower than
1133 750 Nm³/h, the biogas cleaning via membranes has the highest cost while at capacities higher
1134 than 750 Nm³/h, chemical scrubbing results in higher upgrading costs. Another study in 2013
1135 indicated that among PSA, water scrubbing, amine scrubbing, membrane, and Genosorb[®]
1136 scrubbing, the PSA route has the highest investment cost for almost all raw biogas capacities
1137 of 250–2800 Nm³/h. At high raw biogas capacities, the Genosorb[®] approach is the cheapest

1138 one. On the other hand, at low biogas flowrates, amine scrubbing is the cheapest in most of
1139 cases.

1140 For the case that the biogas is upgraded to biomethane for grid injection, the membrane
1141 separation is the cheapest technology at low capacities among water scrubbing, PSA,
1142 membrane, and amine scrubbing. However, the pressurized water scrubbing is the cheapest
1143 technology at high biomethane production capacities. The biomethane injection costs into the
1144 gas network increase noticeably with the capacity reduction.

1145 The next step to this research is to develop an economical model for biogas upgrading
1146 technologies. The inputs to the model will be biogas composition and capacity, local technical
1147 data including electricity, water, gas and utilities price, etc.

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